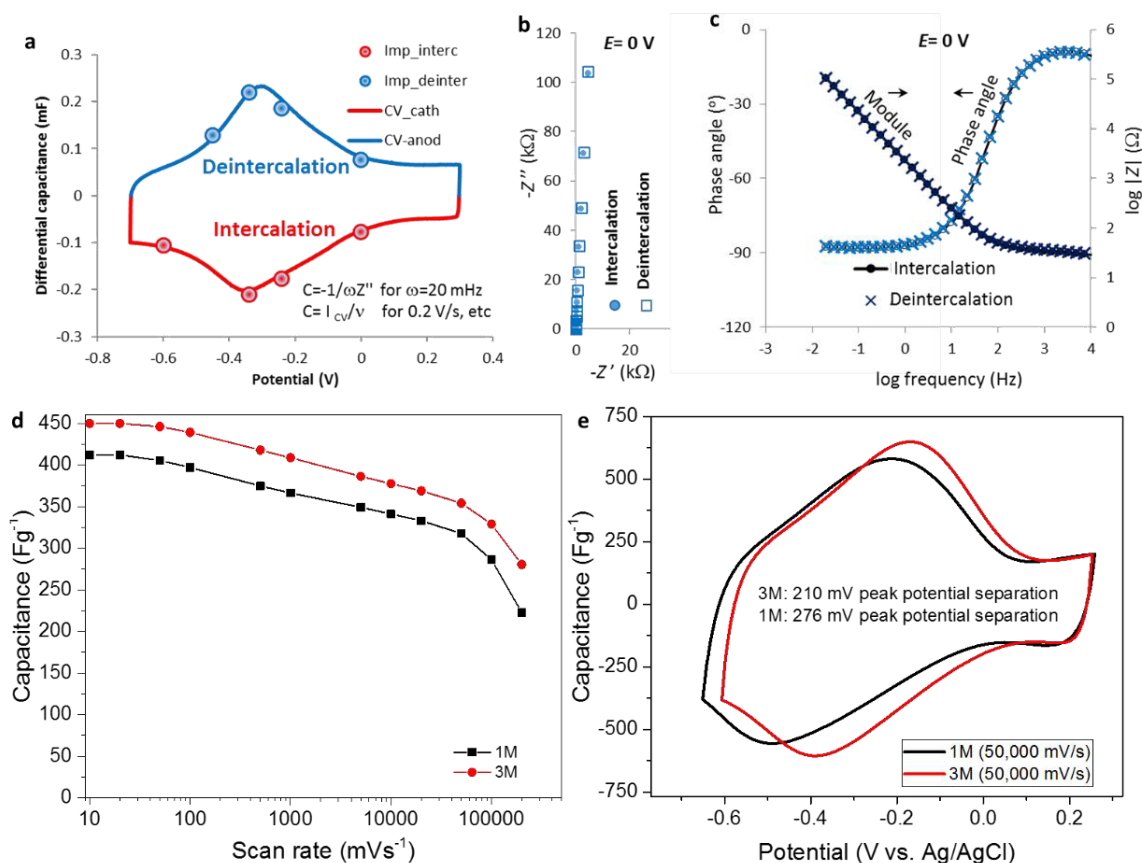
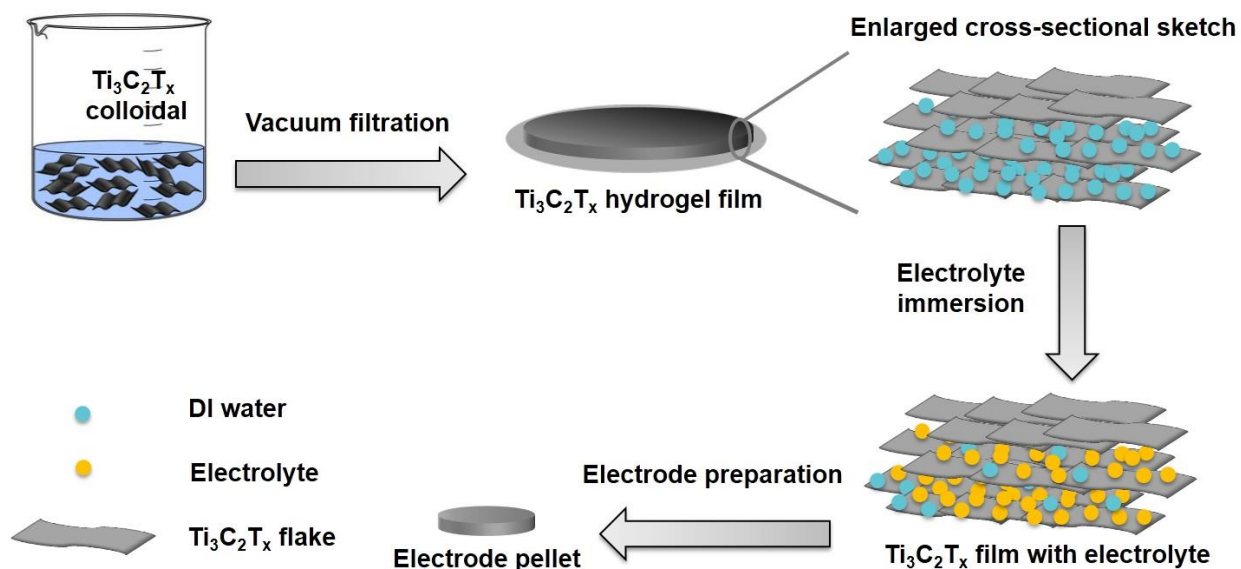


In the format provided by the authors and unedited.

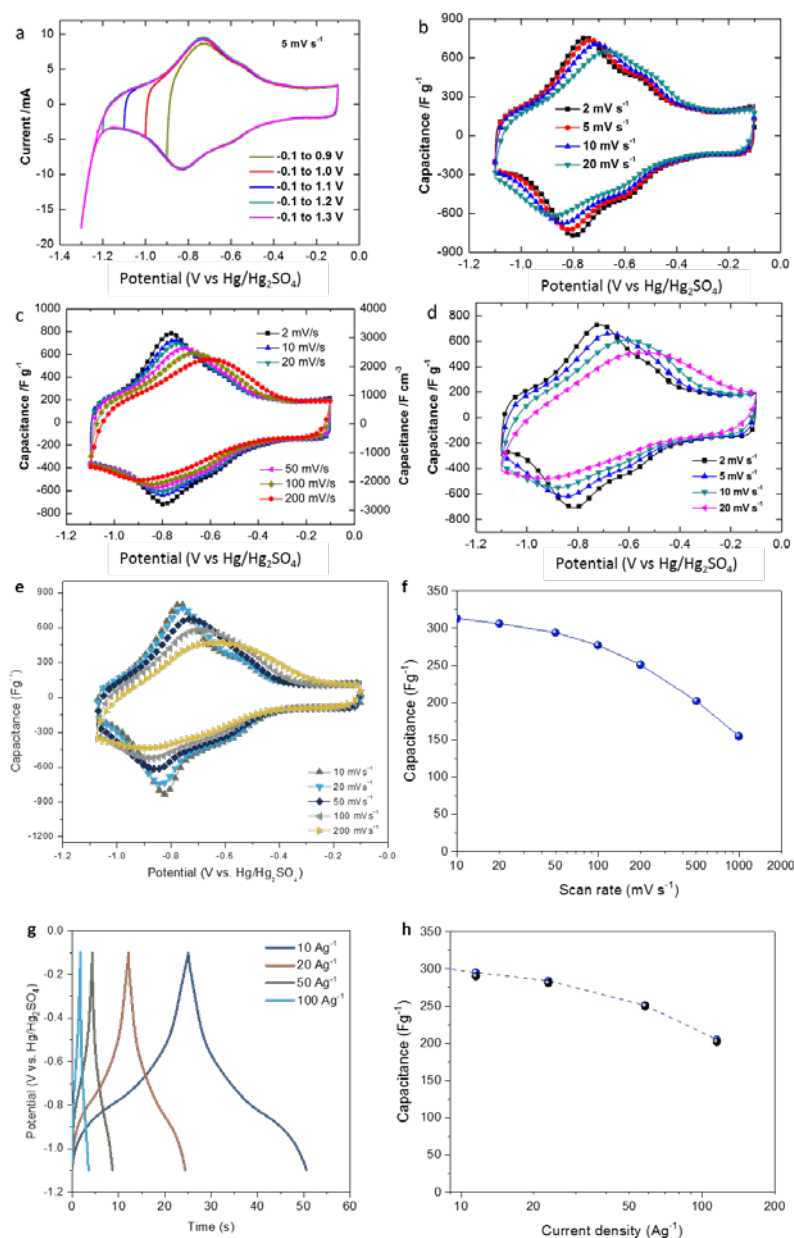
Ultra-high-rate pseudocapacitive energy storage in two-dimensional transition metal carbides



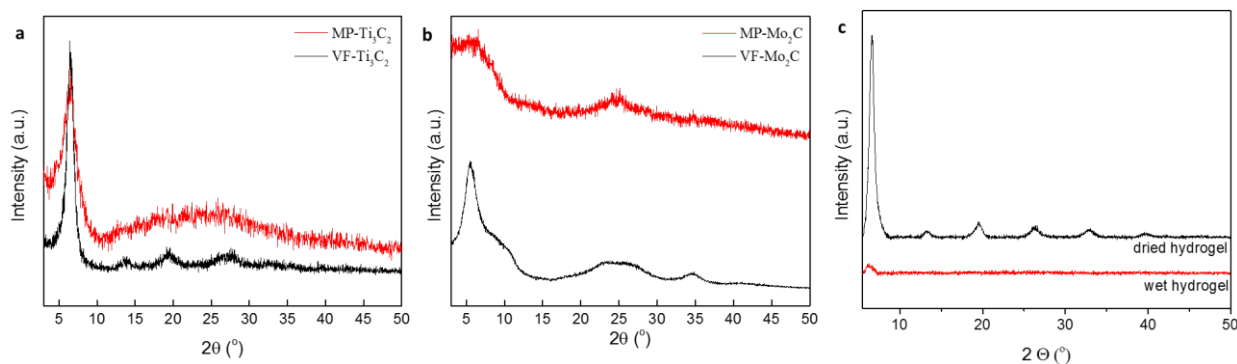
Supplementary Figure 1 | Electrochemical performance of thin $\text{Ti}_3\text{C}_2\text{T}_x$ films. **a**, Cyclic voltammetry profile (line) with over-imposed values of capacitance extracted from EIS at different intercalation (red spheres) and deintercalation (blue spheres) potentials. **b**, EIS data (Nyquist plot) collected at 0 V vs. Ag/AgCl on the intercalation (filled spheres) and deintercalation (hollow squares) branches. **c**, Bode plot collected at 0 V vs. Ag/AgCl. **c**, Gravimetric rate performance extracted from cyclic voltammetry profiles of a 90 nm film on glassy carbon electrode in 1M (black squares) and 3M (red circles) H_2SO_4 . **d**, Cyclic voltammetry profiles collected at 50,000 mV s^{-1} in 1M (black) and 3M H_2SO_4 (red).



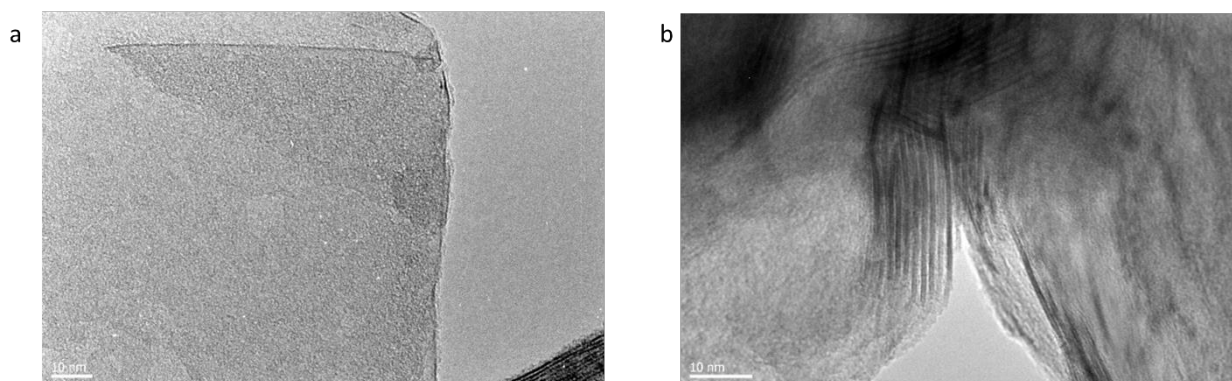
Supplementary Figure 2 | Schematic illustration of $\text{Ti}_3\text{C}_2\text{T}_x/\text{H}_2\text{SO}_4$ hydrogel preparation procedure. $\text{Ti}_3\text{C}_2\text{T}_x$ colloidal solution was taken for vacuum filtration to prepare the $\text{Ti}_3\text{C}_2\text{T}_x$ hydrogel film. The obtained hydrogel film was then immersed in 3 M H_2SO_4 electrolyte to replace the water and increase/ensure accessibility of MXene surface to the electrolytes. Electrode pellet was punched out from the $\text{Ti}_3\text{C}_2\text{T}_x$ film after electrolyte immersion for more than 24 h.



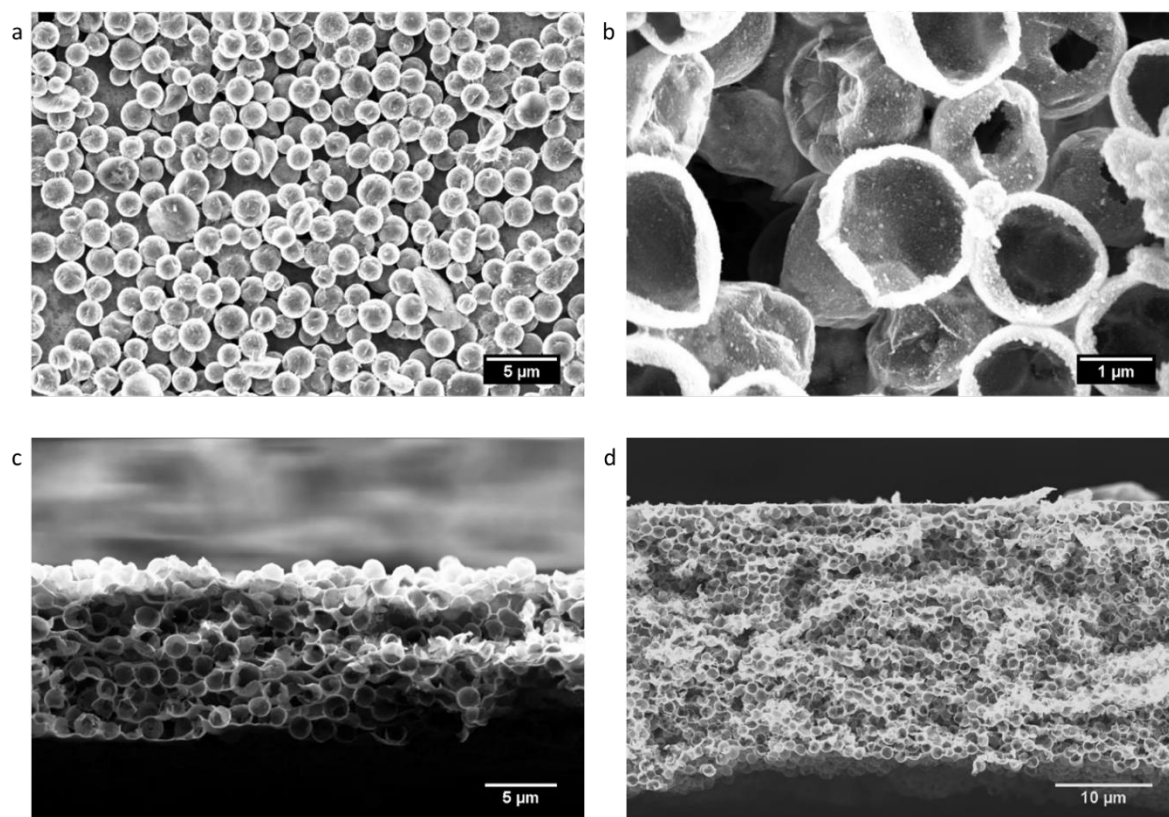
Supplementary Figure 3 | Electrochemical performance of $\text{Ti}_3\text{C}_2\text{T}_x$ hydrogel films. **a**, Cyclic voltammetry profiles collected at 5 mV s^{-1} in different potential ranges (active material mass loading of 5.3 mg/cm^2 , $13 \text{ }\mu\text{m}$ thick). **b**, Cyclic voltammetry profiles collected at scan rates from 2 to 20 mV s^{-1} (active material mass loading of 5.3 mg/cm^2 , $13 \text{ }\mu\text{m}$ thick). **c**, Cyclic voltammetry profiles collected at scan rates from 2 to 200 mV s^{-1} (active material mass loading of 1.2 mg/cm^2 , $3 \text{ }\mu\text{m}$ thick). **d**, Cyclic voltammetry profiles collected at scan rates from 2 to 20 mV s^{-1} (active material mass loading of 11.3 mg/cm^2 , $40 \text{ }\mu\text{m}$ thick). **e-h**: Rate performance of a hydrogel sample with mass loading of 2.65 mg/cm^2 : **e**, Cyclic voltammetry profiles collected at scan rates from 10 to 200 mV s^{-1} . **f**, Gravimetric rate performance (from CV): capacitance vs. scan rate. **g**, Galvanostatic cycling profiles collected at 10, 20, 50 and 100 Ag^{-1} . **h**, Gravimetric rate performance (from GCPL): capacitance vs. charge/discharge current density.



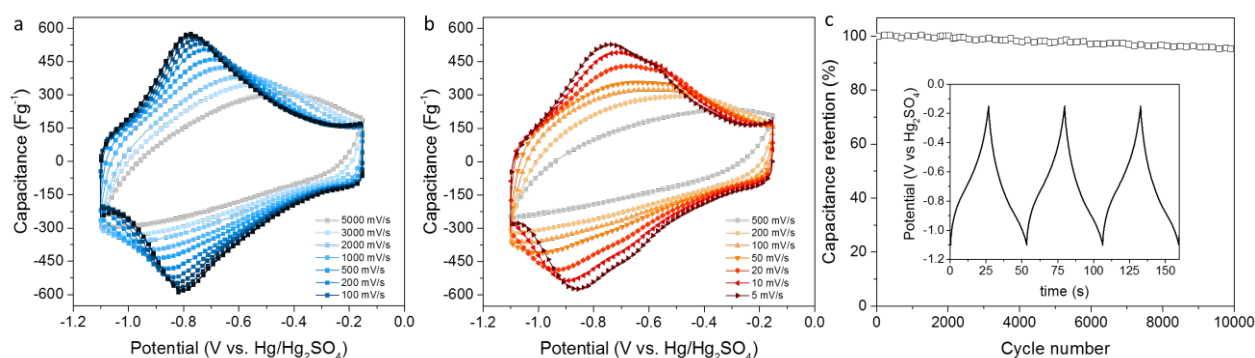
Supplementary Figure 4 | X-ray diffractograms of a, $\text{Ti}_3\text{C}_2\text{T}_x$ MXene macroporous and vacuum-filtered freestanding films. b, Mo_2CT_x MXenes macroporous and vacuum-filtered freestanding films. c, dry and wet hydrogel.



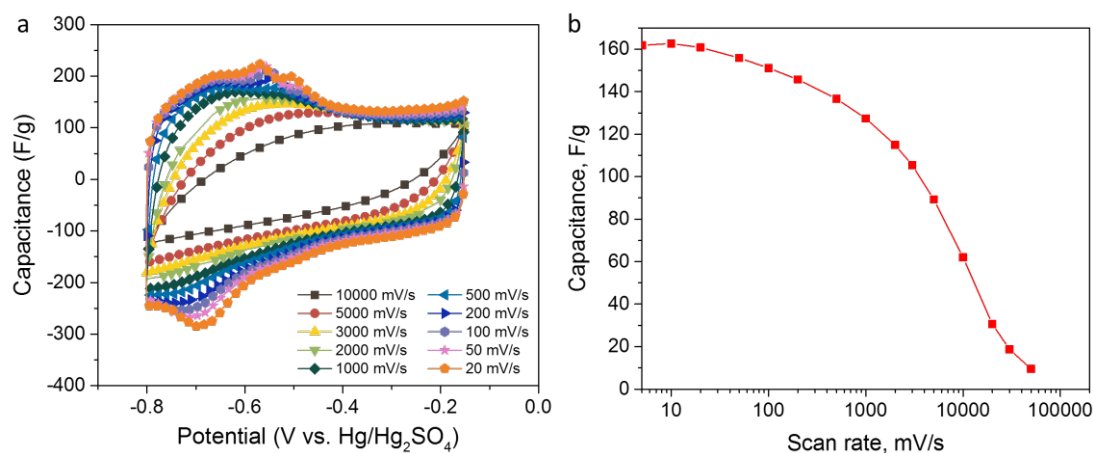
Supplementary Figure 5 | TEM micrographs of hollow Ti_3C_2 spheres. a, edge of an individual $\text{Ti}_3\text{C}_2\text{T}_x$ flake, confirming presence of single-layer flakes and b, the connection point of two $\text{Ti}_3\text{C}_2\text{T}_x$ hollow spheres. They are with few layers and share some of them, indicating a good connection. This is important for the charge transfer throughout the entire macroporous framework.



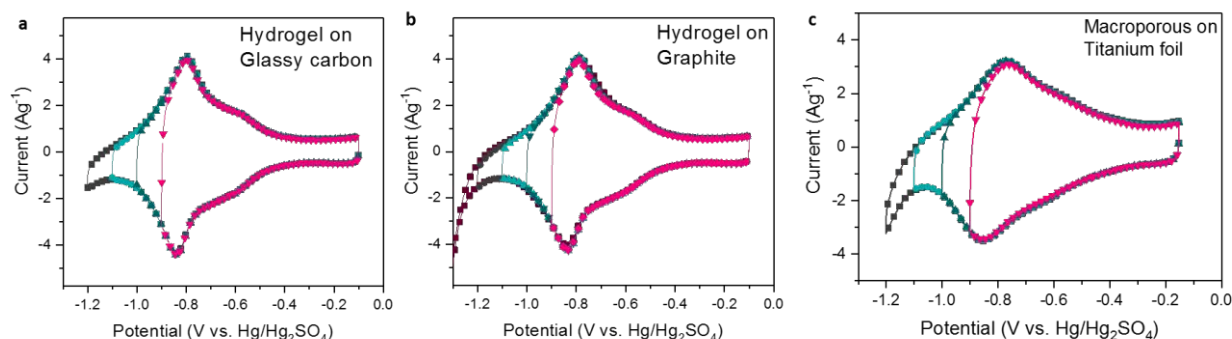
Supplementary Figure 6 | Scanning electron micrographs of macroporous MXene films. a, few-layer $\text{Ti}_3\text{C}_2\text{T}_x$ wrapping PMMA spheres, **b**, $\text{Ti}_3\text{C}_2\text{T}_x$ wrapped PMMA after annealing in Ar at 400 °C, and **c,d**, cross-sections of $\text{Ti}_3\text{C}_2\text{T}_x$ and Mo_2CT_x films, respectively.



Supplementary Figure 7 | a,b, Cyclic voltammetry profiles collected at different scan rates for macroporous $\text{Ti}_3\text{C}_2\text{T}_x$ films: **a**, 25 μm thick, loading of 0.9 mg/cm^2 . **b**, 180 μm thick electrode, loading 4.3 mg/cm^2 . **c**, Capacitance retention test of a 180 μm thick macroporous electrode performed by galvanostatic cycling at 10 Ag^{-1} . Inset depicts galvanostatic cycling profiles collected at 10 Ag^{-1} .



Supplementary Figure 8 | Electrochemical performance of macroporous Mo_2CT_x . **a**, Cyclic voltammetry profiles of a $30 \mu\text{m}$ thick macroporous Mo_2CT_x film with 0.6 mg/cm^2 loading collected in $3 \text{ M H}_2\text{SO}_4$ at scan rates from 20 mV/s to 10000 mV/s , **b**, Rate performance of macroporous Mo_2CT_x MXene film represented as mass-normalized capacitance.



Supplementary Figure 9 | Electrochemical stability window for $\text{Ti}_3\text{C}_2\text{T}_x$ electrodes collected at 5 mV/s^{-1} in $3 \text{ M H}_2\text{SO}_4$ with different current collectors. Cyclic voltammetry profiles for hydrogel $\text{Ti}_3\text{C}_2\text{T}_x$ electrodes (2.65 mg/cm^2) on **a**, glassy carbon and **b**, graphite current collector. **c**, Cyclic voltammetry profiles for macroporous $\text{Ti}_3\text{C}_2\text{T}_x$ electrode (4.3 mg/cm^2) on Ti foil.