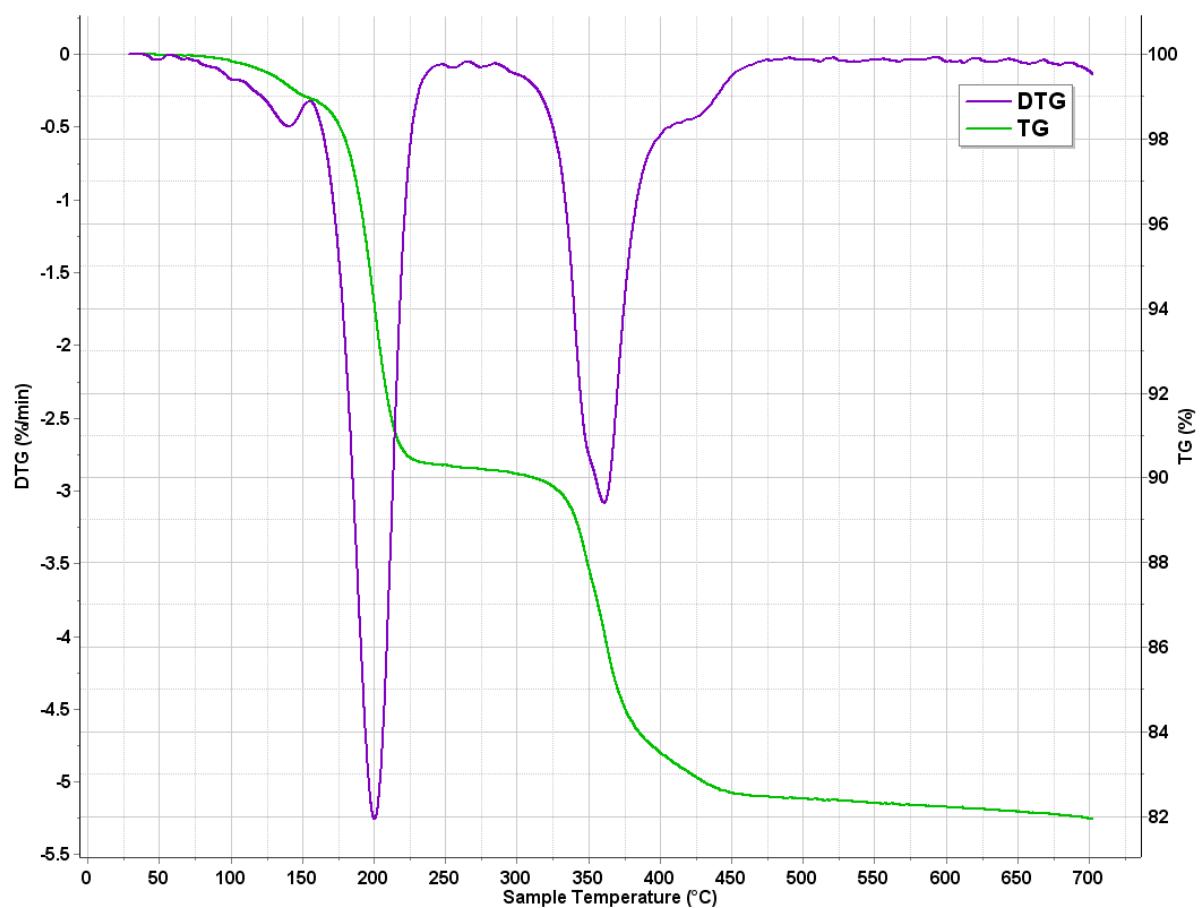
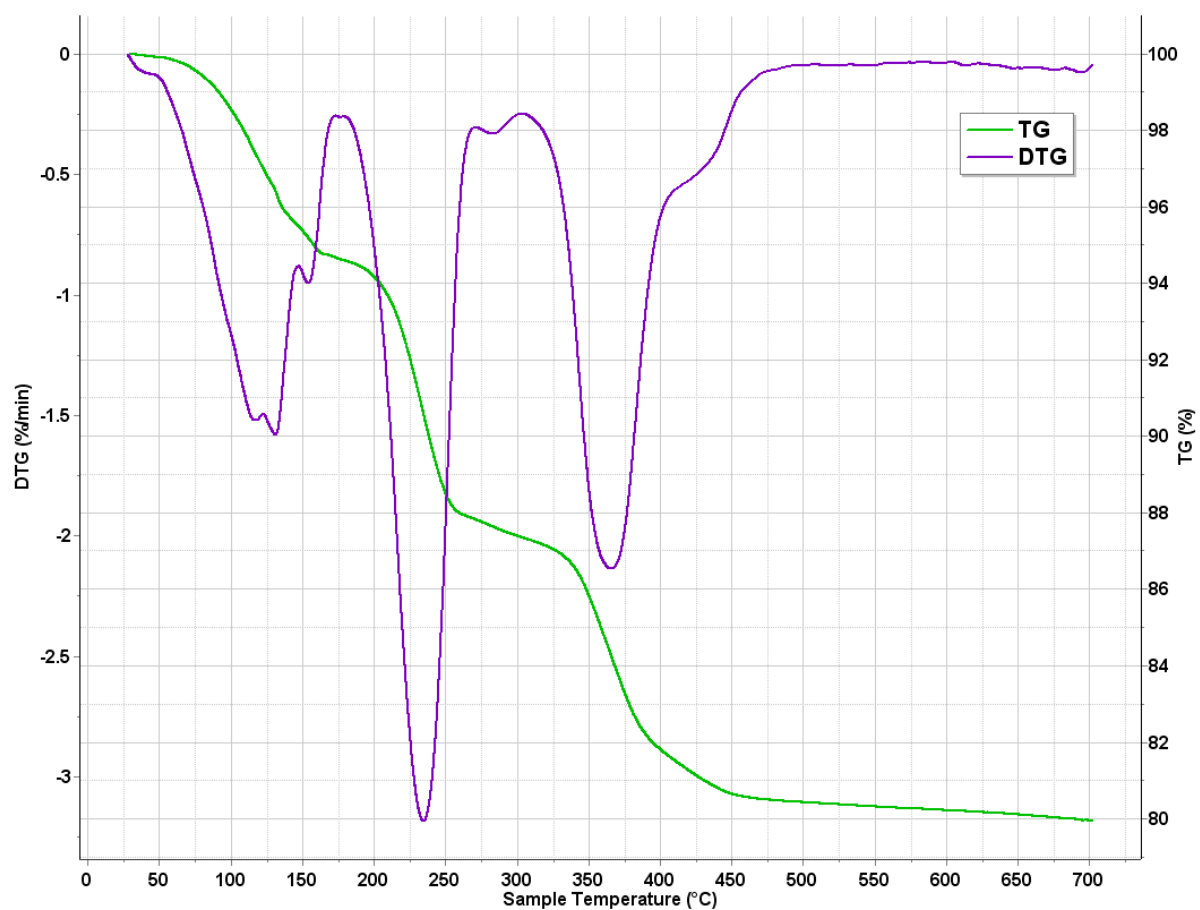


**THERMAL ANALYSIS OF SOLVATOMORPHIC
DECAKIS(DIMETHYLAMMONIUM)
DIHYDRODODECATUNGSTATE HYDRATES**

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Majzik, Laura Bereczki, Imre Miklós Szilágyi, Rajandra P.
Pawar ·László Kótai**

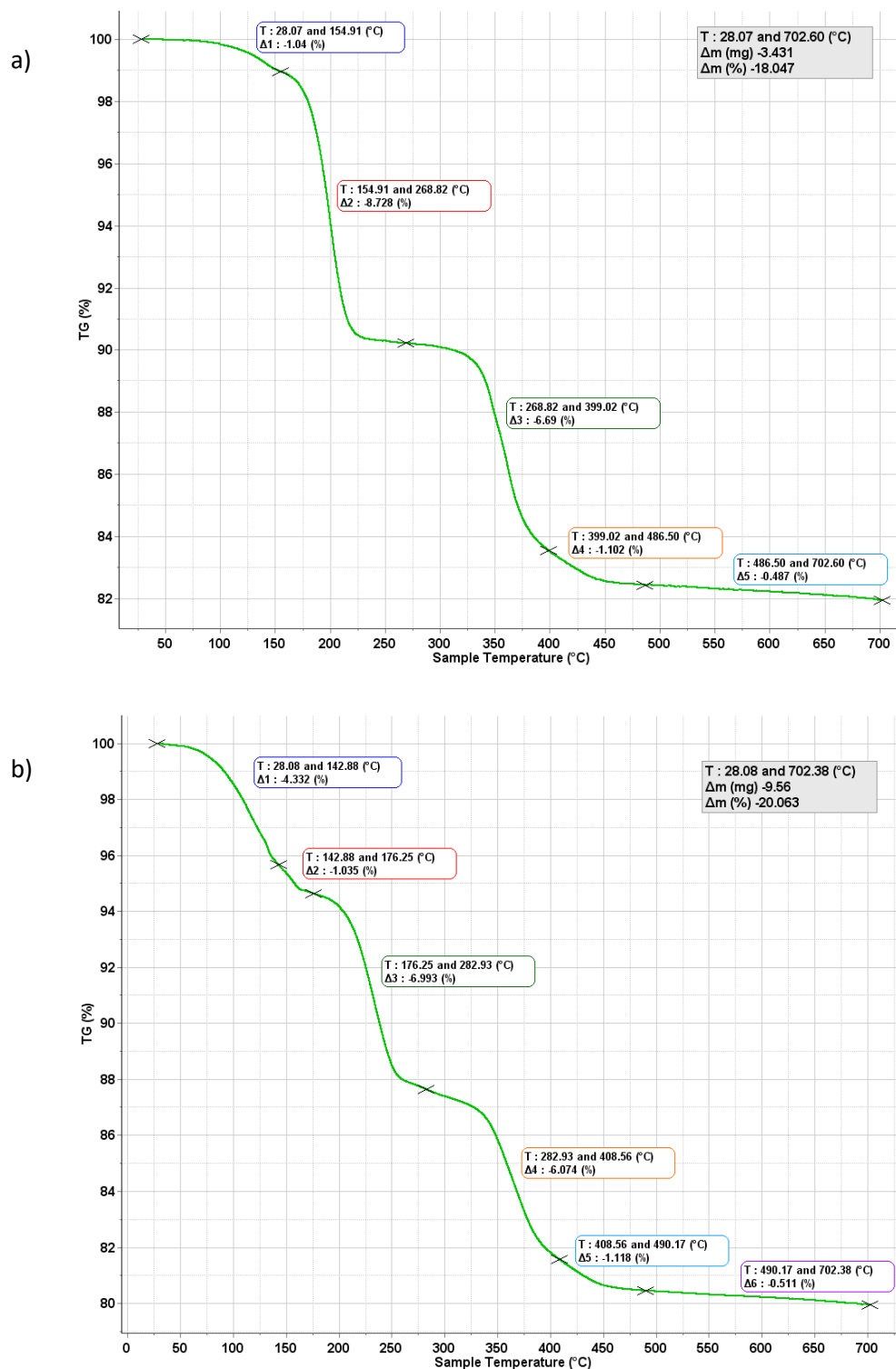


ESI Figure 1. TG and DTG curve of **1a** in He atmosphere

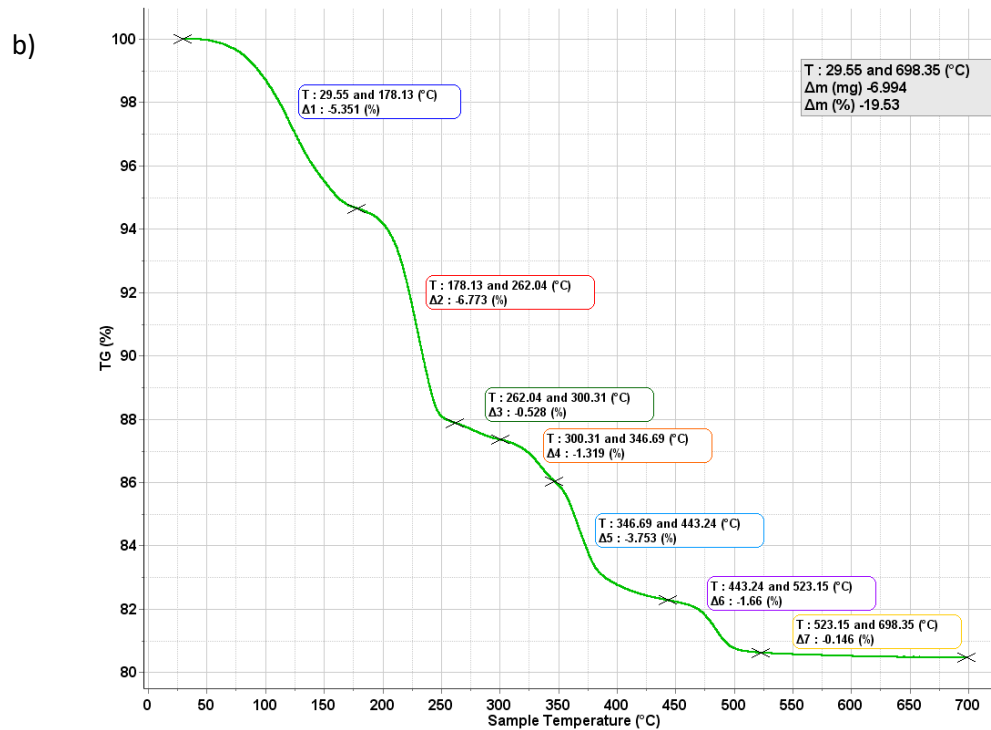
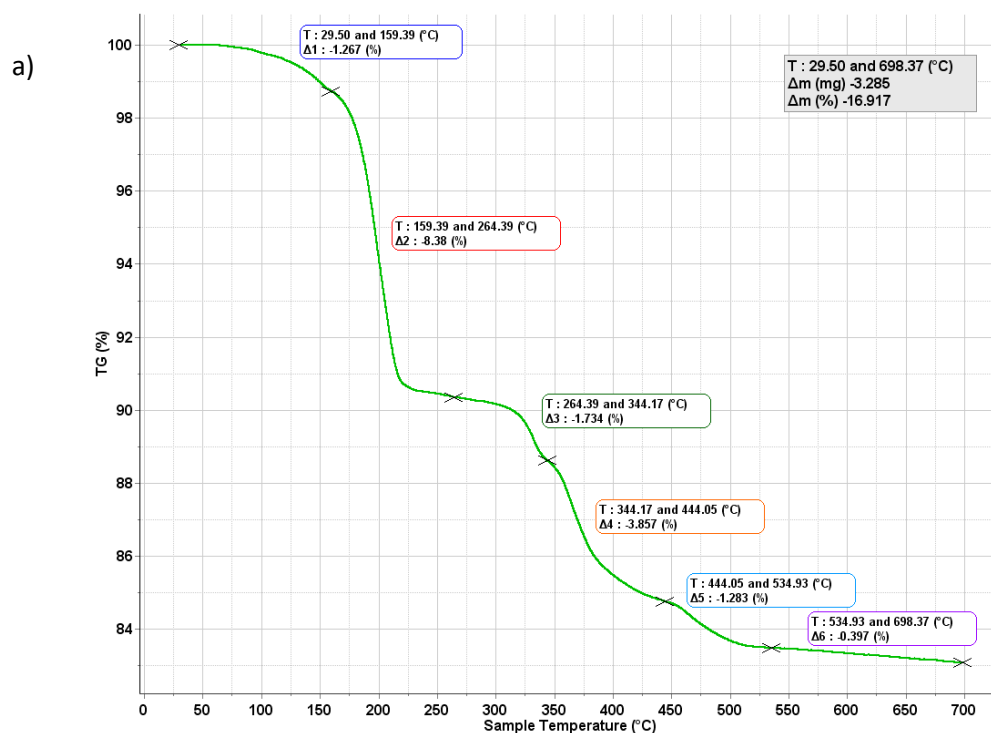


ESI Figure 2. TG and DTG curve of **1a** in He atmosphere

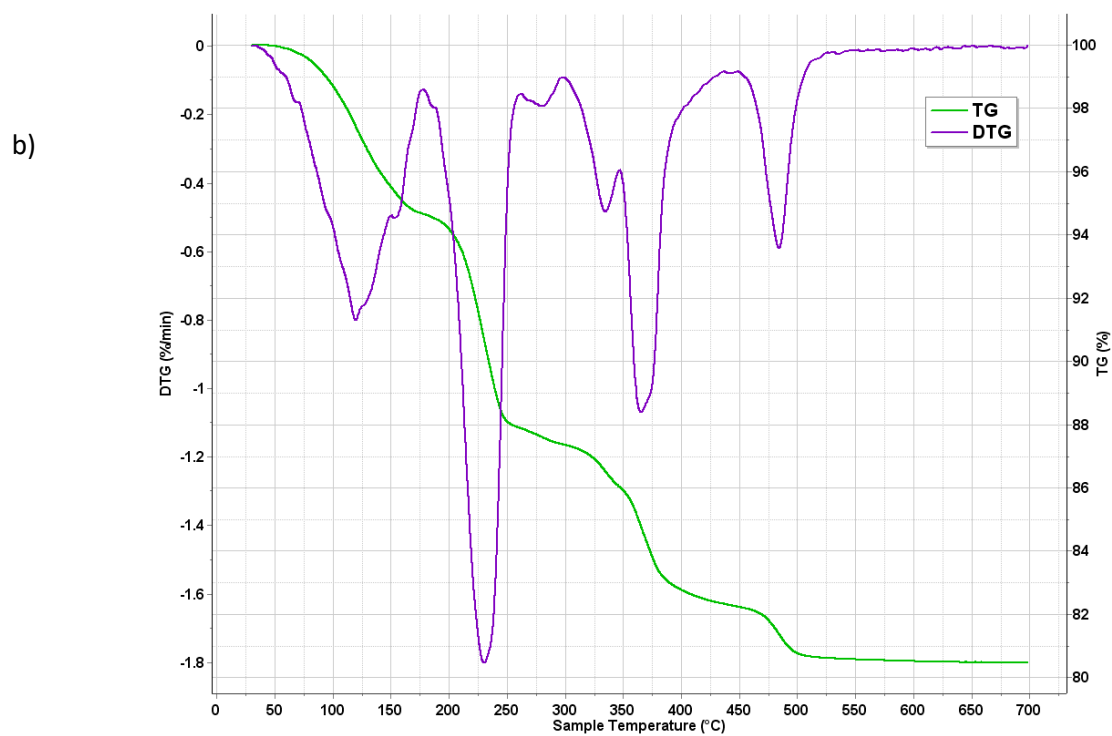
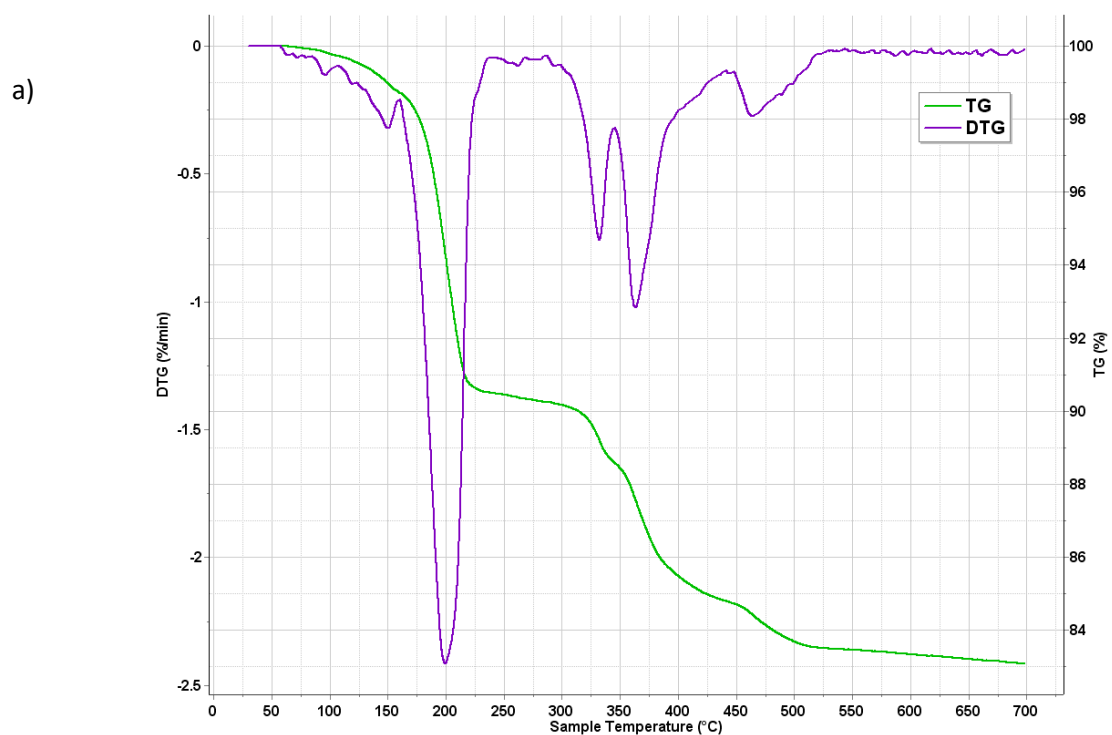
ESI Fig.3. TG curves of compound **1a** (a) and **1b**(b) in He atmosphere



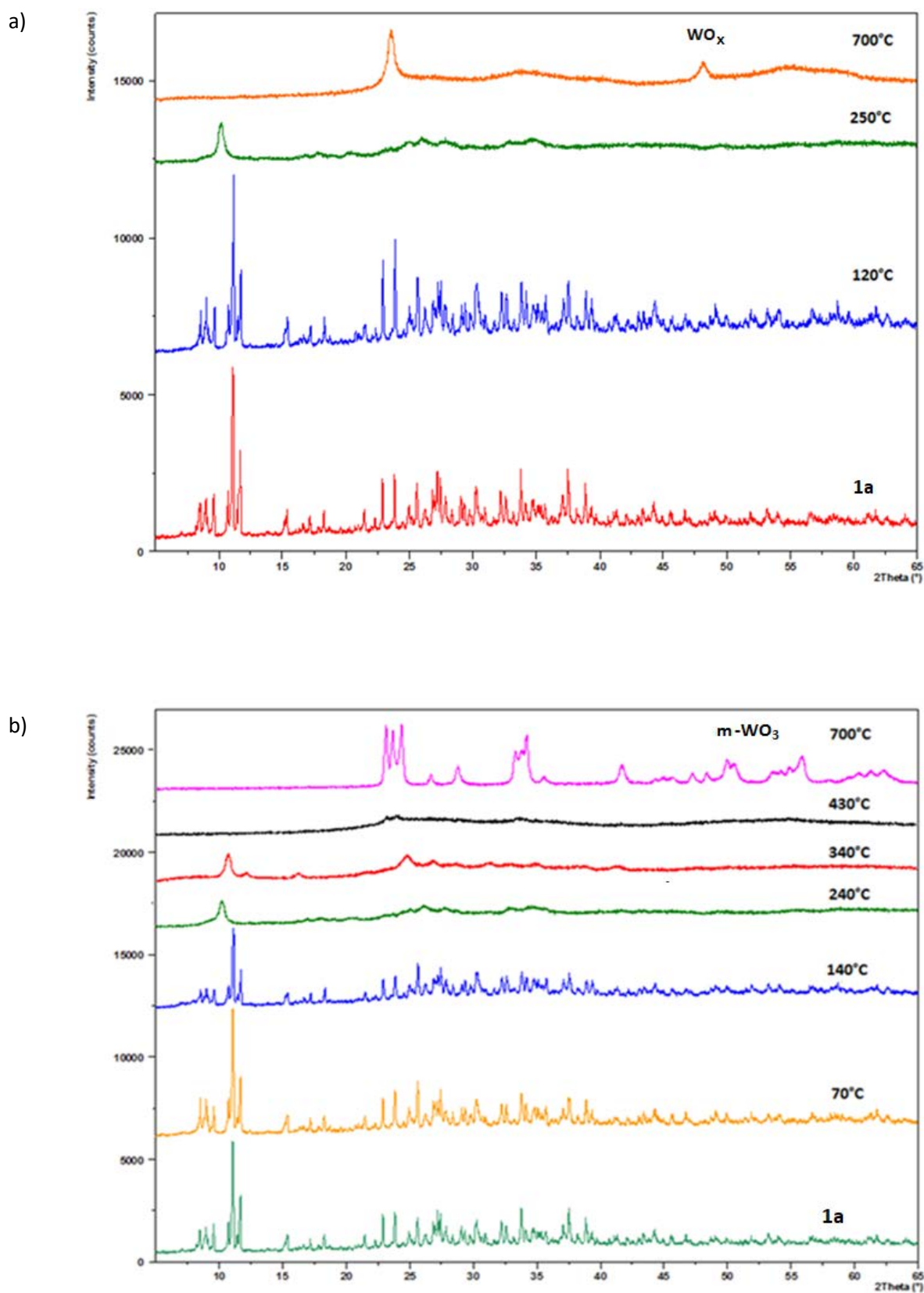
ESI Fig.4. TG curves of compound **1a** (a) and **1b**(b) in synthetic air atmosphere



ESI Fig. 5. TG and DTG curves of compounds **1a** (a) and **1b** (b) in synthetic air atmosphere

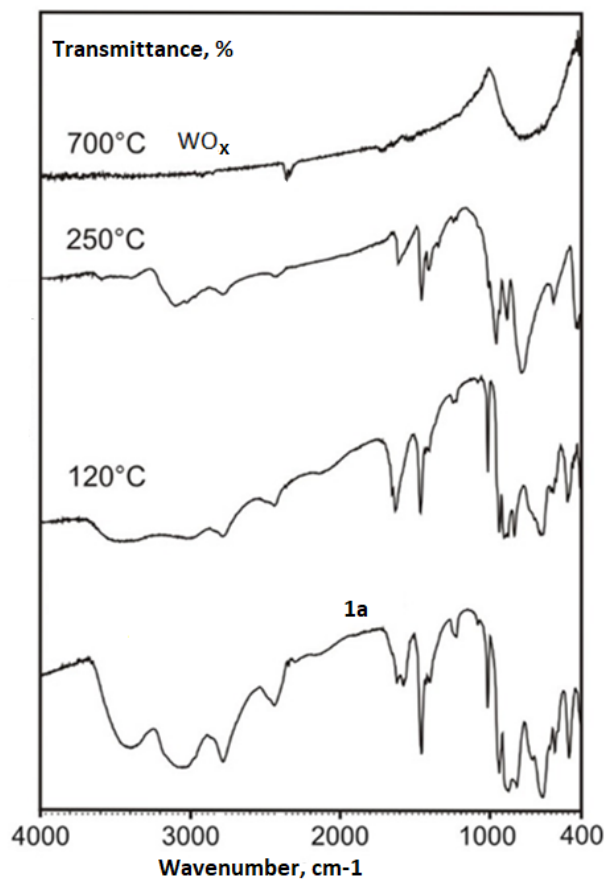


ESI Fig.6. XRD of the **1a** and its thermal decomposition product in inert (a) and synthetic air (b) atmosphere

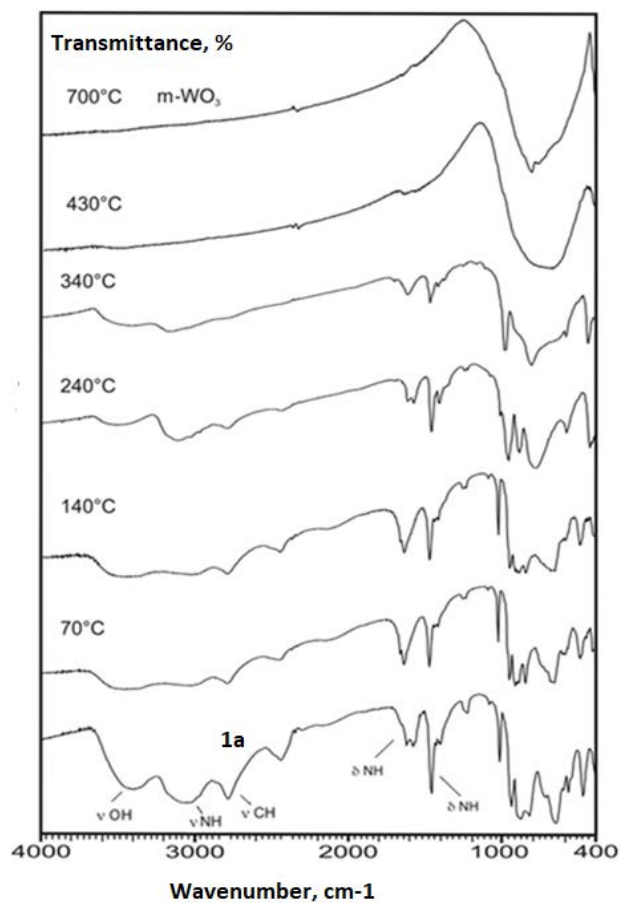


ESI Fig.7. IR of the **1a** and its thermal decomposition product in inert (a) and synthetic air (b) atmosphere

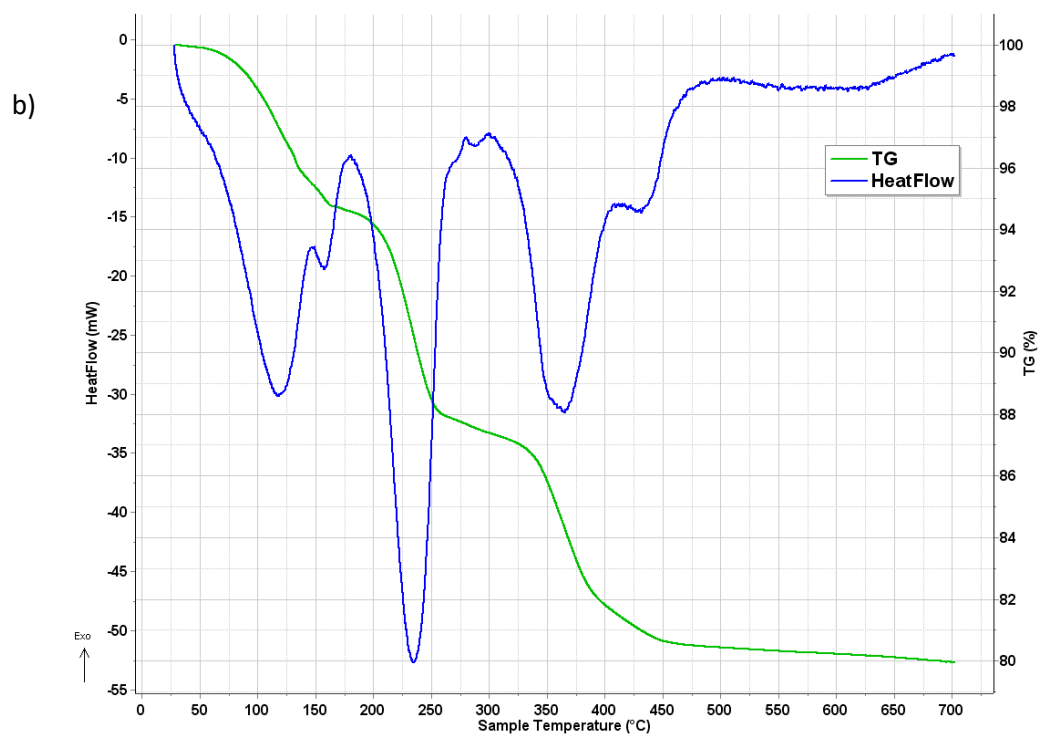
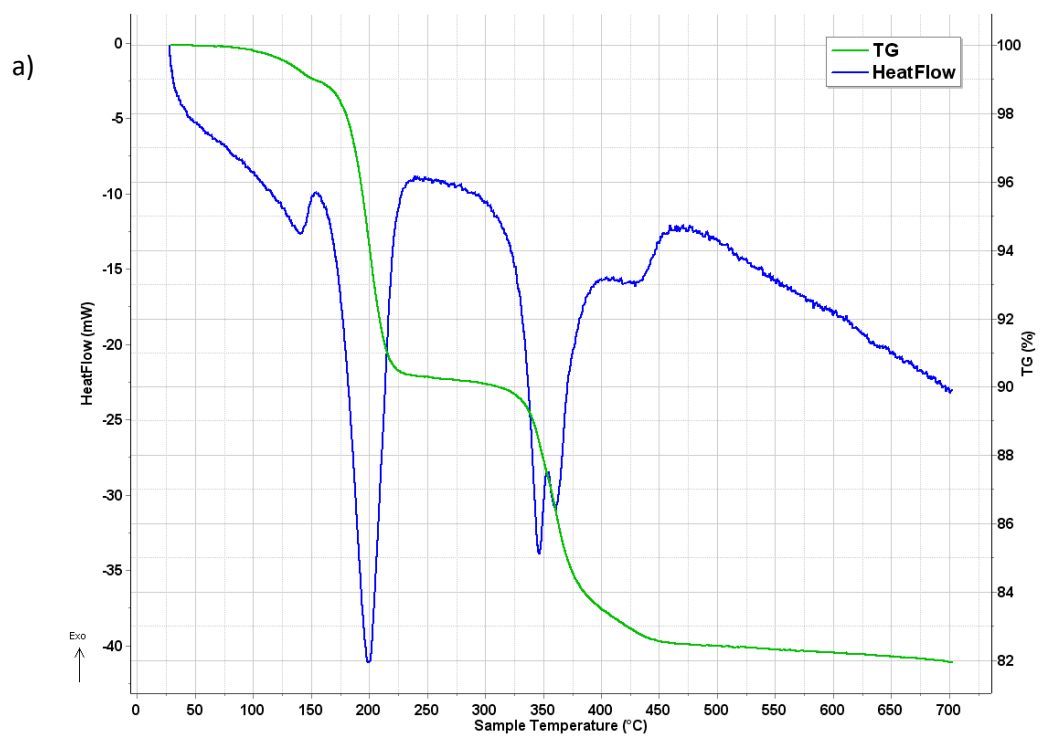
a)



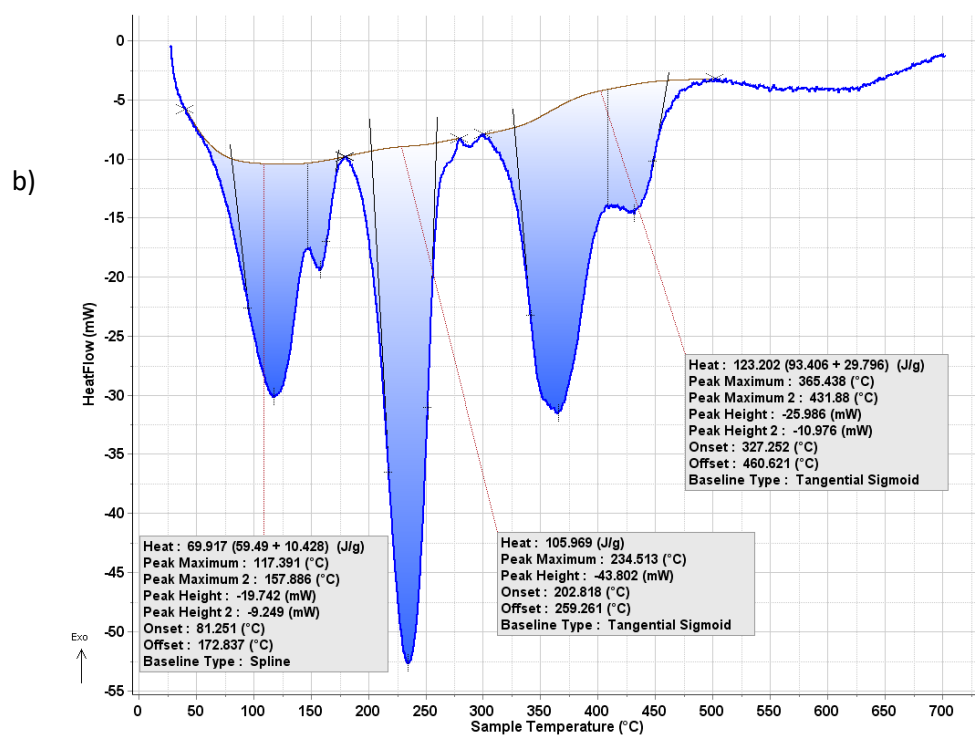
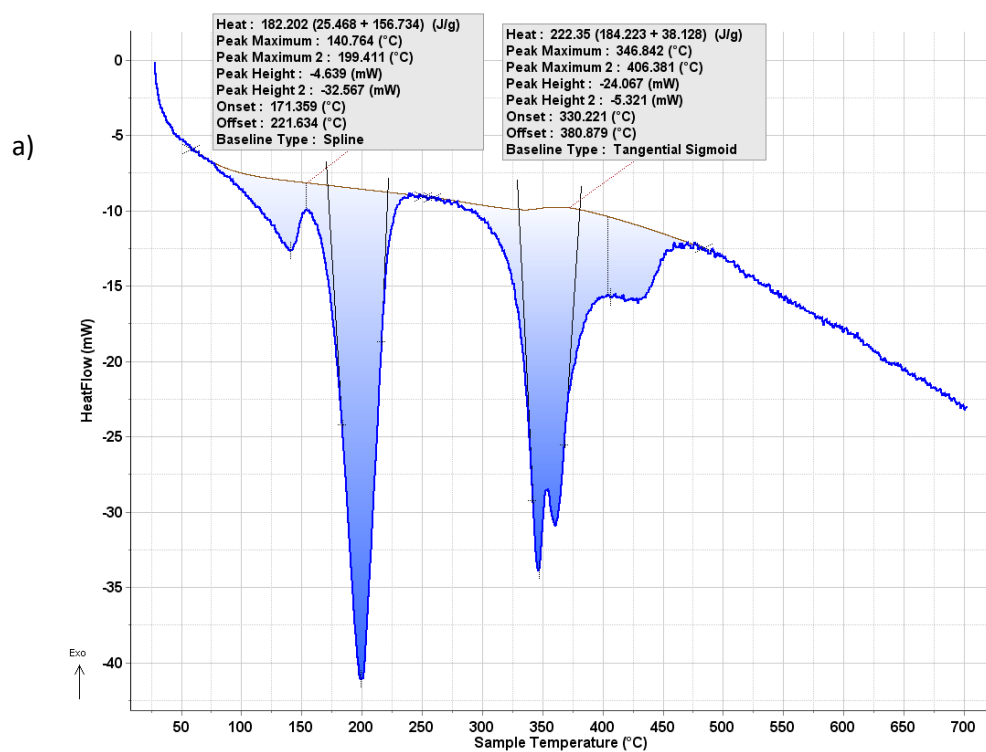
b)



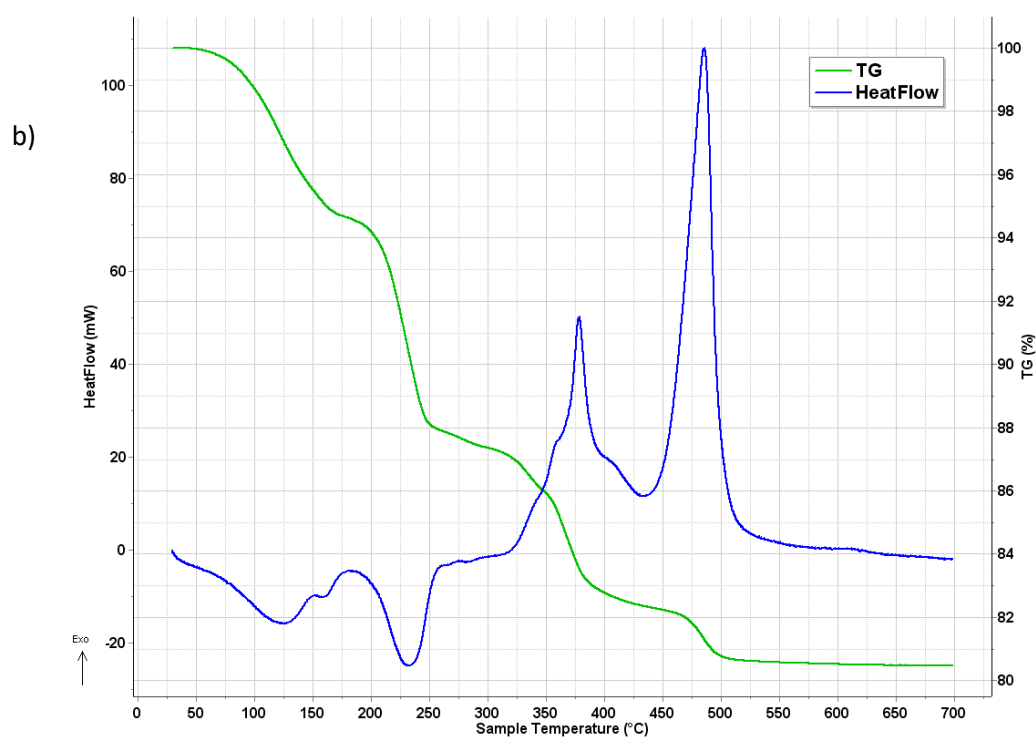
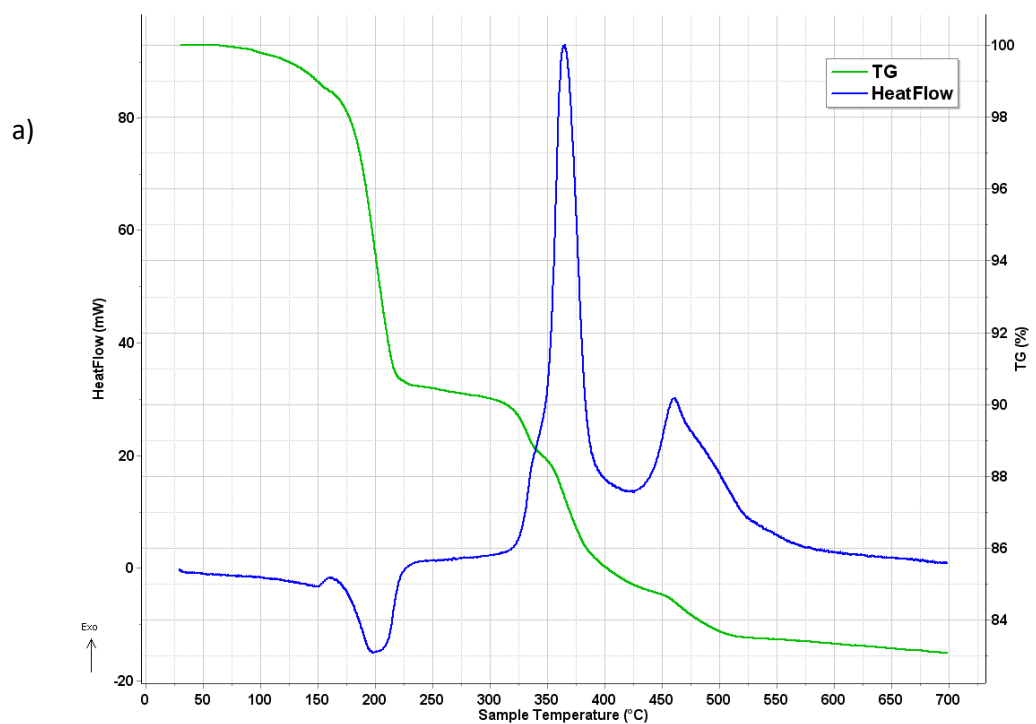
ESI Fig 8. TG and DSC curves of compound **1a** (a) and **1b** (b) in inert atmosphere



ESI Fig.9. DSC results of compound **1a** (a) and **1b** (b) in inert atmosphere

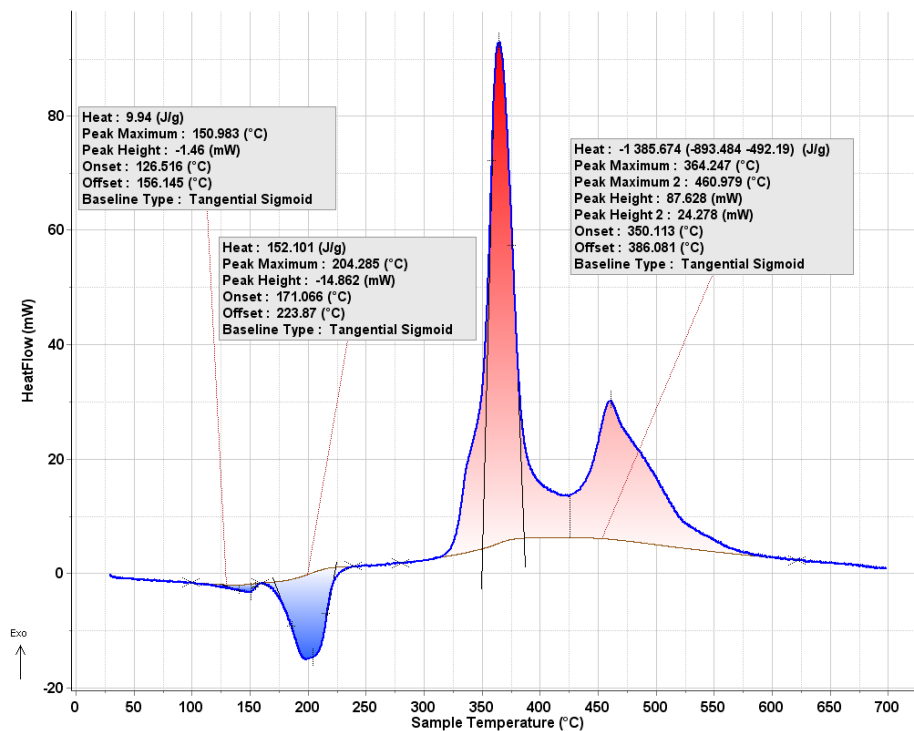


ESI Fig. 10. TG and DSC curves of compound **1a** (a) and **1b** (b) in synthetic air

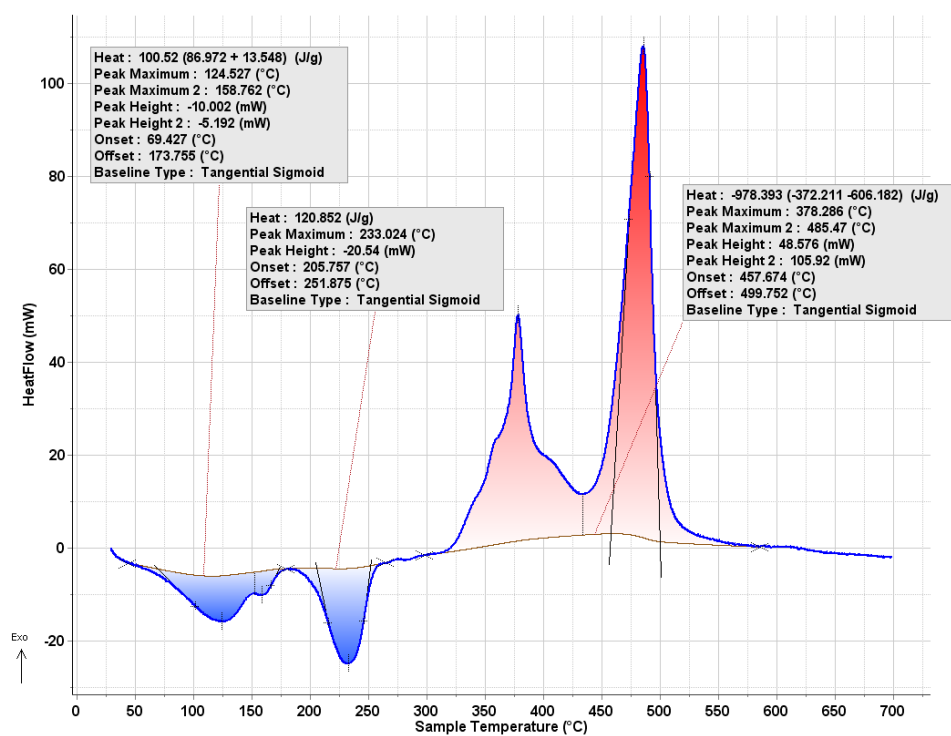


ESI Fig. 11. Quantitative DSC results of compounds **1a** (a) and **1b** (b) on heating under synthetic air atmosphere

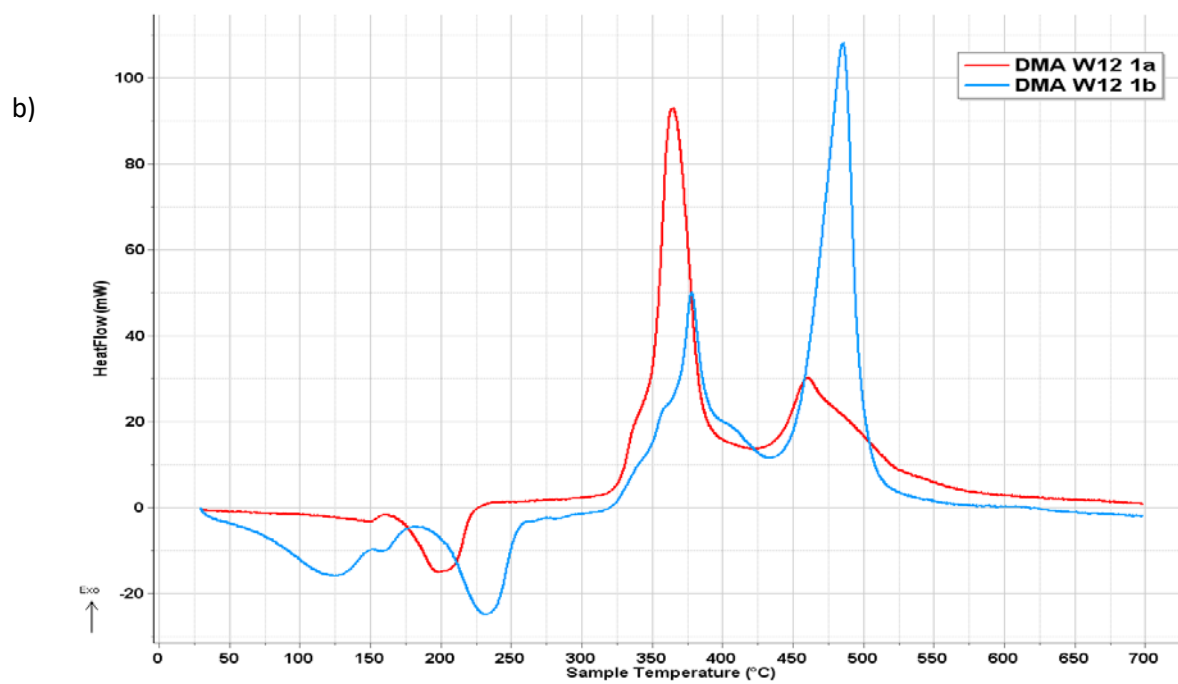
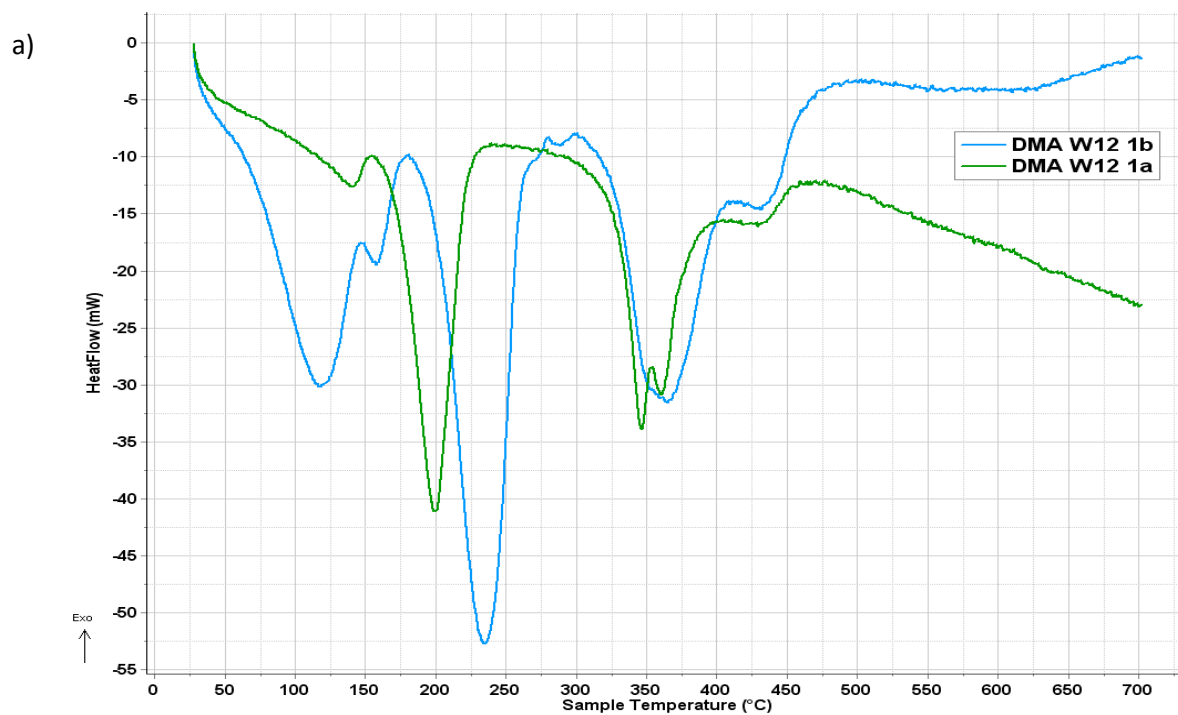
a)



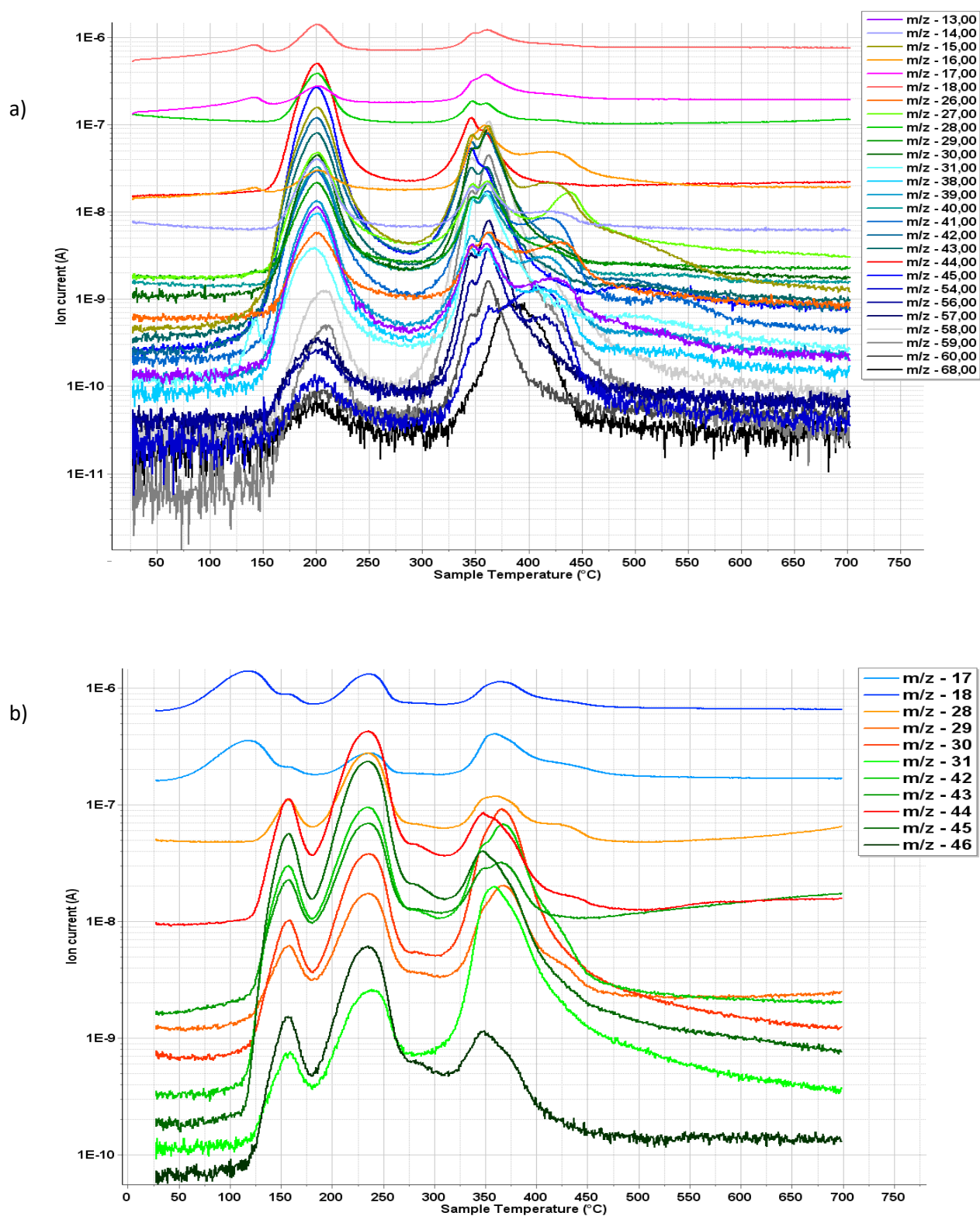
b)



ESI Fig. 12. Comparison of the DSC curves of compounds **1a** and **1b** in inert (a) and synthetic air (b) atmosphere

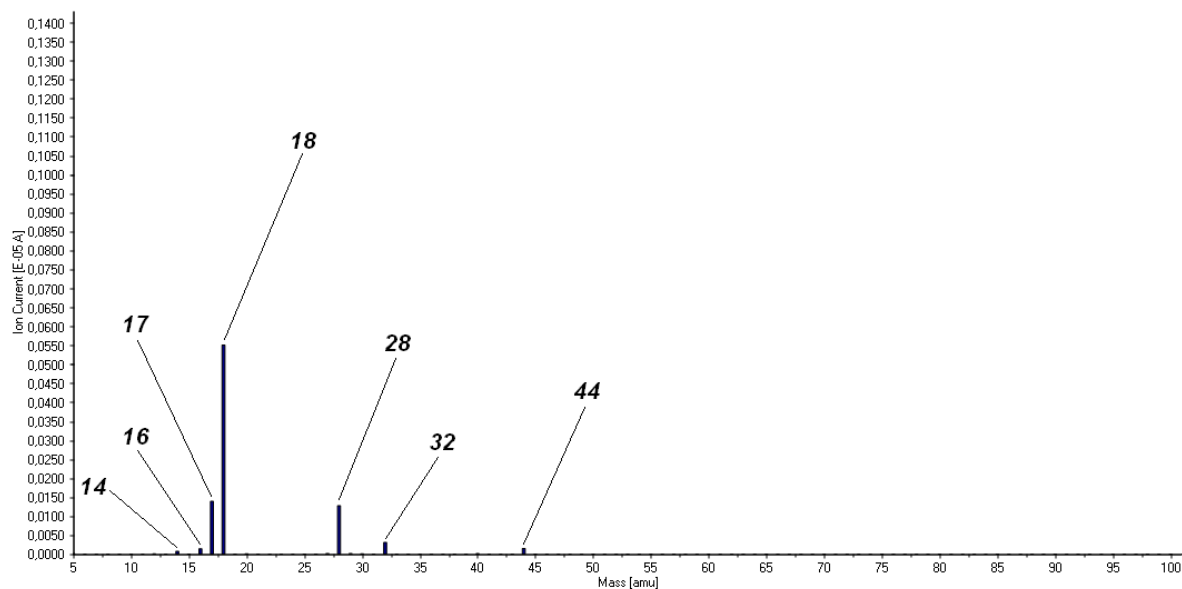


ESI Fig.13. TG-MS curves of the water and DMA fragments measured during thermal decomposition of compound **1a** (a) and **1b** (b) under inert atmosphere

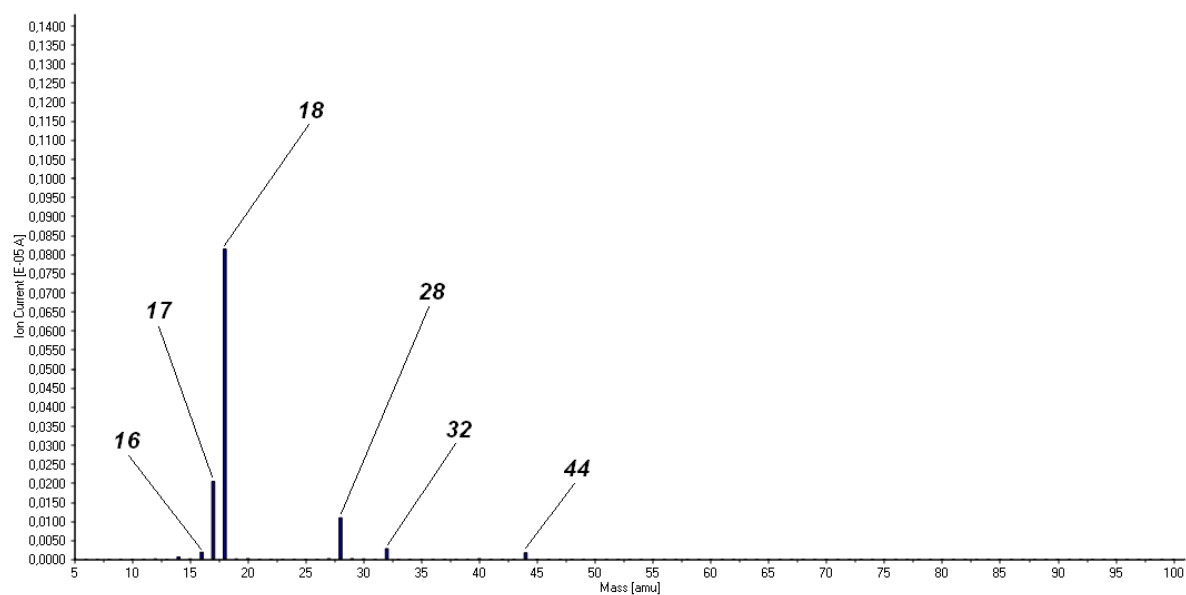


a)

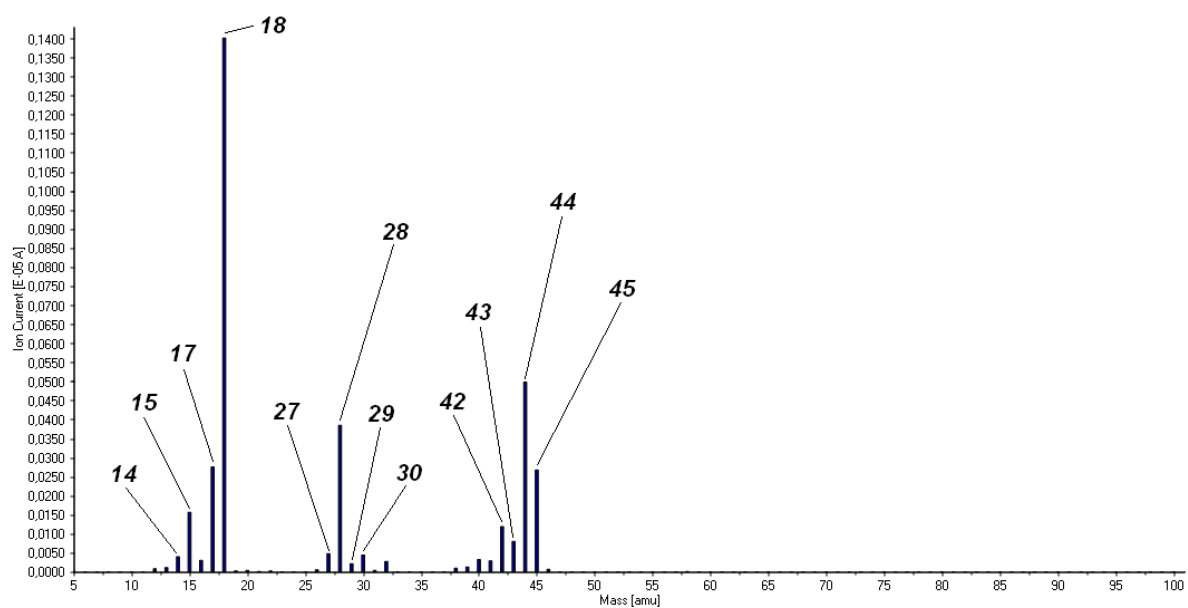
ESI Fig.14. Evolved gas mass spectra of gaseous compounds formed from compound **1a** on heating under inert atmosphere



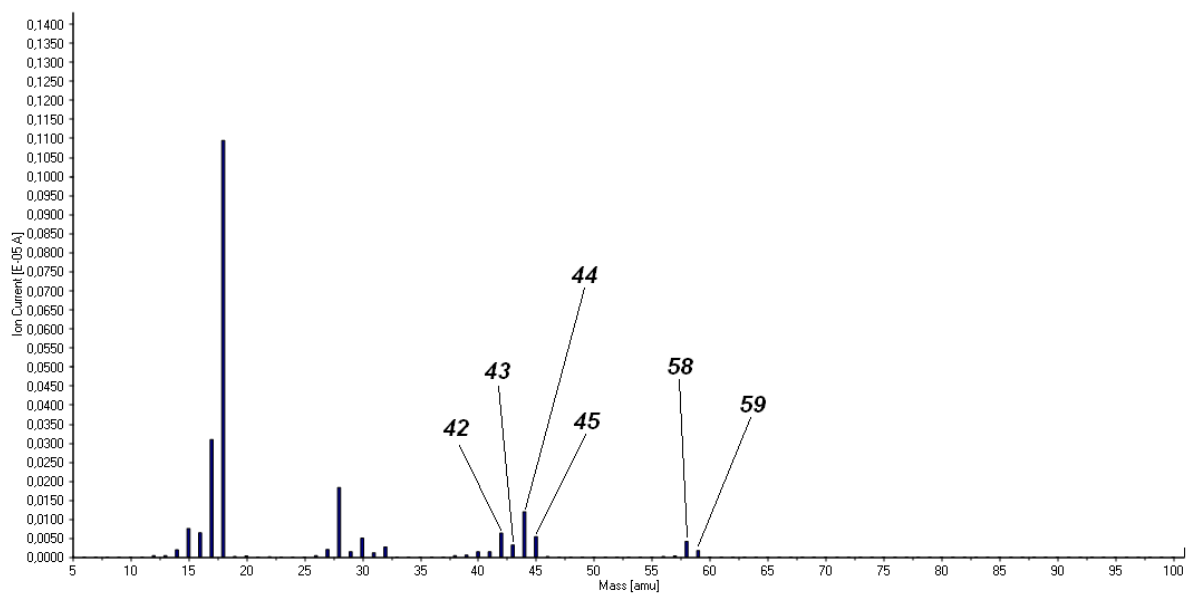
a) T=35 °C



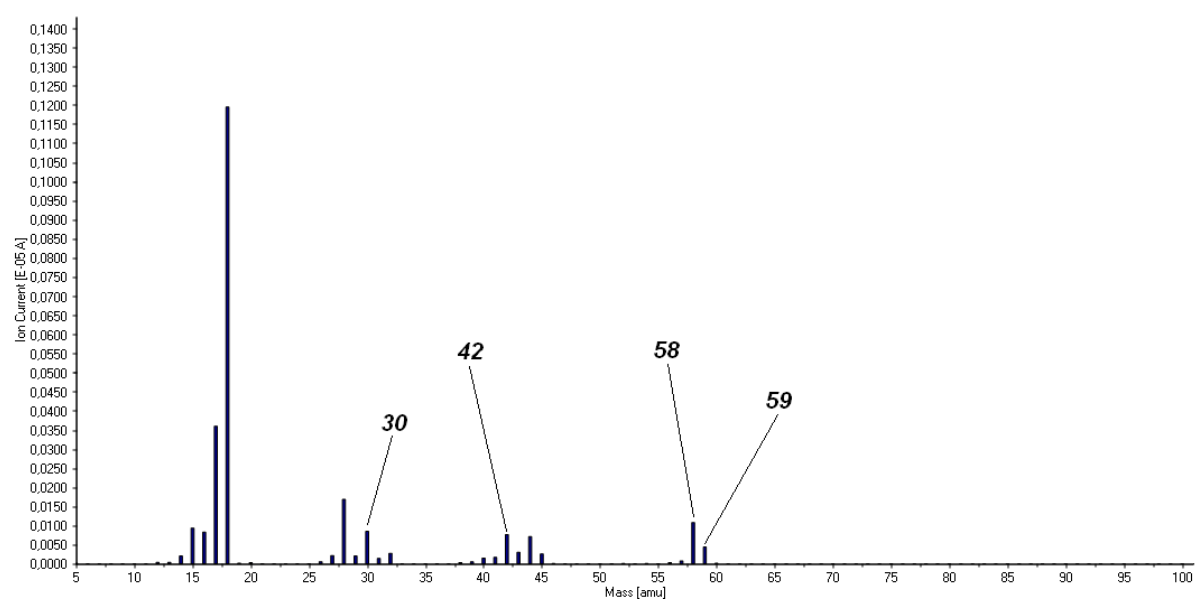
b) 144 °C



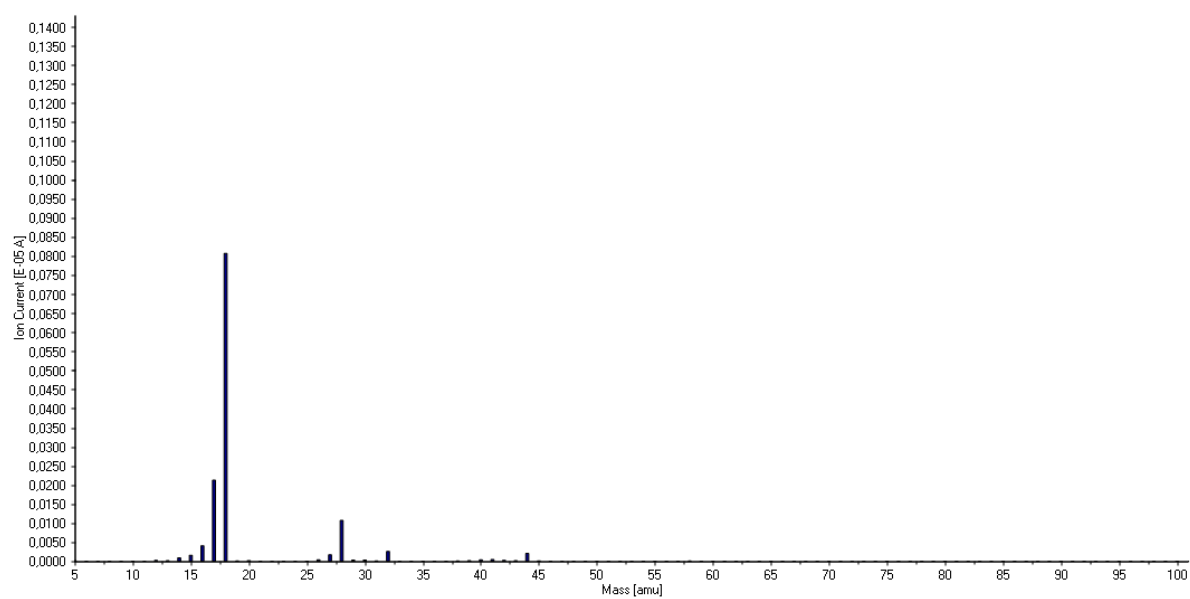
c) 203 °C



d) 348 °C

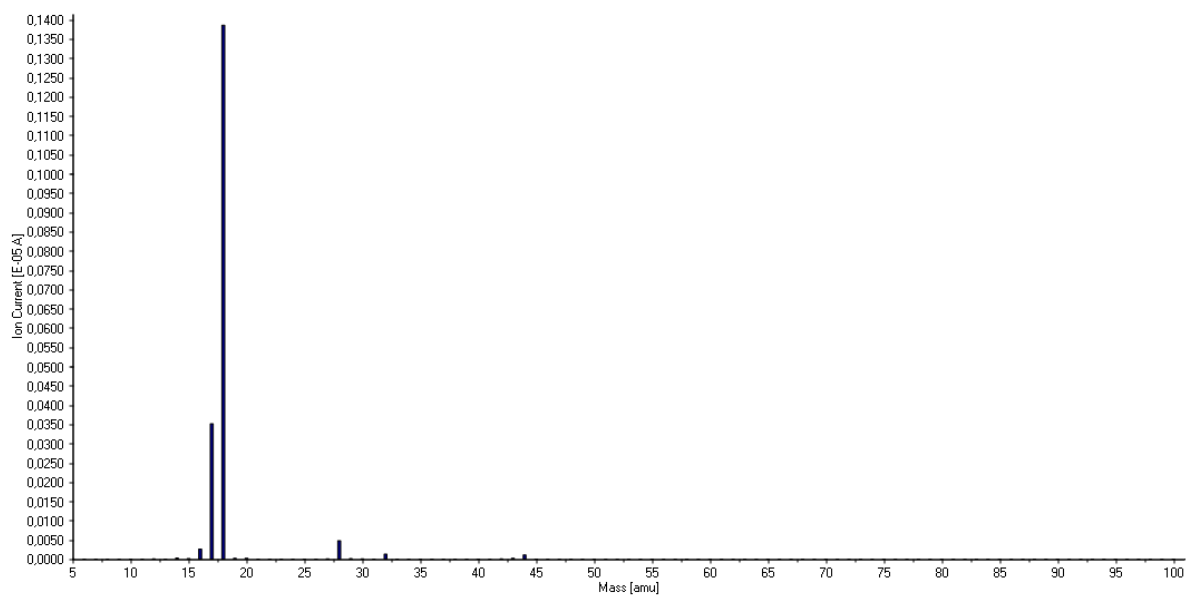


e) 365 °C

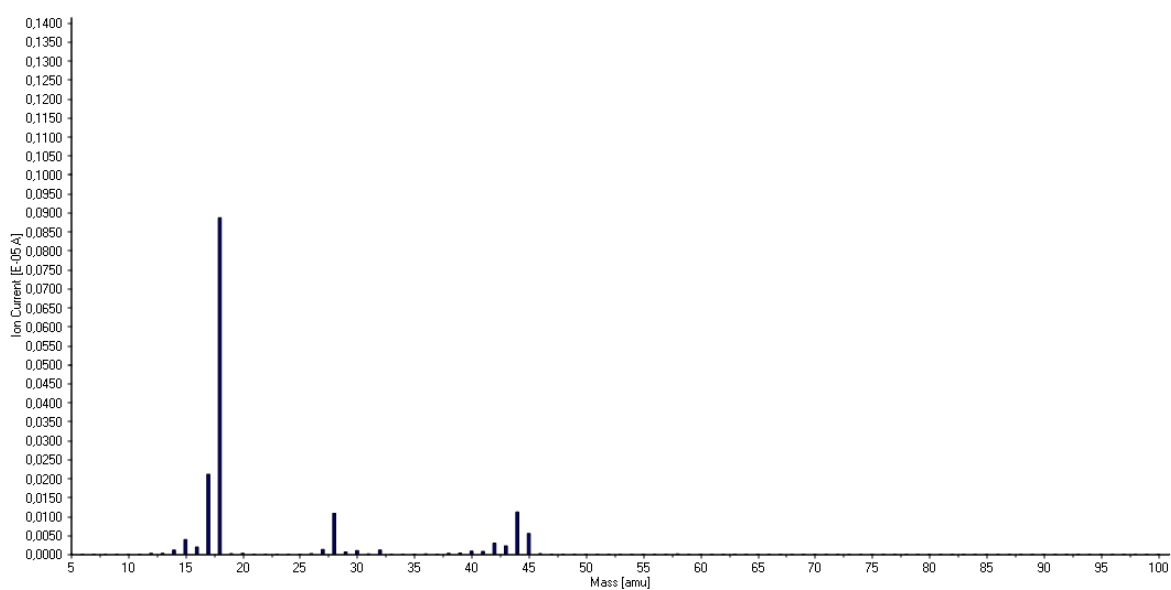


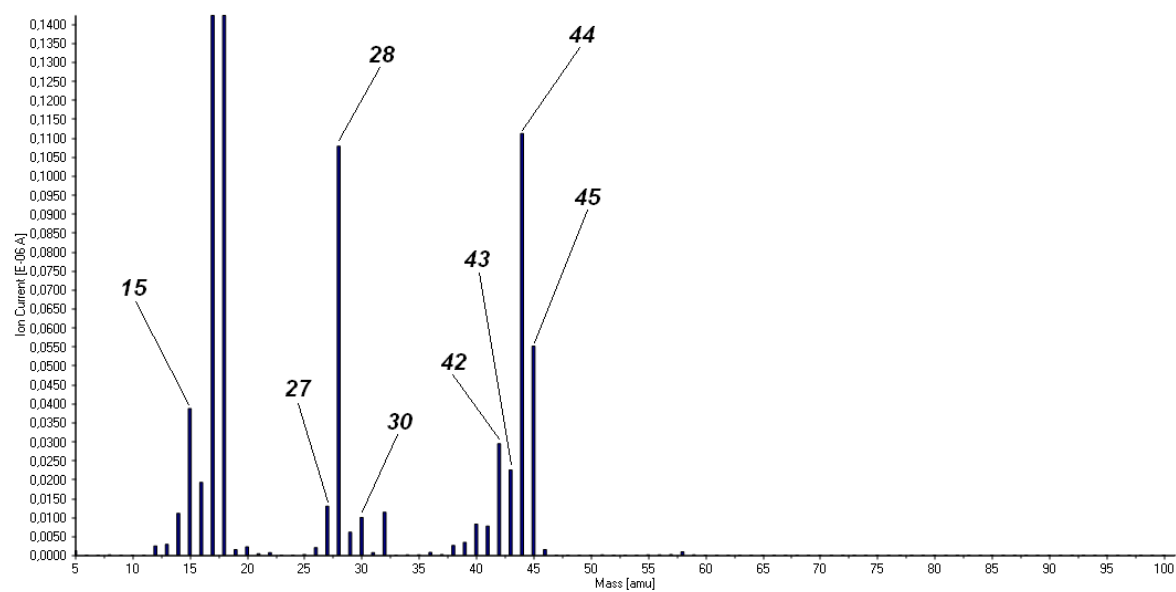
f) 441 °C

ESI Fig.15 . Evolved gas mass spectra of gaseous compounds formed from compound **1b** on heating under inert atmosphere

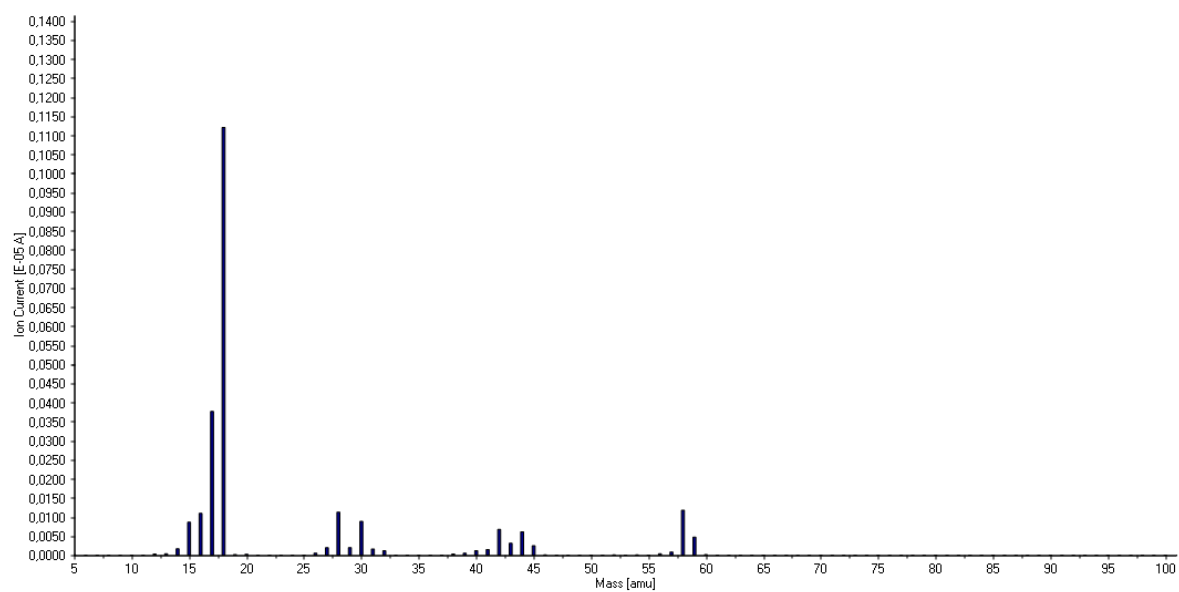


a) T=119 °C

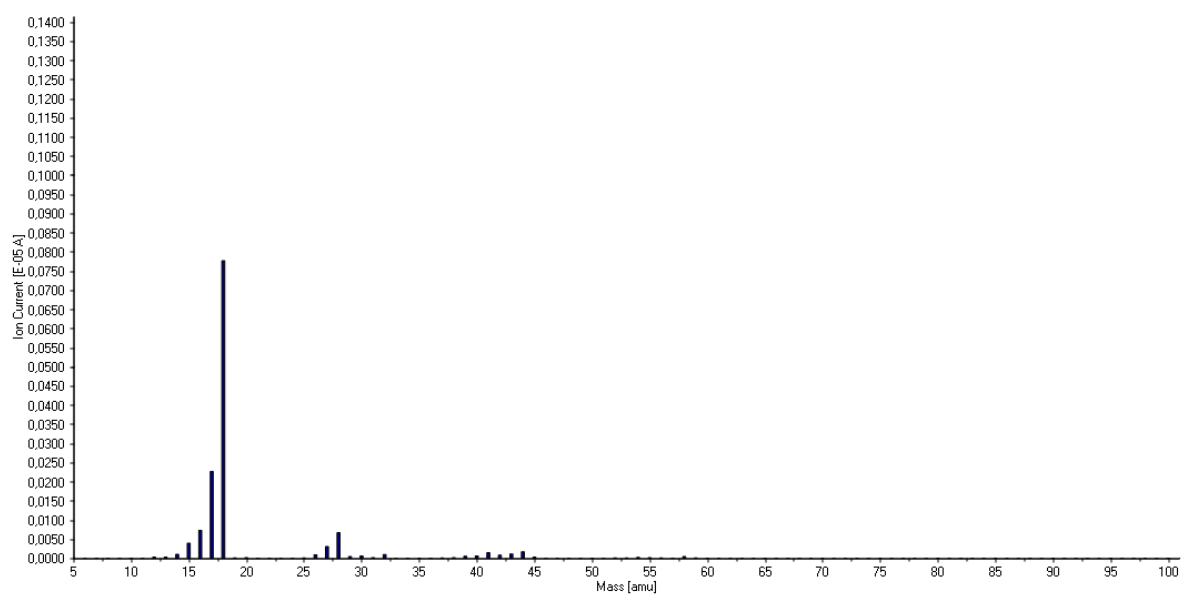




c) 236 °C

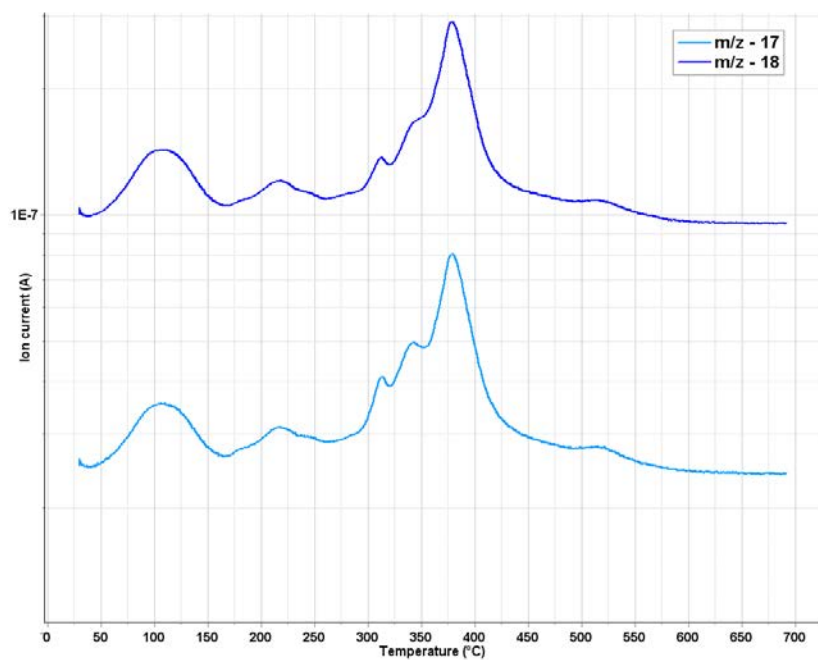


d) T=379 °C

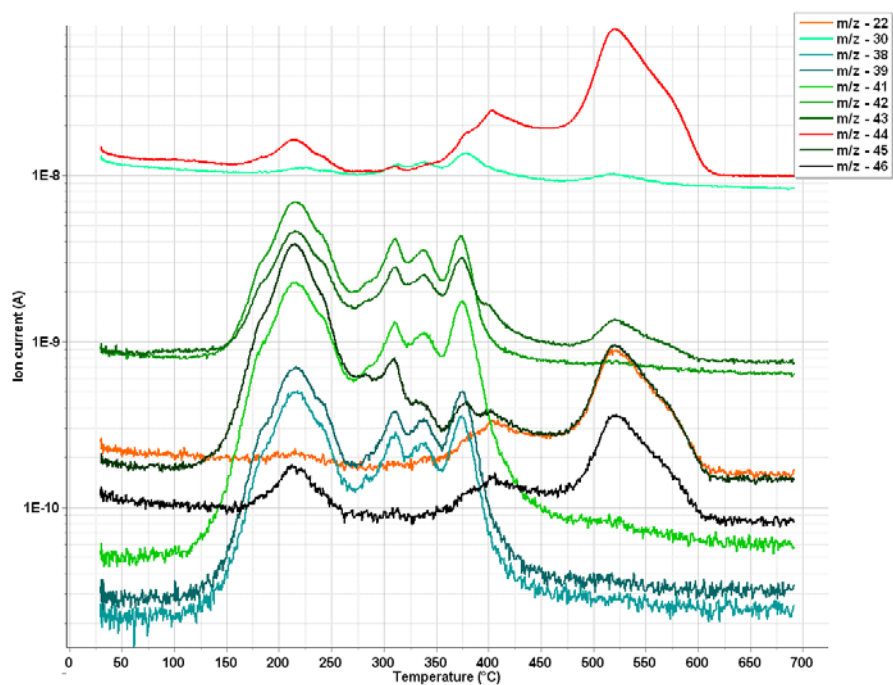


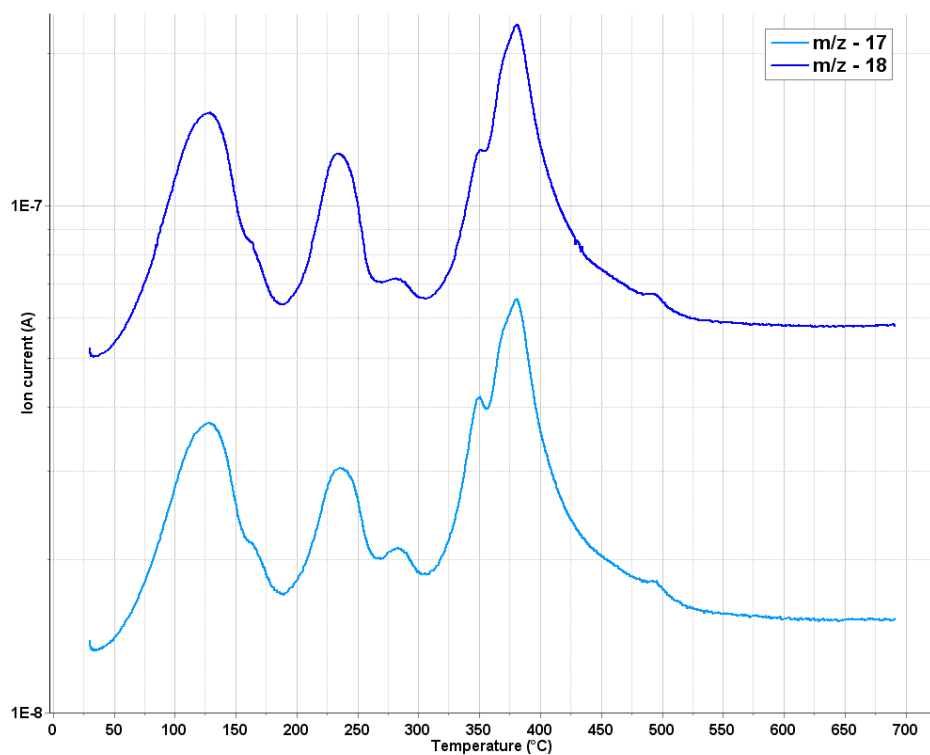
e) 420 °C

ESI Fig. 16. TG-MS curve of compound **1a** in **1b** in air with 5°/h heating rate



a)

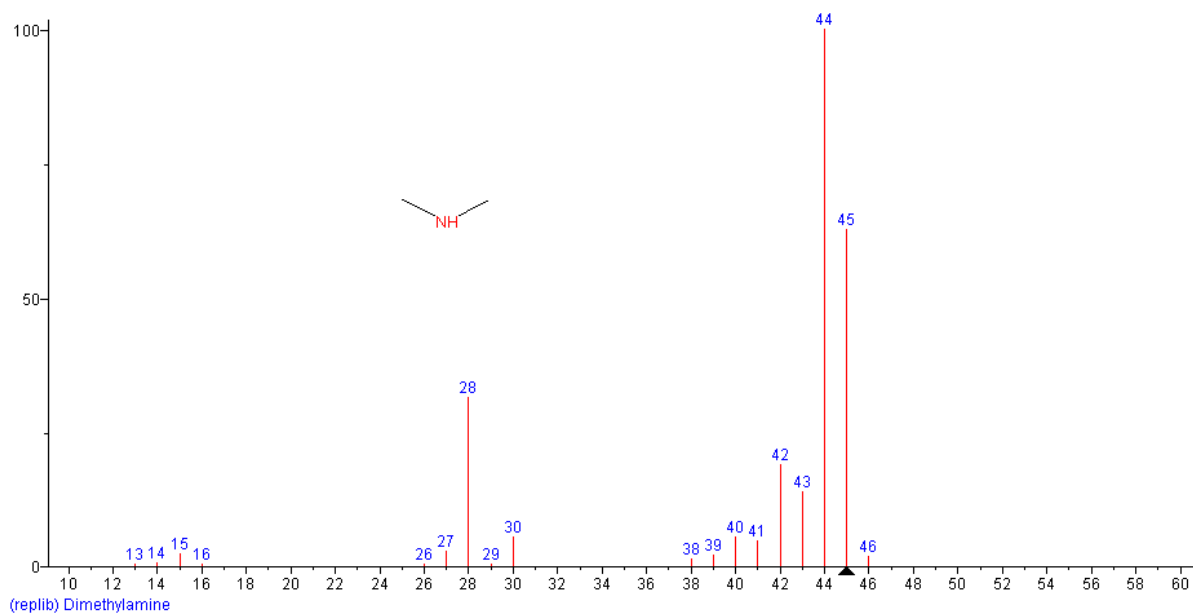




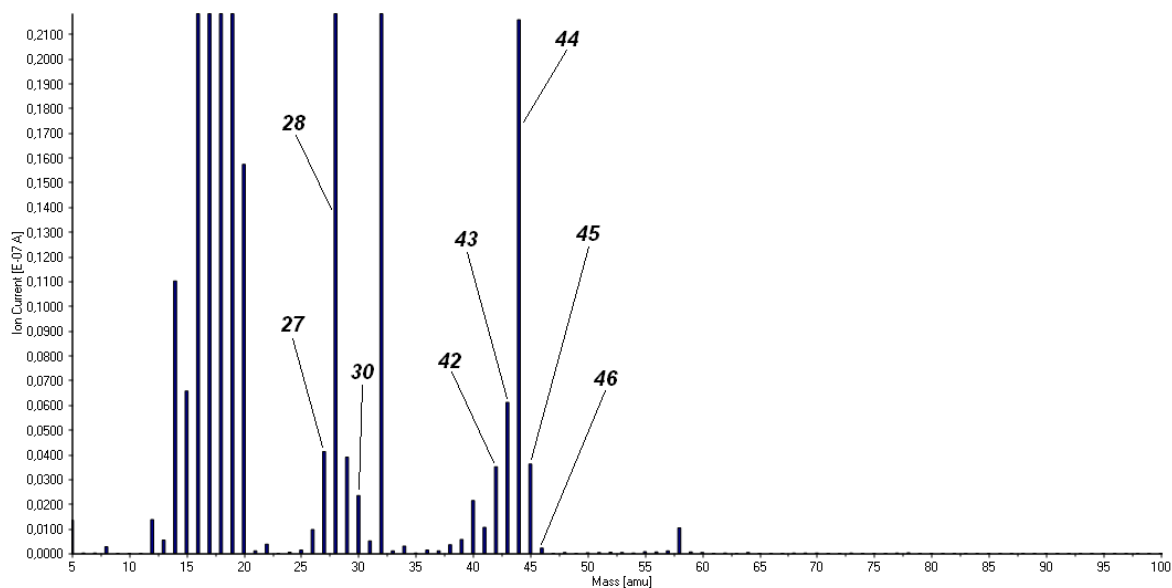
b)



ESI Fig.17. Dimethylamine NIST reference (a) and analog mass spectrum at 98 °C with our instrument (b)

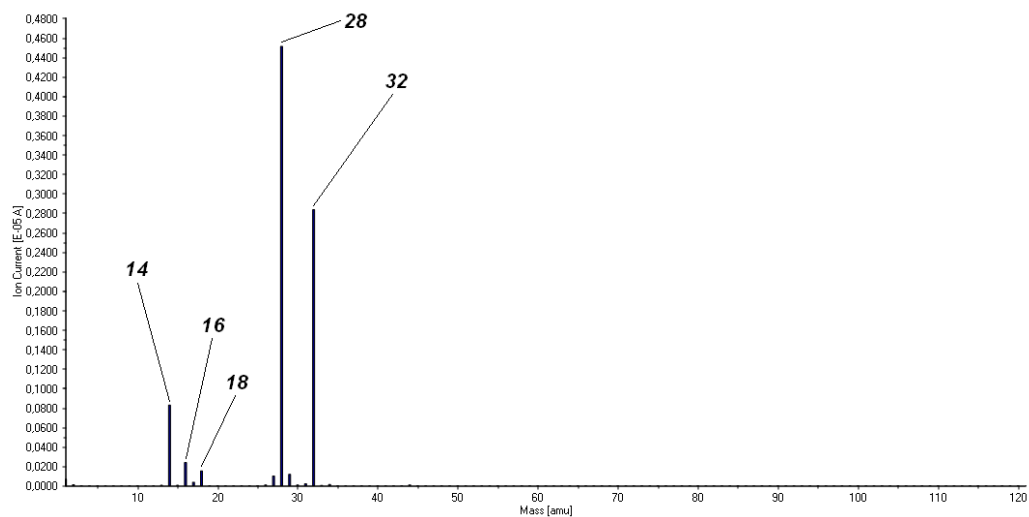


a) Dimethylamine NIST MS spectrum

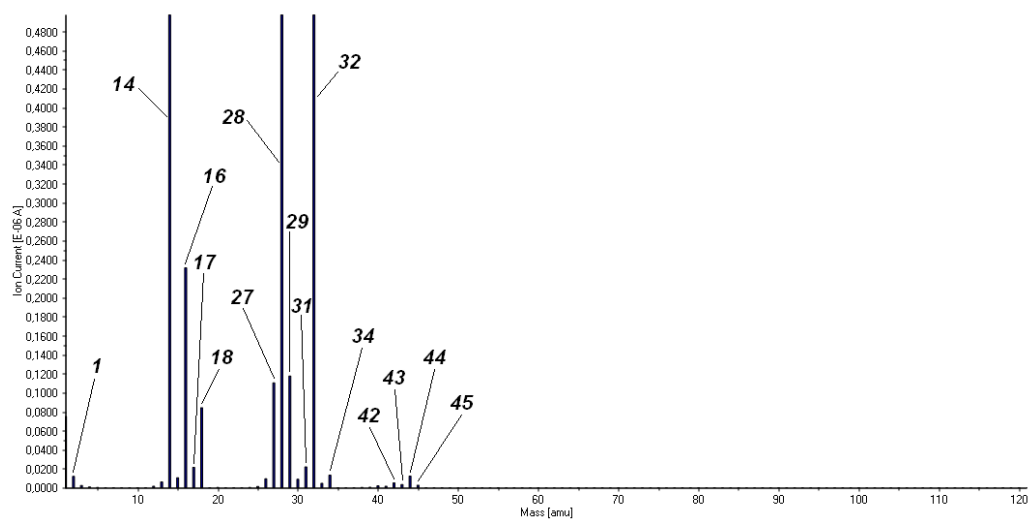


b) Dimethylamine MS spectrum measured by us in inert atmosphere

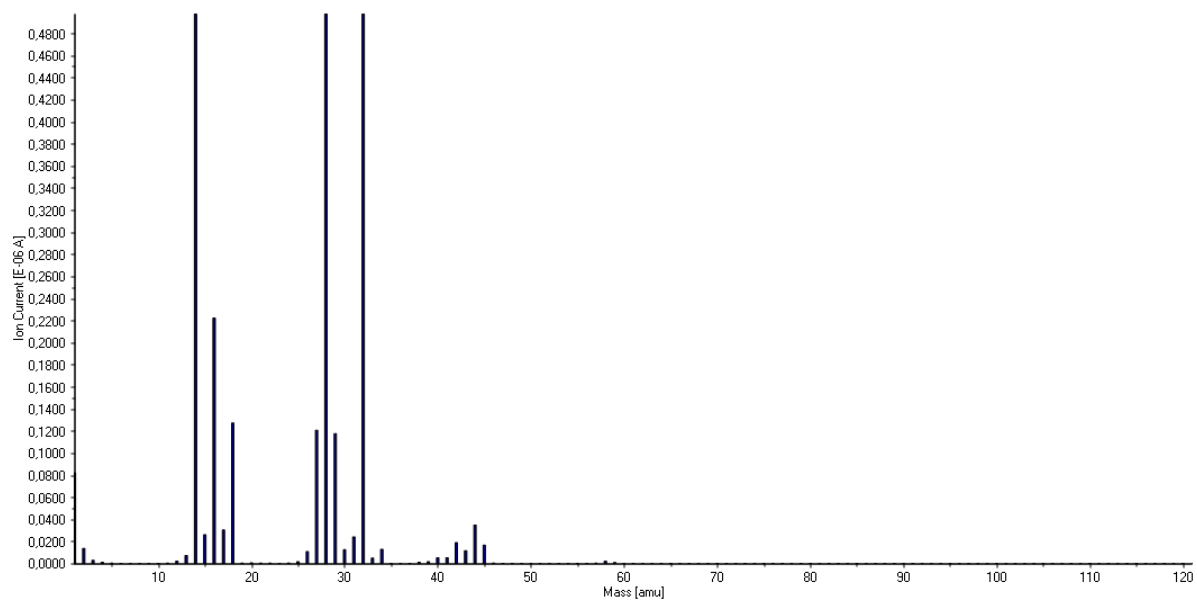
ESI Fig.18 . Evolved gas mass spectra of gaseous compounds formed from compound **1b** on heating in synthetic air atmosphere



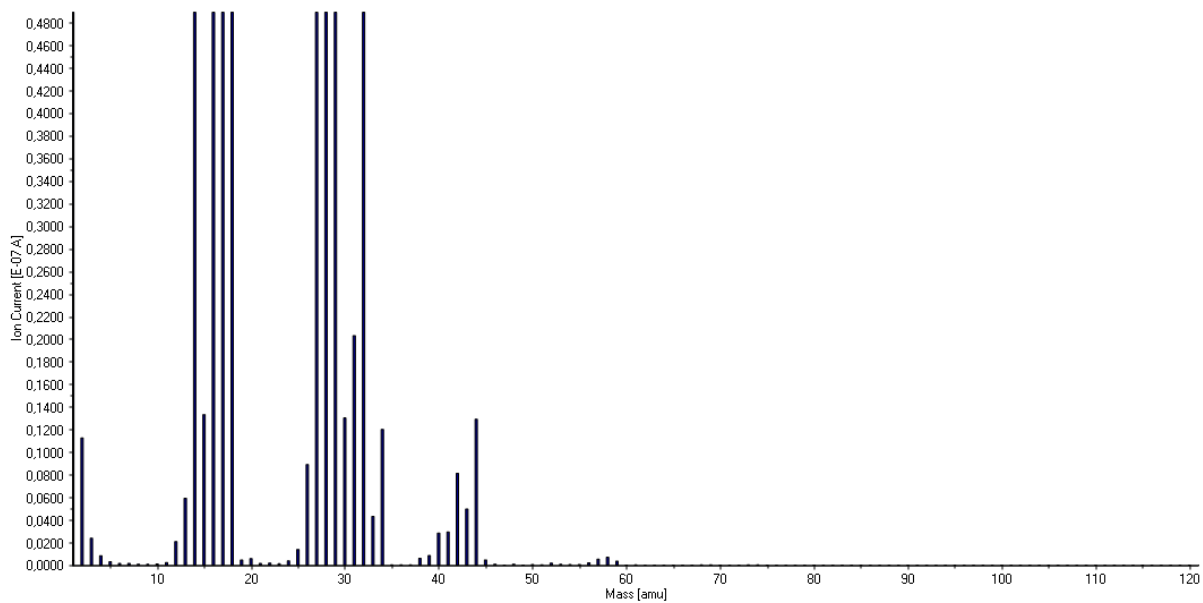
a) 131 °C



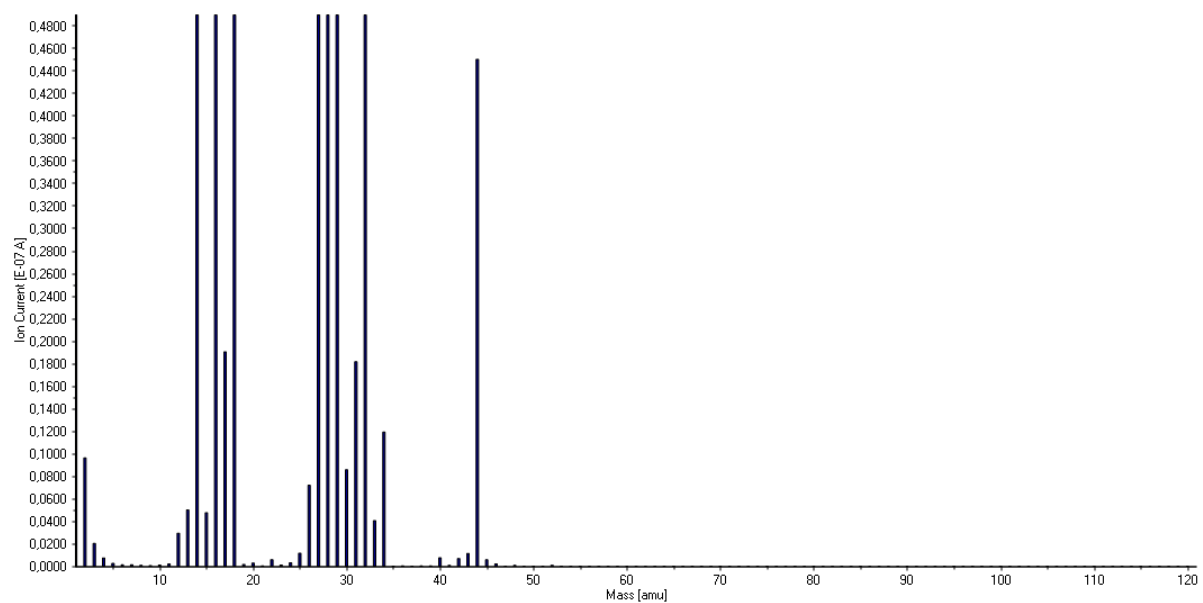
b) 166 °C (zoom 10x)



c) 236 °C (10x zoom)



d) 373 °C (100 x zoom)

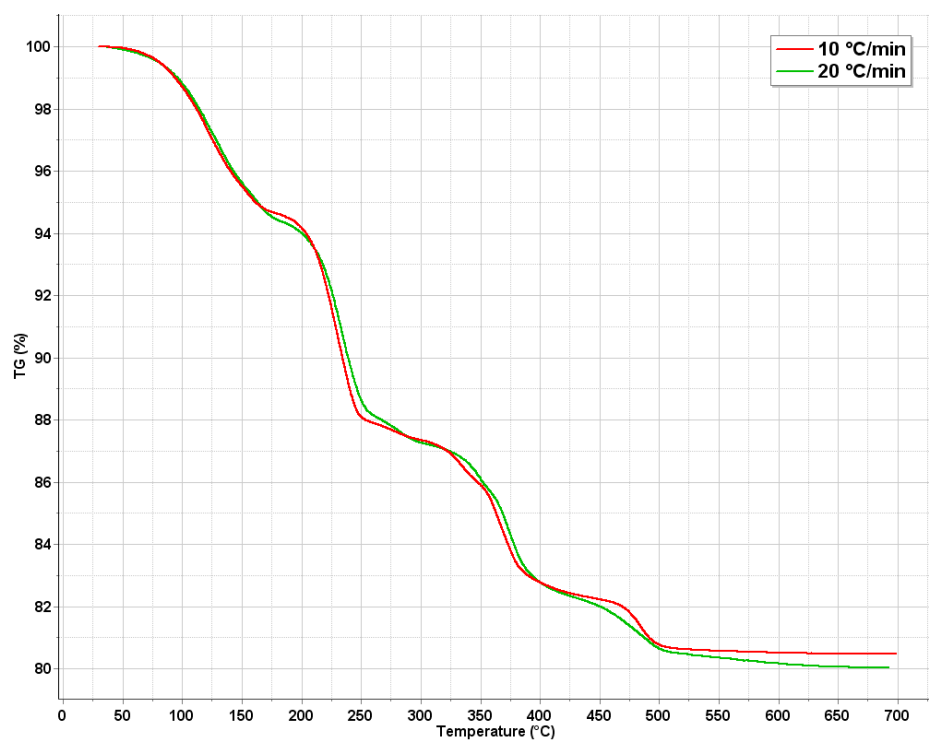


e) 471 °C (100 x zoom)

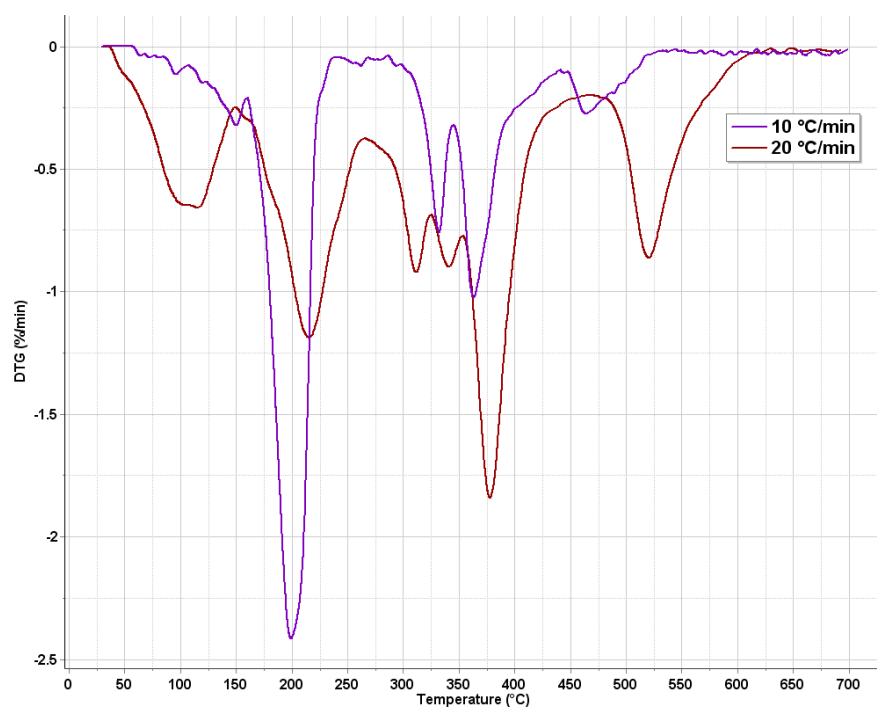
ESI Figure 19. The effect of heating rate (10 and 20 °C/min) on the decomposition characteristics of compounds 1a and 1b in air



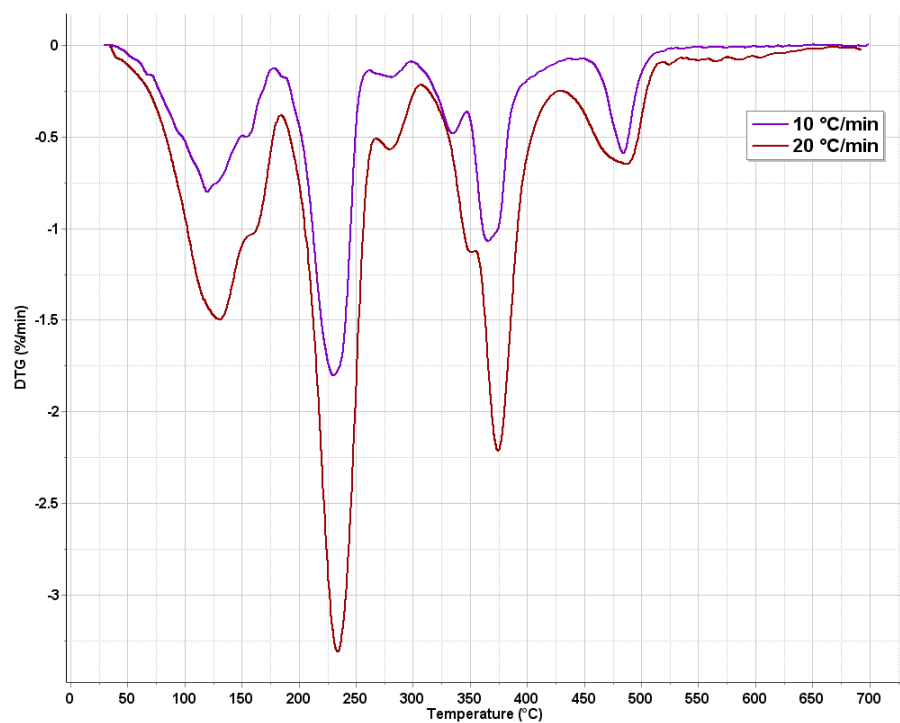
a) TG curves of compound 1a in air



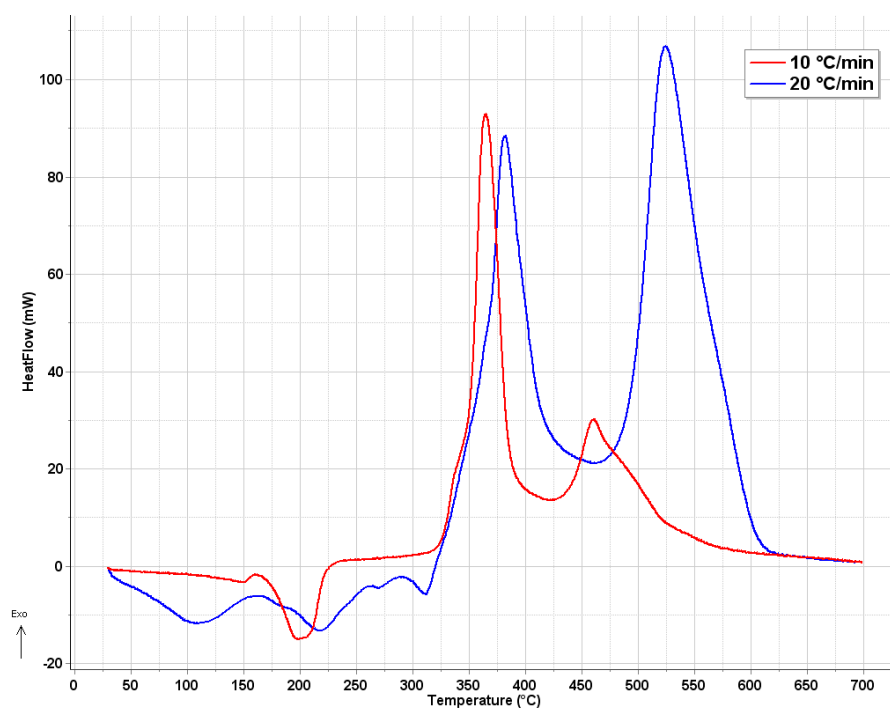
b) TG curves of compound **1b** in air



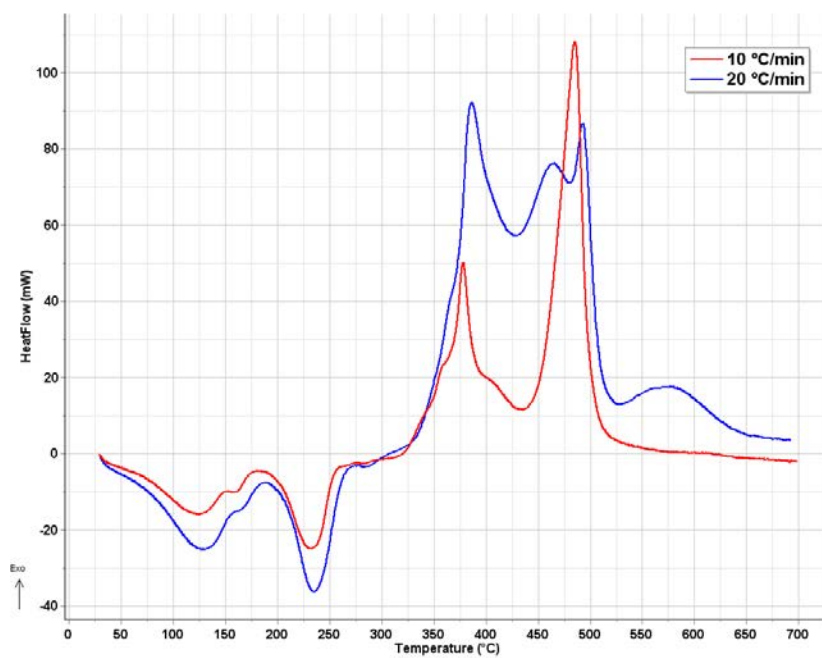
c) DTG of compound **1a** in air



d) DTG of compound **1b** in air



e) DSC of compound **1a** in air



f) DSC of compound **1b** in air