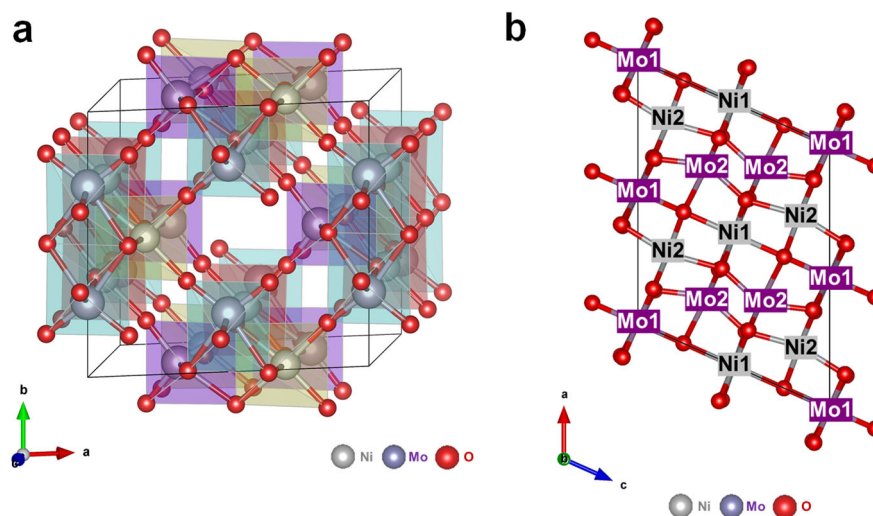


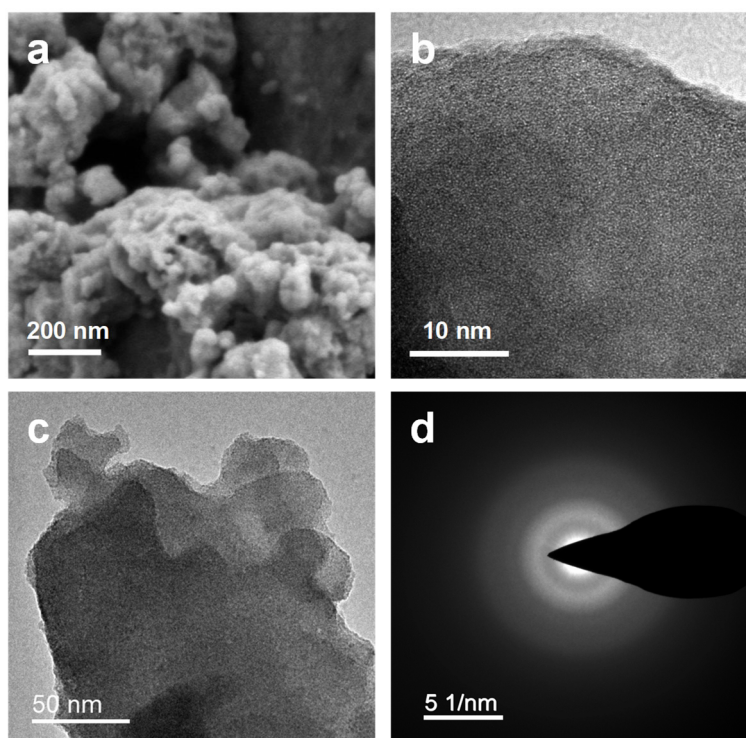
Supporting Information for
Boosting and stabilizing oxygen evolution reaction through
Ru single atoms anchored amorphous NiMoO_x
electrocatalyst

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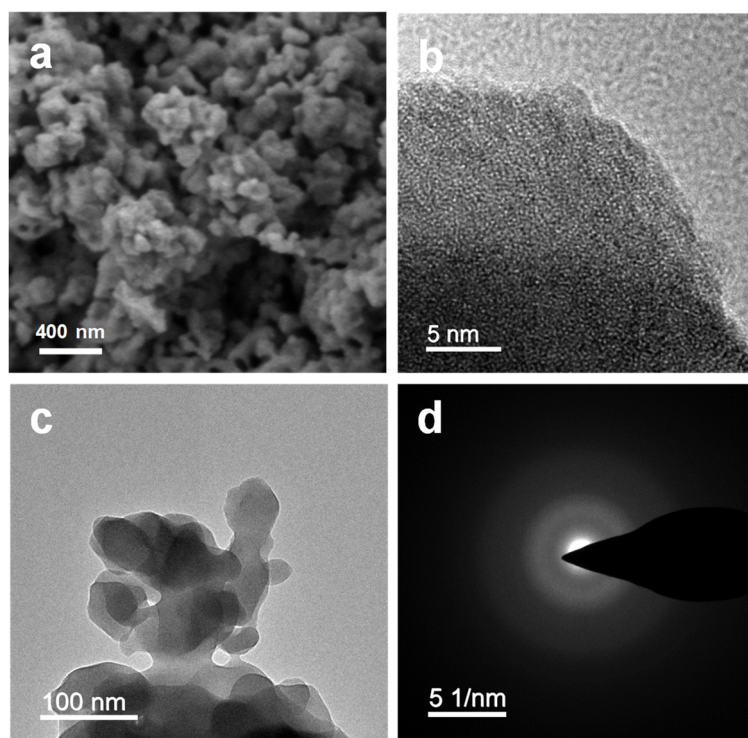
Supplementary Figures



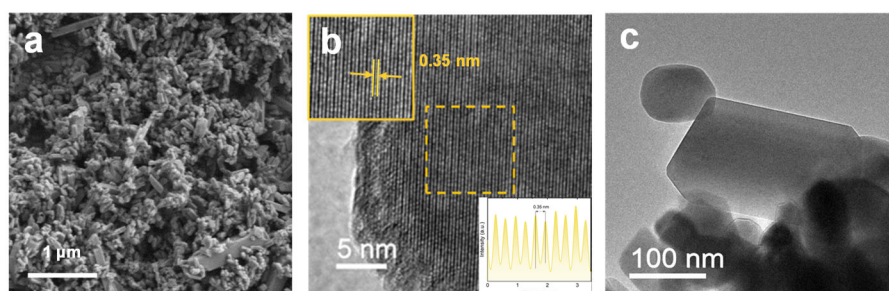
Supplementary Figure 1. Top view (a) and side view (b) of crystalline NiMoO_4 structure (PDF#33-0948).



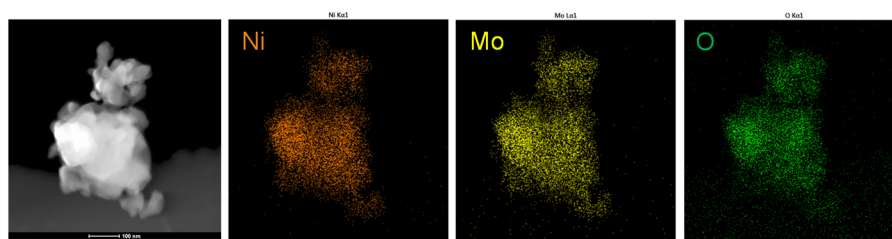
Supplementary Figure 2. (a) SEM, (b, c) HR-TEM and (d) SAED images of a-RNMO.



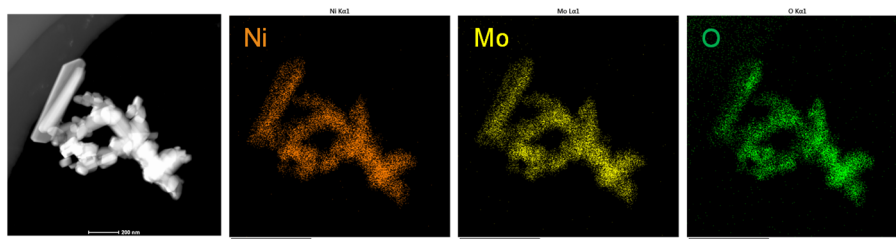
Supplementary Figure 3. (a) SEM, (b, c) HR-TEM and (d) SAED images of a-NMO.



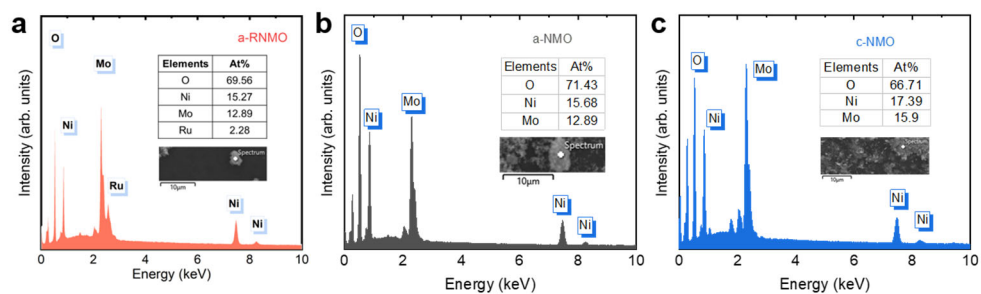
Supplementary Figure 4. (a) SEM, (b, c) HR-TEM images of c-NMO.



Supplementary Figure 5. EDS-mapping of a-NMO.

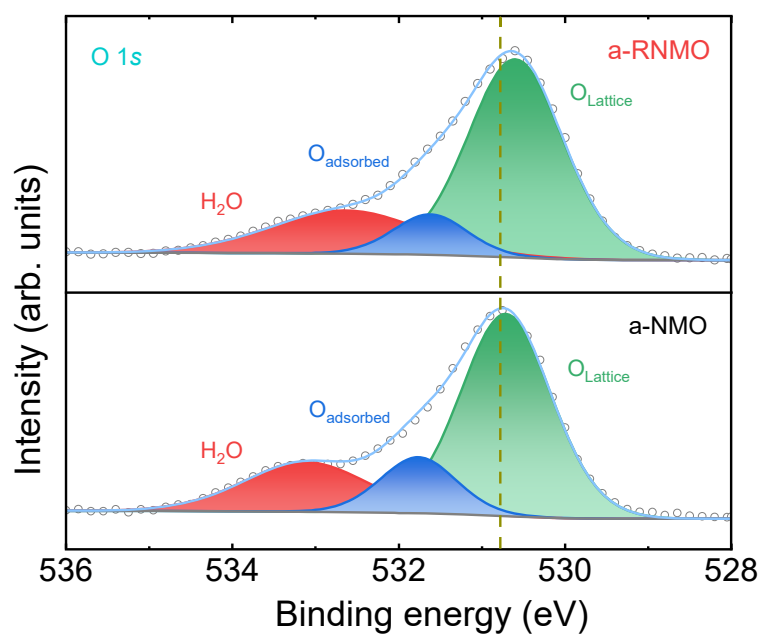


Supplementary Figure 6. EDS-mapping of c-NMO.

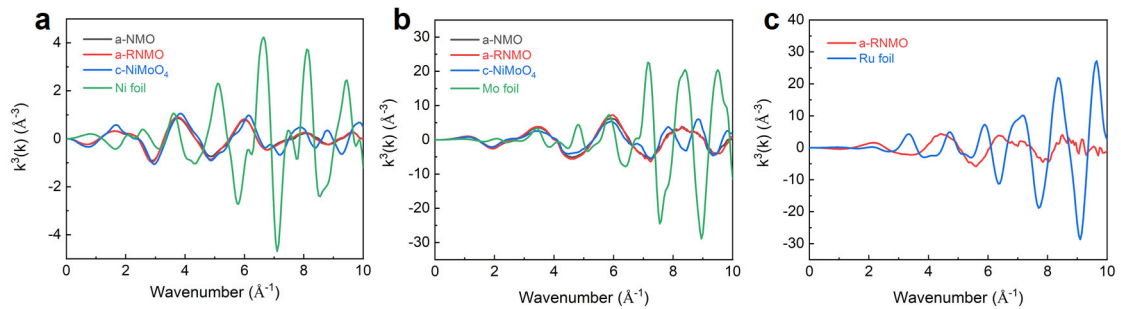


Supplementary Figure 7. SEM-EDS spectrum of (a) a-RNMO, (b) a-NMO, and (c) c-NMO.

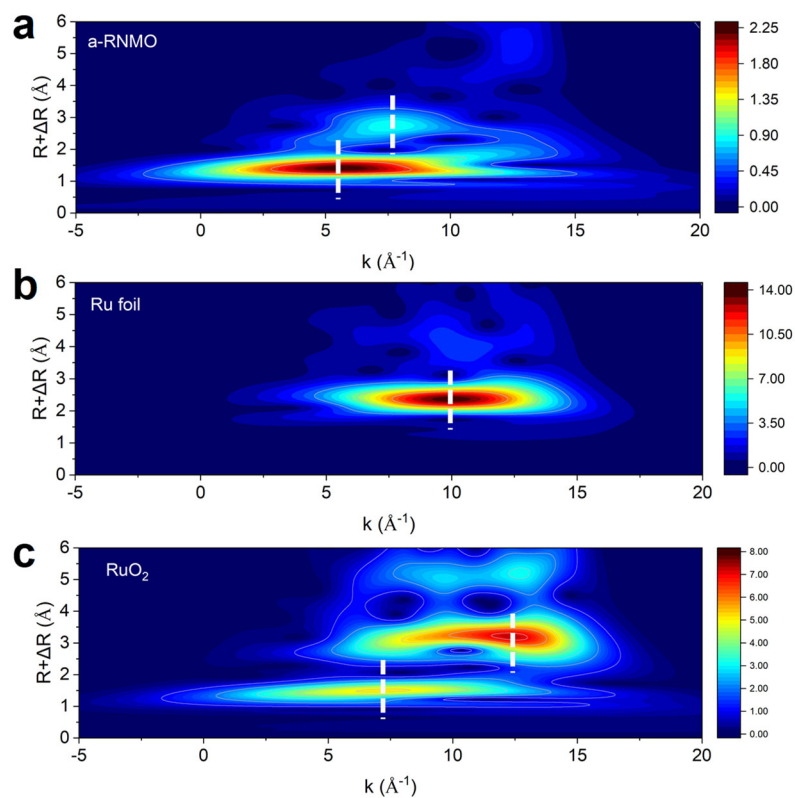
Source data are provided as a Source Data file.



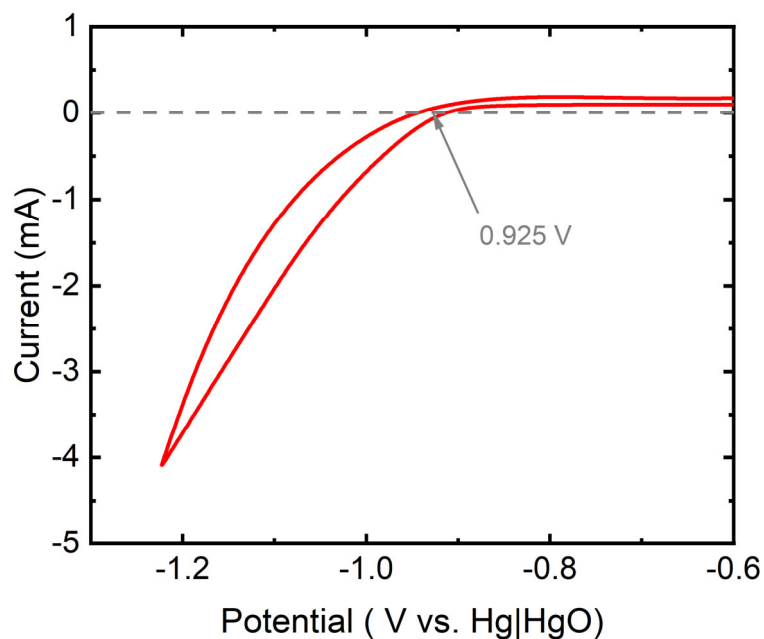
Supplementary Figure 8. O 1s XPS spectra of a-NMO and a-RNMO. Source data are provided as a Source Data file.



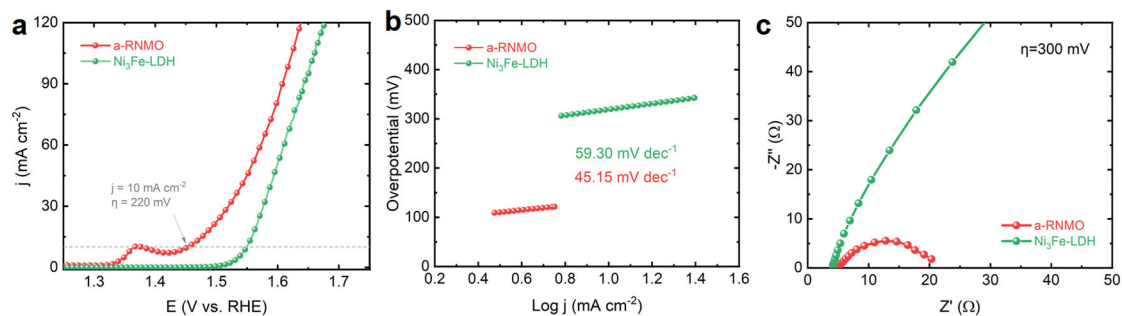
Supplementary Figure 9. The k^3 -weighted k-space of (a) Ni *K*-edge, (b) Mo *K*-edge, and (c) Ru *K*-edge experimental spectra of a-NMO, a-RNMO, and references. Source data are provided as a Source Data file.



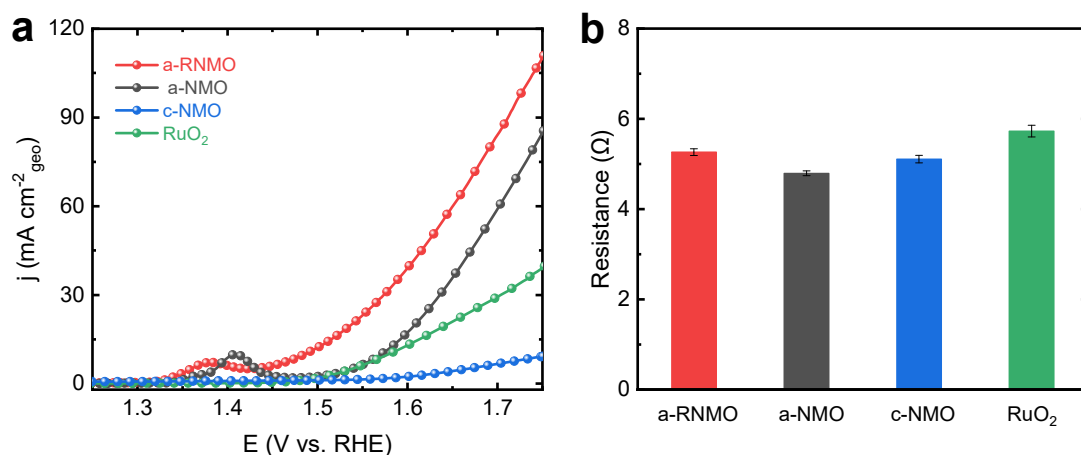
Supplementary Figure 10. WT-EXAFS analysis of Ru *K*-edge of (a) a-RNMO, (b) Ru foil, and (c) RuO₂. Source data are provided as a Source Data file.



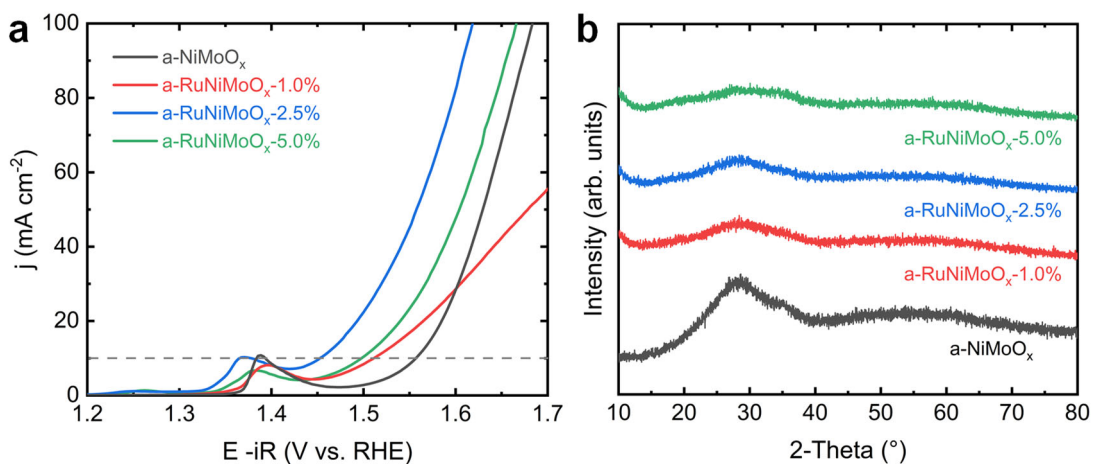
Supplementary Figure 11. Potential (without iR compensation) calibration of the Hg|HgO reference electrode in H₂-saturated 1 M KOH solution at 25 °C. Source data are provided as a Source Data file.



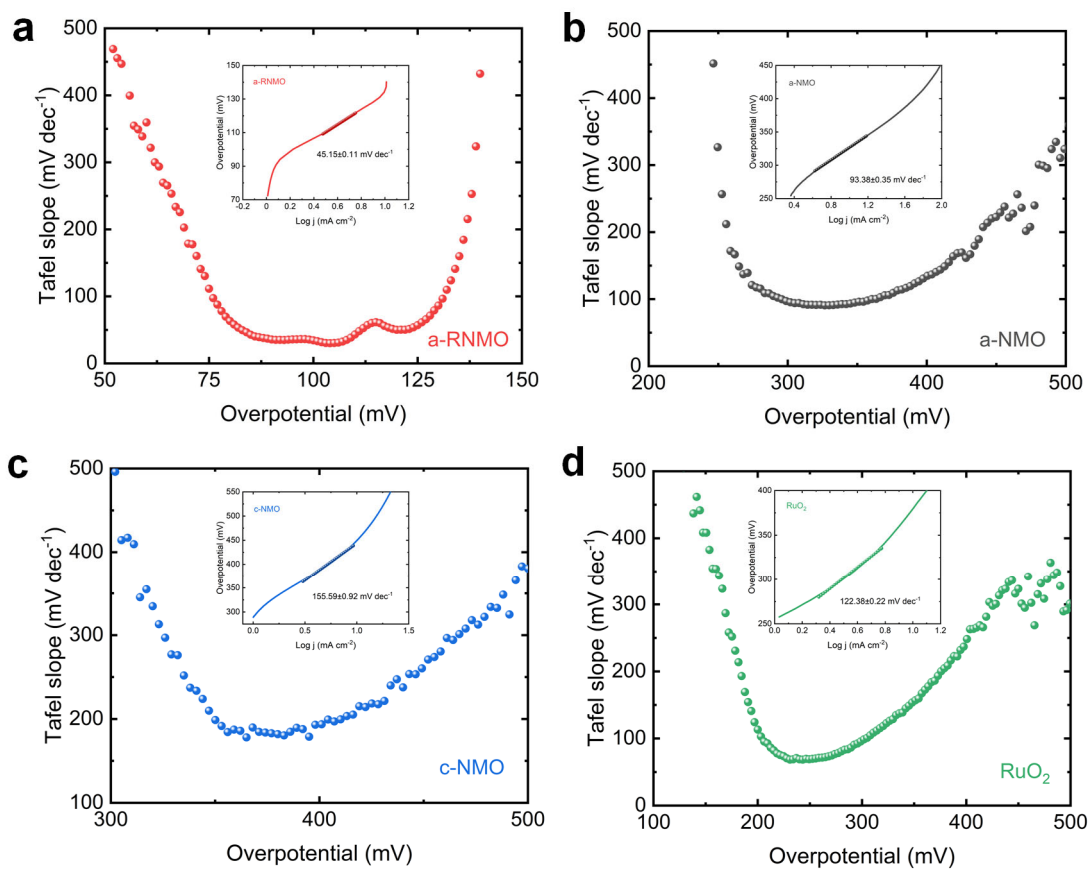
Supplementary Figure 12. Electrochemical performance of a-RNMO and NiFe-LDH. (a) LSV polarization curves. (b) Tafel plots extracted from (a). (c) EIS Nyquist plots. Source data are provided as a Source Data file.



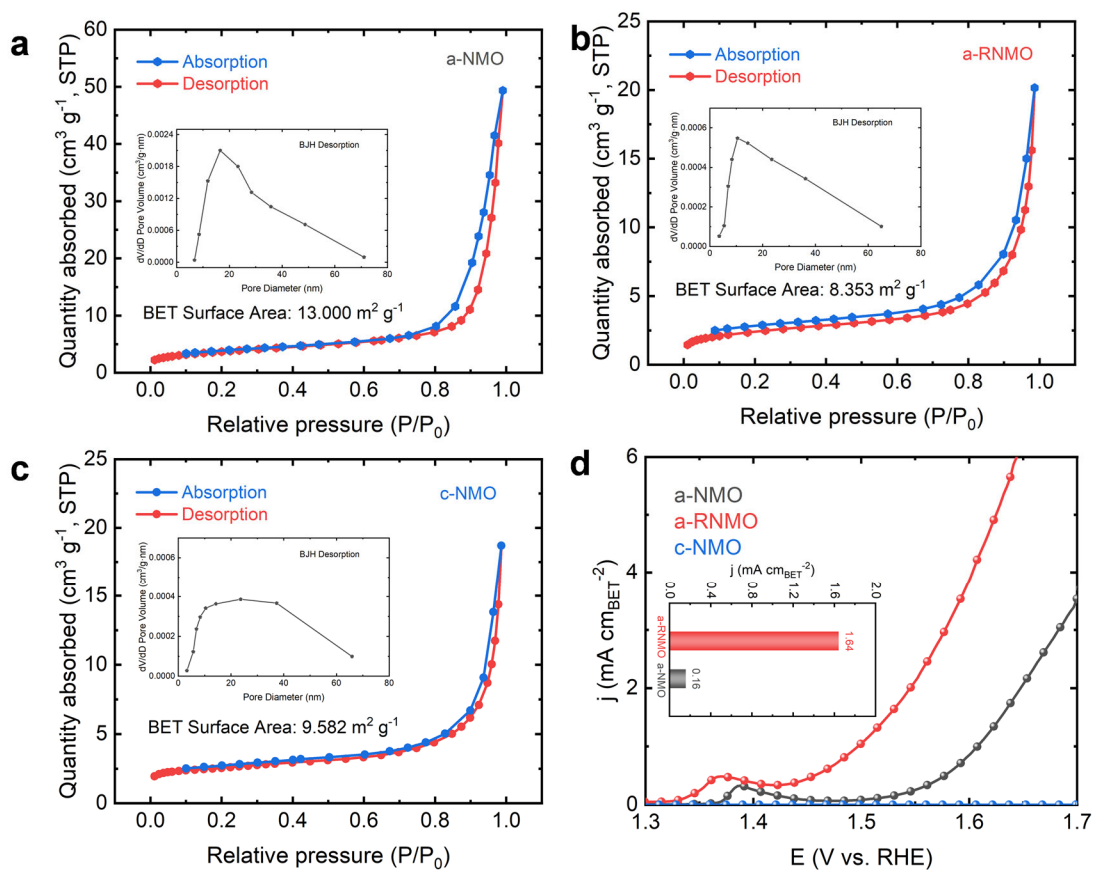
Supplementary Figure 13. (a) LSV polarization curves without iR-compensation of a-RNMO, a-NMO, c-NMO, and RuO₂ in 1 M KOH under 25 °C, at scan rate of 5 mV s⁻¹, using RDE (0.196 cm²) with 1600 rpm. (b) The detected resistance for these electrocatalysts. Note: error bars represent the standard deviation of three independent measurements. Source data are provided as a Source Data file.



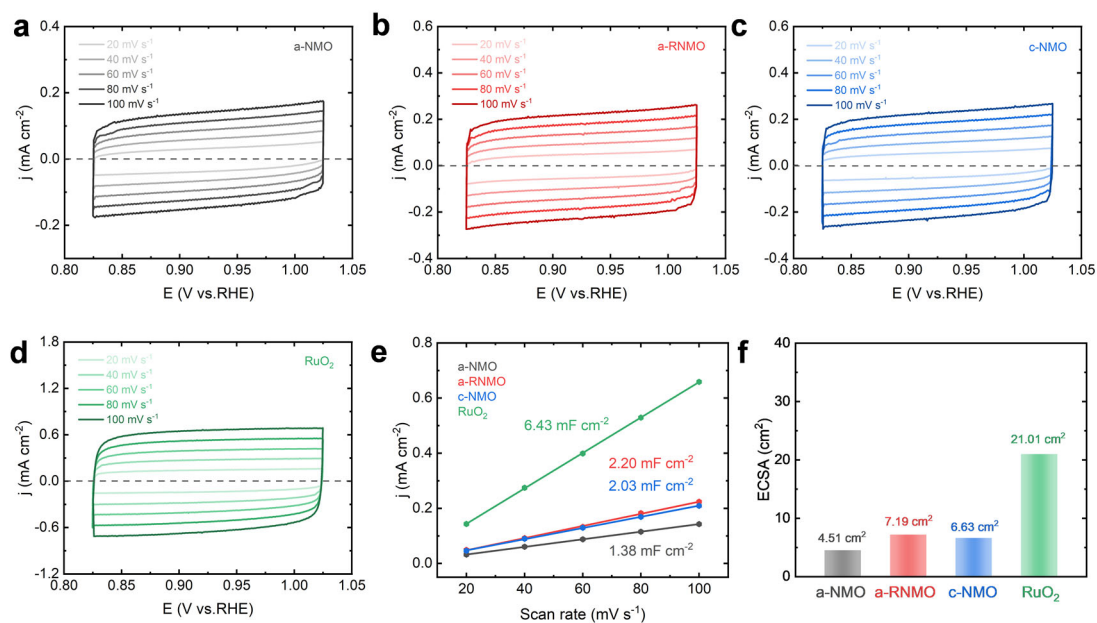
Supplementary Figure 14. (a) LSV polarization curves with iR-compensation with different Ru content, including at. % = 0%, 1%, 2.5%, 5%. (b) XRD patterns of them. Source data are provided as a Source Data file.



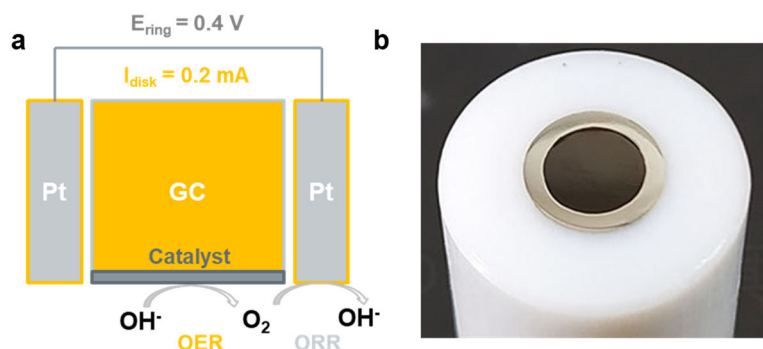
Supplementary Figure 15. The graphs that plot Tafel slope as a function of overpotential on (a) a-RNMO, (b) a-NMO, (c) c-NMO, and (d) RuO_2 . Inset: The Tafel plots of these catalysts. Source data are provided as a Source Data file.



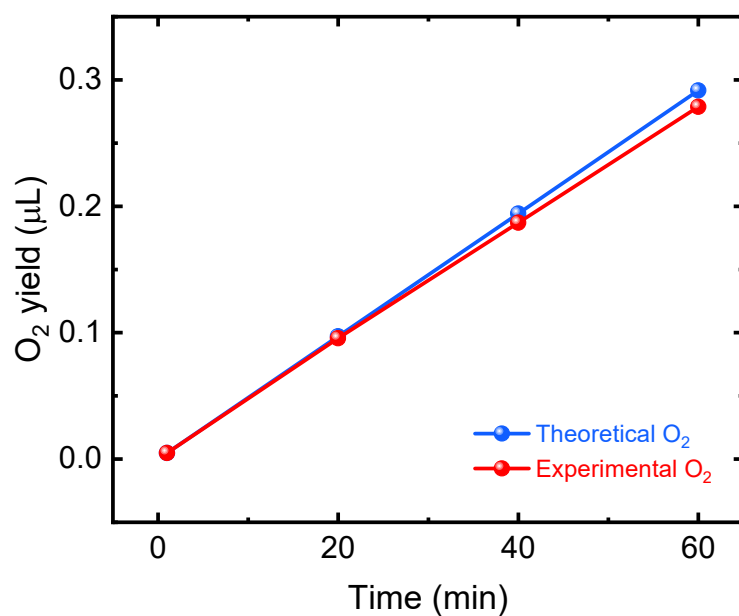
Supplementary Figure 16. N₂ adsorption and desorption isotherms of (a) a-NMO, (b) a-RNMO, and (c) c-NMO. (d) LSV polarization curves normalized to BET specific areas of a-NMO, a-RNMO, and c-NMO. Inset: The normalized current density values of a-NMO and a-RNMO at 1.53 V vs. RHE. Source data are provided as a Source Data file.



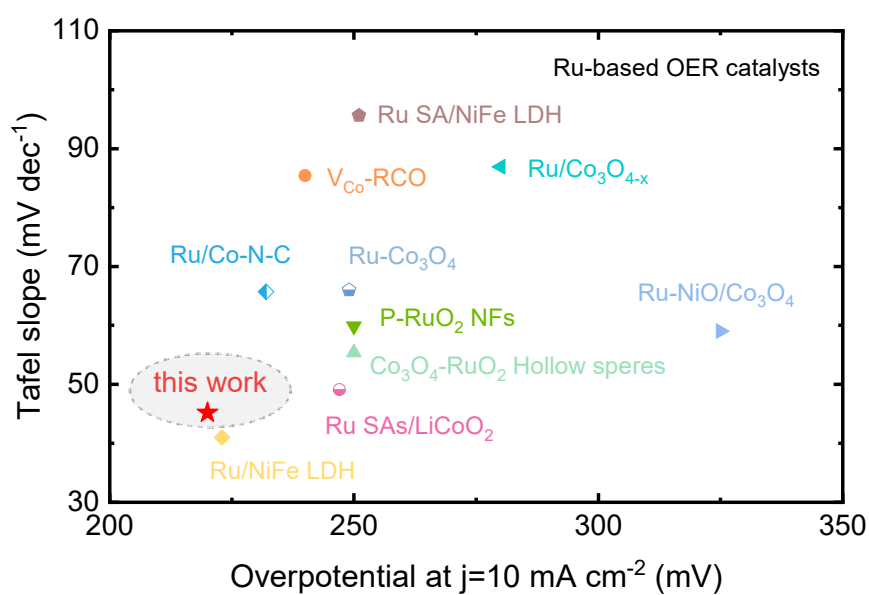
Supplementary Figure 17. Electrochemical CV scans at non-Faradic current for (a) a-NMO, (b) a-RNMO, (c) c-NMO, and (d) RuO₂, with 20, 40, 60, 80, and 100 mV s⁻¹ scan rates. (e) linear fitting between the capacitive currents and CV scan rates for these electrocatalysts. (f) The ECSA values calculated for these electrocatalysts. Source data are provided as a Source Data file.



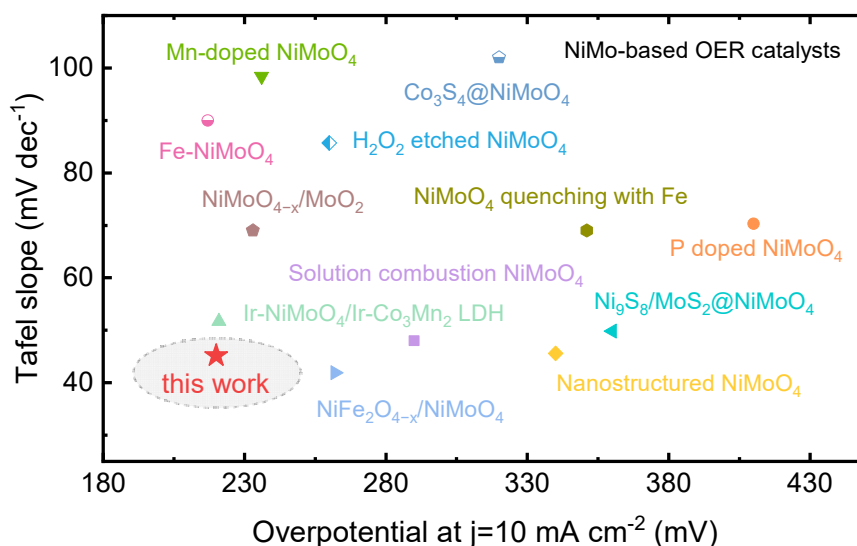
Supplementary Figure 18. The schematic diagram created in Microsoft PowerPoint of the reactions (a) using RRDE (b) during the FE test.



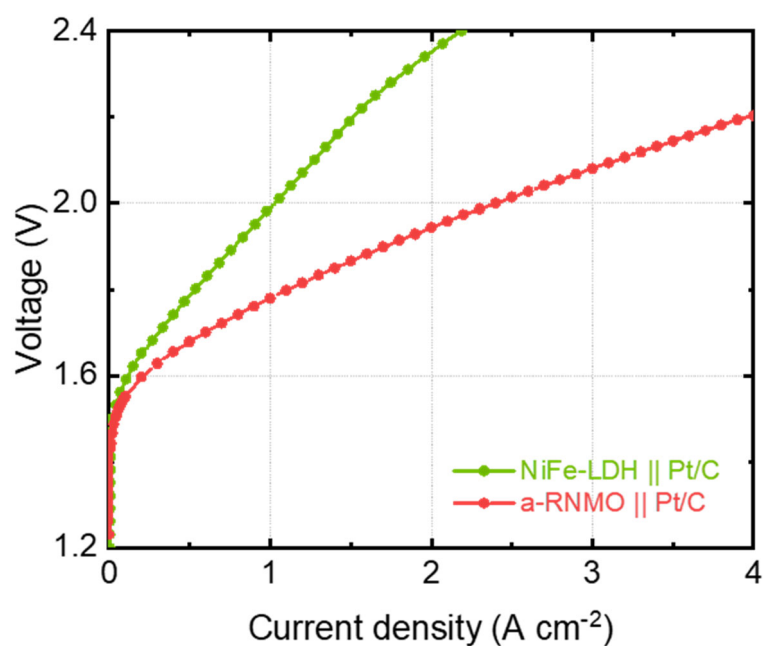
Supplementary Figure 19. The experimental and theoretical oxygen yield, calculated by Faradic efficiency (FE). Source data are provided as a Source Data file.



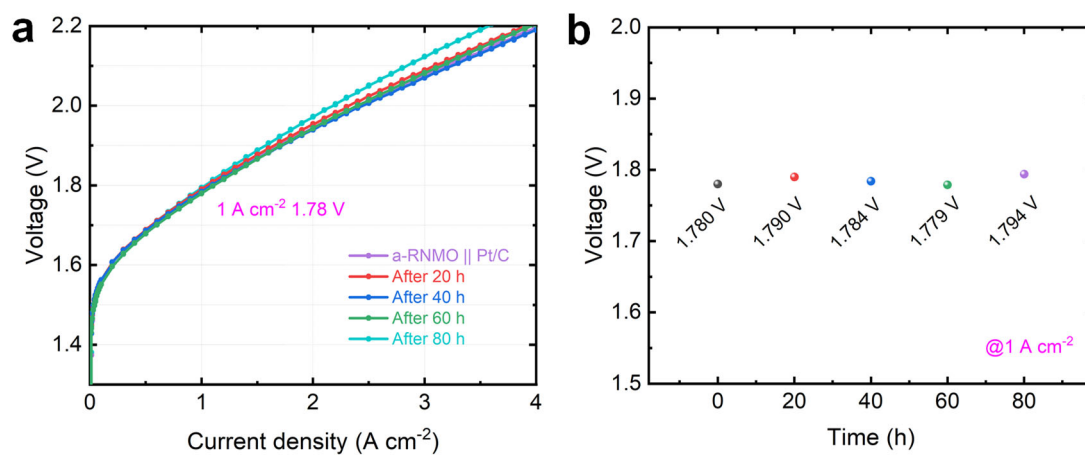
Supplementary Figure 20. Comparison of the Tafel slopes and overpotentials in reported Ru-based OER electrocatalysts.^{1, 2, 3, 4, 5, 6, 7, 8, 9, 10}



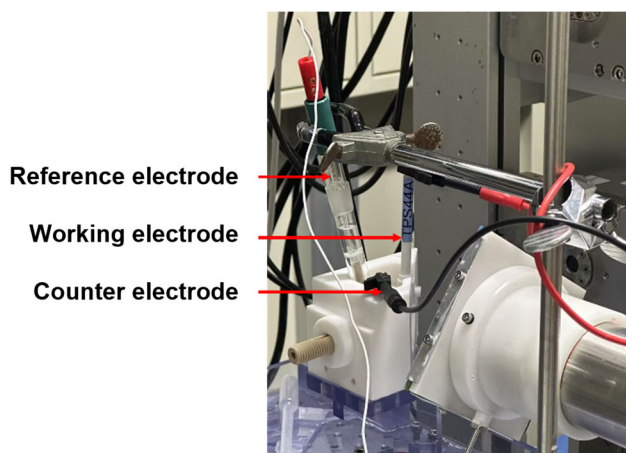
Supplementary Figure 21. Comparison of the Tafel slopes and overpotentials in reported NiMo-based oxide OER electrocatalysts.^{11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23}



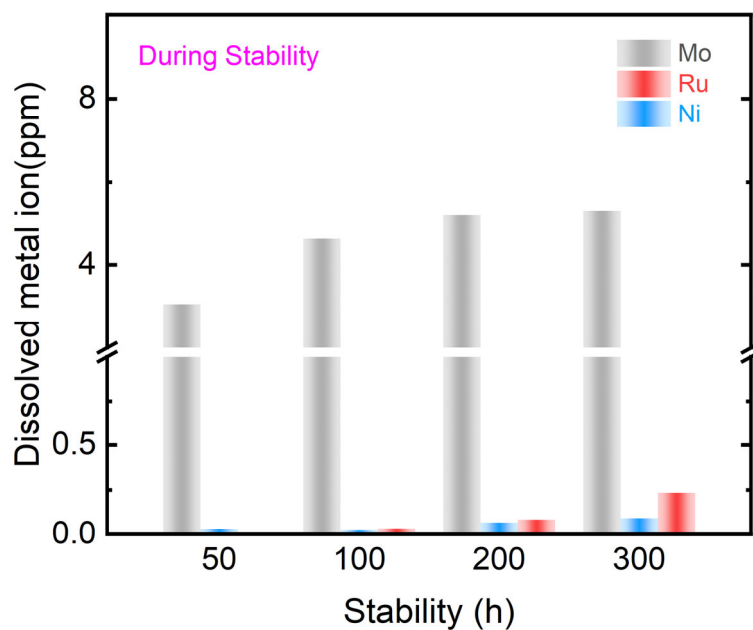
Supplementary Figure 22. AEMWE polarization curve of the electrolyzer without iR correction using a-RNMO or NiFe-LDH as the anode and Pt/C as the cathode. Source data are provided as a Source Data file.



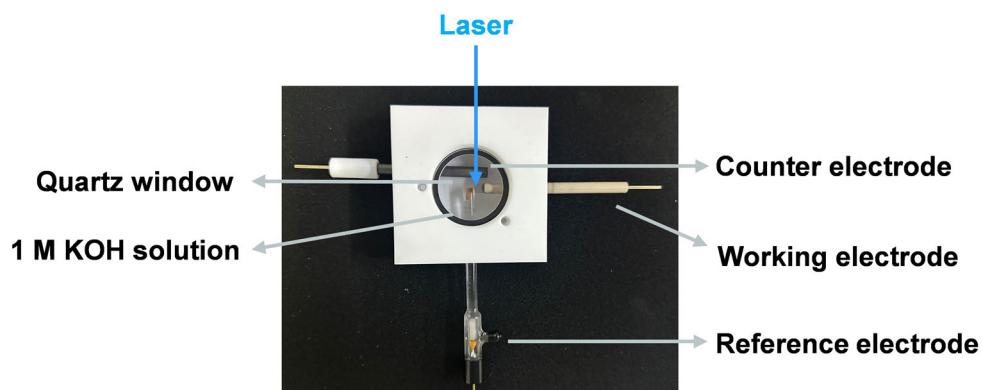
Supplementary Figure 23. AWMWE system of (a) polarization curves without iR correction and (b) voltage at 1 A cm^{-2} during durability with a-RNMO as the anode and Pt/C as the cathode. Source data are provided as a Source Data file.



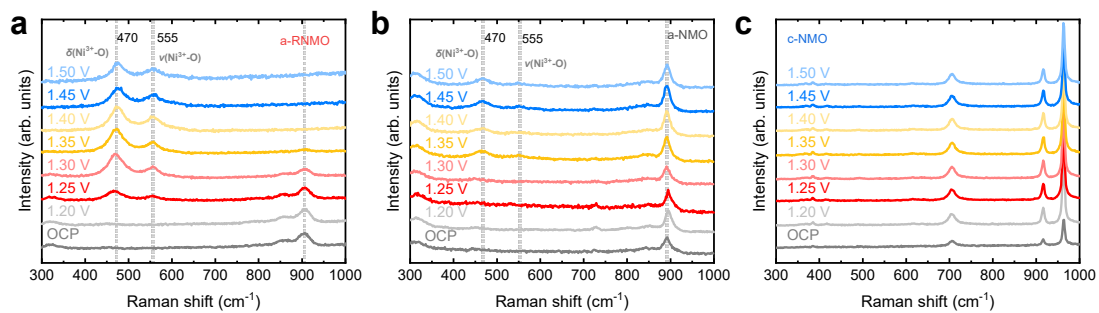
Supplementary Figure 24. The setup of the time-reversed operando XAS test.



Supplementary Figure 25. The dissolved amounts, by ICP-MS measurements, of Mo, Ru, and Ni ions during the durability test. Source data are provided as a Source Data file.

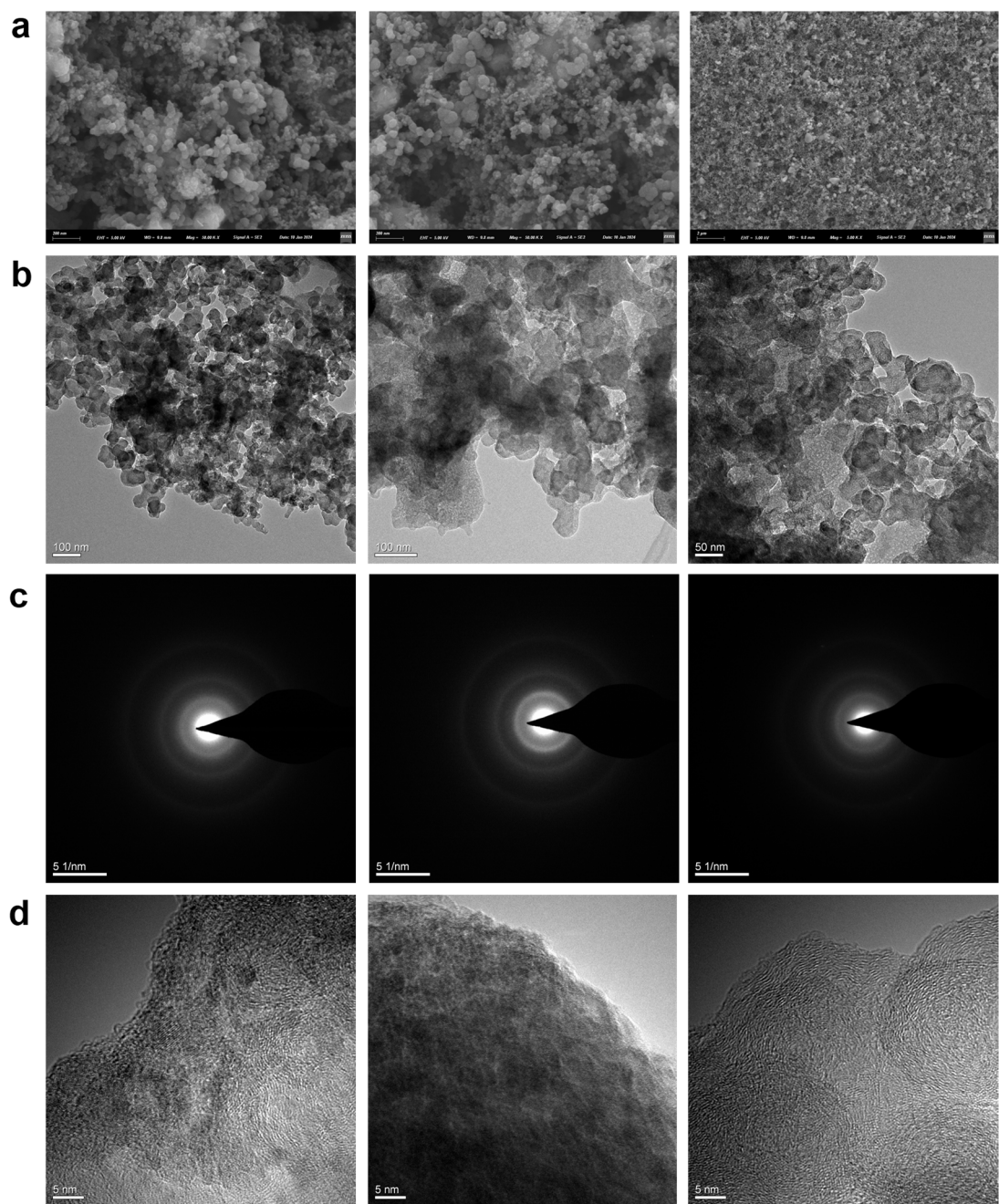


Supplementary Figure 26. The electrochemical cell device of the operando Raman test.

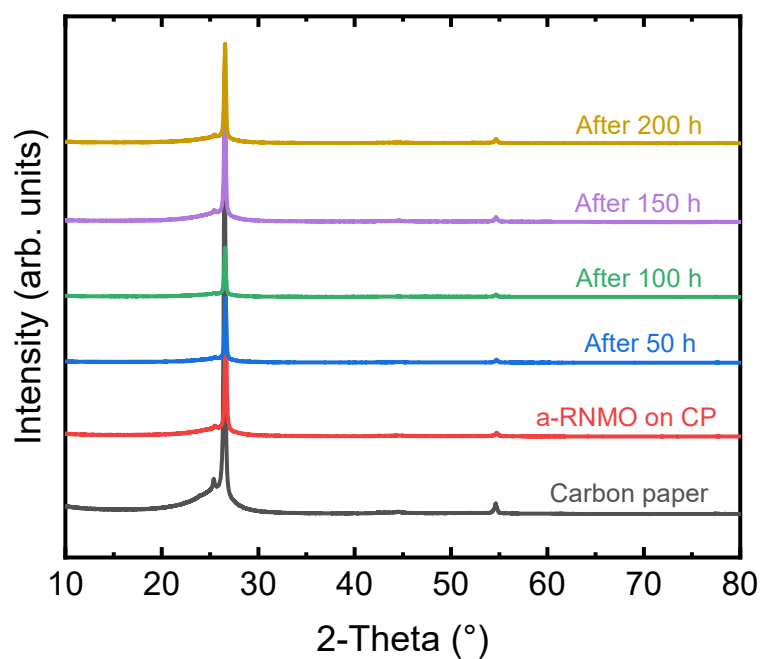


Supplementary Figure 27. Raman spectra of (a) a-RNMO and (b) a-NMO, and (c) c-NMO measured in 1 M KOH under OCP, and OER operating condition (from 1.20 V to 1.50 V vs. RHE).

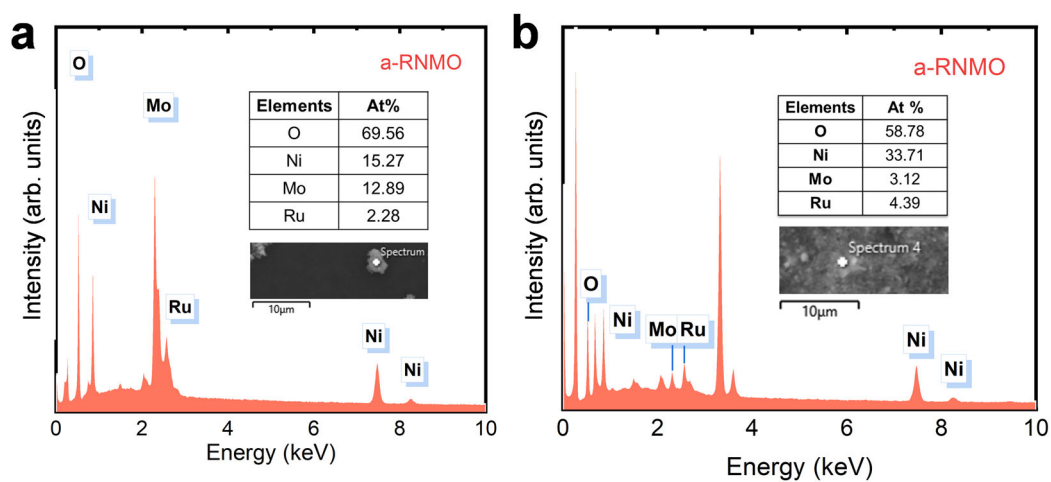
Source data are provided as a Source Data file.



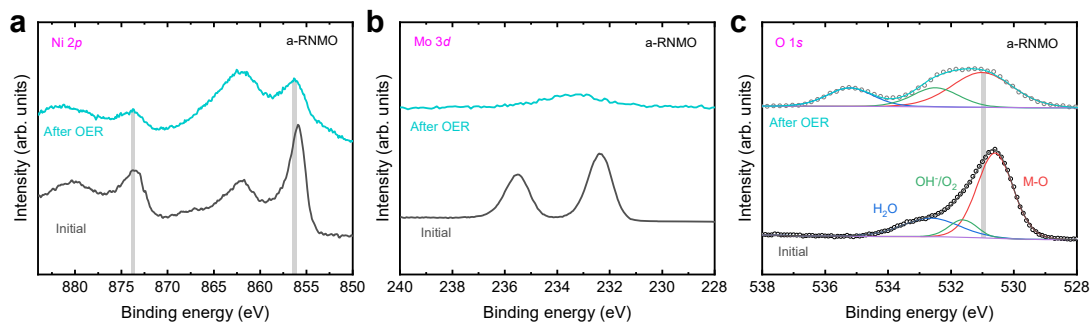
Supplementary Figure 28. (a) SEM, (b) TEM, and (c) SAED, (d) HR-TEM images of a-RNMO after OER operation.



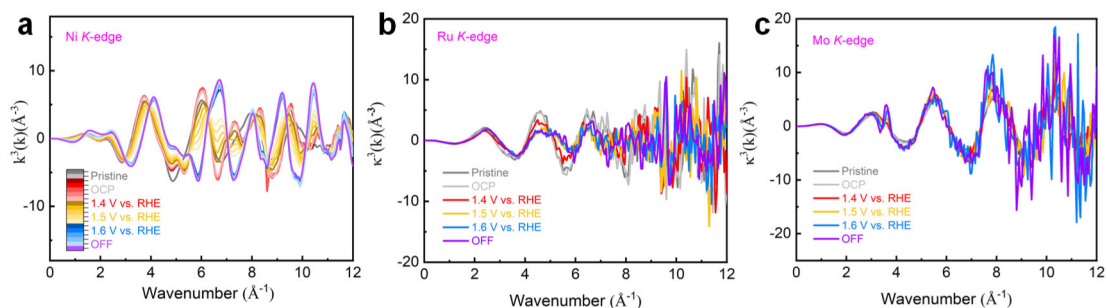
Supplementary Figure 29. XRD patterns of a-RNMO during 200 h OER stability. Source data are provided as a Source Data file.



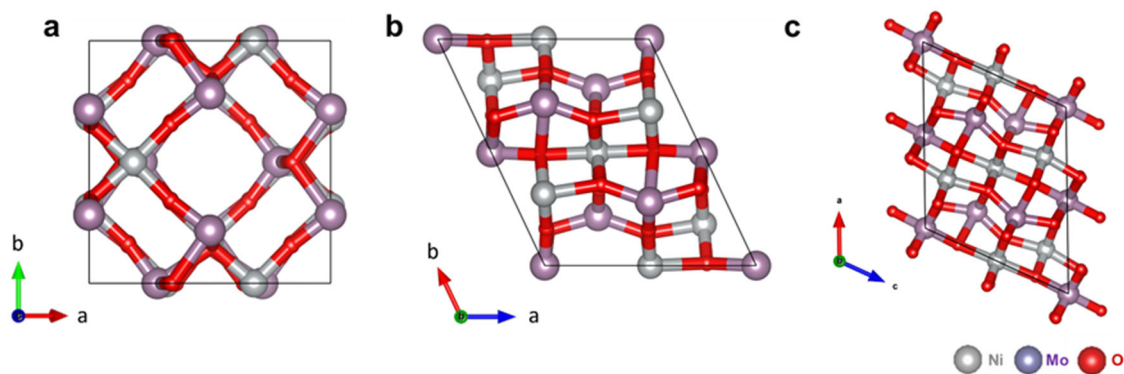
Supplementary Figure 30. SEM-EDS spectrum of (a) pristine a-RNMO and (b) after OER operation. Source data are provided as a Source Data file.



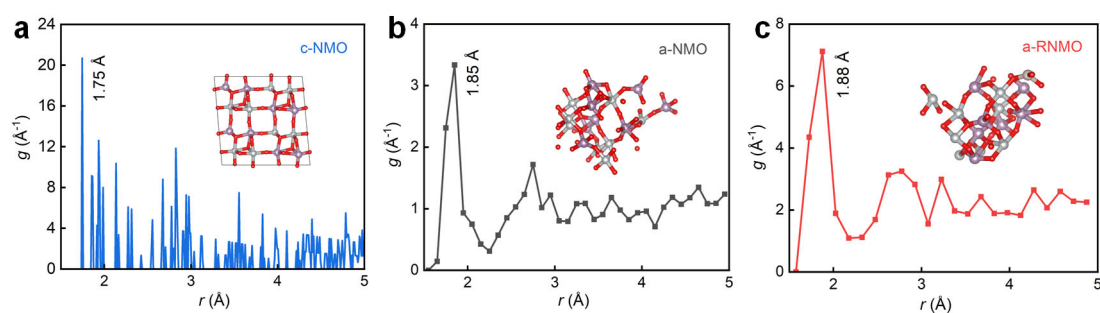
Supplementary Figure 31. XPS spectrum of a-RNMO with (a) Ni 2p, (b) Mo 3d, and (c) O 1s before and after stability test. Source data are provided as a Source Data file.



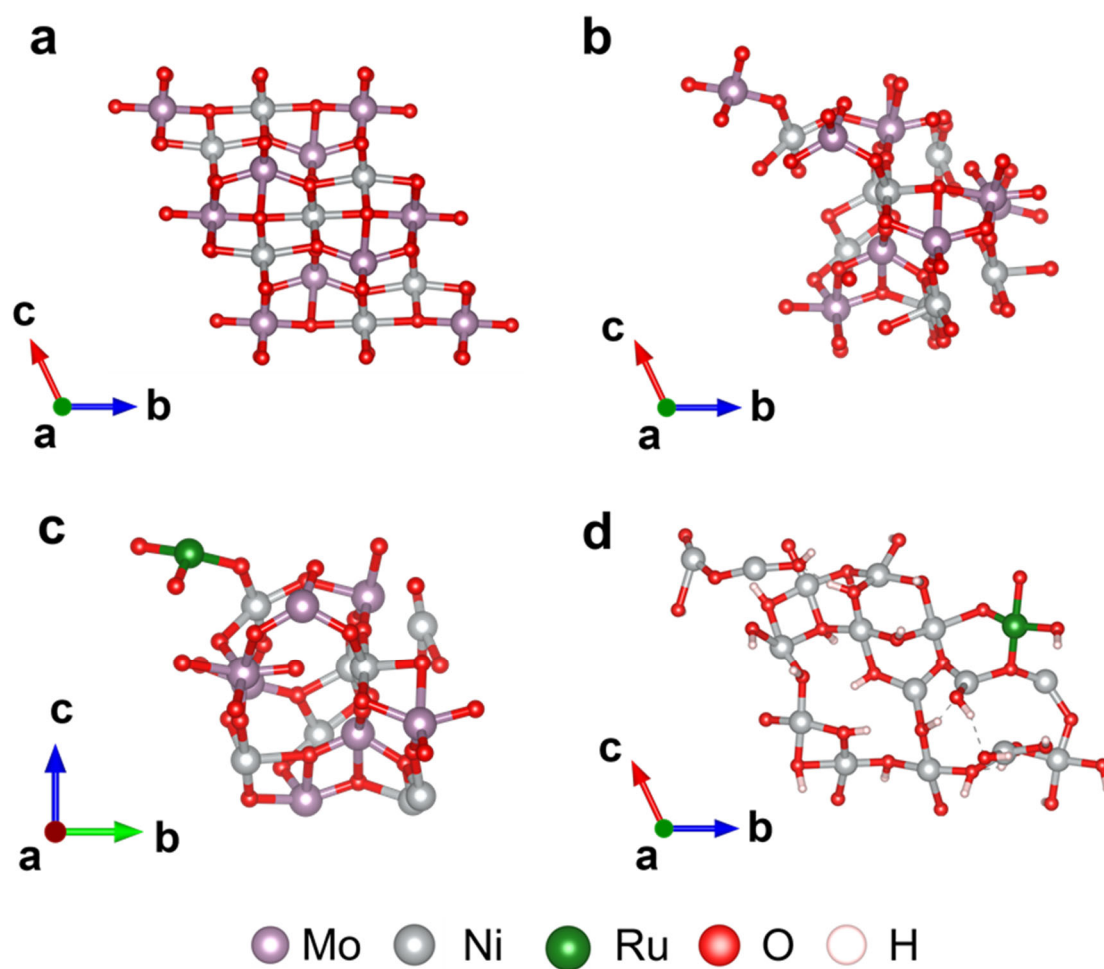
Supplementary Figure 32. The k^3 -weighted k -space of (a) Ni K-edge, (b) Ru K-edge, and (c) Mo K-edge experimental spectra of a-RNMO. Source data are provided as a Source Data file.



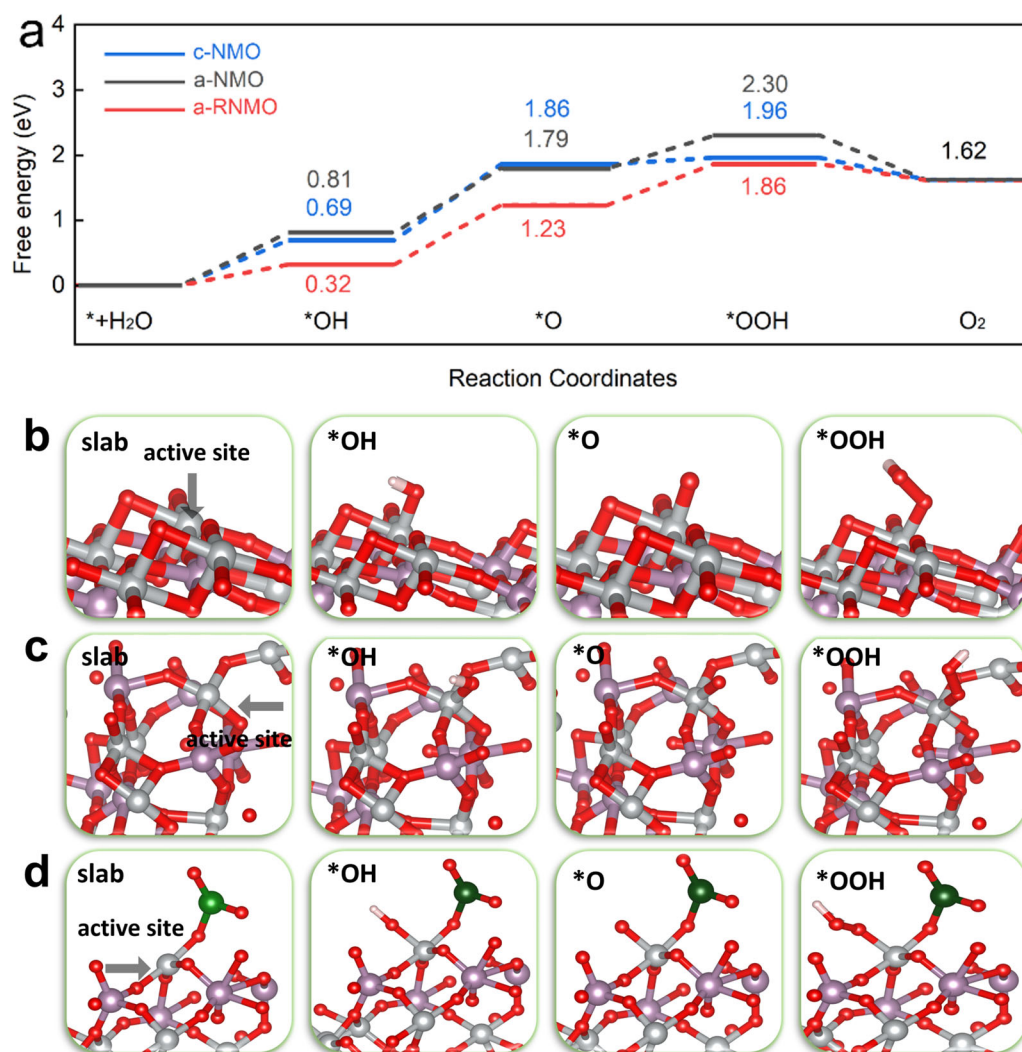
Supplementary Figure 33. Crystal model about c-NMO with three different projection directions.



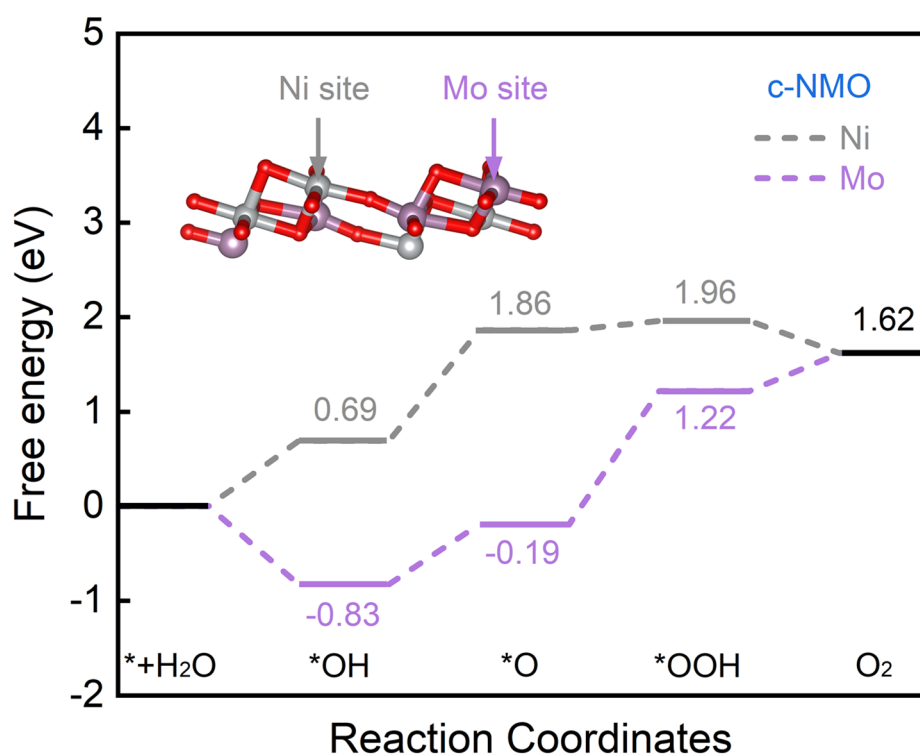
Supplementary Figure 34. PDF results from theoretical simulations, corresponding (a) c-NMO, (b) a-NMO, and (c) a-RNMO. Source data are provided as a Source Data file.



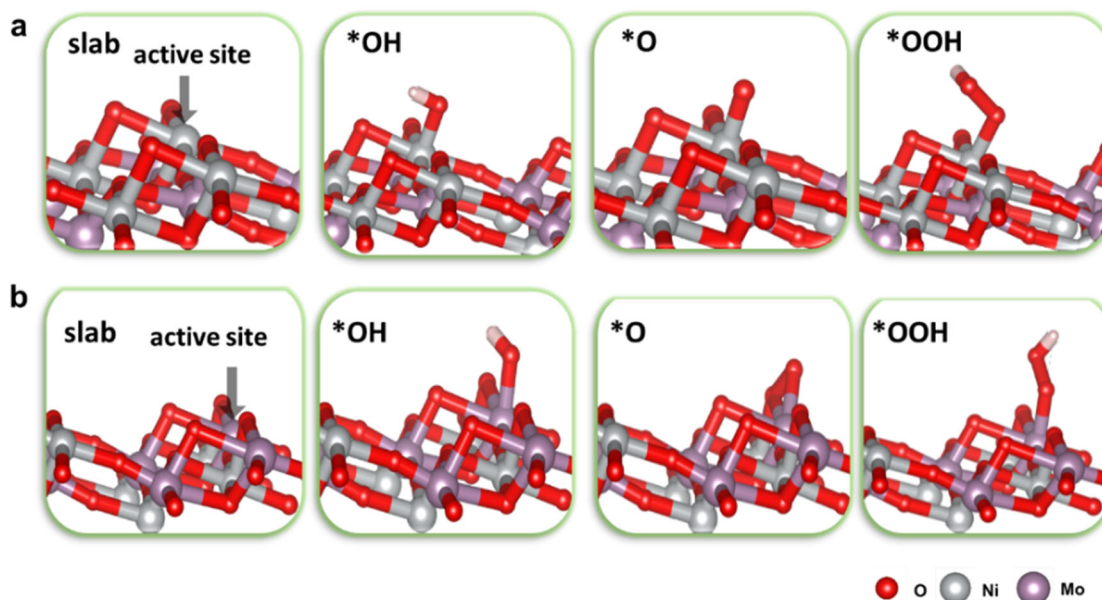
Supplementary Figure 35. The POSCAR details of (a) c-NMO, (b) a-NMO, (c) a-RNMO, and (d) a-Ru/NiOOH.



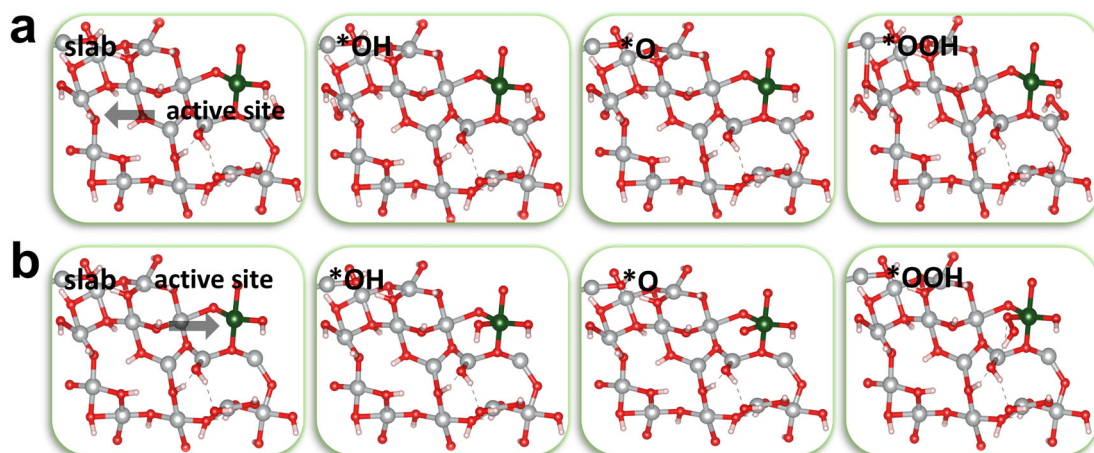
Supplementary Figure 36. (a) Gibbs free energy diagram for the four steps of the OER process (U=0 V). (b-d) The corresponding models of oxygen intermediates during OER process, such as *OH, *O, and *OOH of (b) c-NMO, (c) a-NMO, and (d) a-RNMO. The Ni ion in the (-112) crystal plane was the active site. The red, grey, purple, and green spheres represent O, Ni, Mo, and Ru, respectively. Source data are provided as a Source Data file.



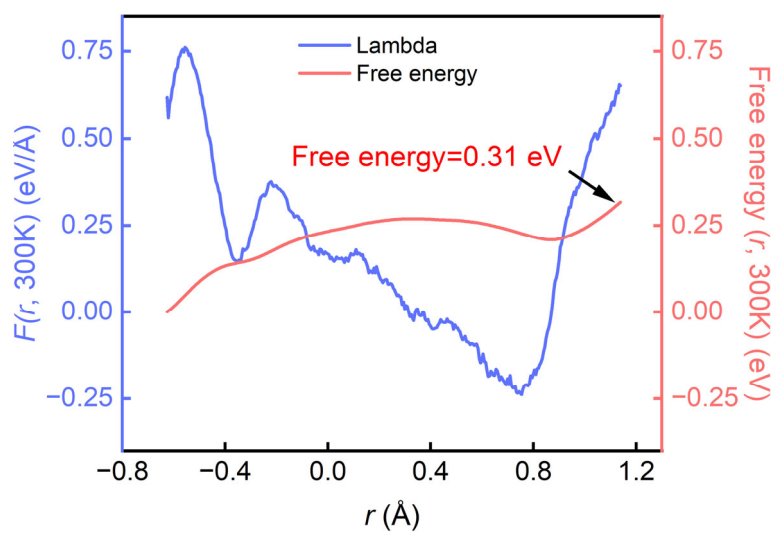
Supplementary Figure 37. Free energy results on the most stable crystal plane (-112) about c-NMO, including Ni sites and Mo sites, respectively ($U=0$ V). Source data are provided as a Source Data file.



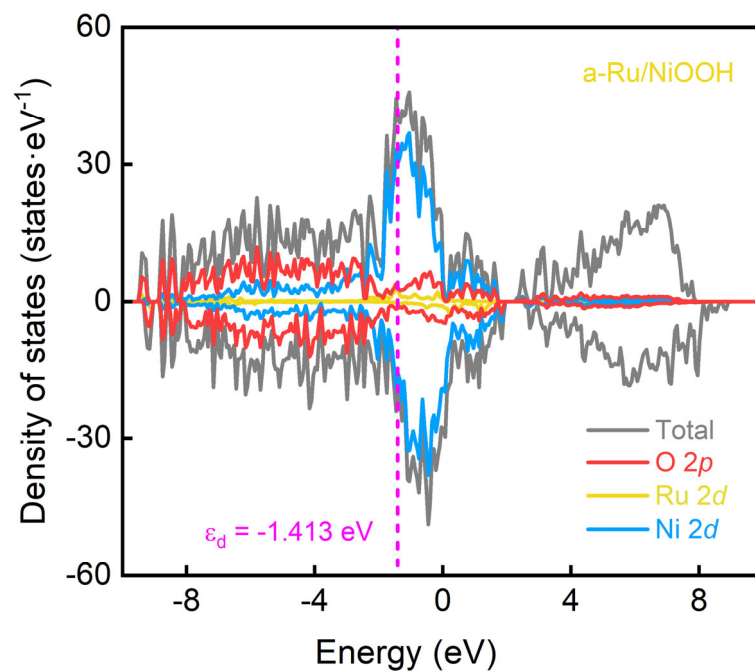
Supplementary Figure 38. The corresponding models of oxygen intermediates during OER process about c-NMO. (a) The Ni ion (grey color) was the active site. (b) The Mo ion (purple color) was the active site.



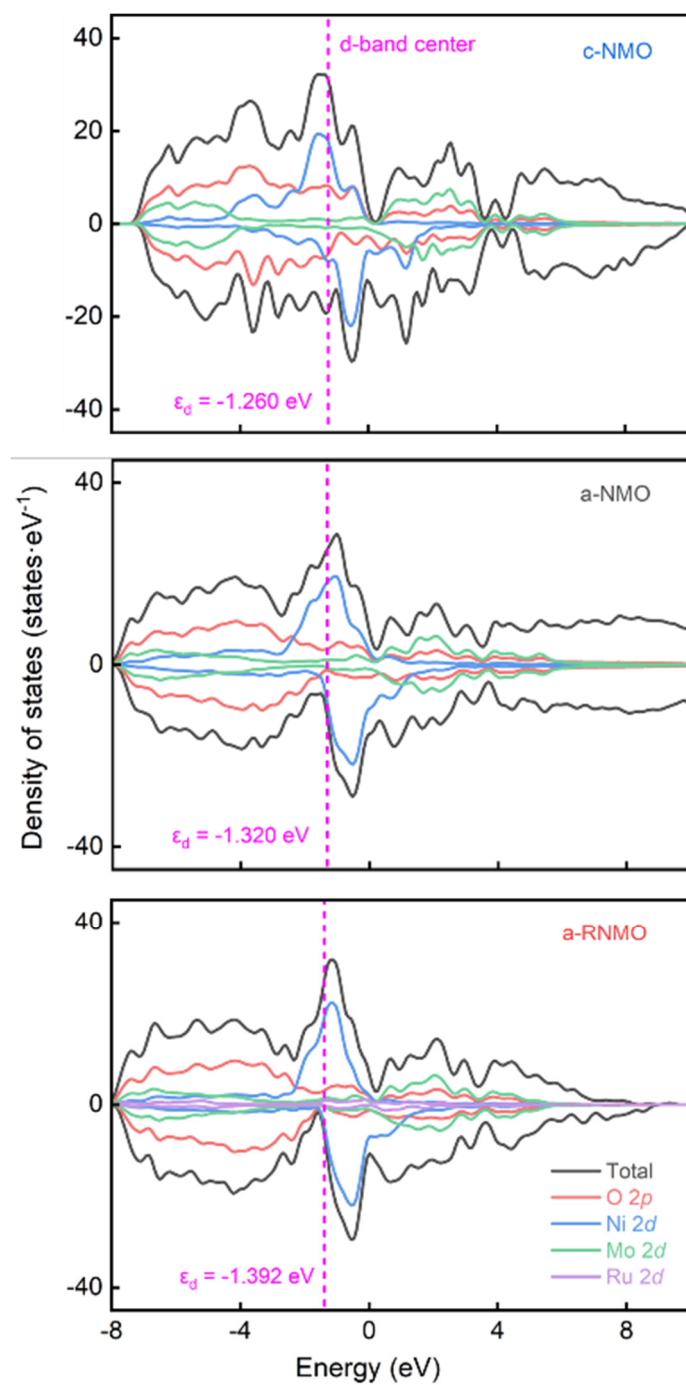
Supplementary Figure 39. The corresponding models of oxygen intermediates during OER process about a-Ru/NiOOH. (a) The Ni ion (grey color) was the active site. (b) The Ru ion (green color) was the active site.



Supplementary Figure 40. Potential of mean force (PMF) profile along the Ru-Ni sites at 300 K. Source data are provided as a Source Data file.

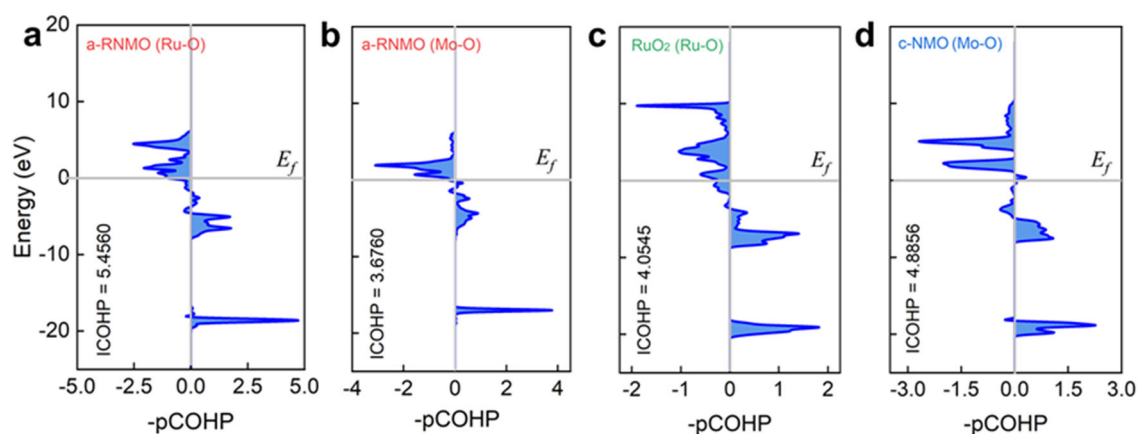


Supplementary Figure 41. Projected density of states (pDOS) curves and *d*-band center of a-Ru/NiOOH. Source data are provided as a Source Data file.

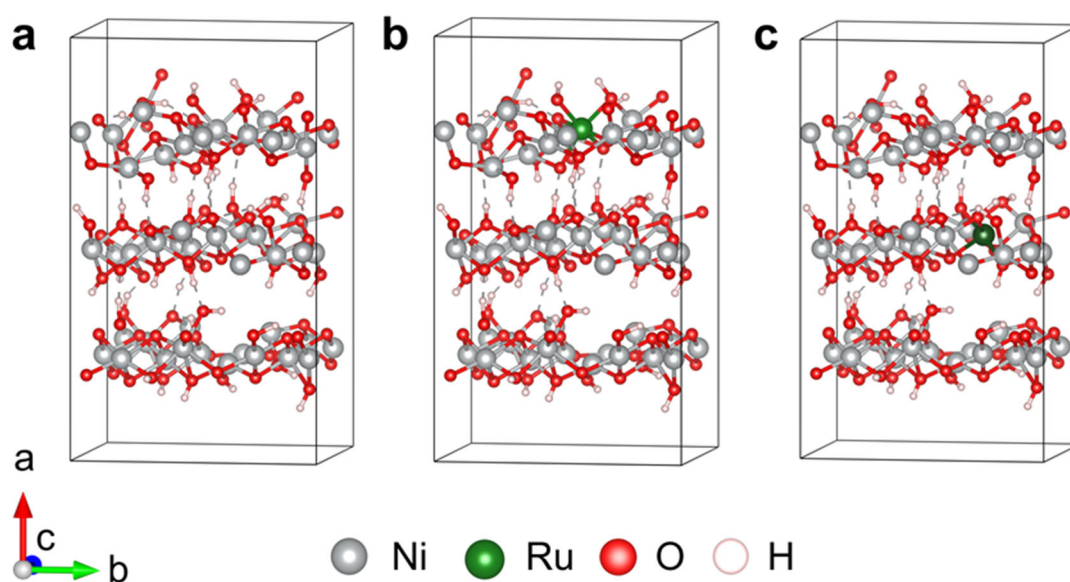


Supplementary Figure 42. DOS curves and d -band center of c-NMO, a-NMO, and a-RNMO.

Source data are provided as a Source Data file.



Supplementary Figure 43. The crystal orbital Hamilton populations (COHPs) of (a-b) a-RNMO (Ru-O, and Mo-O), (c) RuO₂ (Ru-O), and (d) c-NMO (Mo-O). Source data are provided as a Source Data file.



Supplementary Figure 44. The structure models of (a) a-NiOOH, (b) a-Ru/NiOOH, and (c) Ru dopant in a-NiOOH.

Supplementary Tables

Supplementary Table 1. The X-ray PDF refinement results of a-NMO, a-RNMO, and c-NMO. The fitting range is from 1.5 to 10 Å. R_w is the goodness-of-fit parameter. U_{iso} (units of Å²) is the isotropic atomic displacement parameter (ADP). SPD (units of Å) is the domain size of coherent scattering in the material.

		a-NMO	a-RNMO	c-NMO
R_w		0.2898	0.2497	0.1736
a (Å)		9.7098	8.7164	9.6347
b (Å)		8.1479	8.9606	8.7162
c (Å)		7.6016	7.3537	7.7298
	Ni#1	0.0197	0.0205	0.0024
	Ni#2	0.0073	0.0004	0.0152
U_{iso} (Å ²)	Mo#1	0.0031	0.0055	0.0112
	Mo#2	0.0016	0.0035	0.0018
	O	0.0002	0.0002	0.0006
SPD (Å)		7.8673	8.3899	/

Supplementary Table 2. The Mo/Ni and Ru/Ni molar ratio results by inductively coupled plasma- optical emission spectrometer (ICP-OES) measurement.

Samples (x value)	a-NMO	a-RNMO
Mo/Ni molar ratio	~0.78	~0.75
Ru/Ni molar ratio	N.A.	~0.15

Supplementary Table 3. Fitting parameters of EIS for a-RNMO, a-NMO, c-NMO, and RuO₂.

	R_s (Ohm)	CPE-T	CPE-P	R_{ct} (Ohm)
a-RNMO	5.266 ± 0.075	0.01063 ± 0.00048	0.80525 ± 0.01493	15.33 ± 0.36
a-NMO	4.796 ± 0.054	0.01287 ± 0.00018	0.94058 ± 0.00674	47.82 ± 0.76
c-NMO	5.111 ± 0.084	0.00050 ± 0.000005	0.92376 ± 0.00254	890.70 ± 9.86
RuO ₂	5.731 ± 0.129	0.00129 ± 0.00003	0.82796 ± 0.00661	101.80 ± 1.18

Supplementary Table 4. The Tafel slopes and overpotentials at 10 mA cm⁻² of reported Ru-based OER electrocatalysts.

Overpotential at 10 mA cm ⁻²	Tafel slope	Ru-based electrocatalysts	Reference
232	65.7	Ru/Co-N-C	1
240	85.4	V _{Co} -RCO	2
247	49.1	Ru SAs/LiCoO ₂	3
249	66	Ru-Co ₃ O ₄	4
250	55.4	Co ₃ O ₄ -RuO ₂ Hollow speres	5
250	59.9	P-RuO ₂ NFs	6
251	95.6	Ru SA/NiFe LDH	7
280	86.9	Ru/Co ₃ O _{4-x}	8
325	59	Ru-NiO/Co ₃ O ₄	9
223	41	Ru/NiFe LDH	10
220	45.15	a-RNMO	This work

Supplementary Table 5. The Tafel slopes and overpotentials at 10 mA cm⁻² of reported NiMo-based oxide OER electrocatalysts.

Overpotential at 10 mA cm ⁻²	Tafel slope	NiMo-based oxide electrocatalysts	Reference
217	90	Fe-NiMoO ₄ nanotube	11
221	51.7	Ir-NiMoO ₄ /Ir-Co ₃ Mn ₂ LDH	12
233	69	NiMoO _{4-x} /MoO ₂	13
236	98.4	Mn-doped NiMoO ₄	14
260	85.7	NiMoO ₄ nanorods etched by H ₂ O ₂	15
262	41.9	NiFe ₂ O _{4-x} /NiMoO ₄	16
283	44.3	N-NiMoO ₄ /NiS ₂	17
290	48	NiMoO ₄ with solution combustion method	18
320	102	Co ₃ S ₄ @NiMoO ₄ nanotube	19
340	45.6	Nanostructured NiMoO ₄	20
351	69	NiMoO ₄ quenching in Fe(NO ₃) ₃ solutions	21
360	49.83	Ni ₉ S ₈ /MoS ₂ @NiMoO ₄	22
410	70.3	P doped NiMoO ₄	23
220	45.15	a-RNMO	This work

Supplementary Table 6. The Mo/Ni and Ru/Ni molar ratio results, by ICP-MS measurement, before and after OER process.

Samples (x value)	a-RNMO pristine	a-RNMO after OER
Mo/Ni molar ratio	~0.75	~0.14
Ru/Ni molar ratio	~0.15	~0.14

Supplementary Table 7. The formation enthalpy (E_f) and dissolution potential (U_{diss}) of RuO_2 with Ru, c-NMO with Mo, a-RNMO with Ru and Mo, Ru SAs in a-NiOOH with Ru, and Ru doped in a-NiOOH with Ru.

Ru	Ru ₂ O ₄	Ru2	O2	Ef		Udiss (Ru)	Ne	Udiss	
	-44.6142	-18.4695	-9.8577	-6.42935		0.46	2	3.674674	
Mo	c-NMO	O2	Ni4	Mo2	Ef	Udiss (Mo)	Ne	Udiss	
	-360.883	-9.8577	-21.7879	-21.8608	-9.017579	-0.2	3	2.80586	
Ru	a-RNMO	O2	Ni4	Mo2	Ru2	Ef	Udiss (Ru)	Ne	Udiss
	-308.392	-9.8577	-21.7879	-21.8608	-18.4695	-9.49843	0.46	2	5.209216
Mo	a-RNMO	O2	Ni4	Mo2	Ru2	Ef	Udiss (Mo)	Ne	Udiss
	-308.392	-9.8577	-21.7879	-21.8608	-18.4695	-9.49843	-0.2	3	2.966144
Ru	Ru SAs in a-NiOOH	Ru2	Total	Ef		Udiss (Ru)	Ne	Udiss	
	-884.4657	-18.4695	-899.925	-6.22405		0.46	2	3.572025	
Ru	Ru doped in a-NiOOH	Ru2	Total	Ef		Udiss (Ru)	Ne	Udiss	
	-884.9567	-18.4695	-898.838	-4.64695		0.46	2	2.783475	

Supplementary Table 8. Zero-point energies about c-NMO, a-NMO, a-RNMO, and a-Ru/NiOOH.

c-NMO			
	*OH	*O	*OOH
Ni active site	0.31	0.03	0.36
Mo active site	0.27	0.03	0.33
a-NMO			
	*OH	*O	*OOH
Ni active site	0.3	0.01	0.34
a-RNMO			
	*OH	*O	*OOH
Ni active site	0.29	0.01	0.35
Ru active site	-0.27	0.05	0.35
a-Ru/NiOOH			
	*OH	*O	*OOH
Ni active site	0.29	0.01	0.35
Ru active site	-0.28	0.05	0.35

Supplementary References

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