

Supplementary Information for

**Synthesis and characterisation of iodobismuthates containing N-
substituted 1,4-diazabicyclo[2.2.2]octane**

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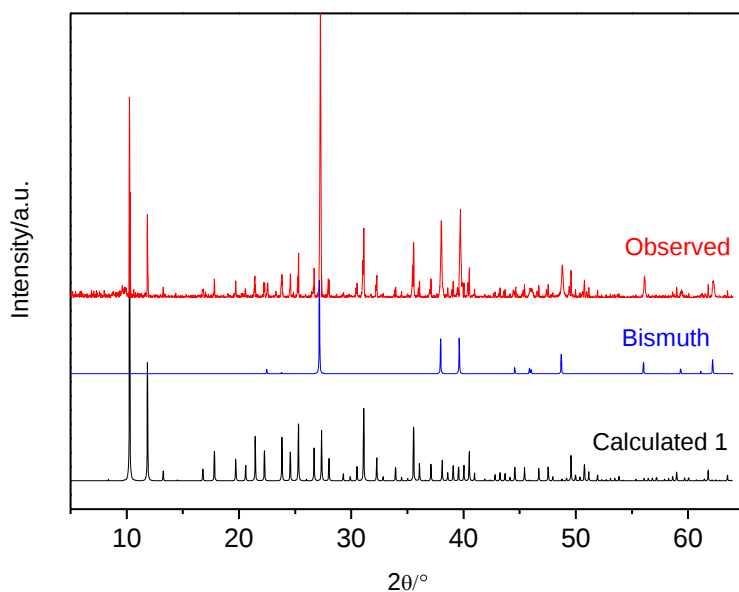


Figure S1. Powder diffraction pattern of the solid product of the reaction producing compound **1** (red line). The pattern calculated using the structure for **1**, determined by single-crystal X-ray diffraction, is shown in black and that calculated for bismuth metal is shown as a blue line.

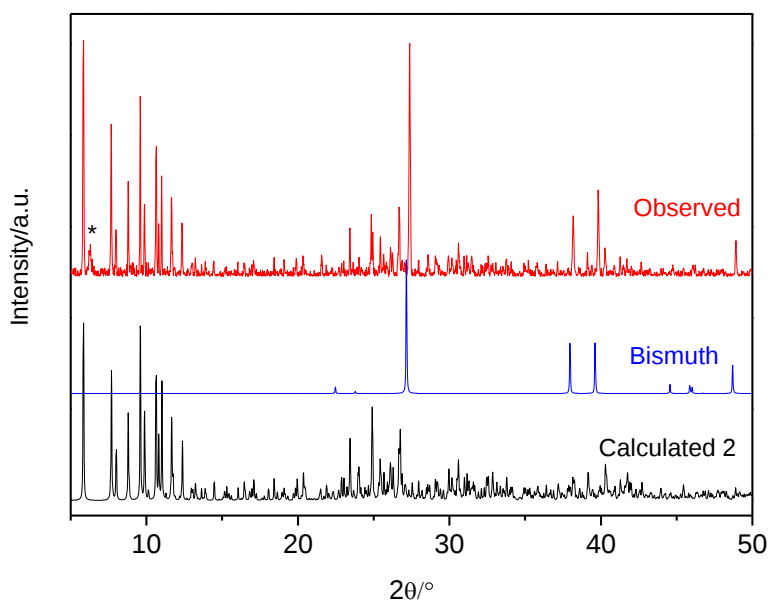


Figure S2. Powder diffraction pattern of the solid product of the reaction producing compound **2** (red line). The pattern calculated using the structure for **2**, determined by single-crystal X-ray diffraction, is shown in black and that calculated for bismuth metal is shown as a blue line. The asterisk marks peaks possibly arise from compound **3**.

Table S1. Lattice parameters of compounds **1** and **2** determined using powder X-ray diffraction data collected at room temperature.

Compound	1	2
$a/\text{\AA}$	14.998(82)	8.970(2)
$b/\text{\AA}$	14.998(82)	40.10(5)
$c/\text{\AA}$	14.998(82)	22.97(5)
$\beta/^\circ$	90	92.5(3)

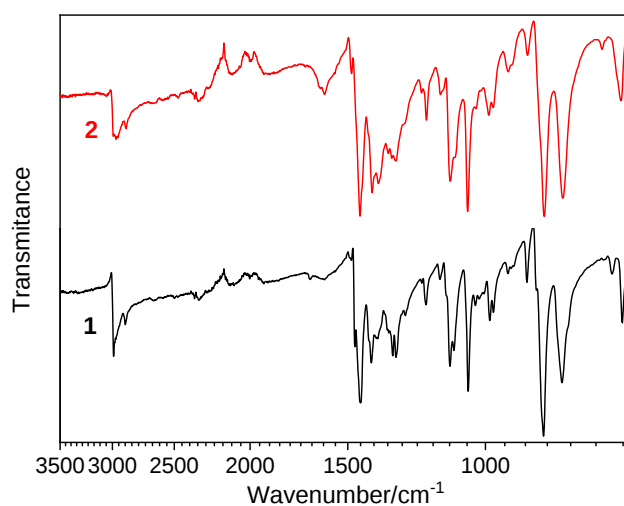


Figure S3. FTIR data collected on hand-picked single crystals of compounds **1** (black line) and **2** (red line).

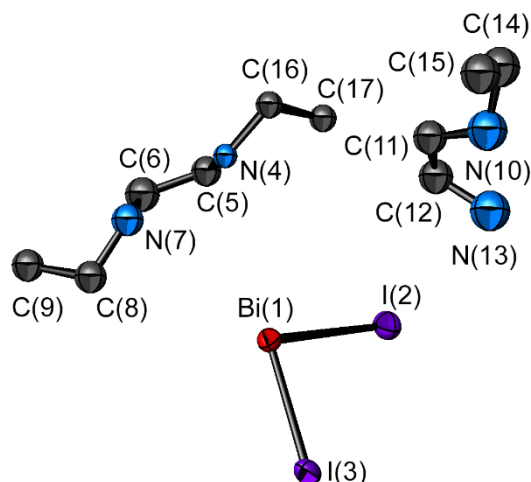


Figure S4. Asymmetric unit of compound **1** with displacement ellipsoids drawn at the 50% probability level. H atoms have been omitted for clarity.

Table S2 Short C-H $\cdots$ I contacts (Å) for compound **1**.

I(2) $\cdots$ H(111) ⁱ	3.075(28)
I(2) $\cdots$ H(112) ⁱⁱ	3.298(26)
I(2) $\cdots$ H(121) ⁱⁱⁱ	3.284(25)
I(2) $\cdots$ H(122) ^{iv}	3.080(28)
I(2) $\cdots$ H(142) ⁱ	3.118(2)
I(3) $\cdots$ H(51) ^v	3.144(24)
I(3) $\cdots$ H(52) ^{vi}	3.069(25)
I(3) $\cdots$ H(61) ^{vii}	3.392(25)
I(3) $\cdots$ H(62) ^{viii}	3.145(28)
I(3) $\cdots$ H(81) ^{viii}	3.141(2)
I(3) $\cdots$ H(82) ^{ix}	3.056(2)
I(3) $\cdots$ H(93) ^{vii}	3.254(2)
I(3) $\cdots$ H(161) ^x	3.384(2)
I(3) $\cdots$ H(172) ^{vi}	3.374(2)

Symmetry codes: (i) $1/2 + x, 3/2 - y, 1 - z$; (ii) $z, 1 + x, y$; (iii) $y - 1/2, 3/2 - z, -x$; (iv) $-x, 1 + y, 1/2 - z$; (v) $1 - z, x + 1/2, 1/2 - y$; (vi) $1 - y, 1/2 + z, 1/2 - x$; (vii) $1/2 - x, 1 - y, z - 1/2$; (viii) $1/2 - z, 1 - x, y - 1/2$; (ix) $1/2 - y, 1 - x, z - 1/2$; (x) $1 - x, 1/2 + y, 1/2 - z$.

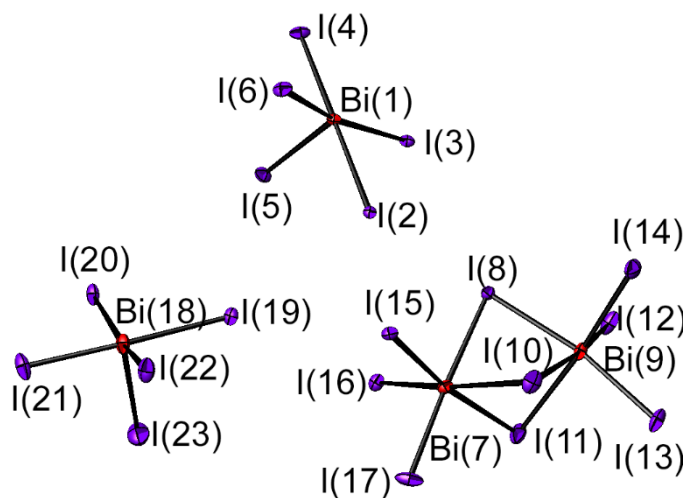


Figure S5. The inorganic anions in the asymmetric unit of compound **2** with displacement ellipsoids drawn at the 50% probability level.

Table S3 Short intermolecular I $\cdots$ I contacts (Å) for compound **2**.

I(2) $\cdots$ I(8)	4.131(1)
I(4) $\cdots$ I(5)	4.183(1)
I(4) $\cdots$ I(8) ⁱ	3.910(1)
I(5) $\cdots$ I(21) ⁱⁱ	4.187(1)
I(8) $\cdots$ I(10)	4.166(1)
I(12) $\cdots$ I(14)	4.176(1)
I(20) $\cdots$ I(20) ⁱⁱⁱ	4.147(1)

Symmetry codes: (i) $1-x, 1-y, 2-z$; (ii) $1-x, 1-y, 1-z$; (iii) $2-x, 1-y, 1-z$.