

Supplementary Information

CVD-Grown Monolayer MoS₂ in Bioabsorbable Electronics and Biosensors

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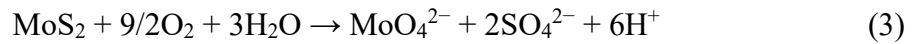
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Supplementary Notes

Supplementary Note 1. The possible reactions during MoS₂ biodegradation process in PBS solution

Since the dissolution tests of monolayer MoS₂ in this work involved PBS solution, which contains disodium hydrogen phosphate (Na₂HPO₄), sodium chloride (NaCl), potassium chloride (KCl), potassium dihydrogen phosphate (KH₂PO₄), O₂, and H₂O, likely reactions are^{1,2}:



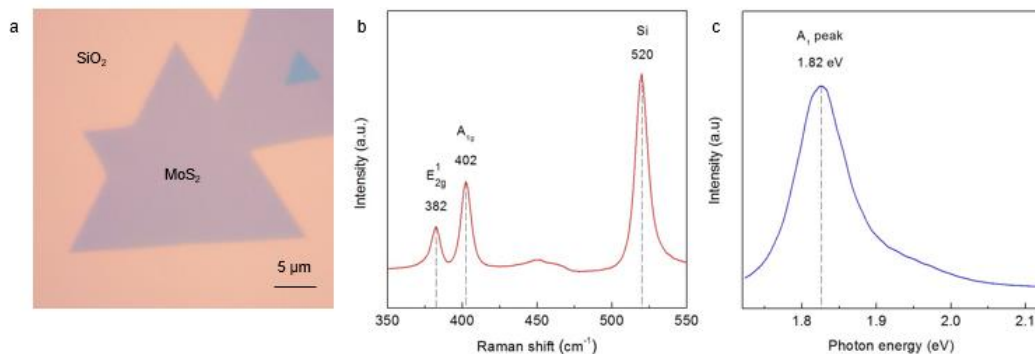
Where M = Na or K.

Besides, the empirical kinetic equation affected by pH level in PBS solution is³:

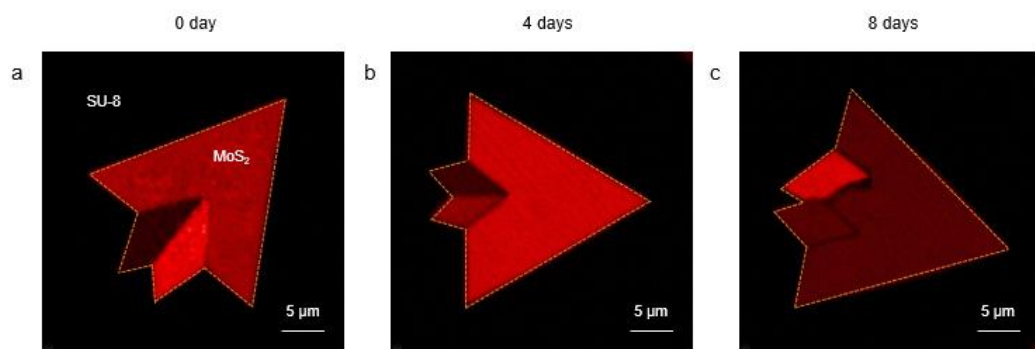
$$d[MoS_2]/[MoS_2]dt = 1.9*[OH^-]^{0.17}, 1/d \quad (4)$$

where the units of the dissolution rate and OH⁻¹ concentration are days⁻¹ and mol/L, respectively.

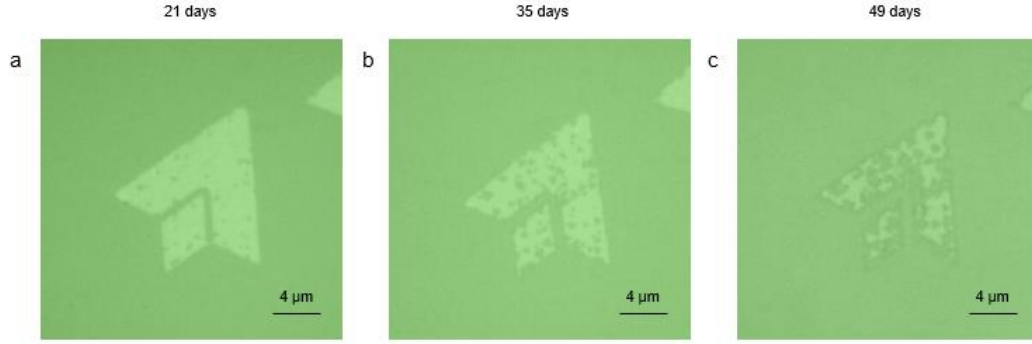
Supplementary Figures



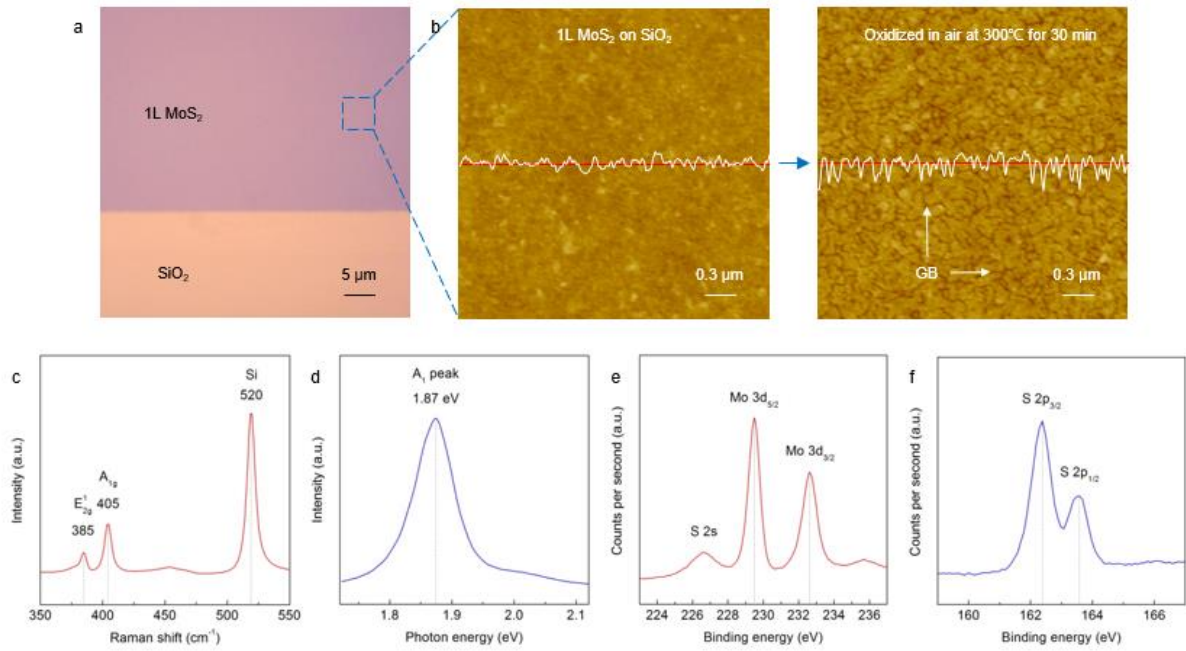
Supplementary Figure 1 | The optical characterization of as-grown monolayer MoS₂ crystals. **a**, OM image of isolated polycrystalline MoS₂ monolayer fabricated on SiO₂ substrate by using APCVD method; **b,c**, Raman (**b**) and PL (**c**) spectra of the corresponding MoS₂ monolayer.



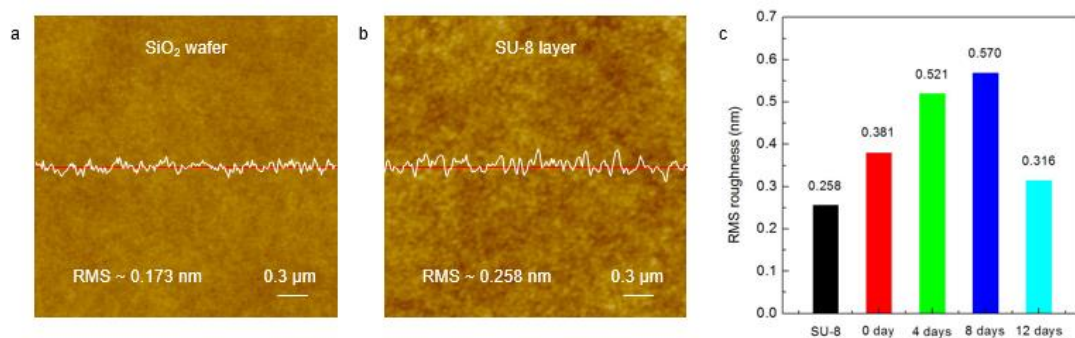
Supplementary Figure 2 | SHG characterization of monolayer MoS₂ crystals with different dissolution time. **a-c** SHG mapping images of MoS₂ crystals on SU-8/sapphire substrate kept in PBS solution at 75 °C within 0 day (**a**), 4 days (**b**) and 8 days (**c**).



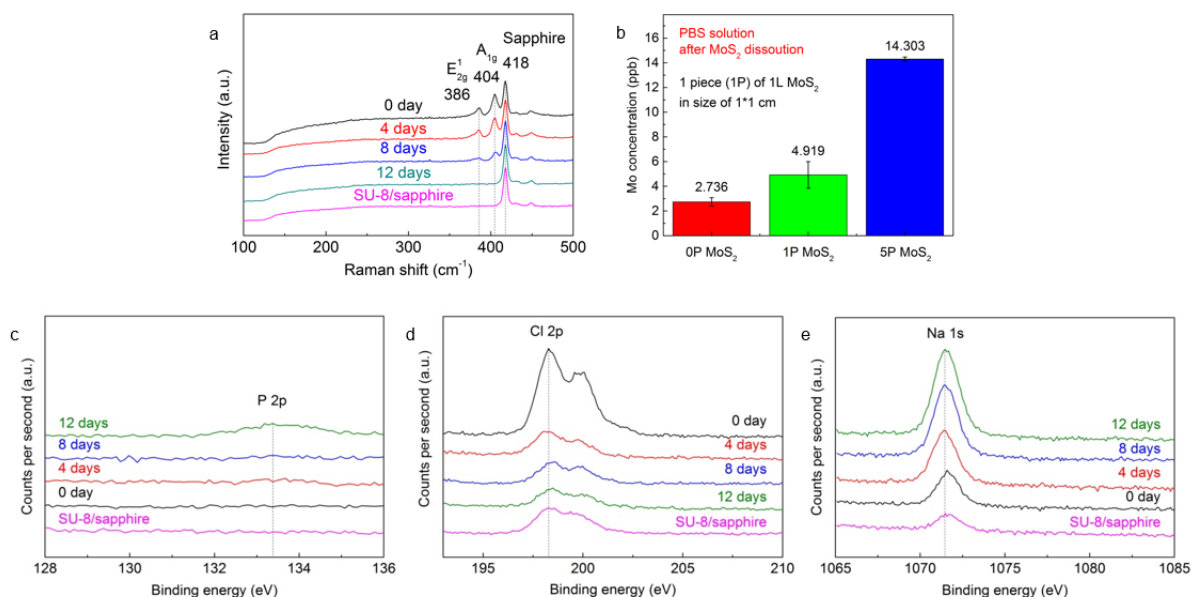
Supplementary Figure 3 | Morphology evolution of monolayer MoS₂ crystals with different dissolution time. a-c OM images of MoS₂ crystals on SU-8/sapphire substrate after kept in PBS solution at 75 °C for 21 (a), 35 (b) and 49 (c) days, respectively.



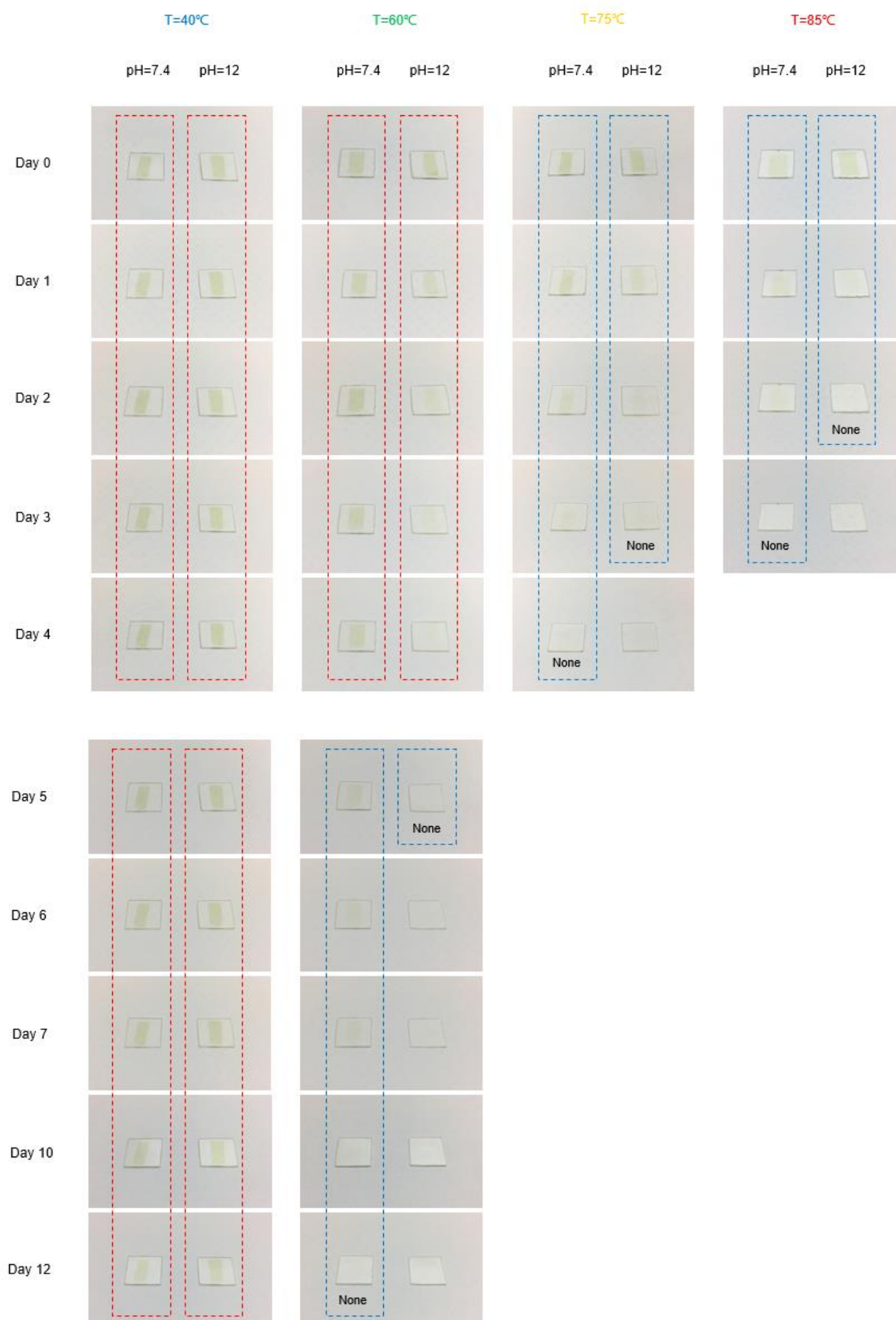
Supplementary Figure 4 | Optical and structural characterization of monolayer MoS₂ continuous film. a, OM image of monolayer MoS₂ continuous film on SiO₂ substrate synthesized by using LPCVD method; b, AFM images of the related MoS₂ film before and after oxidation to visualize its grain boundary; c-f Raman (c), PL(d) and XPS (e,f) spectra of the corresponding MoS₂ film.

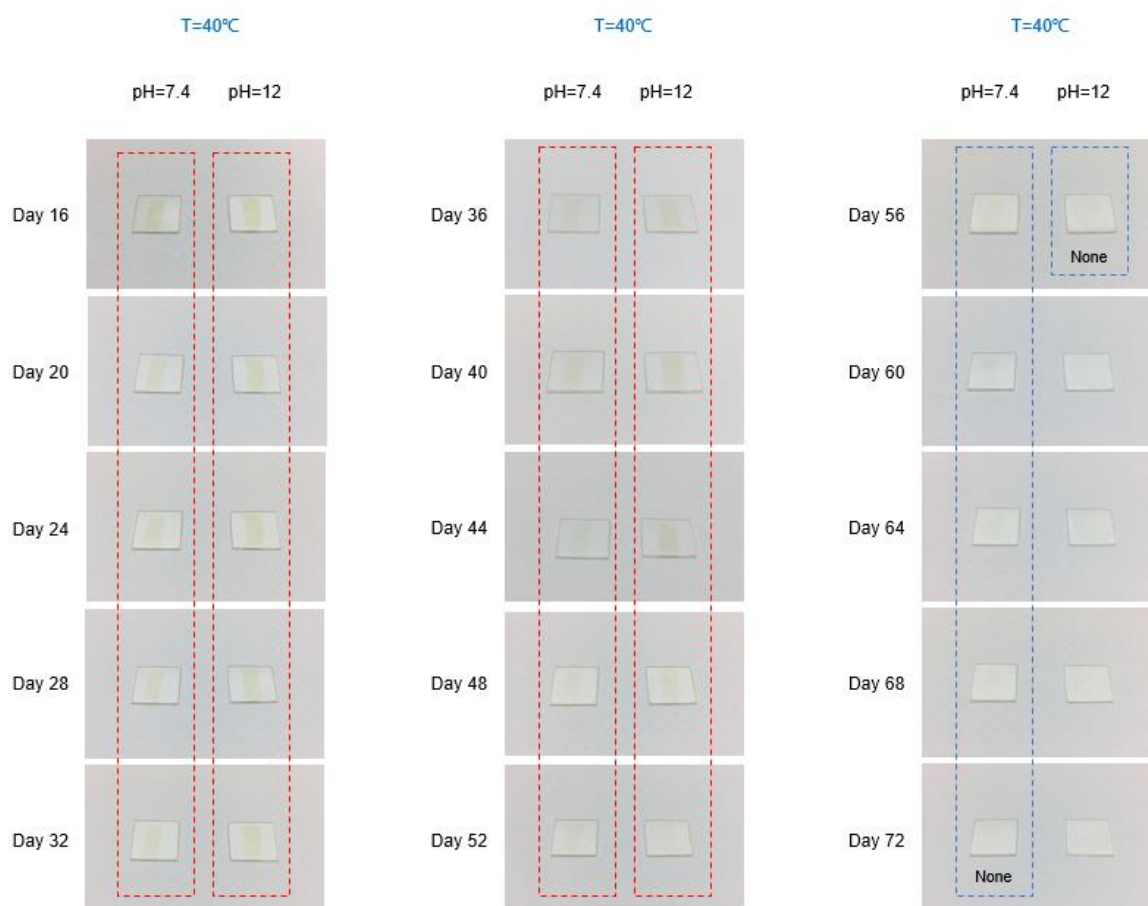


Supplementary Figure 5 | Surface roughness of SU-8 layer and monolayer MoS₂ film with different dissolution time. a,b, AFM images of SiO₂ (a) and SU-8 (b) substrates; c, RMS roughness evolution of SU-8 substrate and monolayer MoS₂ on SU-8 with different dissolution times from 0 day to 12 days in PBS solution at 60 °C.



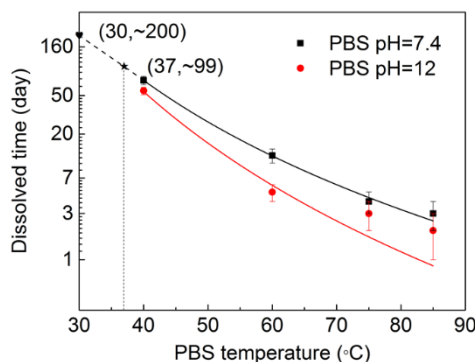
Supplementary Figure 6 | Raman, ICP-MS, and XPS study of monolayer MoS₂ film with different dissolution time or PBS solution. a, Time-dependent Raman spectra of the dissolved MoS₂ monolayers, with Raman shift ranging from 100 to 500 cm⁻¹; b, Mo concentration in PBS solution before and after monolayer MoS₂ dissolution, measured by using inductively coupled plasma mass spectrometry (ICP-MS); c-e, XPS spectra of P (c), Cl (d) and Na (e) from the monolayer MoS₂ film on SU-8/sapphire with different dissolution time from 0 day to 12 days in PBS solution at 60 °C.



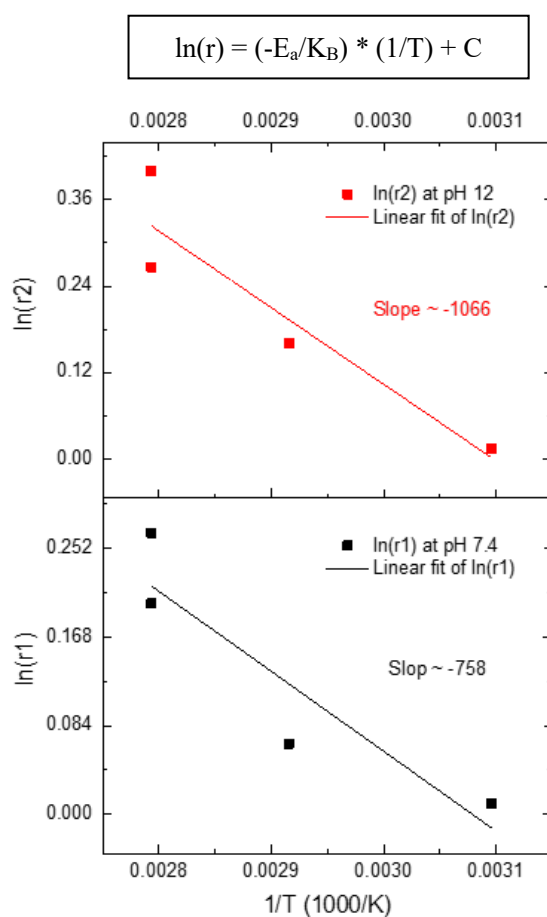


Supplementary Figure 7 | Temperature-, pH- and time-dependent dissolution of monolayer MoS₂ film.

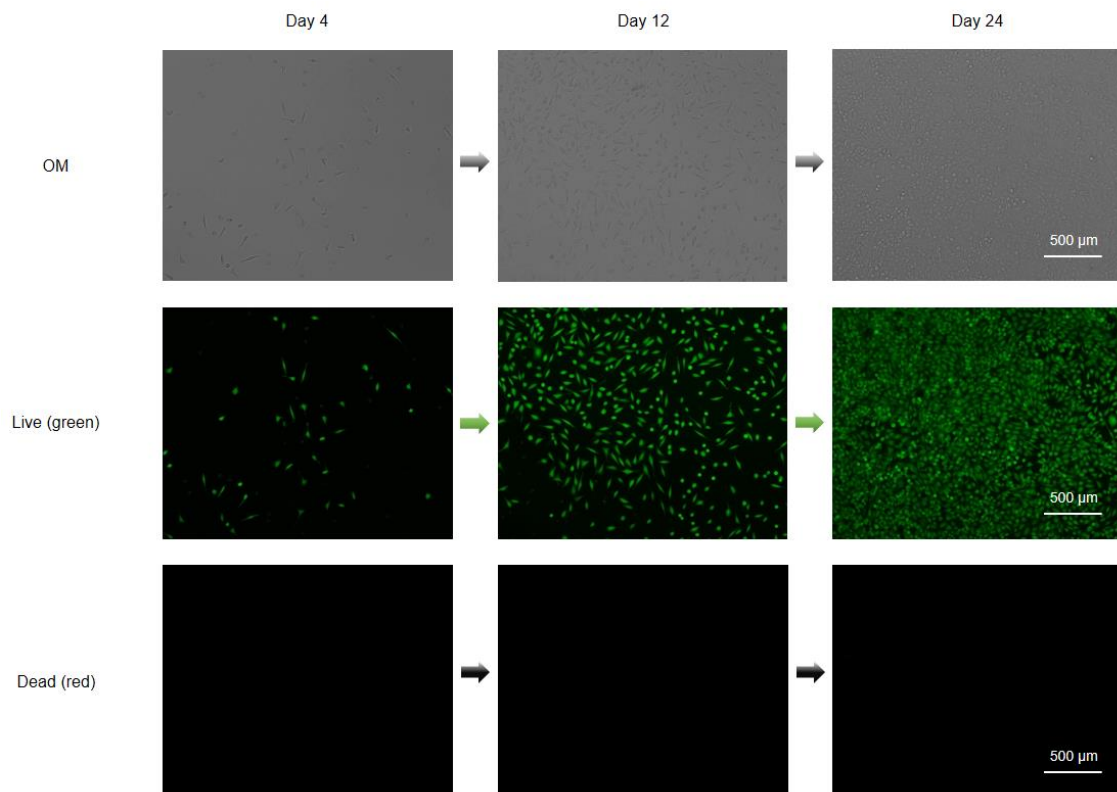
Optical images of large-scale MoS₂ monolayers on SU-8/sapphire substrate (1×1 cm) kept in PBS solution with different temperature (40-85 °C), pH level (7.4-12) and time (0-72 days). After 12 days, monolayer MoS₂ films in PBS solution at 60, 75 and 85 °C were all degraded except the one at 40 °C. Since then, we observed the samples at 40 °C until the end (72 days).



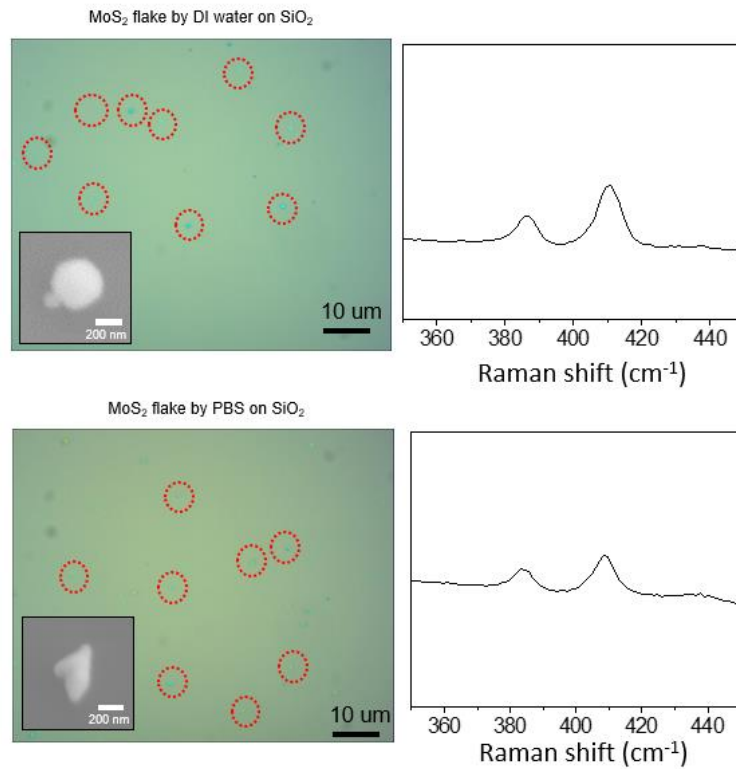
Supplementary Figure 8 | Temperature- and pH-dependent speed at which the MoS₂ monolayer dissolved in PBS solution. The linear scale of y axis has been converted into ln scale. Error bars of the experimental data represent standard error of mean.



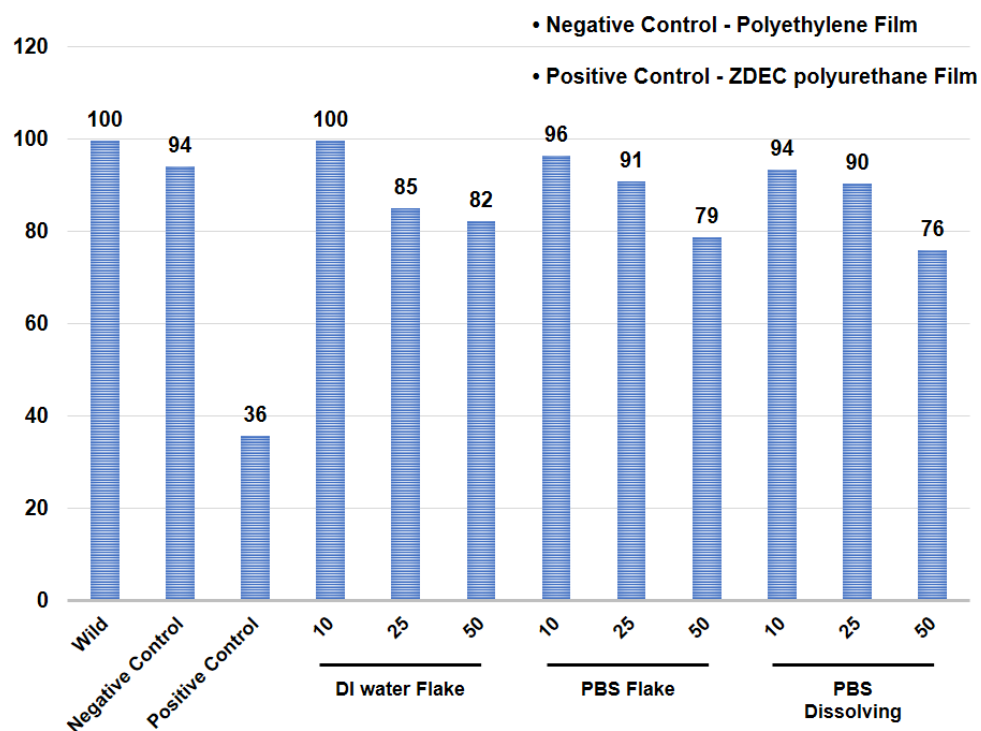
Supplementary Figure 9 | Temperature- and pH-dependent dissolution rate of monolayer MoS₂ film. Plots of the natural logarithm of the monolayer MoS₂ dissolution rates (r) as a function of the inverse of the temperature (T) at pH 7.4 and 12. The slope of the linear fitting estimates the activation energy as shown in the equation (Top of the figure). The equation was derived from the previous reported research (Ref. 14 in main text).



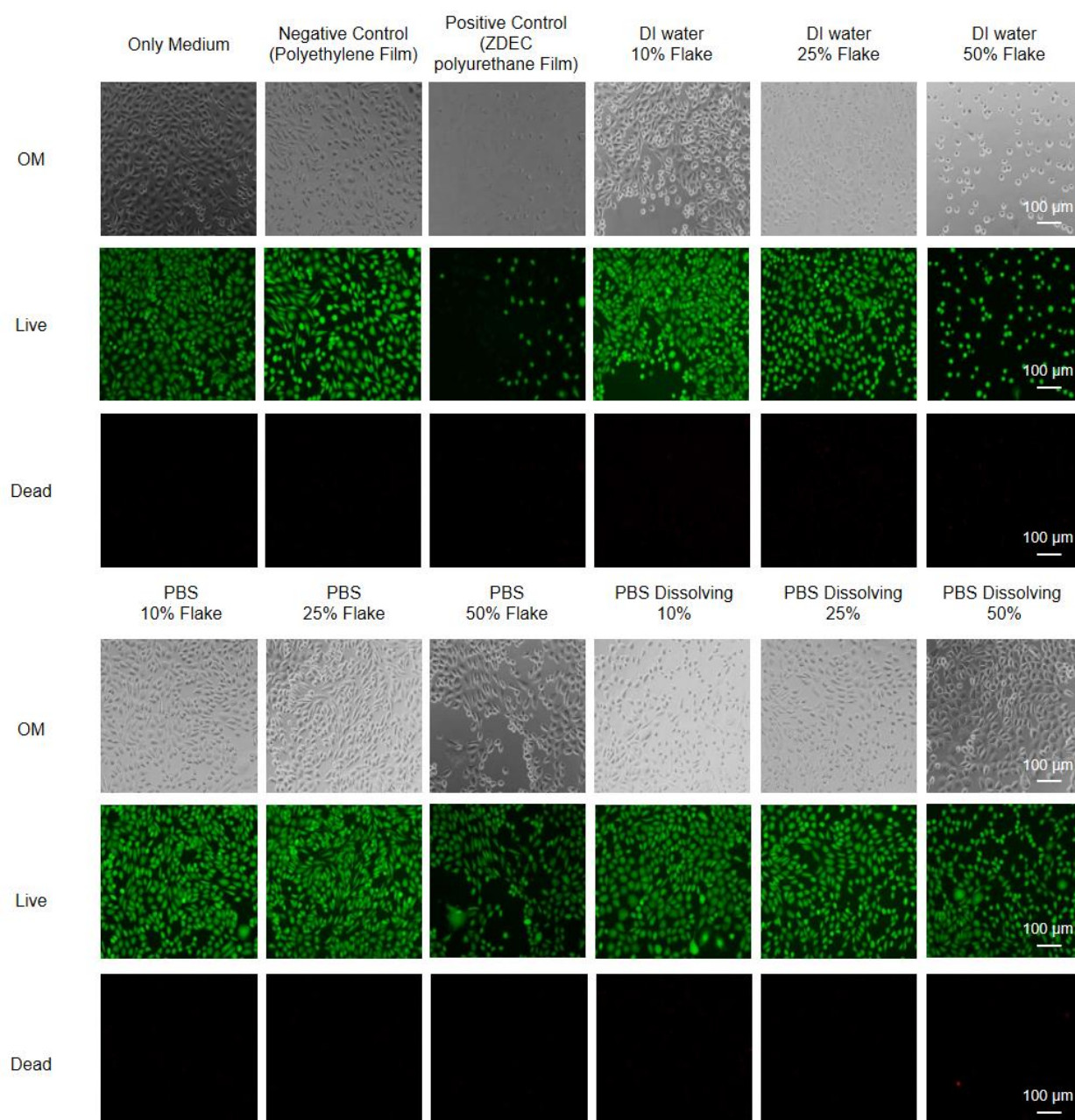
Supplementary Figure 10 | Time-dependent cell culture on monolayer MoS₂ film. OM and fluorescent images of cell culturing on monolayer MoS₂ film for 4, 12 and 24 days.



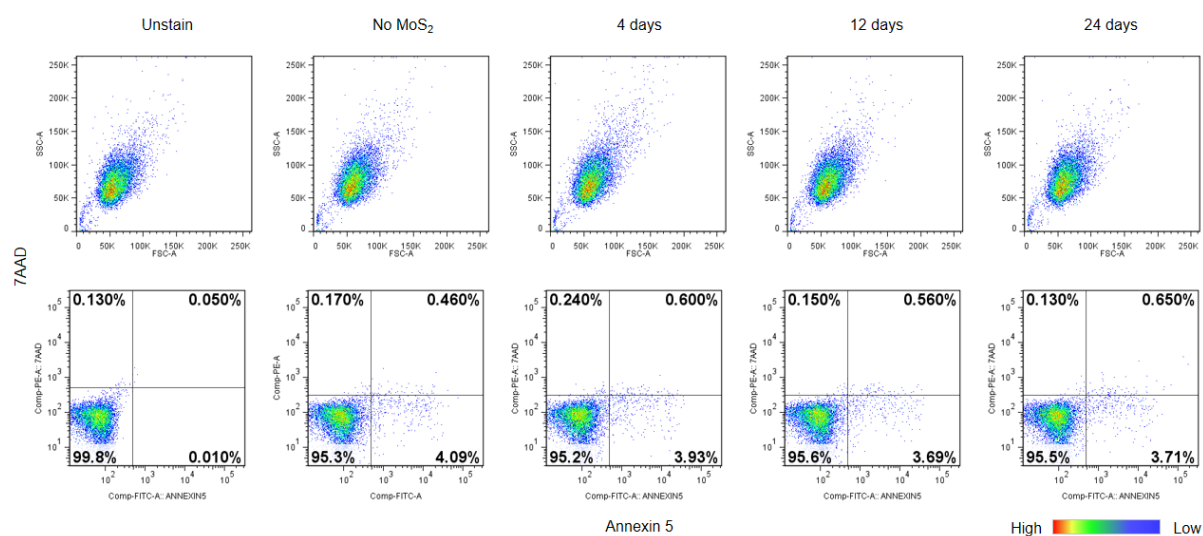
Supplementary Figure 11 | Morphological and structural characterization of MoS₂ flake obtained from the continuous film. OM images of dispersed MoS₂ flake in DI water and PBS solution and corresponding Raman spectra on SiO₂ substrate. Insets are the SEM image of the MoS₂ flakes in size of 300~400 nm.



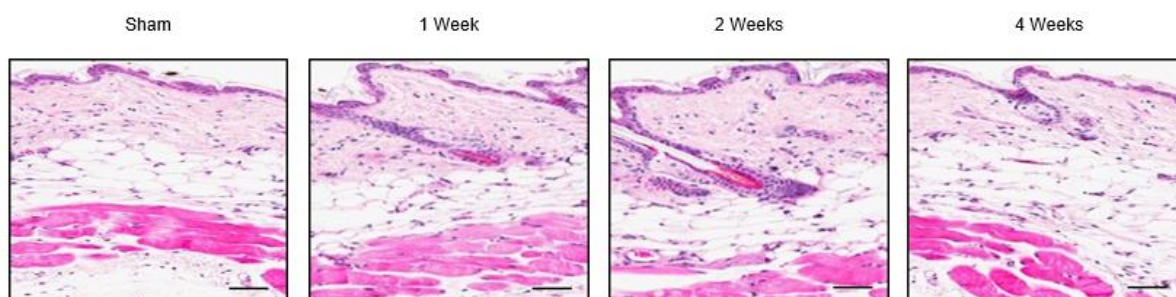
Supplementary Figure 12 | The statistical results of cell viability. Cell viability of L-929 cells on MoS₂ flake in DI water and PBS solution, dissolved MoS₂ in PBS, Culture medium, HDPE and ZDEC by CCK-8 assay.



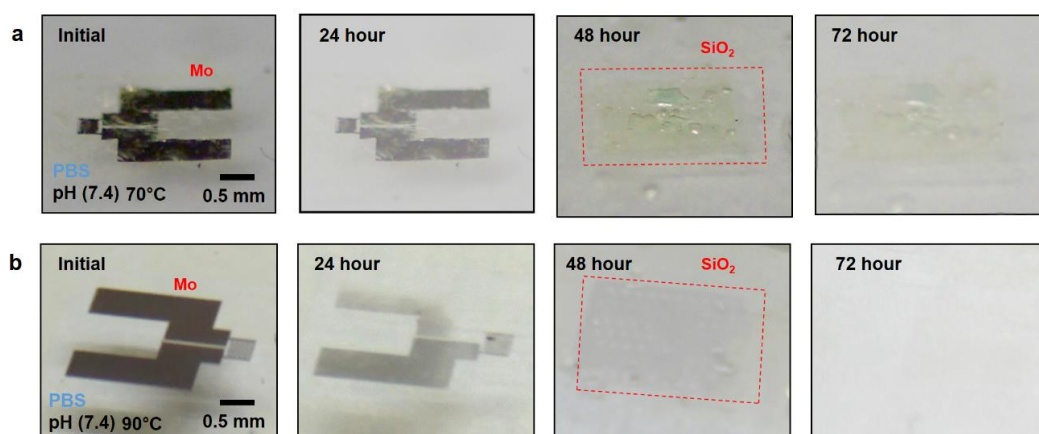
Supplementary Figure 13 | Cell culture results in different mediums. Set of differential interference contrast images and fluorescent images showing the cell viability of culture medium, HDPE, ZDEC, MoS₂ flake in DI water (10%, 25%, 50%), MoS₂ flake in PBS (10%, 25%, 50%), and dissolved MoS₂ in PBS (10%, 25%, 50%).



Supplementary Figure 14 | Apoptotic population study based on 2D MoS₂ film. Estimation of apoptotic populations in L929 of culture on MoS₂ film over 24 days using by Annexin V and 7-AAD, stained using flow cytometry.



Supplementary Figure 15 | *In vivo* biocompatibility studies. Representative microscope images (scale bar: 100 μ m) of hematoxylin and eosin (H&E) stained tissue sections after 0-4 weeks post-implantation.



Supplementary Figure 16 | Time-dependent dissolution process of the 2D MoS₂ bioabsorbable device. Dissolution images collected at 4 stages of the accelerated condition of a system immersed in an aqueous buffer solution (pH 7.4) at 70 °C (a) and 90 °C (b).

Supplementary References:

1. Wang, X., Shen, X., Wang, Z., Yu, R. & Chen, L. Atomic-scale clarification of structural transition of MoS₂ upon sodium intercalation. *ACS nano* **8**, 11394-11400 (2014).
2. Zhang, L. *et al.* In situ TEM observing structural transitions of MoS₂ upon sodium insertion and extraction. *RSC Adv.* **6**, 96035-96038 (2016).
3. Wang, Z. *et al.* Chemical Dissolution Pathways of MoS₂ Nanosheets in Biological and Environmental Media. *Environ. Sci. Technol.* **50**, 7208-7217 (2016).