

In the format provided by the authors and unedited.

Introducing curcumin biosynthesis in *Arabidopsis* enhances lignocellulosic biomass processing

Paula Oyarce^{1,2,8}, Barbara De Meester^{1,2,8}, Fernando Fonseca ^{1,2}, Lisanne de Vries ^{1,2},
Geert Goeminne^{1,2,3}, Andreas Pallidis^{1,2}, Riet De Rycke^{1,2,3,4}, Yukiko Tsuji^{5,6}, Yanding Li ^{5,6},
Sander Van den Bosch ⁷, Bert Sels⁷, John Ralph ^{5,6}, Ruben Vanholme^{1,2,3,9} and Wout Boerjan ^{1,2,3,9*}

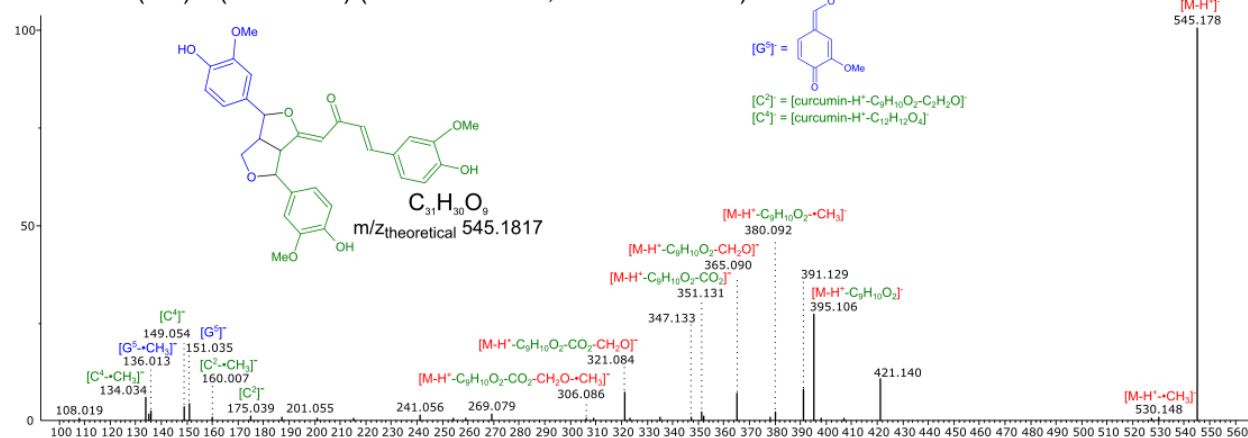
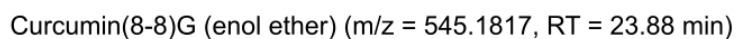
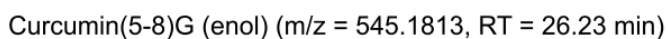
¹Department of Plant Biotechnology and Bioinformatics, Ghent University, Ghent, Belgium. ²VIB Center for Plant Systems Biology, Ghent, Belgium.

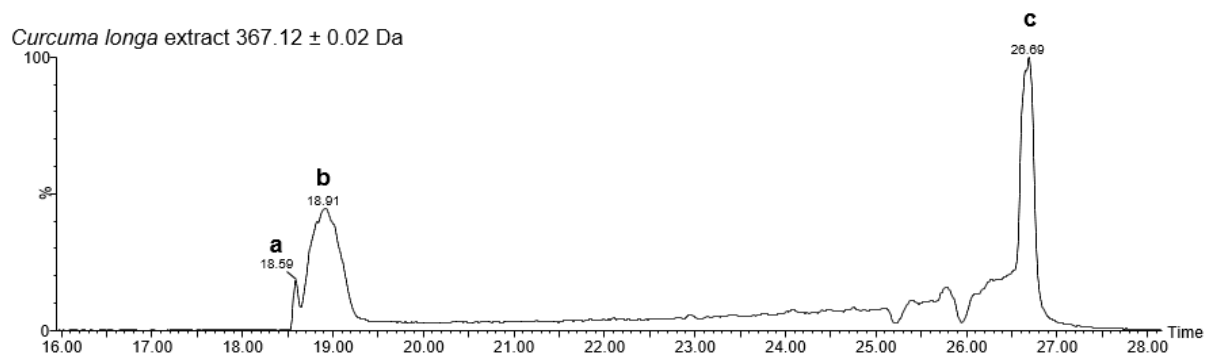
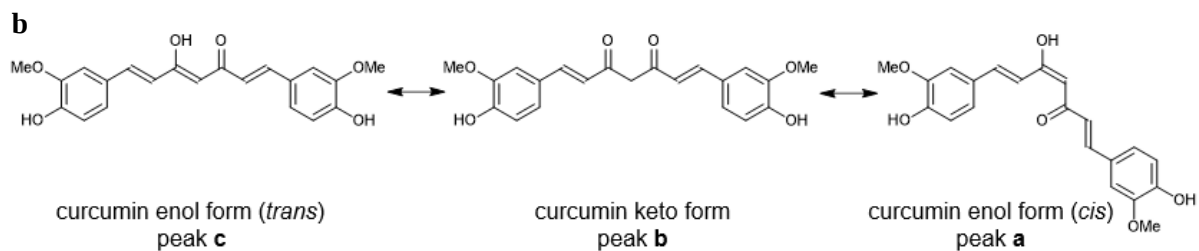
³VIB Metabolomics Core, Ghent, Belgium. ⁴Ghent University Expertise Centre for Transmission Electron Microscopy and VIB BioImaging Core, Ghent, Belgium. ⁵Department of Biochemistry, University of Wisconsin, Madison, WI, USA. ⁶US Department of Energy, Great Lakes Bioenergy Research Center, Wisconsin Energy Institute, Madison, WI, USA. ⁷Center for Surface Chemistry and Catalysis, KU Leuven, Heverlee, Belgium. ⁸These authors contributed equally:

Paula Oyarce, Barbara De Meester. ⁹These authors jointly supervised this work: Ruben Vanholme, Wout Boerjan. *e-mail: woboe@psb.vib-ugent.be

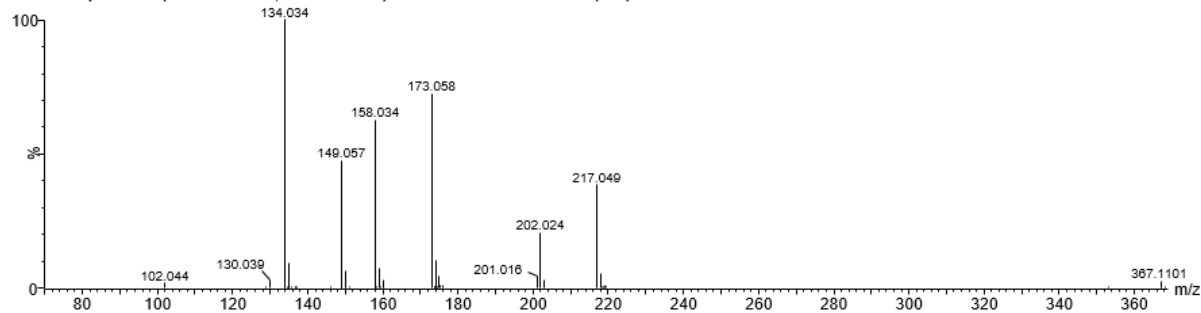
Supplementary Figure 1. MS/MS spectra and structural characterization of curcumin coupling products in oxidative coupling assays. (a) MS/MS spectra of curcumin-coniferyl alcohol coupling products observed in in vitro oxidative coupling assays. A representative MS/MS spectrum from seven, ten, and five recorded spectra for curcumin(8-O-4)G, curcumin(8-5)G and curcumin(8-8)G, respectively, is shown. Fragments and neutral losses that originate from the coniferyl alcohol moiety are given in blue, neutral losses that originate from the linkage type^{28,29} are given in red. The annotation of the curcumin fragments (in green) is explained in panels (b) to (f). Curcumin(8-O-4)G and curcumin(8-5)G were assigned as enol tautomers, based on the relative signal intensities of m/z 158.0, 160.0, 173.1 and 175.0 in the respective MS/MS spectra. For curcumin(8-8)G, a keto tautomer is impossible because the compound has an enol ether functionality. (b) Fragmentation of curcumin isomers and tautomers observed in *Curcuma longa*. The differences in MS/MS spectra between keto and enol tautomers is explained by the fragmentation pathways in panel (e) and (f). The *cis* and *trans* annotation is based on the observation that peak a and peak c have similar MS/MS spectra, but different retention times. In the *trans* conformation, the alcohol and ketone can interact with an intramolecular hydrogen bridge, making the molecule less polar and thus more retentive in the reversed-phase separation. In the *cis* conformation, the alcohol and ketone can not interact, making the molecule more polar as compared with the *trans* conformation, and thus less retentive in the reversed-phase separation. Upon ionization, the *cis* and *trans* equilibrate instantly, explaining the similar MS/MS spectra. (c) Fragmentation of demethoxycurcumin isomers and tautomers observed in *C. longa*. The differences in MS/MS spectra between keto and enol tautomers is explained by the fragmentation pathways similar to those of curcumin given in panel (e) and (f). (d) Fragmentation of bisdemethoxycurcumin isomers and tautomers observed in *C. longa*. The differences in MS/MS spectra between keto and enol tautomers is explained by the fragmentation pathways similar to those of curcumin given in panel (e) and (f). For (b), (c) and (d) one *Curcuma longa* sample was analyzed and one MS/MS spectrum was generated for each compound. (e) Fragmentation pathway of curcumin keto form. (f) Fragmentation pathway of curcumin enol form.

Curcumin(4-O-8)G (enol) (m/z = 563.1920, RT = 22.78 min)

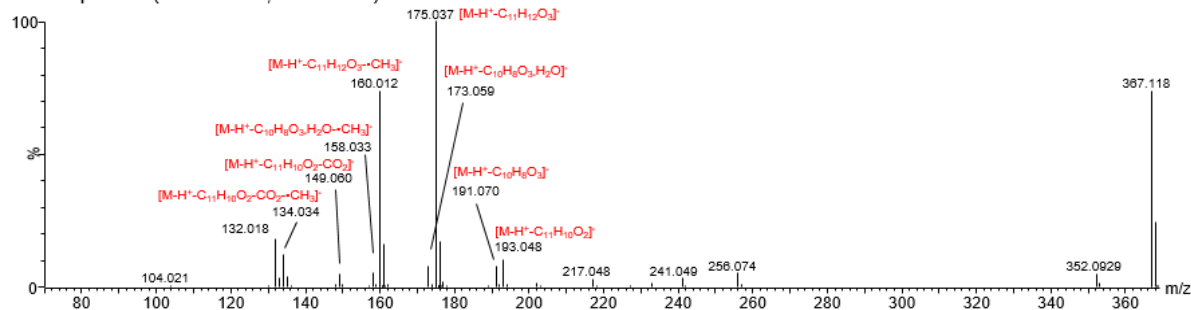




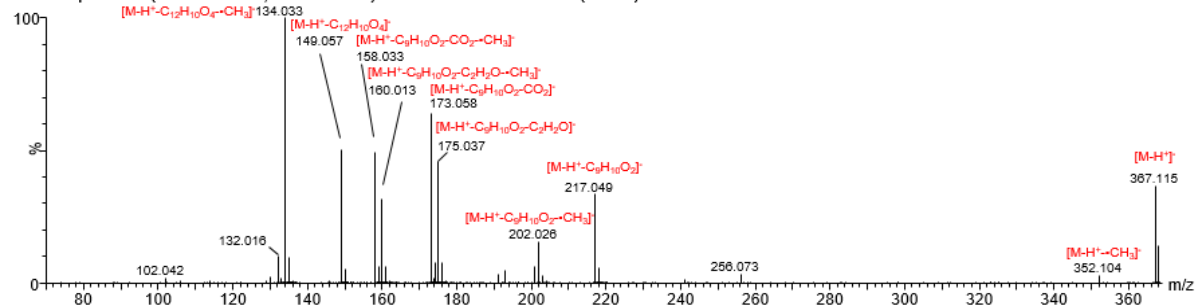
MS/MS peak **a** (m/z 367.12, 18.59 min): curcumin enol form (*cis*)

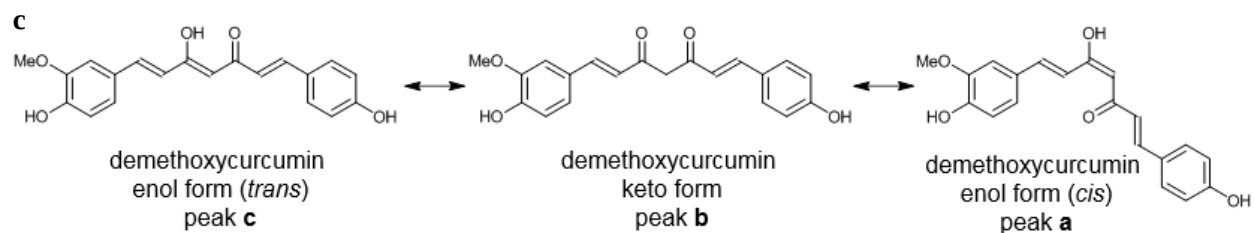


MS/MS peak **b** (m/z 367.12, 18.91 min): curcumin keto form

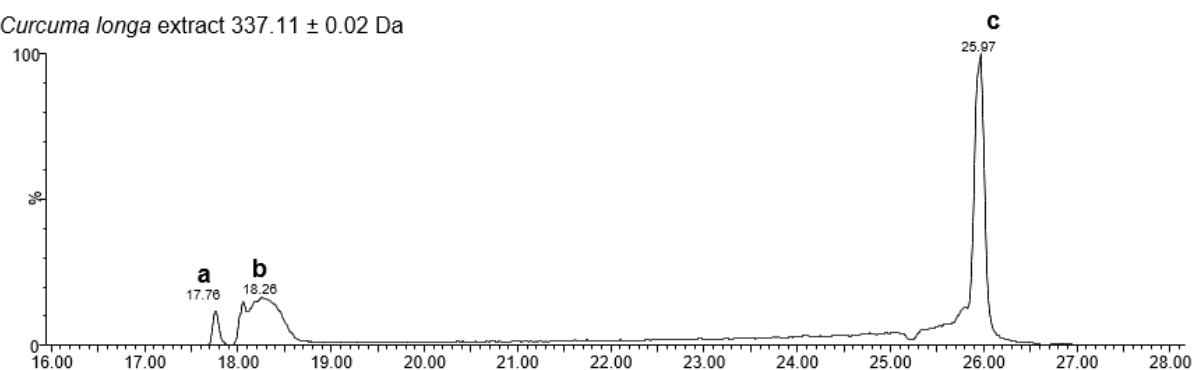


MS/MS peak **c** (m/z 367.12, 26.69 min): curcumin enol form (*trans*)

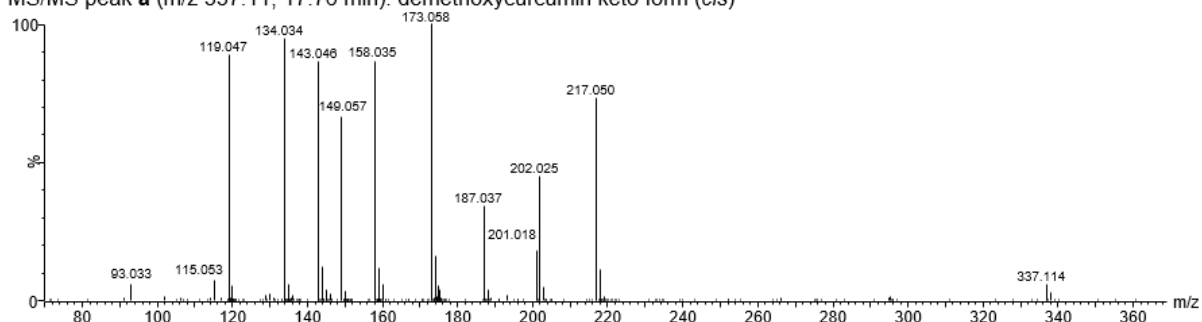




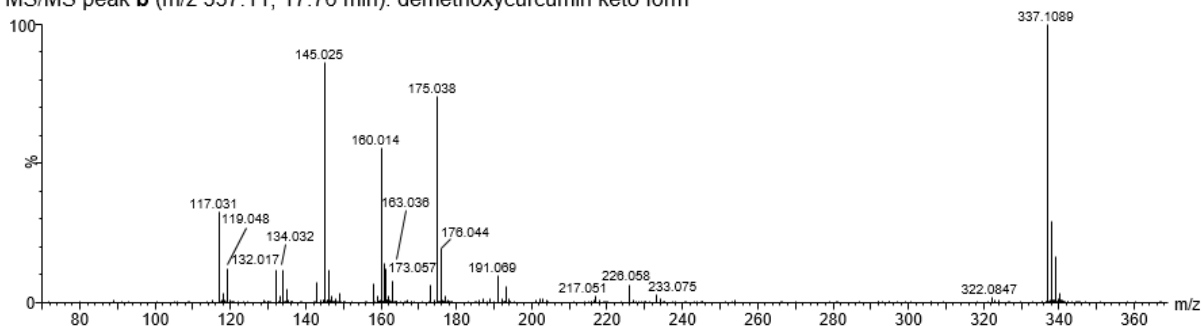
Curcuma longa extract 337.11 ± 0.02 Da



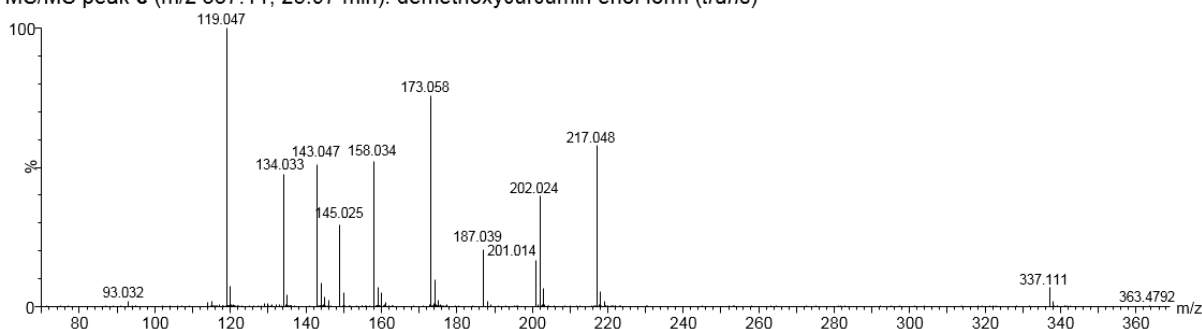
MS/MS peak **a** (m/z 337.11, 17.76 min): demethoxycurcumin keto form (*cis*)

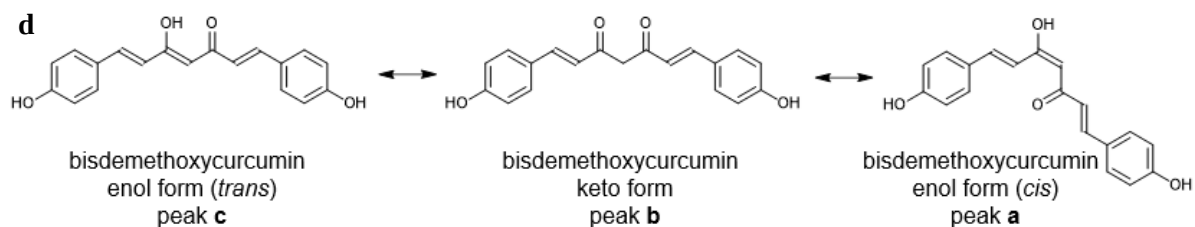


MS/MS peak **b** (m/z 337.11, 17.76 min): demethoxycurcumin keto form

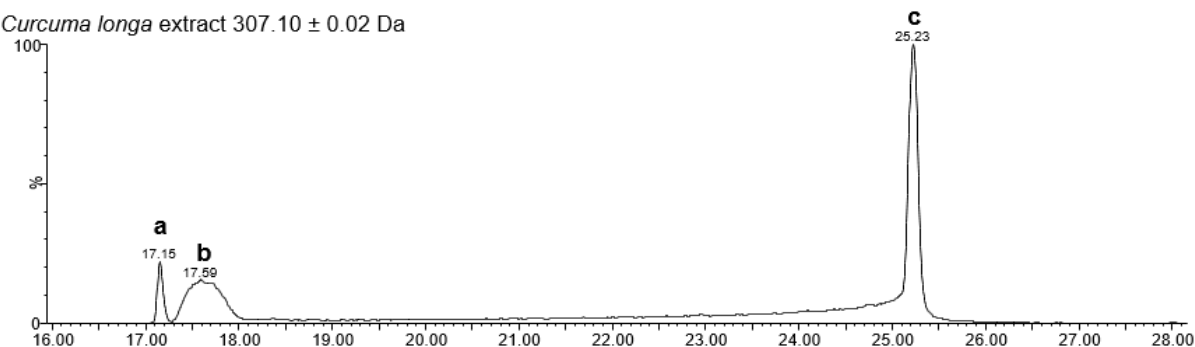


MS/MS peak **c** (m/z 337.11, 25.97 min): demethoxycurcumin enol form (*trans*)

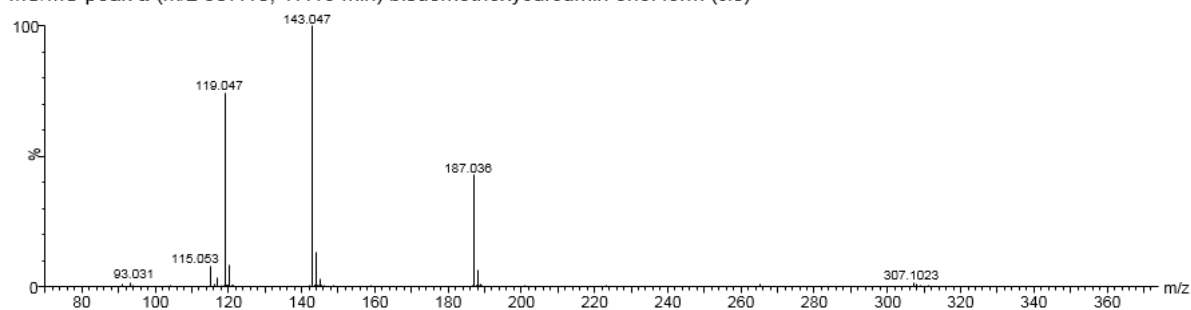




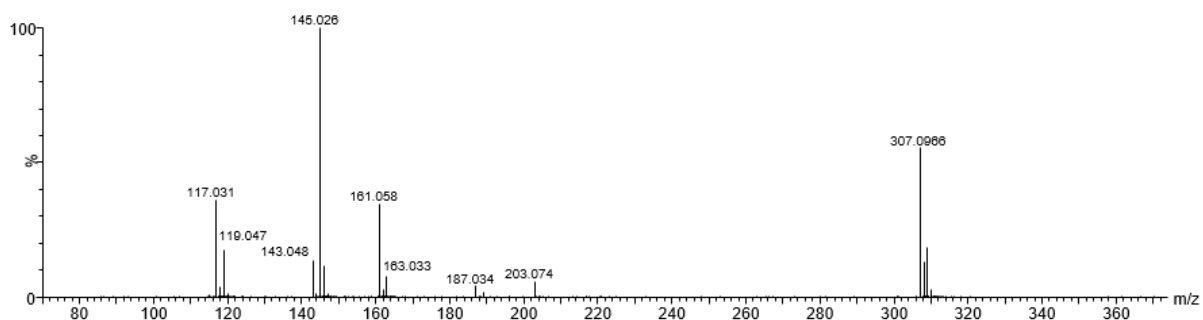
Curcuma longa extract 307.10 ± 0.02 Da



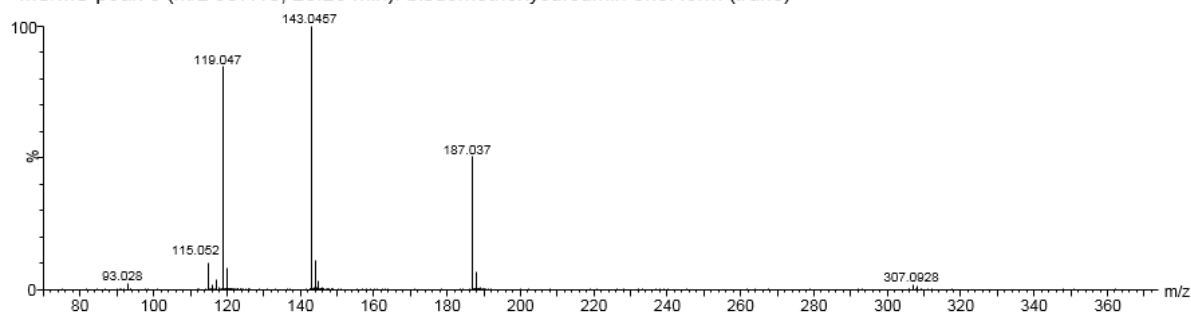
MS/MS peak **a** (m/z 307.10, 17.15 min) bisdemethoxycurcumin enol form (*cis*)



MS/MS peak **b** (m/z 307.10, 17.59 min): bisdemethoxycurcumin keto form

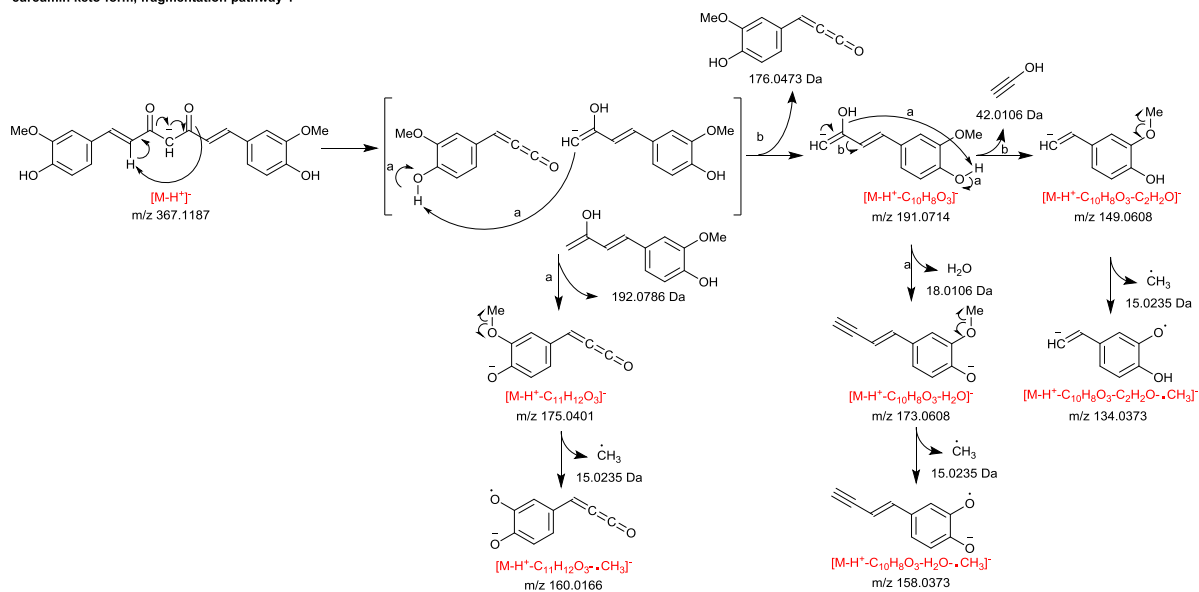


MS/MS peak **c** (m/z 307.10, 25.23 min): bisdemethoxycurcumin enol form (*trans*)

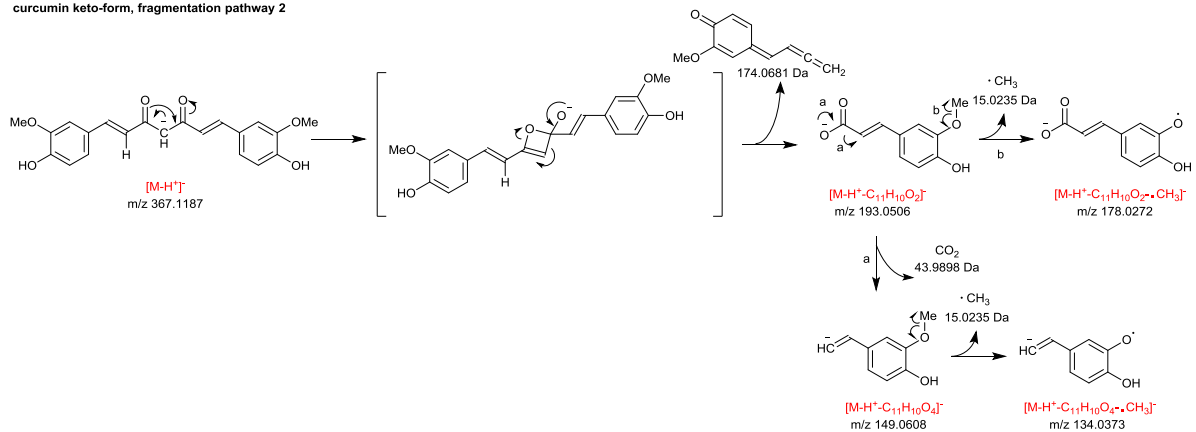


e

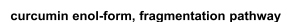
curcumin keto-form, fragmentation pathway 1



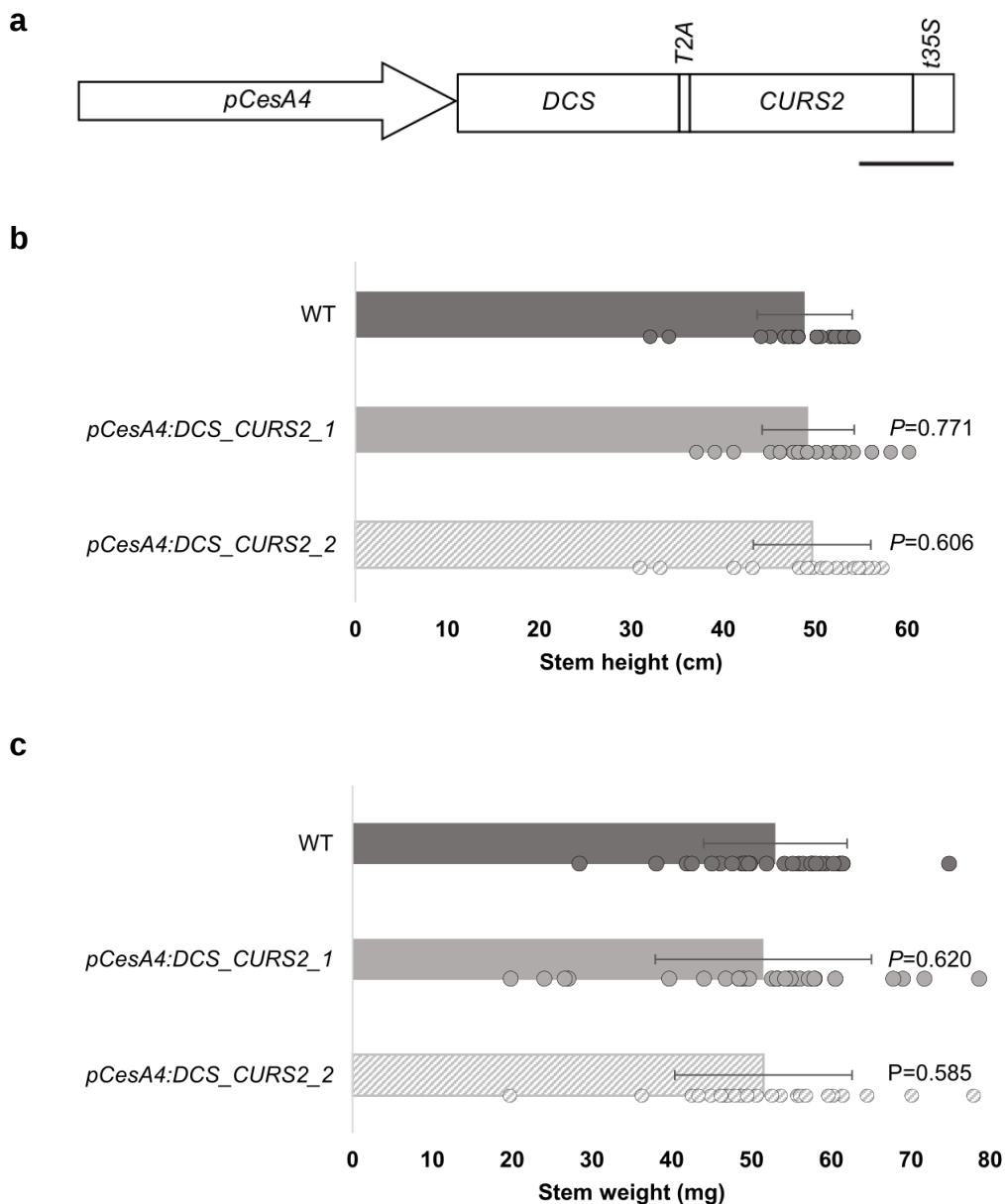
curcumin keto-form, fragmentation pathway 2



f

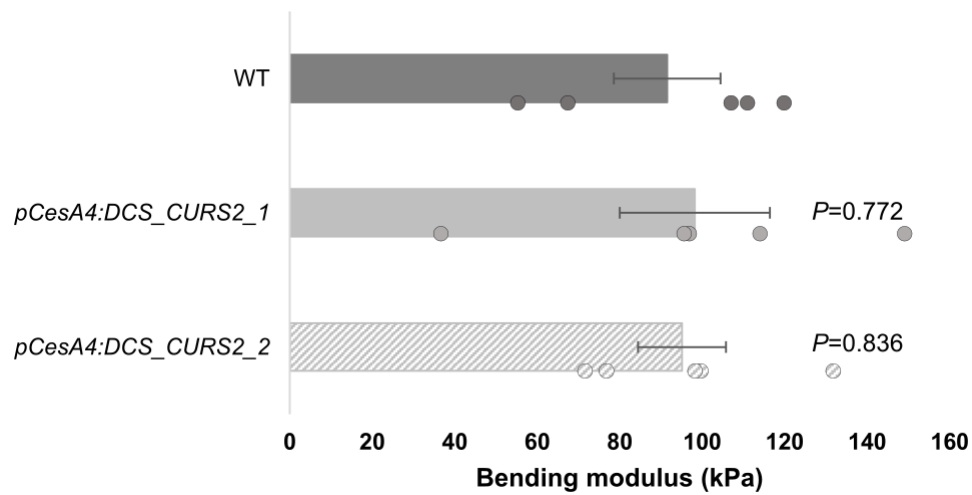


Supplementary Figure 2. Gene construct and biomass measurements of *pCesA4:DCS_CURS2* transgenic *Arabidopsis* lines. (a) Graphical representation of the *pCesA4:DCS_CURS2* construct. *DCS* and *CURS2* genes were linked together by the *T2A* sequence, controlled by the *CesA4* promoter and terminated with a 35S terminator sequence (t35S). The different elements are drawn to scale. Scale bar: 500 bp. (b) Height measurements of the *pCesA4:DCS_CURS2* lines upon senescence. (c) Dry weight of the senesced main inflorescence stem of which siliques, cauline leaves and side branches were removed. Individual values (dots) and means (bars) of 29, 30 and 26 independent biological replicates for WT, *pCesA4:DCS_CURS2_1* and *pCesA4:DCS_CURS2_2*, respectively, are shown. No significant differences were observed for stem height and weight between the lines ($P > 0.05$; two-tailed Student's *t*-test). Error bars indicate standard deviation.



Supplementary Figure 3. Bending modulus determined by two-point bending tests.

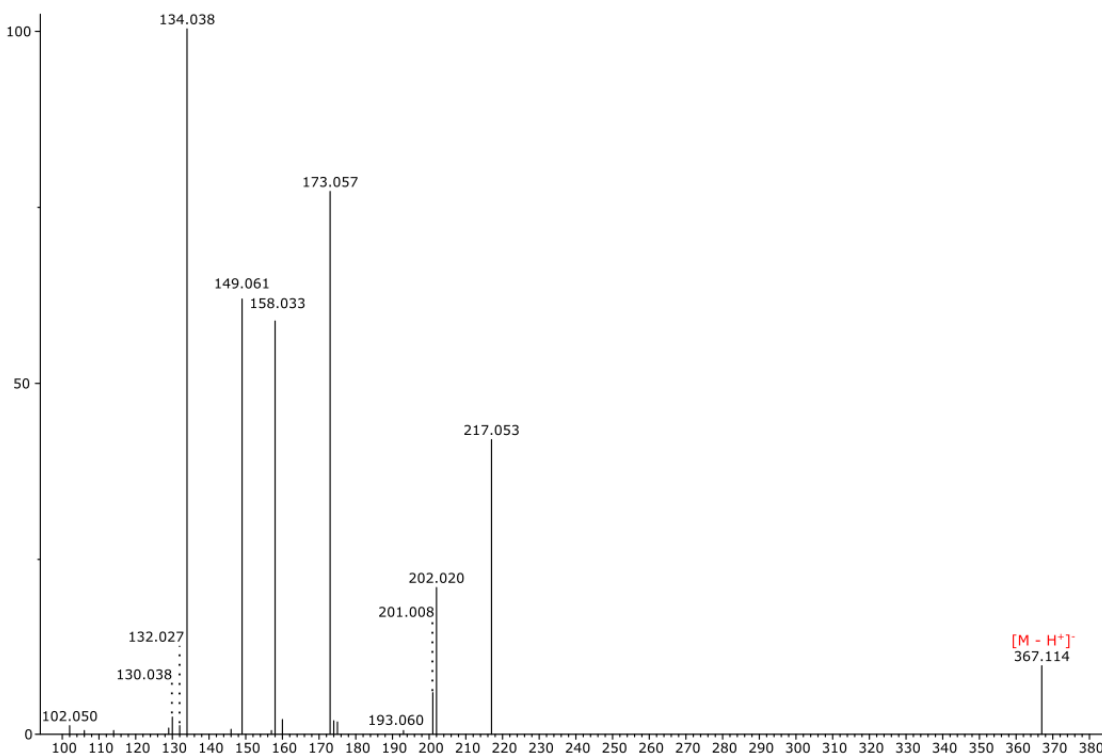
The bending modulus was determined to estimate the stiffness of the stems ($n=5$). Bars and error bars indicate means and standard errors, respectively. No significant differences were observed between the lines ($P > 0.05$; two-tailed Student's t -test).



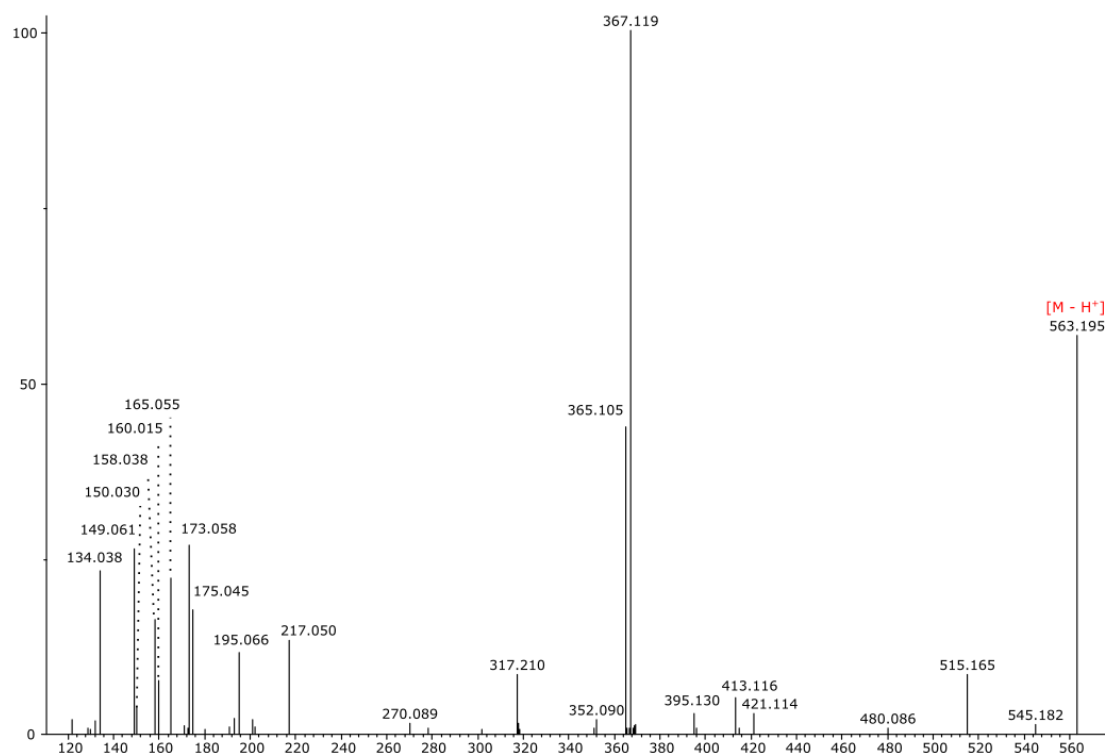
Supplementary Figure 4. Phenolic profiling of *pCesA4:DCS_CURS2* plants. MS/MS fragmentation spectra for compounds detected in extracts of *pCesA4:DCS_CURS2* plants. For each compound, a representative MS/MS spectrum is shown from at least four independent MS/MS spectra. The interpretation of the MS/MS spectra of compound **1-5** is given in Supplementary Figure 1. The fragments indicated with a red triangle in the MS/MS spectra of compound **6-21**, **26**, **27**, **32-38** and **40** refer to fragments from dihydroferuloyl- β -keto acid, as explained by the fragmentation routes shown for compound **6** (dihydroferuloyl- β -keto acid). In the same spectra, neutral losses of $C_9H_{10}O_2$ and $C_{10}H_{10}O_3$ are also explained by the fragmentation routes shown for compound **6** (dihydroferuloyl- β -keto acid). For compound **7**, the glycerol is thought to be on the 4-O position of dihydroferuloyl- β -keto acid, because the observed $[M - H^+ - C_2H_4O_2 - H_2O]^-$ and $[M - H^+ - CO_2 - C_2H_4O_2]^-$ fragments can only be formed if the 9CHOH and 11COOH positions are free. The fragments indicated with a red circle in the MS/MS spectra of compound **17**, **22-25** and **41** refer to fragments from tetrahydroferuloyl- β -keto acid, as explained by the fragmentation routes shown for compound **22** (tetrahydroferuloyl- β -keto acid). In the same spectra, the neutral loss corresponding to $C_2H_4O_2$ and $C_{10}H_{12}O_3$ is also explained by the fragmentation routes shown for compound **22** (tetrahydroferuloyl- β -keto acid). For compound **28**, fragmentation of the 8-5 (phenylcoumaran) structure resulting into H_2O , CH_2O and $^{1,2}A$ neutral losses and $^{1,2}B^-$ fragments, is as described in Morreel et al., 2004³⁵. In the same MS/MS spectrum, CO_2 , C_2H_4O , $C_2H_4O_2$ and $C_4H_6O_3$ neutral losses are derived from the dihydroferuloyl- β -keto acid, as shown for compound **6** (dihydroferuloyl- β -keto acid). For compound **30** and **31**, fragments indicated with a red star are explained in the fragmentation route for compound **30** (dihydroferuloyl- β -keto acid(8-5)G + glutamate). For compound **32** [dihydroferuloyl- β -keto acid(δ -lactone β -O-4)dihydroferuloyl- β -keto acid], the peaks indicated with the red diamond in the MS/MS spectrum are explained in the fragmentation route. For compound **33**, fragmentation resulting into the $^{2,5}X$ fragment, is as described in Morreel et al., 2004³⁵. In the same spectrum, CO_2 and $C_2H_4O_2$ neutral losses are derived from the dihydroferuloyl- β -keto acid, as shown for compound **6**. The $C_3H_4O_3$ neutral loss is also derived from the dihydroferuloyl- β -keto acid moiety, as explained in the fragmentation route. Compound **39** is characterized as dihydroferuloyl- β -keto acid sinapic acid cyclobutane dimer, because the peak with m/z 223.0614 co-elutes with a peak with m/z 461.1476 (see insert). The latter indicates that m/z 223.0614 is an in-source fragment of m/z 461.1476. There is no MS/MS spectrum available for m/z 461.1476, but the MS/MS of m/z 223.0614 is highly similar to that of the sinapic acid standard. In addition, the experimental m/z 461.1476 closely matches the theoretical m/z of dihydroferuloyl- β -keto acid sinapic acid cyclobutane dimer (m/z 461.1476, Δppm 5.0). The observed neutral losses of $C_2H_4O_2$, $C_3H_6O_3$, $C_4H_8O_4$ in the spectra of compound **13**, **14**, **29** are derived from the hexose cross-ring cleavages. For compound **41**, fragmentation of the 4-O-8 (β -aryl ether) structure resulting into H_2O and CH_2O neutral losses and $[G^1]^-$ and $[G^2]^-$ fragments, is as described in Morreel et al., 2004³⁵. In the same MS/MS spectrum, the $C_2H_4O_2$ neutral loss is derived from the tetrahydroferuloyl- β -keto acid, as shown for compound **22** (tetrahydroferuloyl- β -

keto acid). Compound **42-49** matched retention time, m/z and MS/MS spectra as described in De Meester et al., 2018³⁶.

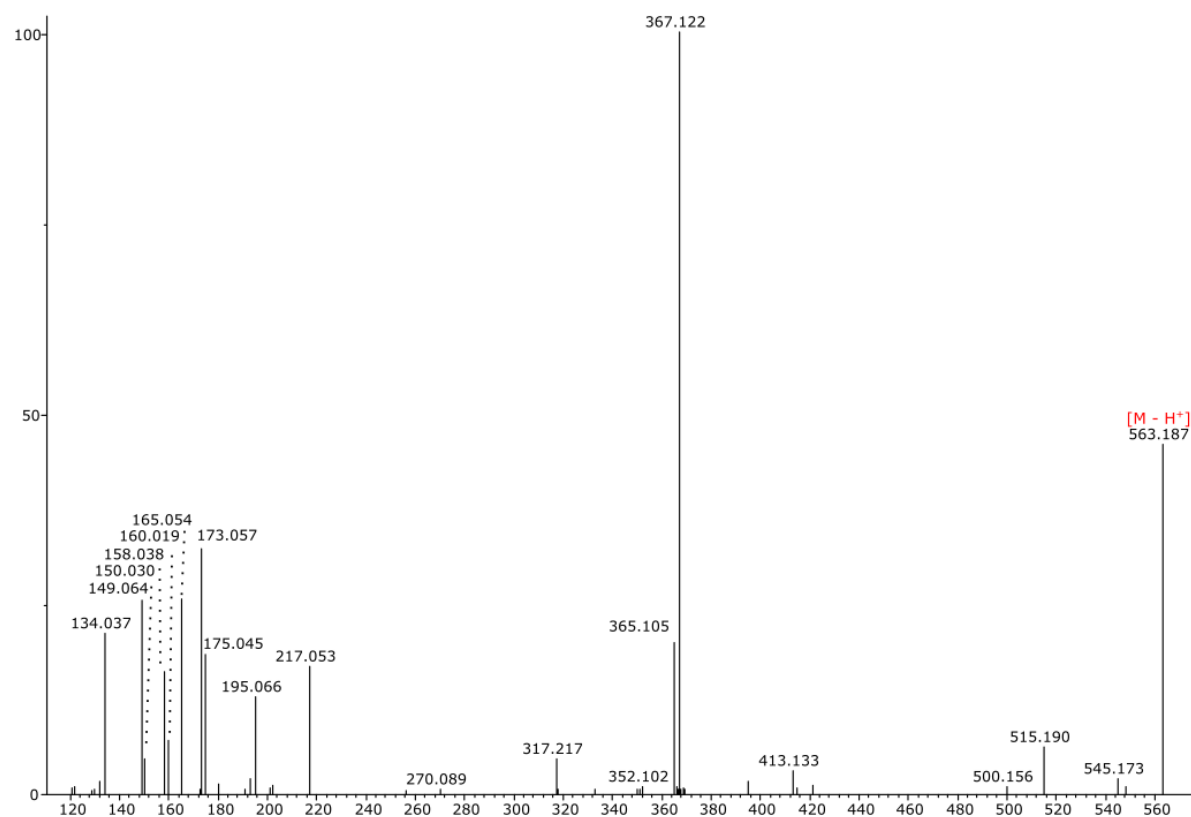
1. MS/MS curcumin (enol) (26.46 min, m/z 367.1202)



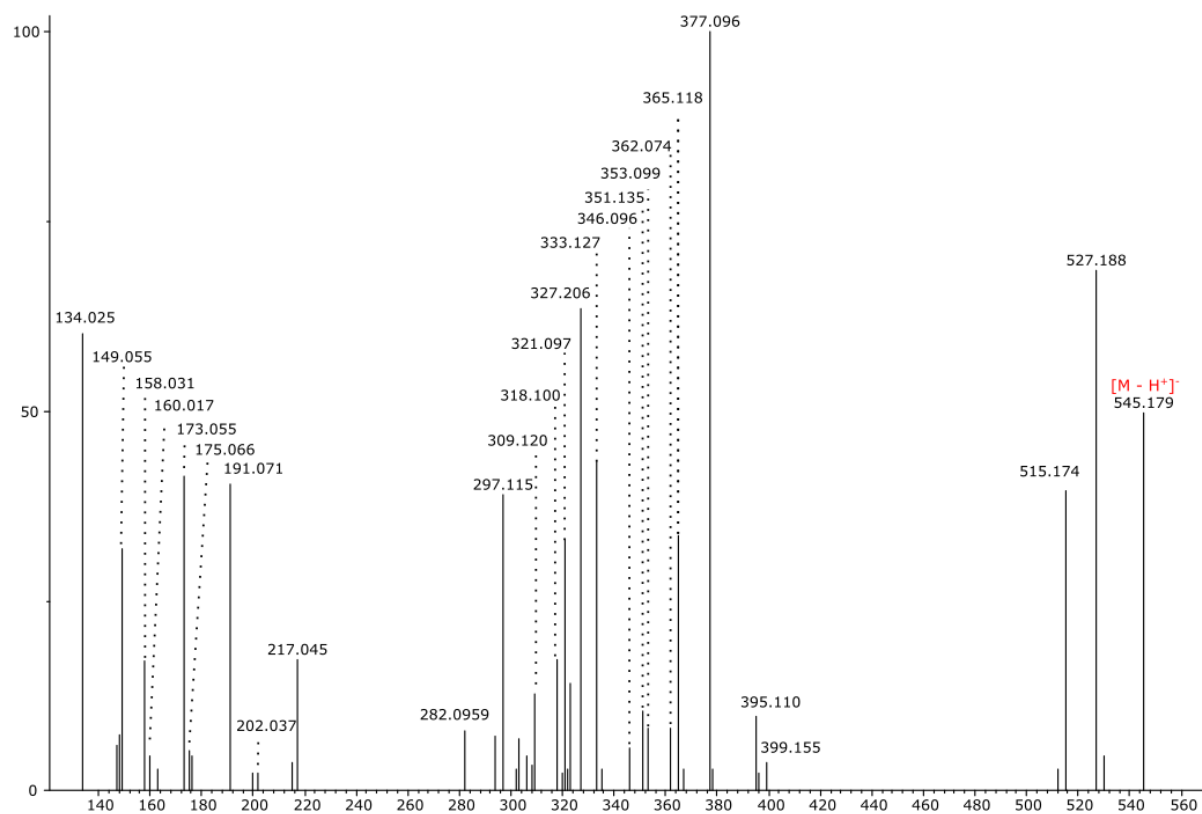
2. MS/MS curcumin(4-O-8)G (enol) 1 (23.83 min, m/z 563.1925)



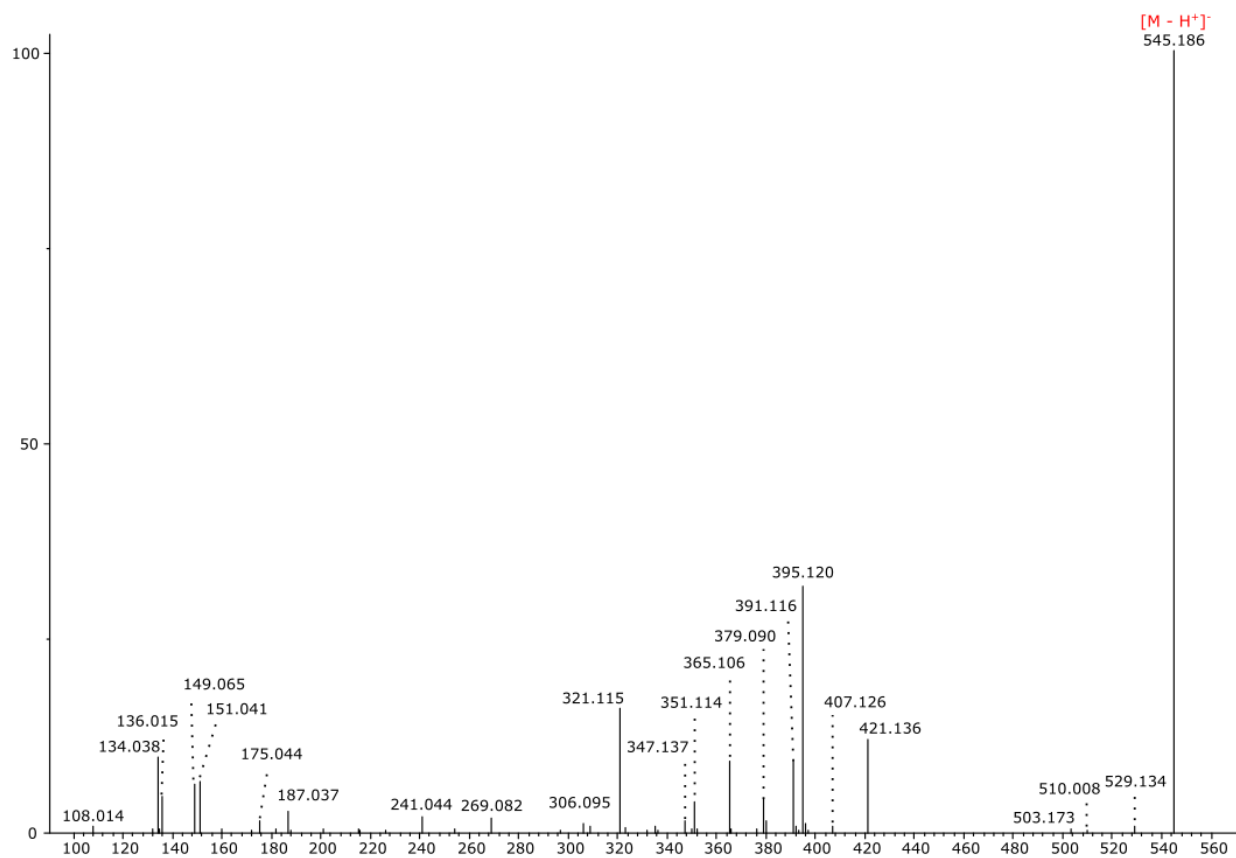
3. MS/MS curcumin(4-O-8)G (enol) 2 (23.93 min, m/z 563.1928)



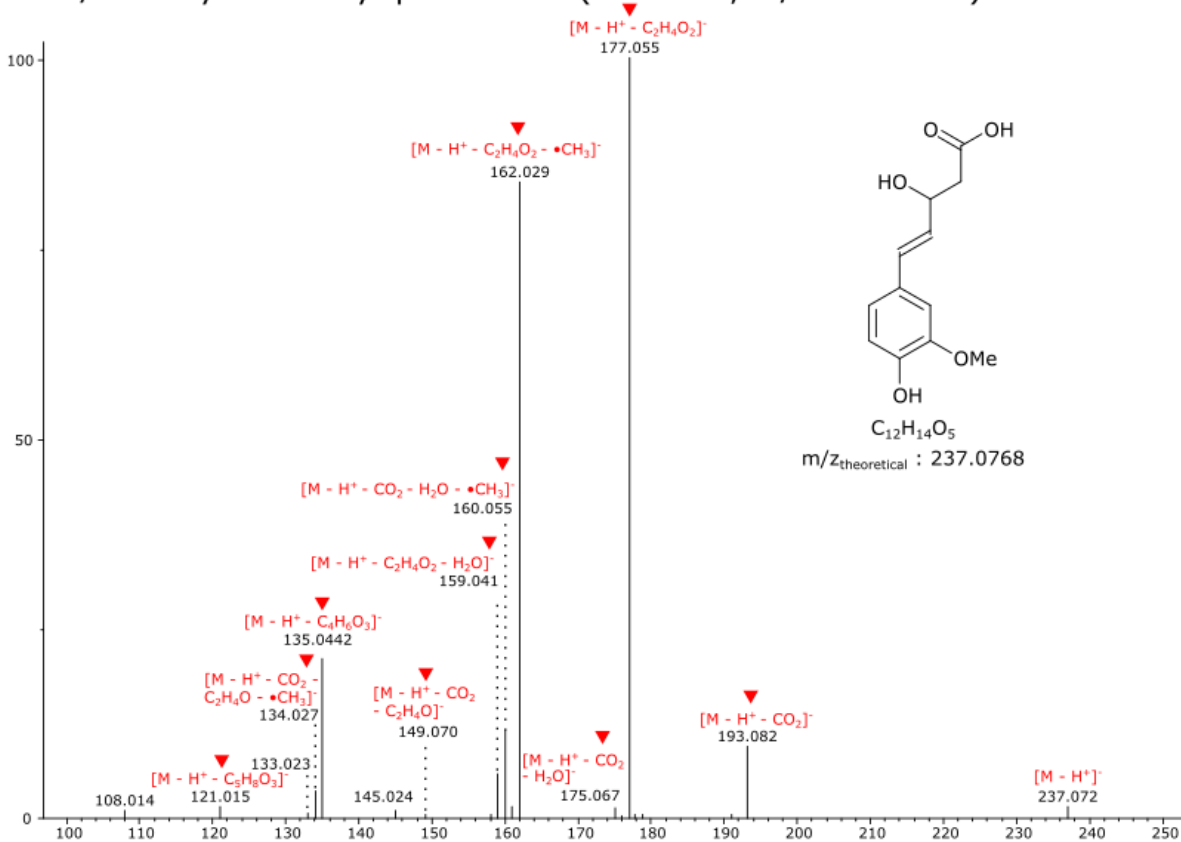
4. MS/MS curcumin(5-8)G (enol) (27.55 min, m/z 545.1822)

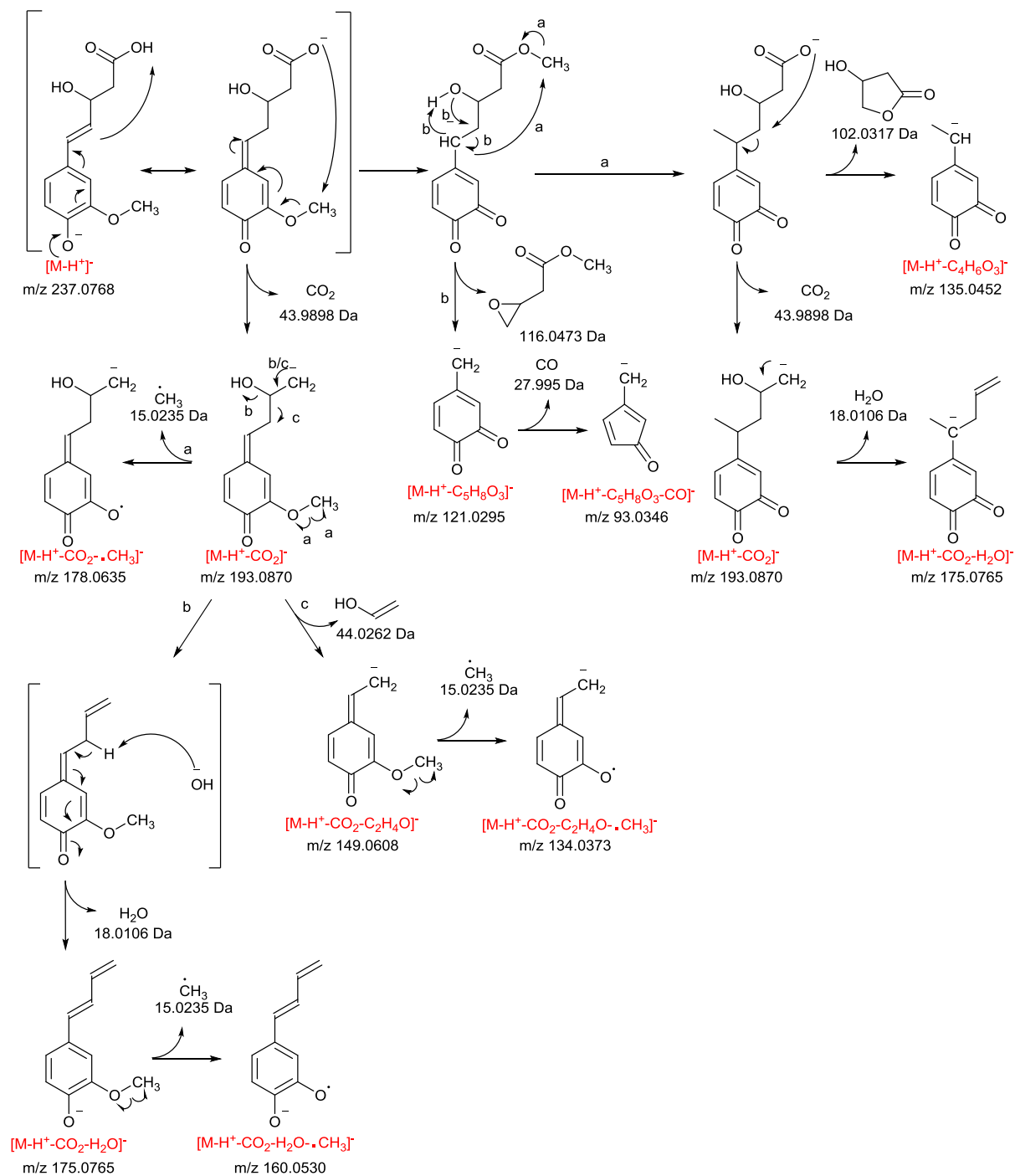


5. MS/MS curcumin(8-8)G (enol ether) (25.09 min, m/z 545.1828)

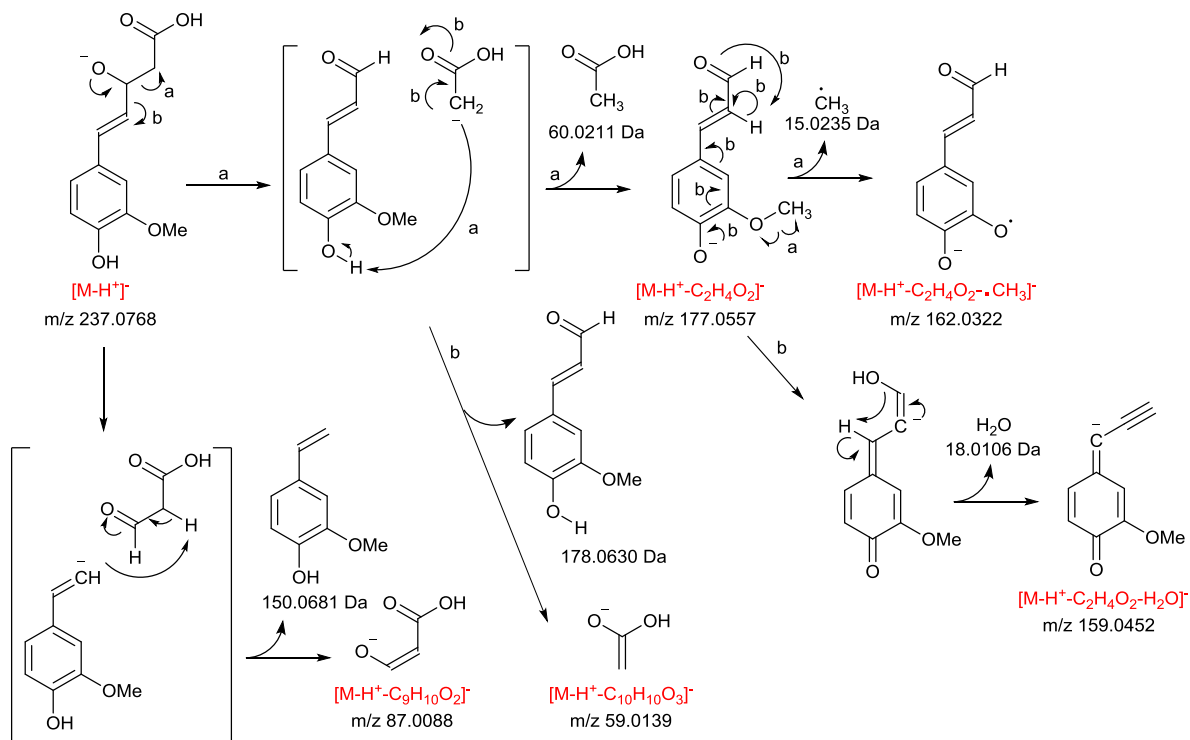


6. MS/MS dihydroferuloyl- β -keto acid (6.28 min, m/z 237.0770)

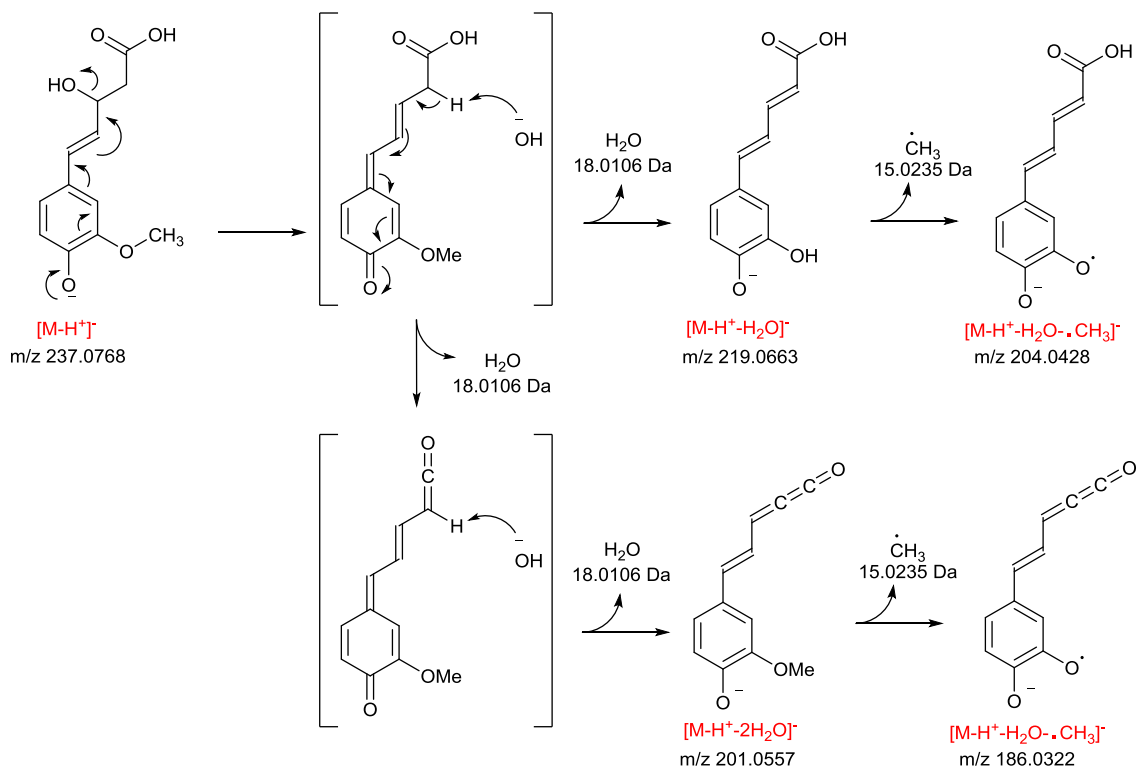




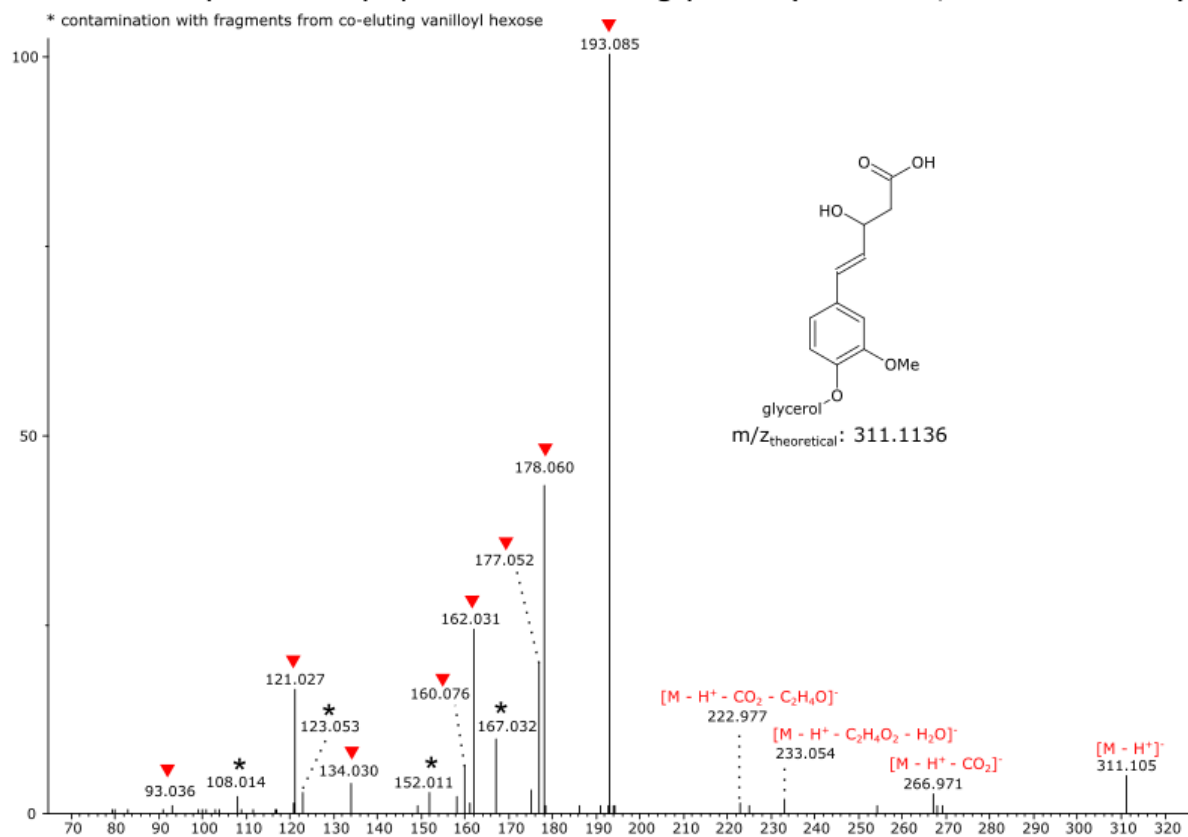
dihydroferuloyl- β -keto acid fragmentation pathway 2



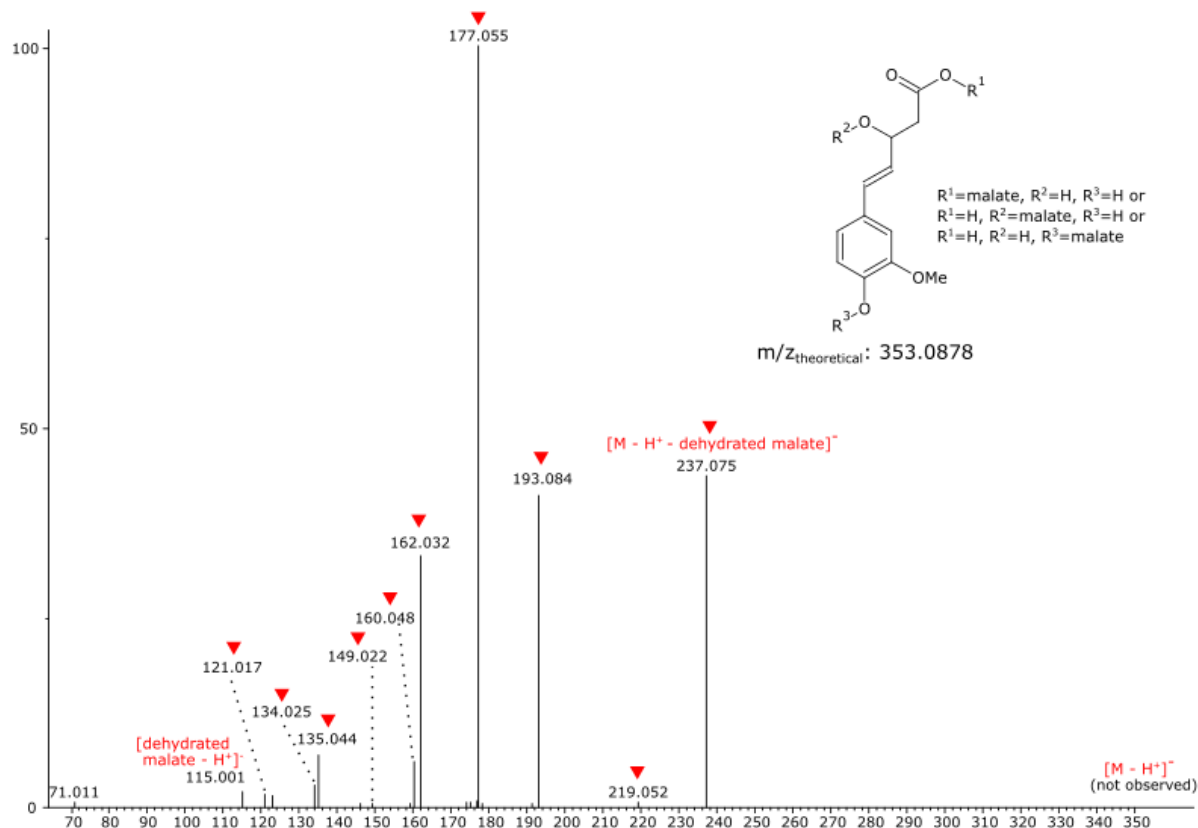
dihydroferuloyl- β -keto acid fragmentation pathway 3



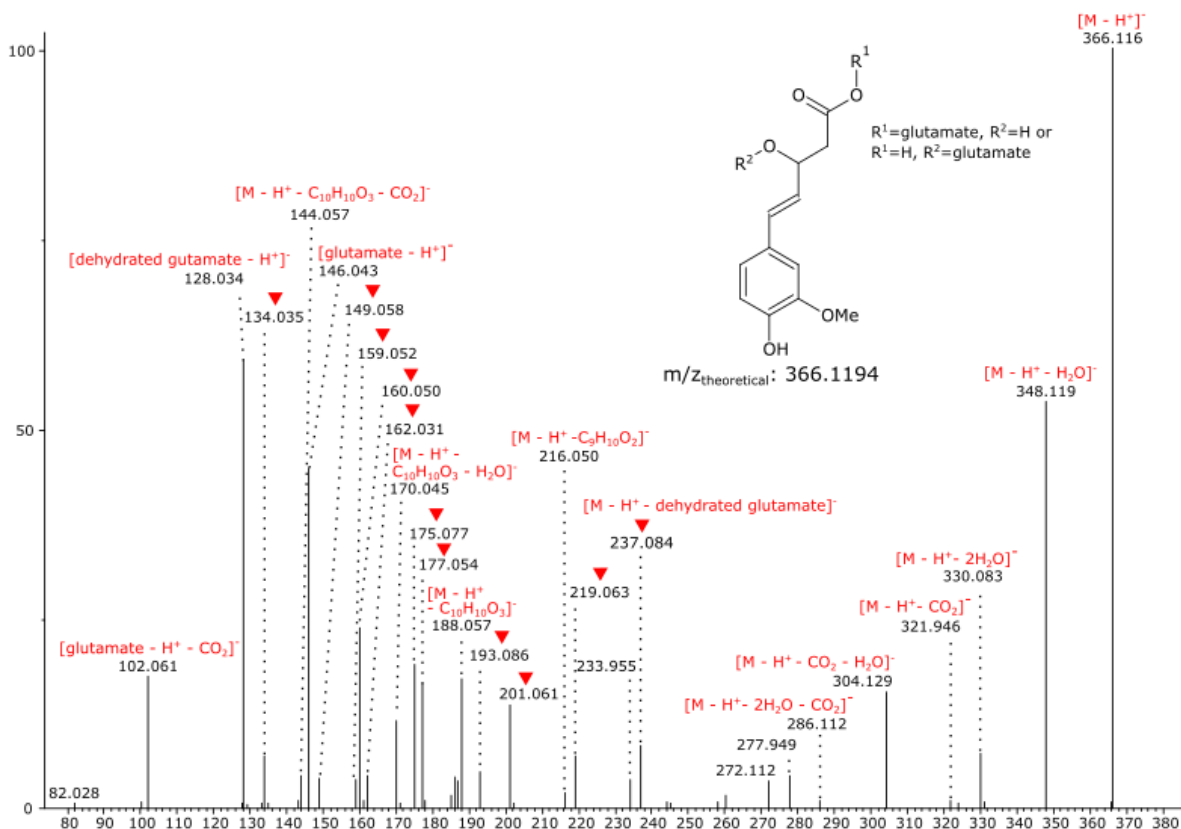
7. MS/MS dihydroferuloyl- β -keto acid-4-*O*-glycerol (4.90 min, m/z 311.1148)



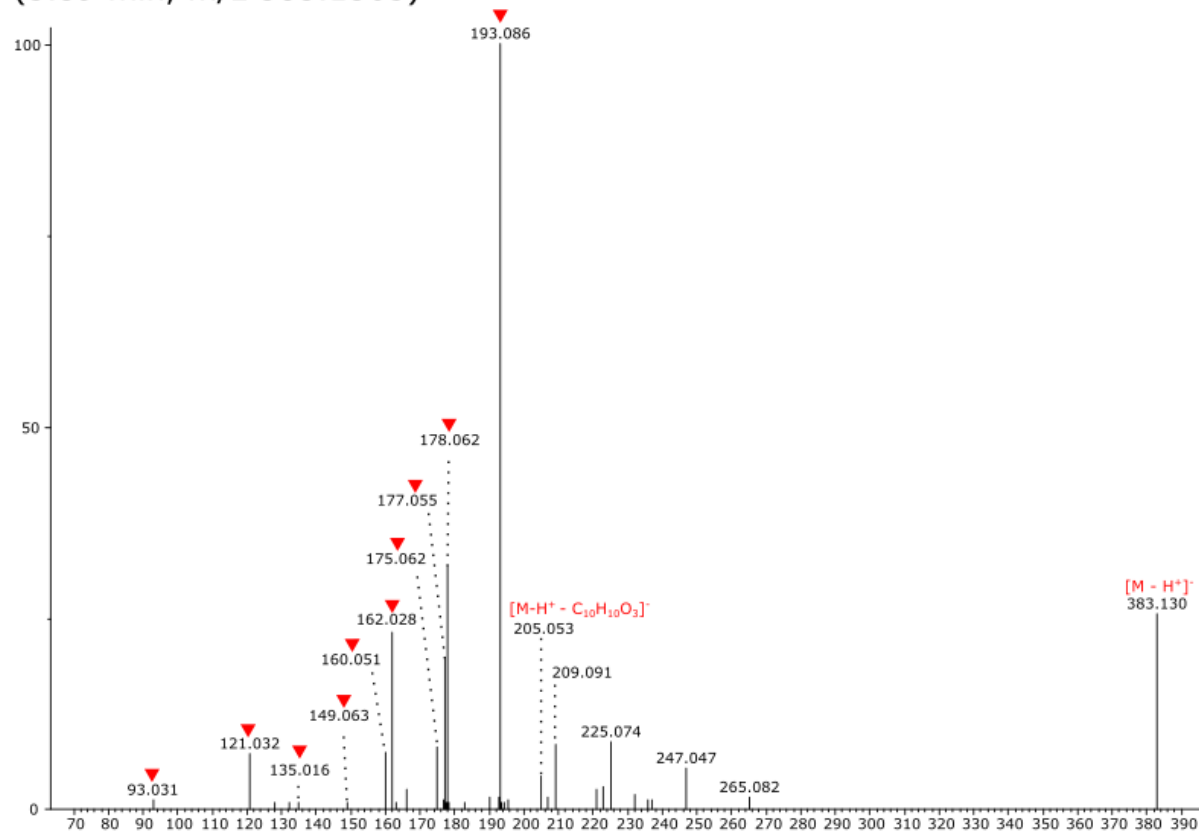
8. MS/MS dihydroferuloyl- β -keto acid + malate (7.00 min, m/z 353.0894)



9. MS/MS dihydroferuloyl- β -keto acid + glutamate (4.80 min, m/z 366.1210)

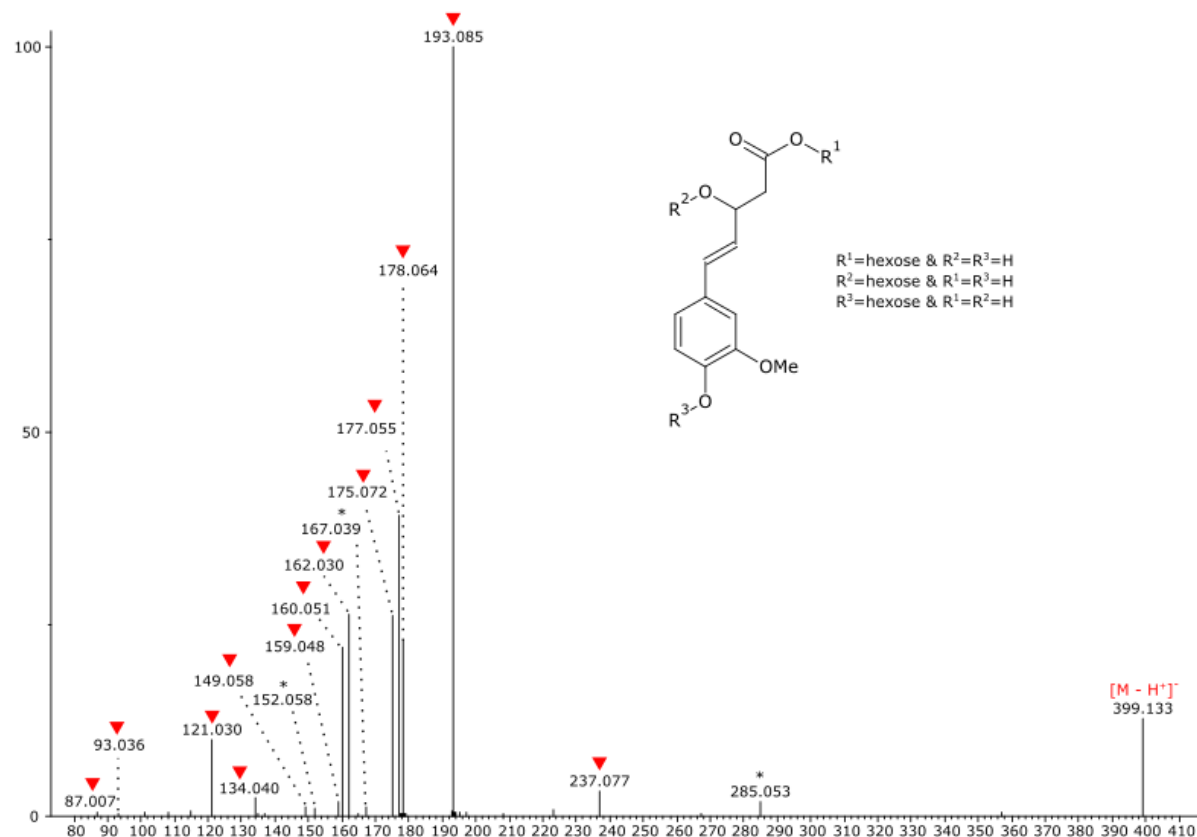


10. MS/MS dihydroferuloyl- β -keto acid + deoxyhexose (5.39 min, m/z 383.1365)

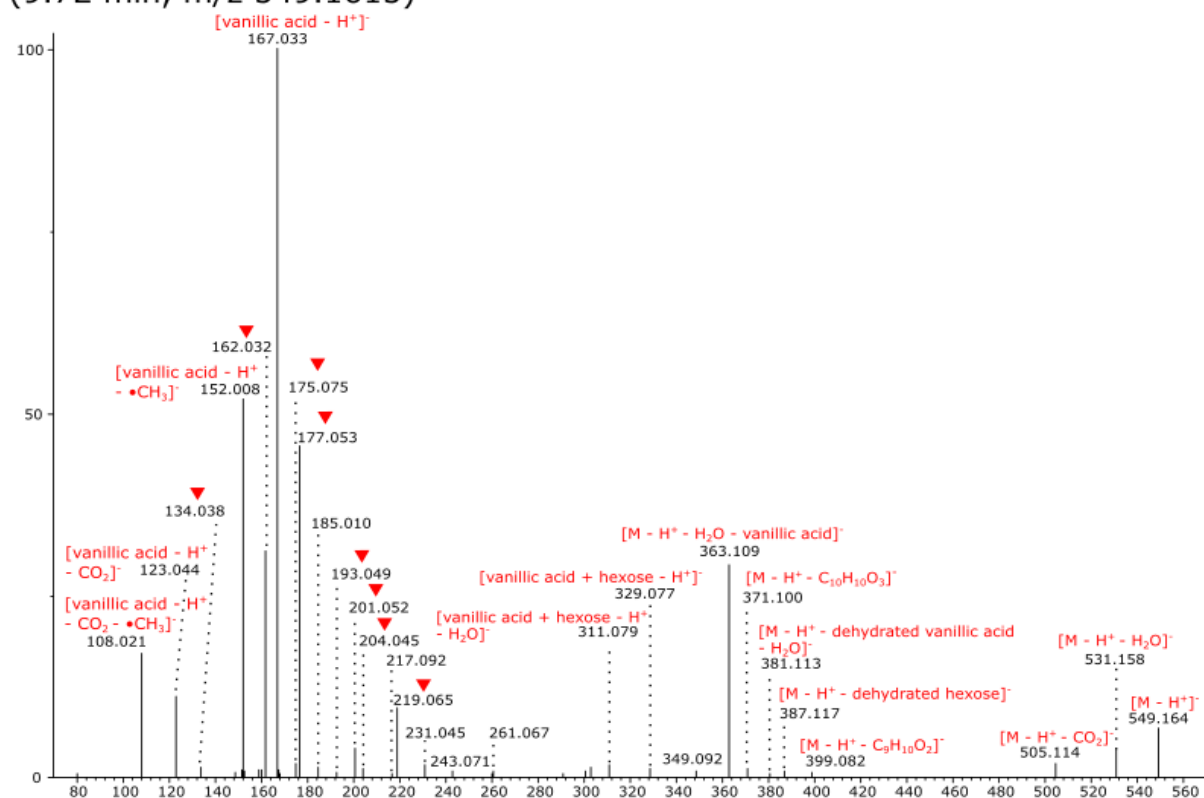


11. MS/MS dihydroferuloyl- β -keto acid + hexose (3.15 min, m/z 399.1306)

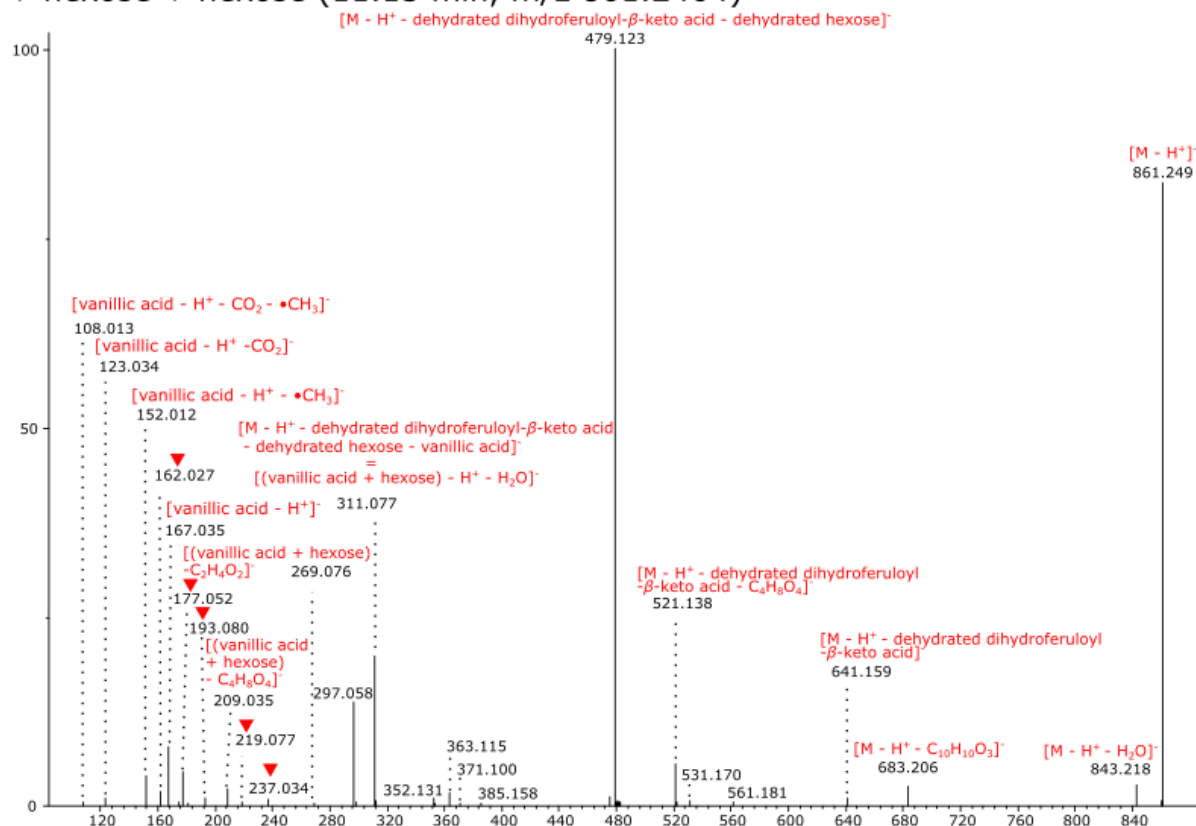
* contamination with fragments from co-eluting dihydroxybenzoic acid pentoside



12. MS/MS dihydroferuloyl- β -keto acid + vanillic acid + hexose (9.72 min, m/z 549.1615)

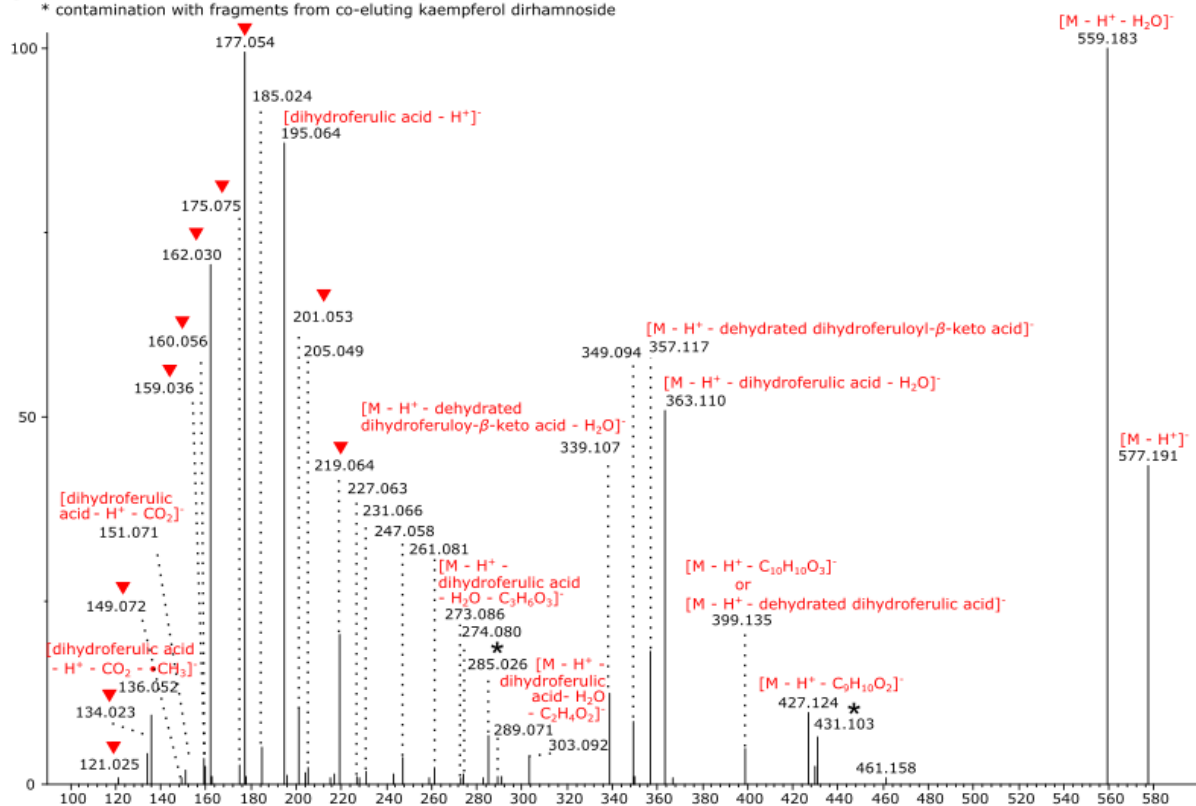


13. MS/MS dihydroferuloyl- β -keto acid + vanillic acid + vanillic acid + hexose + hexose (11.13 min, m/z 861.2464)

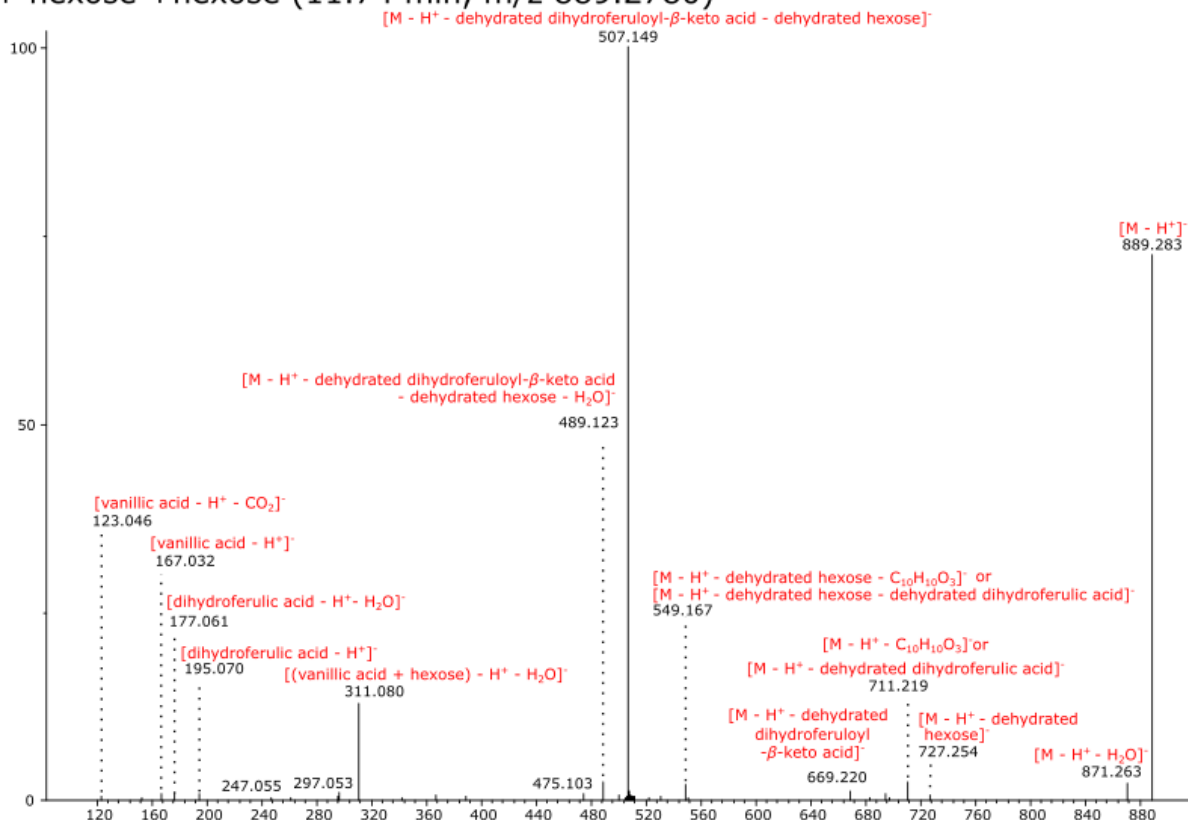


14. MS/MS dihydroferuloyl- β -keto acid + dihydroferulic acid + hexose (11.87 min, m/z 577.1909)

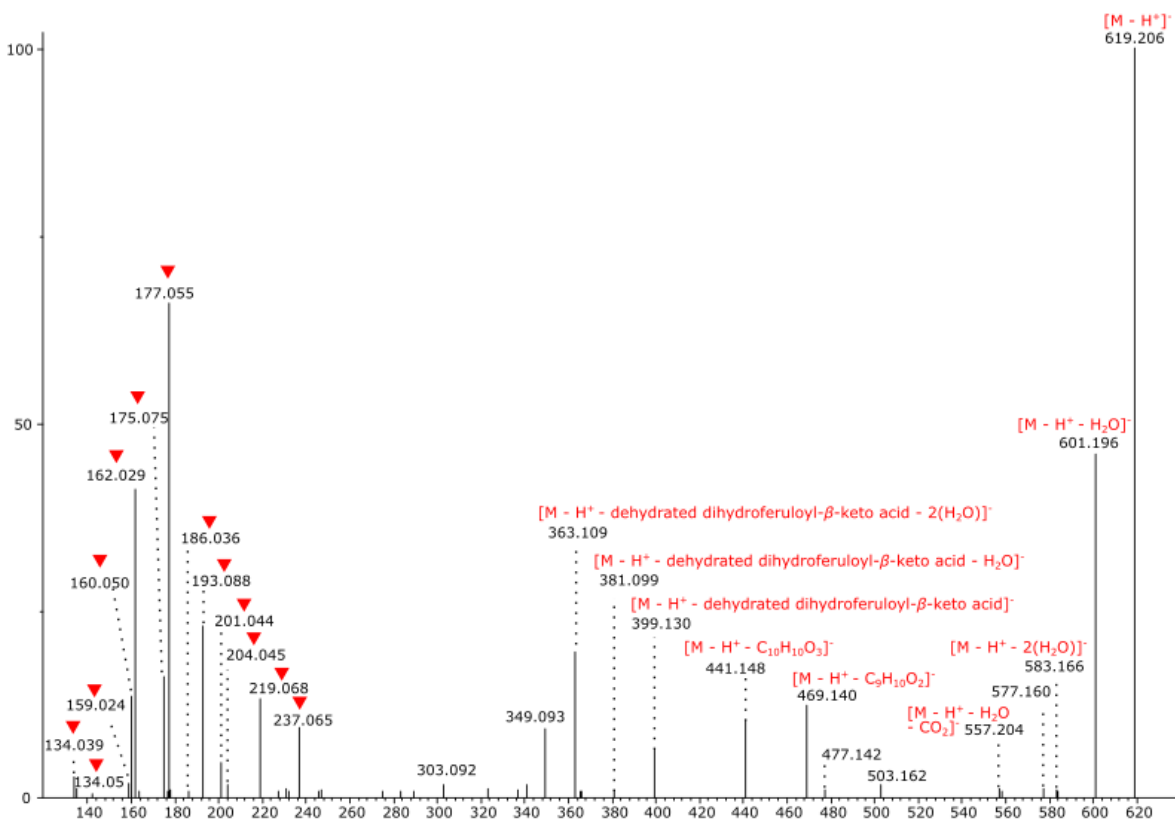
* contamination with fragments from co-eluting kaempferol dirhamnoside



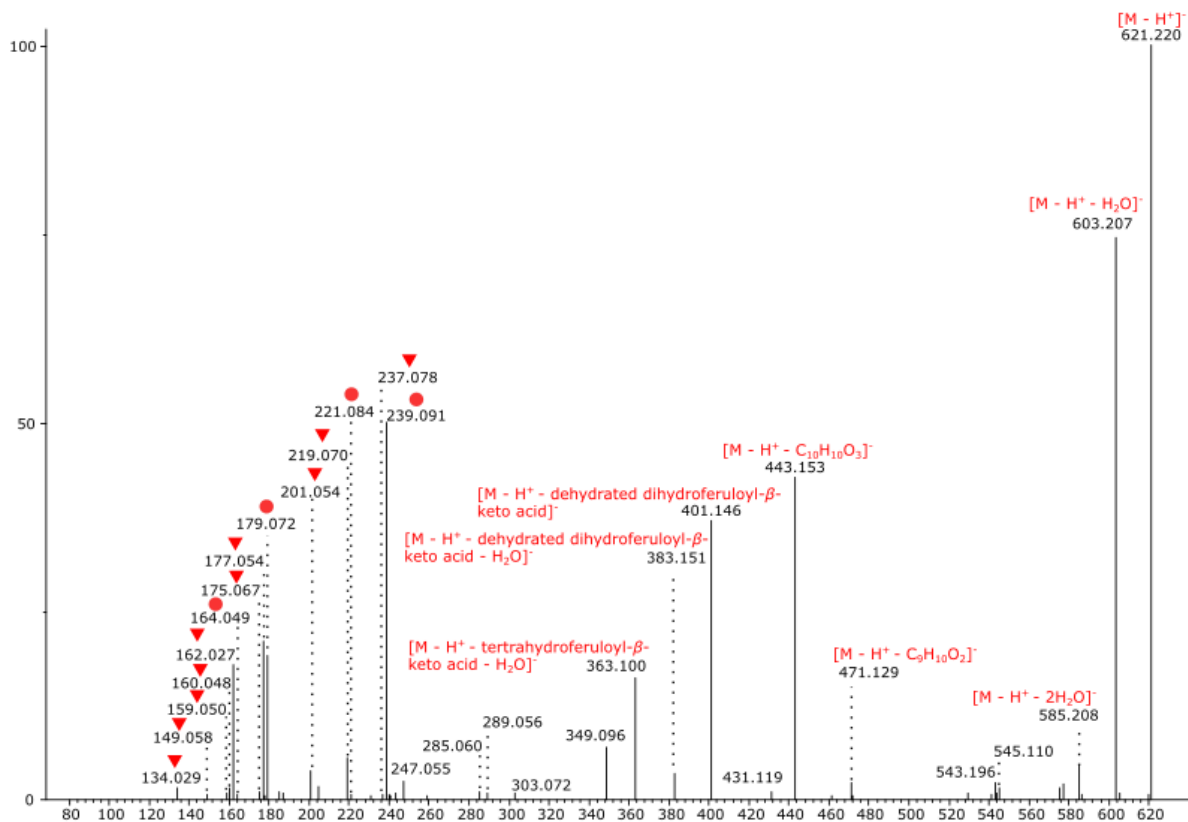
15. MS/MS dihydroferuloyl- β -keto acid + dihydroferulic acid + vanillic acid + hexose + hexose (11.74 min, m/z 889.2780)



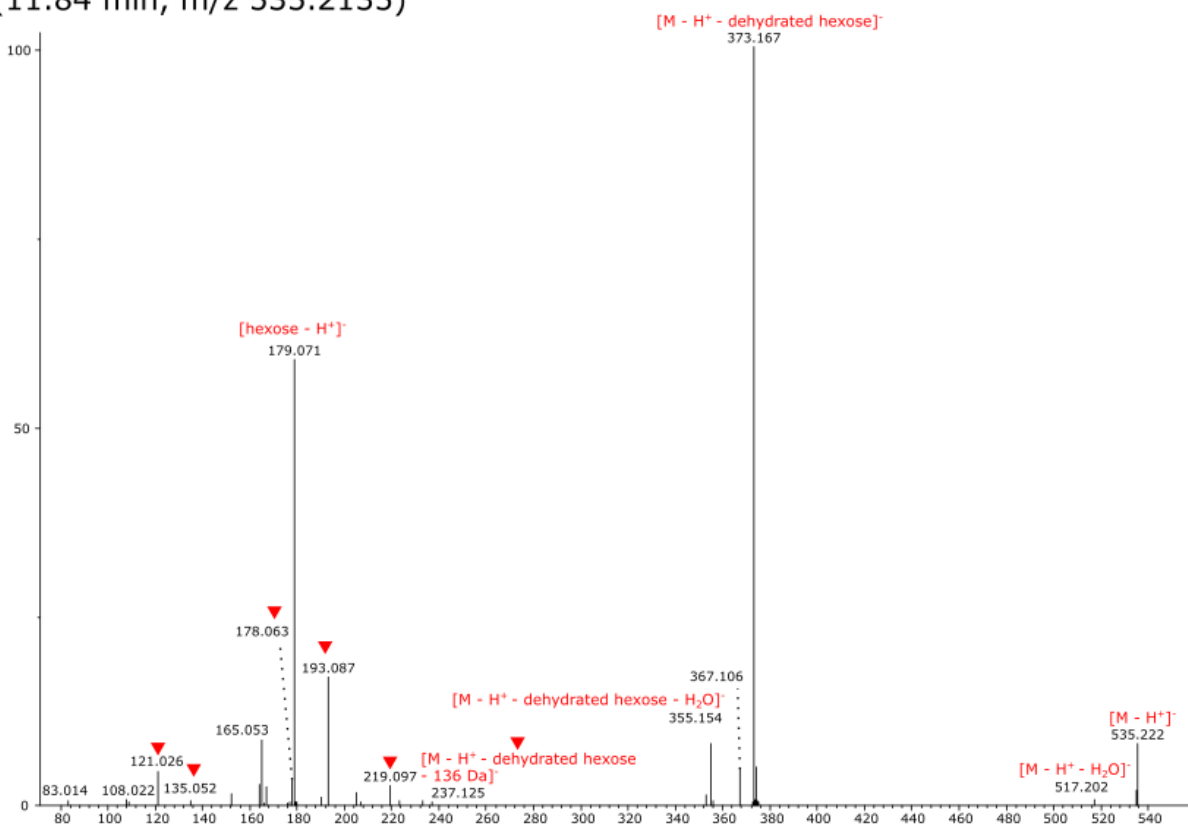
16. MS/MS dihydroferuloyl- β -keto acid + dihydroferuloyl- β -keto acid + hexose (9.96 min, m/z 619.2030)



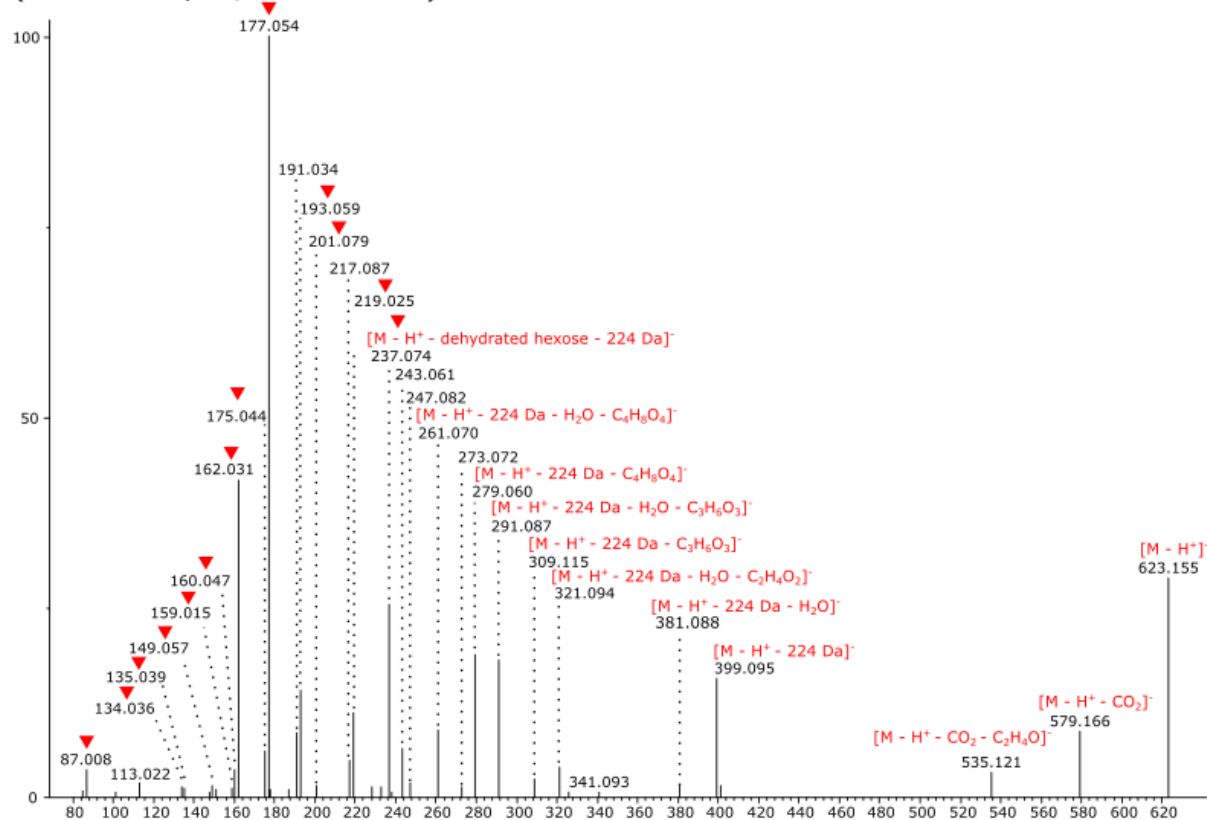
17. MS/MS dihydroferuloyl- β -keto acid + tetrahydroferuloyl- β -keto acid + hexose (10.76 min, m/z 621.2186)



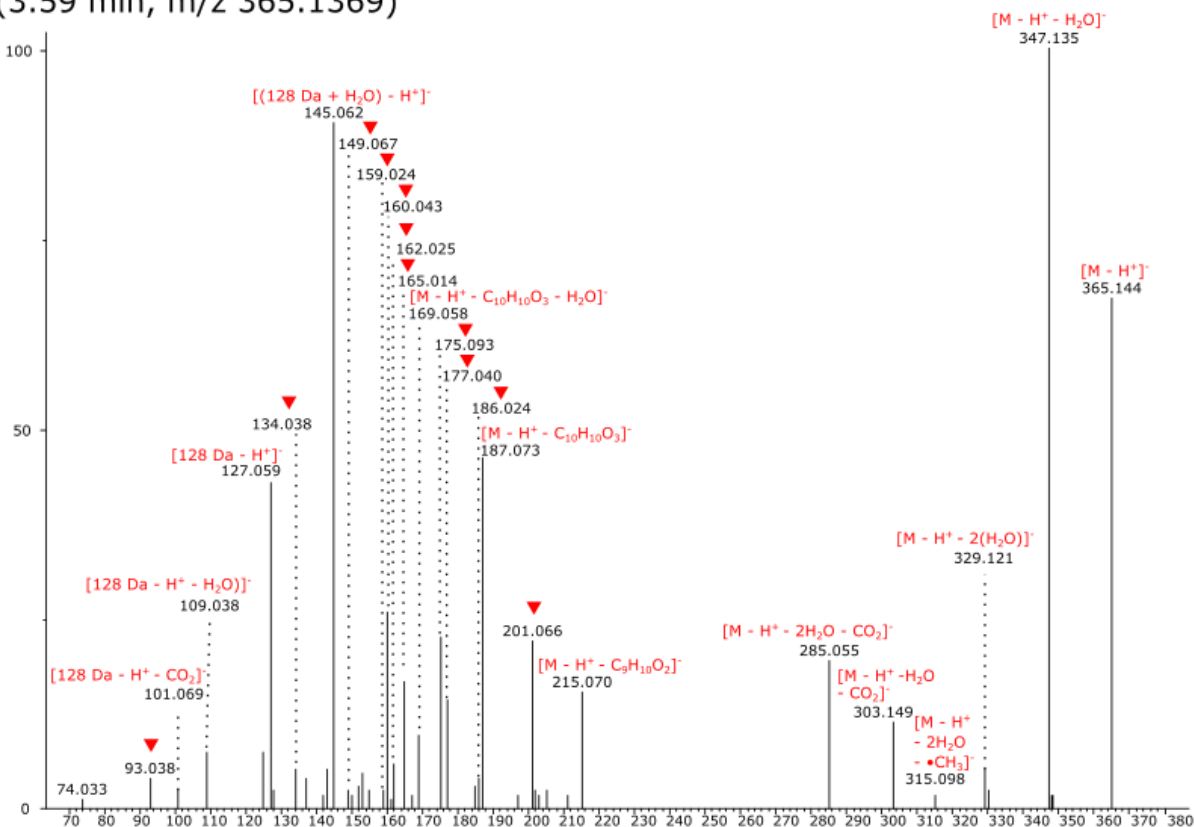
18. MS/MS dihydroferuloyl- β -keto acid + hexose + 136 Da (11.84 min, m/z 535.2135)



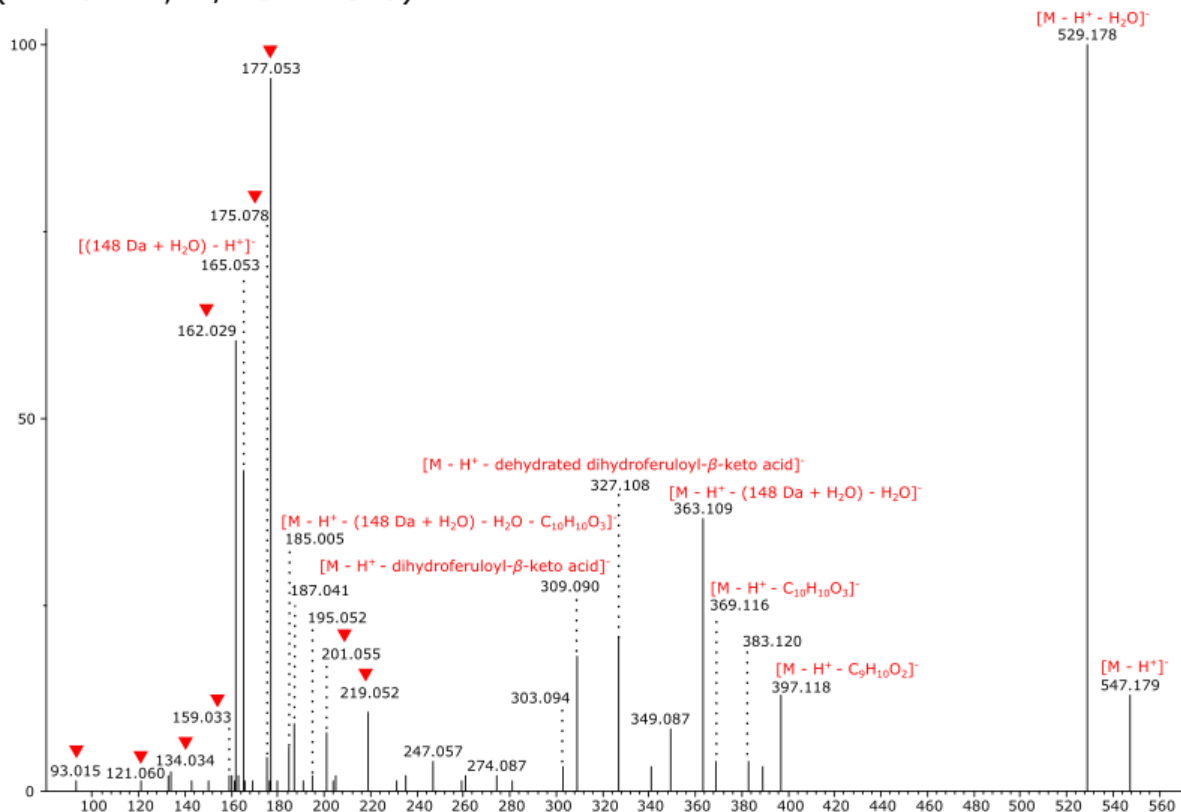
19. MS/MS dihydroferuloyl- β -keto acid + hexose + 224 Da
(11.74 min, m/z 623.1903)



20. MS/MS dihydroferuloyl- β -keto acid + 128 Da
(3.59 min, m/z 365.1369)



21. MS/MS dihydroferuloyl- β -keto acid + hexose + 148 Da
(11.76 min, m/z 547.1825)



22. MS/MS tetrahydroferuloyl- β -keto acid (7.00 min, m/z 239.0930)

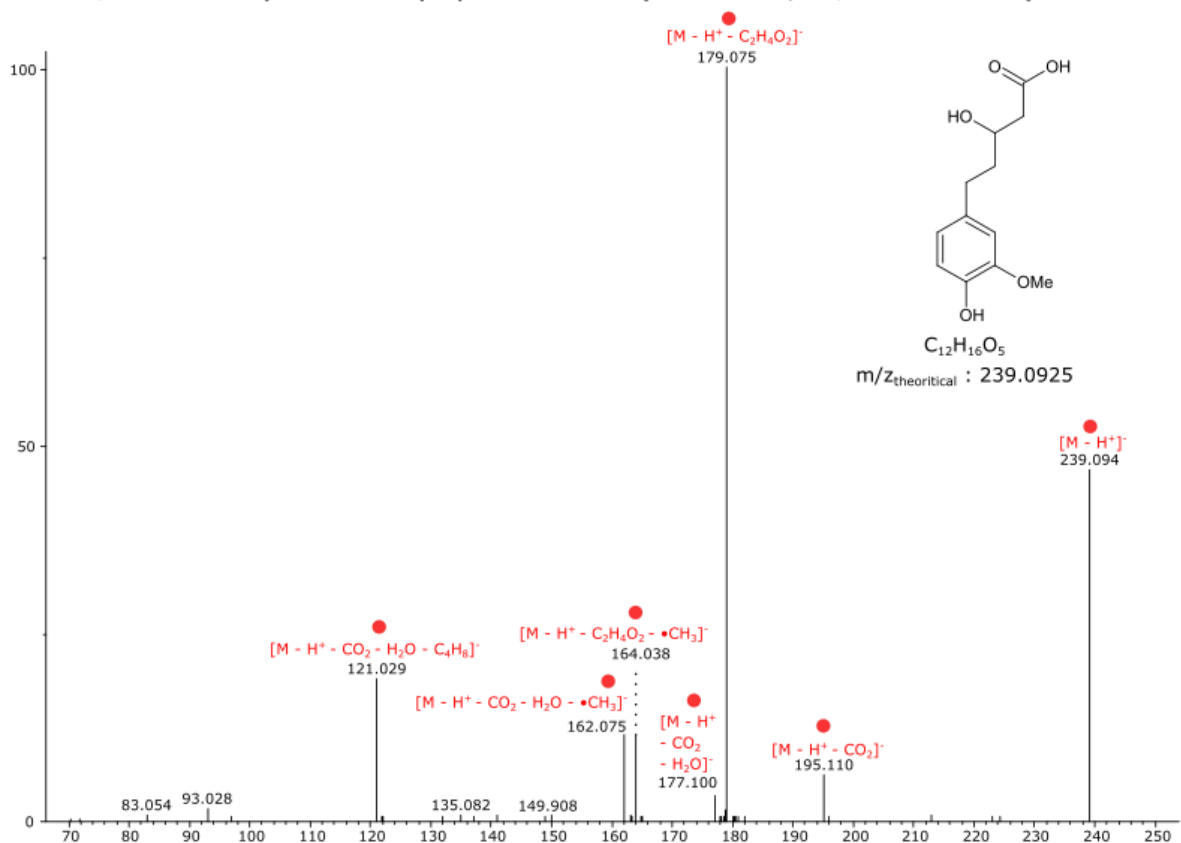


Figure 2 illustrates the fragmentation pathways of the $[M-H]^+$ ion of 1-methoxy-3-(3-hydroxypropyl)benzene. The initial ion is at m/z 239.0925. Two main fragmentation routes are shown:

- Top Pathway:**
 - Initial ion (m/z 239.0925) loses CO_2 (43.9898 Da) to form $[M-H^+-CO_2]^+$ at m/z 195.1027.
 - Loss of a methoxy group (a) leads to a resonance-stabilized cation.
 - Loss of H_2O (18.0106 Da) forms $[M-H^+-CO_2-H_2O]^+$ at m/z 177.0921.
 - Loss of a methyl group (15.0235 Da) forms $[M-H^+-CO_2-H_2O-CH_3]^+$ at m/z 162.0686.
- Bottom Pathway:**
 - Initial ion (m/z 239.0925) loses a methoxy group (b) to form a resonance-stabilized cation.
 - Loss of H_2O (18.0106 Da) forms $[M-H^+-H_2O]^+$ at m/z 221.0891.
 - Loss of a methyl group (15.0235 Da) forms $[M-H^+-CO_2-CH_3]^+$ at m/z 180.0792.
 - Loss of C_4H_8 (56.0626 Da) forms $[M-H^+-CO_2-H_2O-C_4H_8]^+$ at m/z 121.0295.

The diagram includes chemical structures for each ion and transition state, with arrows indicating the fragmentation steps and the loss of neutral molecules.

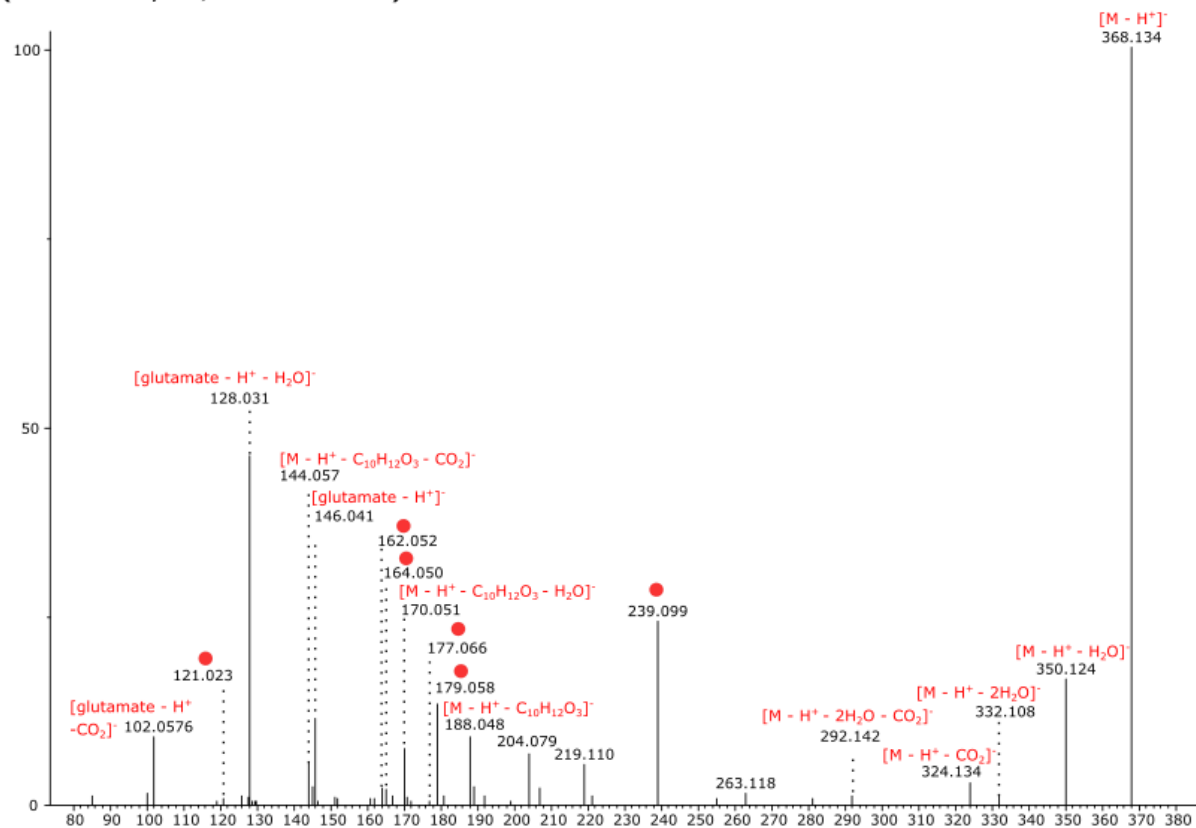
Chemical reaction scheme showing the fragmentation pathways of the $[M-H]^+$ ion of 3-methoxy-4-hydroxybenzaldehyde.

The starting ion is $[M-H]^+$ at m/z 239.0925.

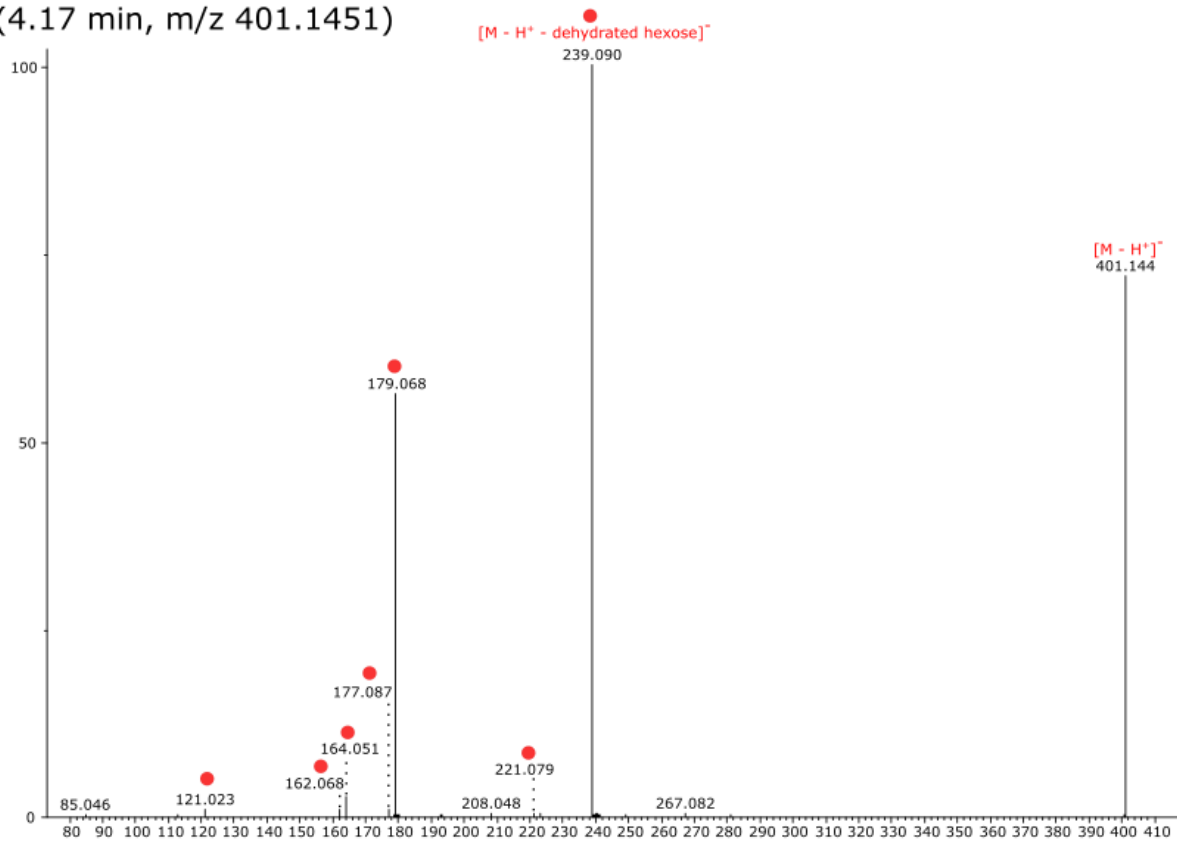
Fragmentation pathways shown:

- Pathway 1: Loss of a water molecule (a) to form a resonance-stabilized cation at m/z 179.0714 ($[M-H^+-C_2H_4O_2]^+$). Subsequent loss of a methyl radical (b) leads to a phenoxide ion at m/z 164.0479 ($[M-H^+-C_2H_4O_2-\bullet CH_3]^+$).
- Pathway 2: Loss of a methoxy group (b) to form a phenoxide ion at m/z 180.0786 Da.
- Pathway 3: Loss of a methoxy group (b) to form a phenoxide ion at m/z 59.0139 ($[M-H^+-C_{10}H_{12}O_3]^+$).

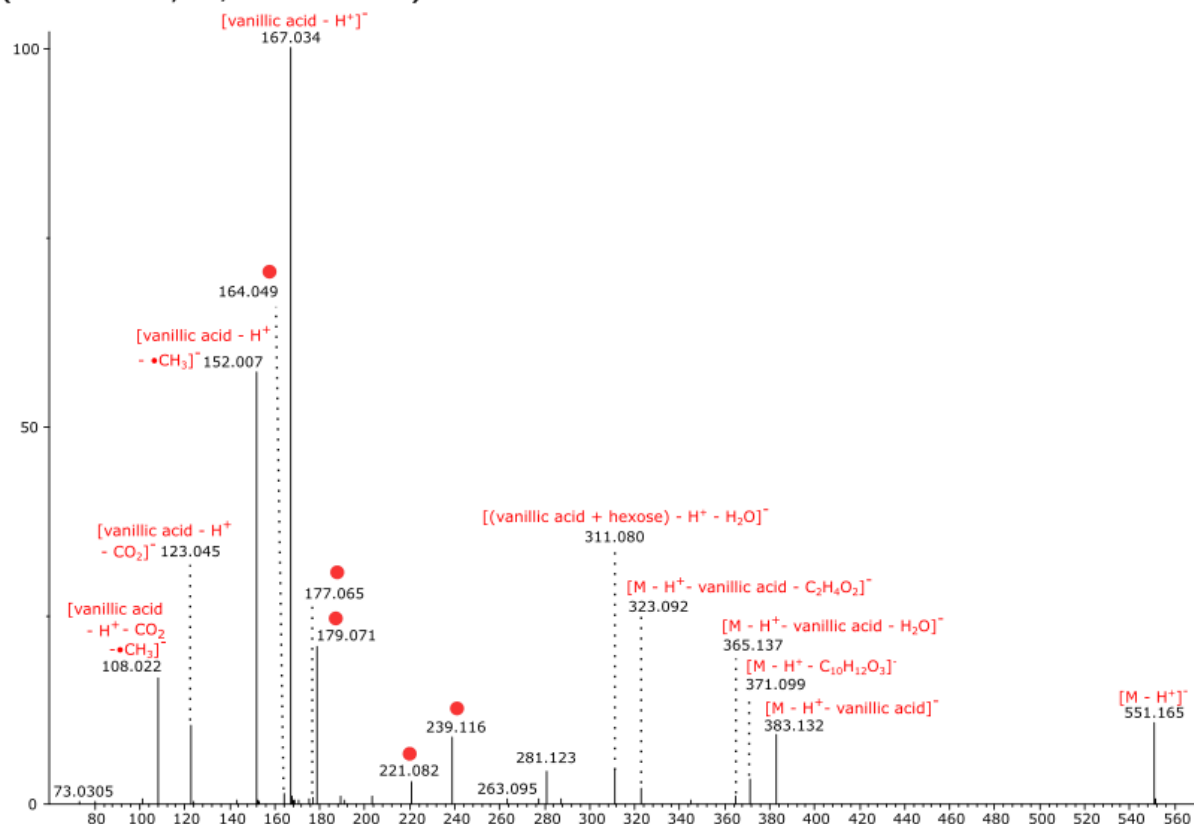
23. MS/MS tetrahydroferuloyl- β -keto acid + glutamate
(6.07 min, m/z 368.1367)



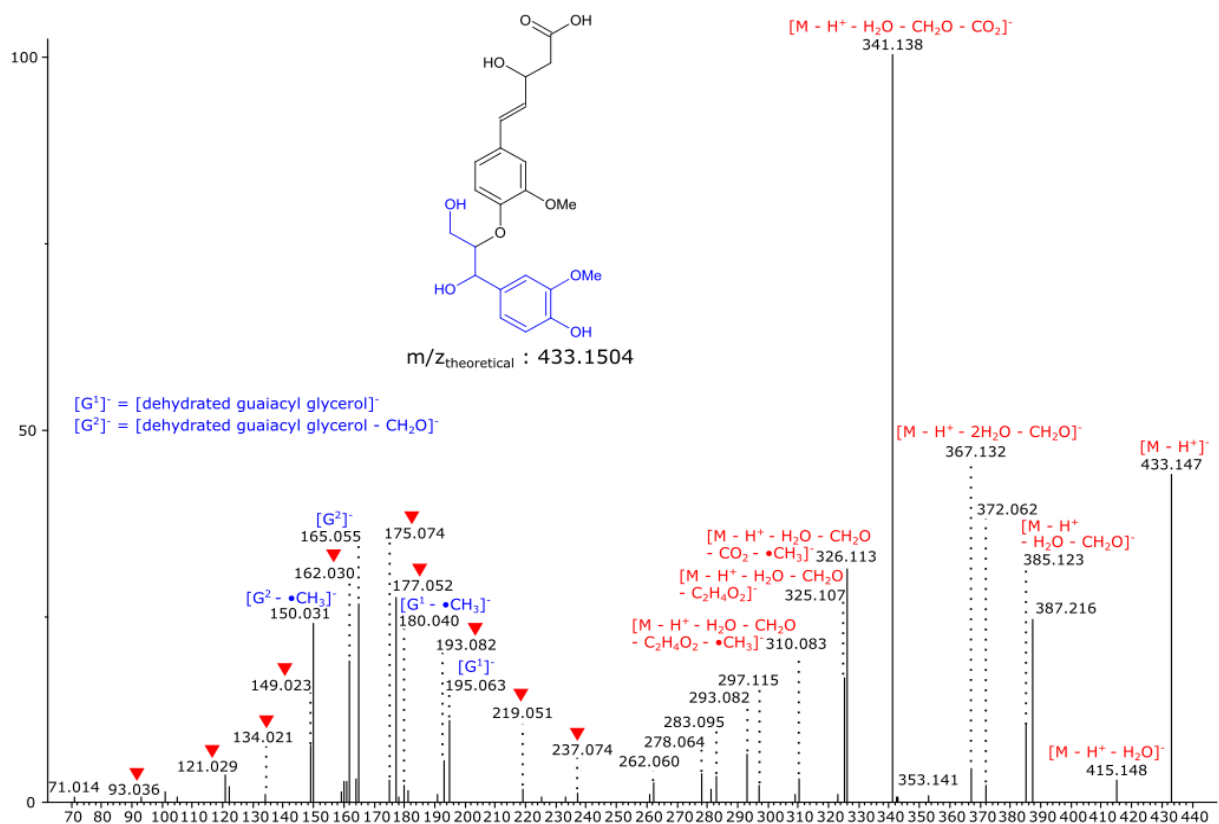
24. MS/MS tetrahydroferuloyl- β -keto acid + hexose
(4.17 min, m/z 401.1451)



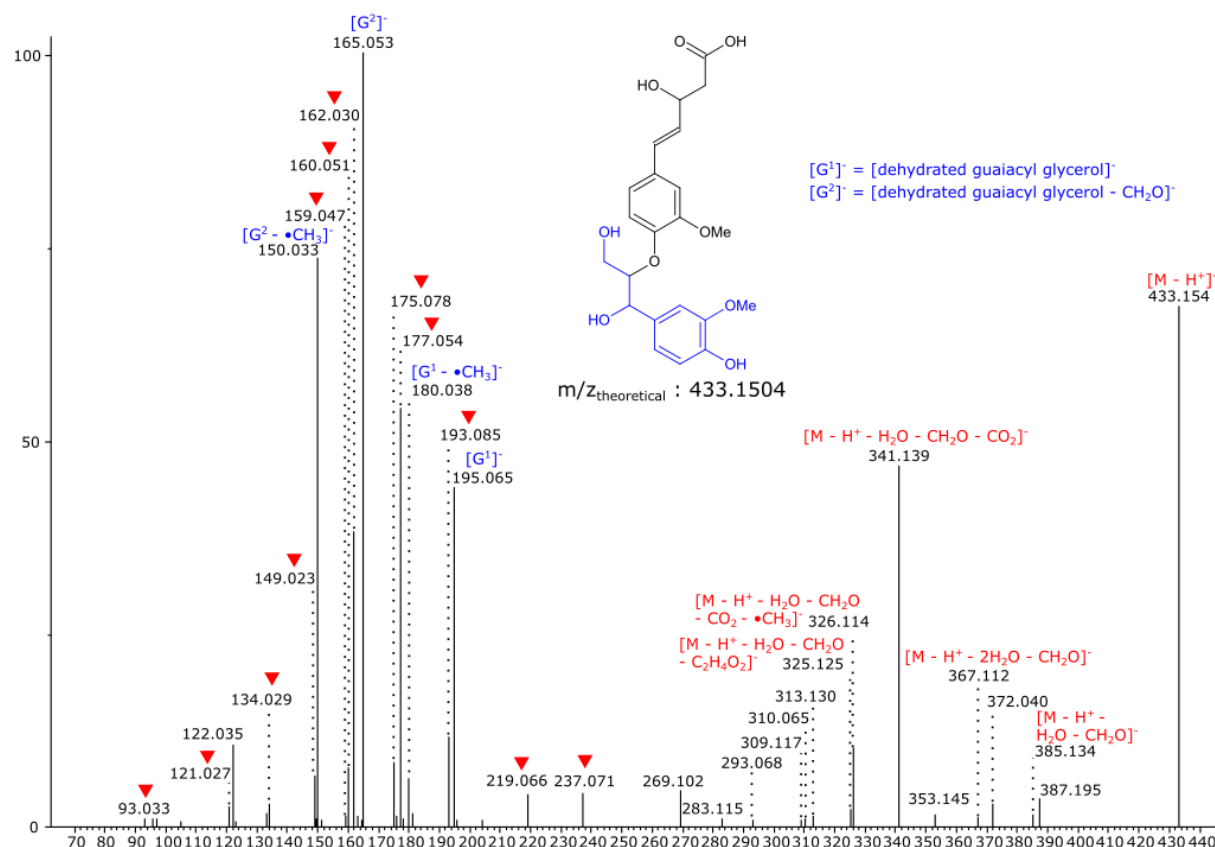
25. MS/MS tetrahydroferuloyl- β -keto acid + vanillic acid + hexose
(10.22 min, m/z 551.1775)



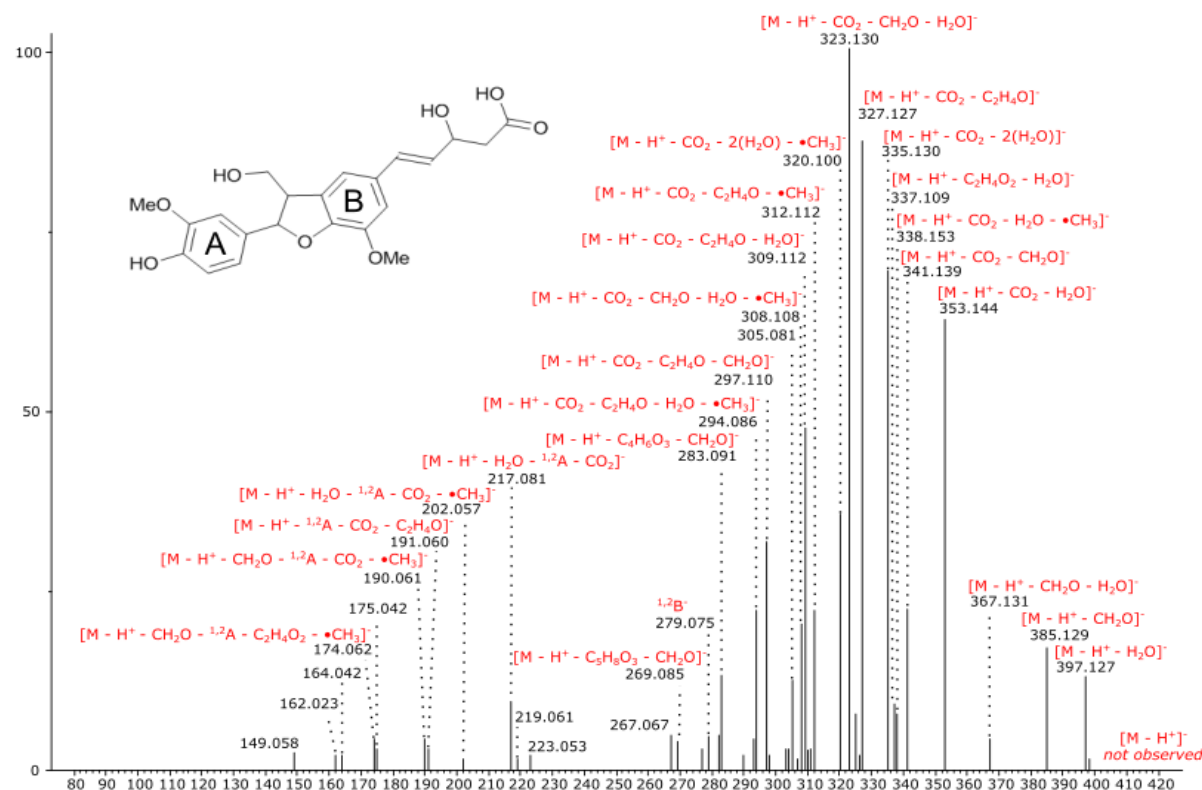
26. MS/MS dihydroferuloyl- β -keto acid(4-O-8)G 1 (7.80 min, m/z 433.1521)



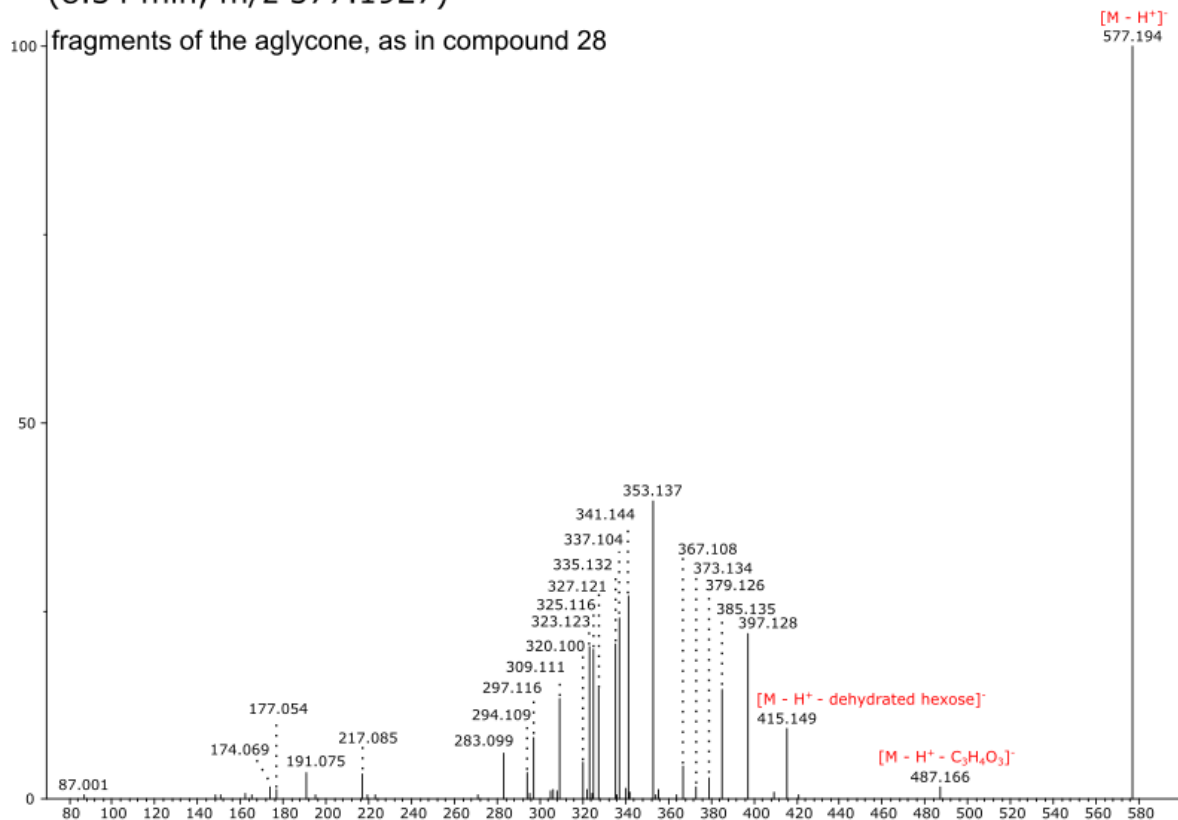
27. MS/MS dihydroferuloyl- β -keto acid(4-O-8)G 2 (8.13 min, m/z 433.1544)



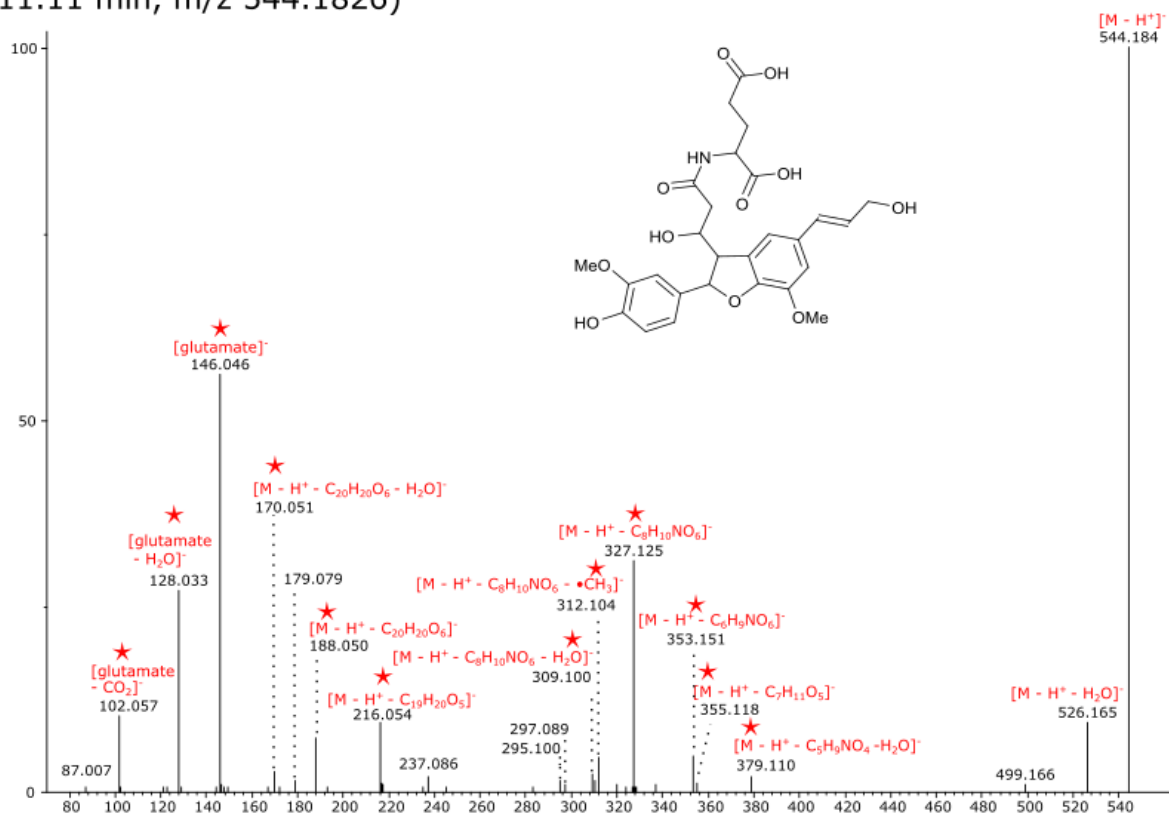
28. MS/MS dihydroferuloyl- β -keto acid(5-8)G (12.0 min, m/z 415.1418)

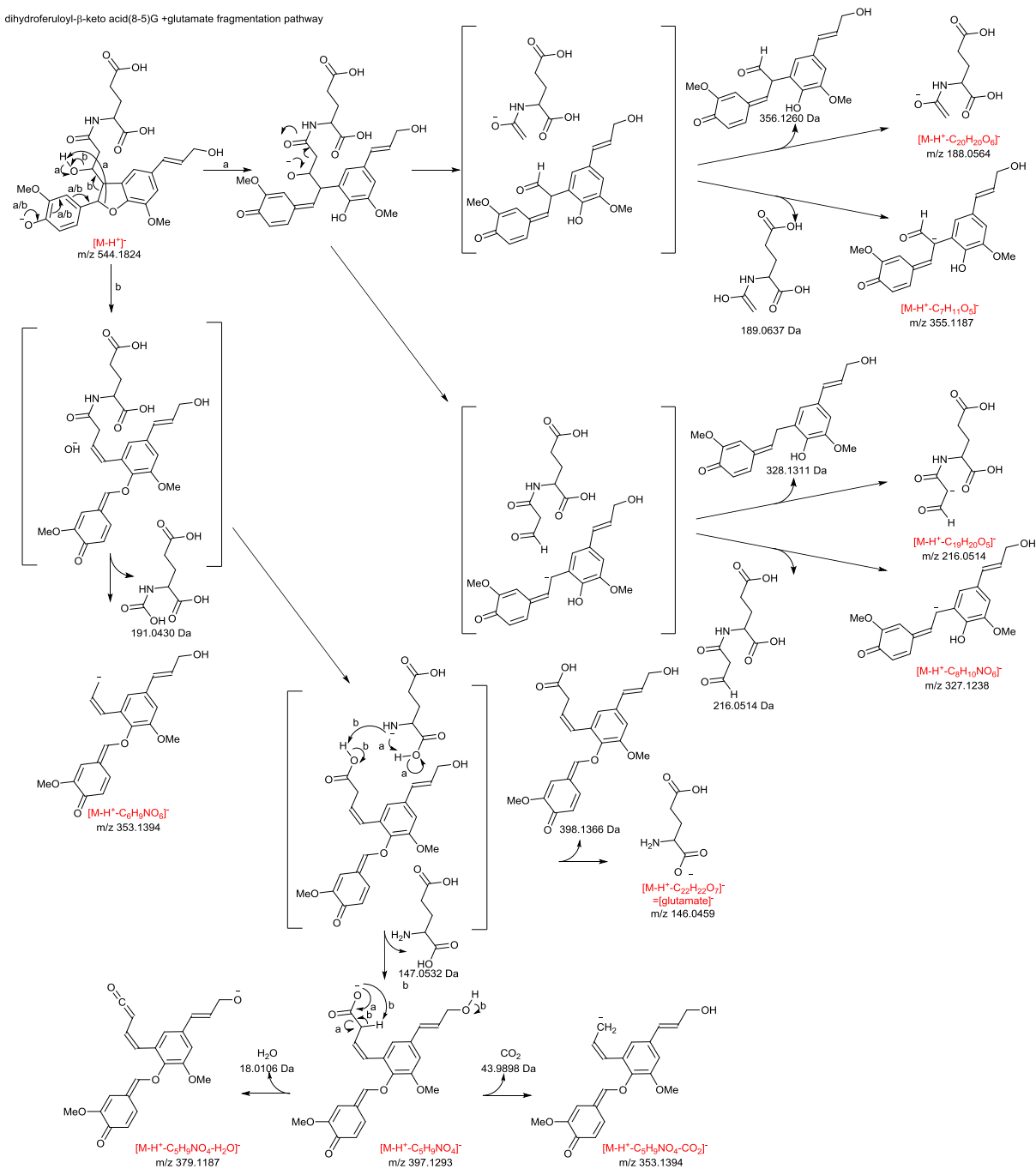


29. MS/MS dihydroferuloyl- β -keto acid(5-8)G + hexose
(8.34 min, m/z 577.1927)

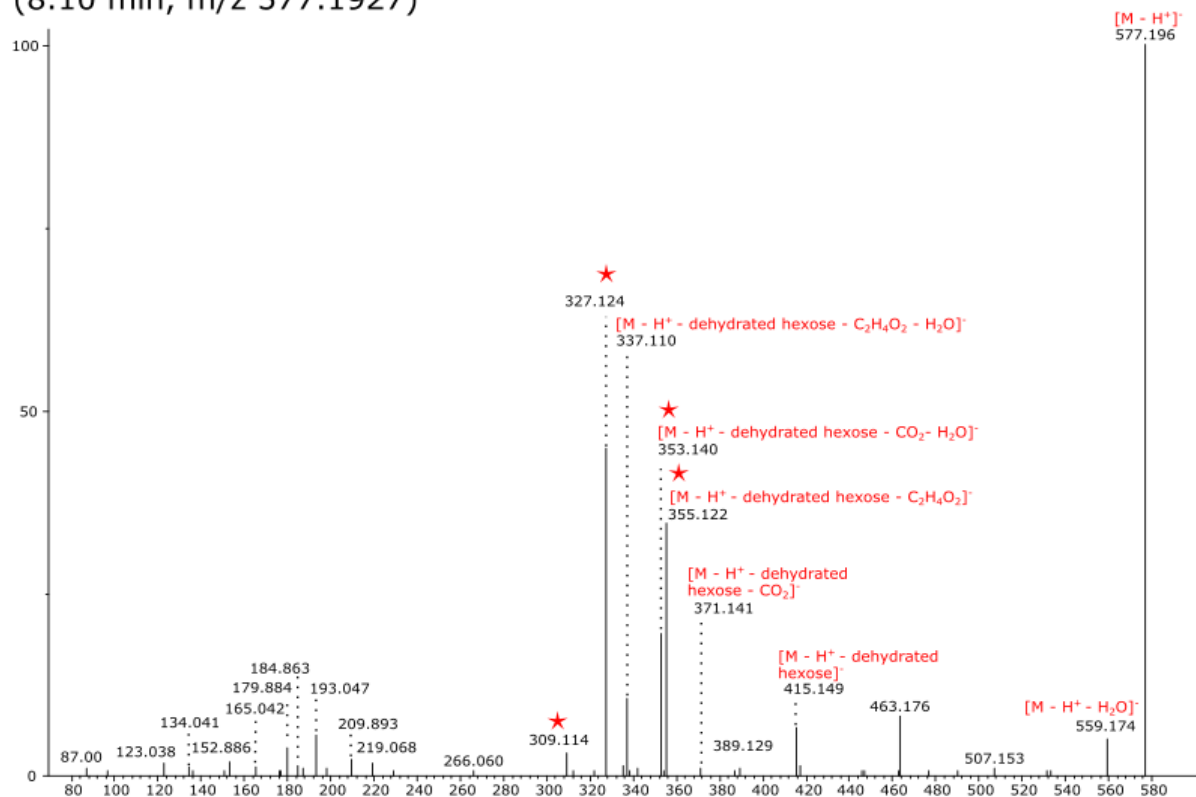


30. MS/MS dihydroferuloyl- β -keto acid(8-5)G + glutamate
(11.11 min, m/z 544.1826)

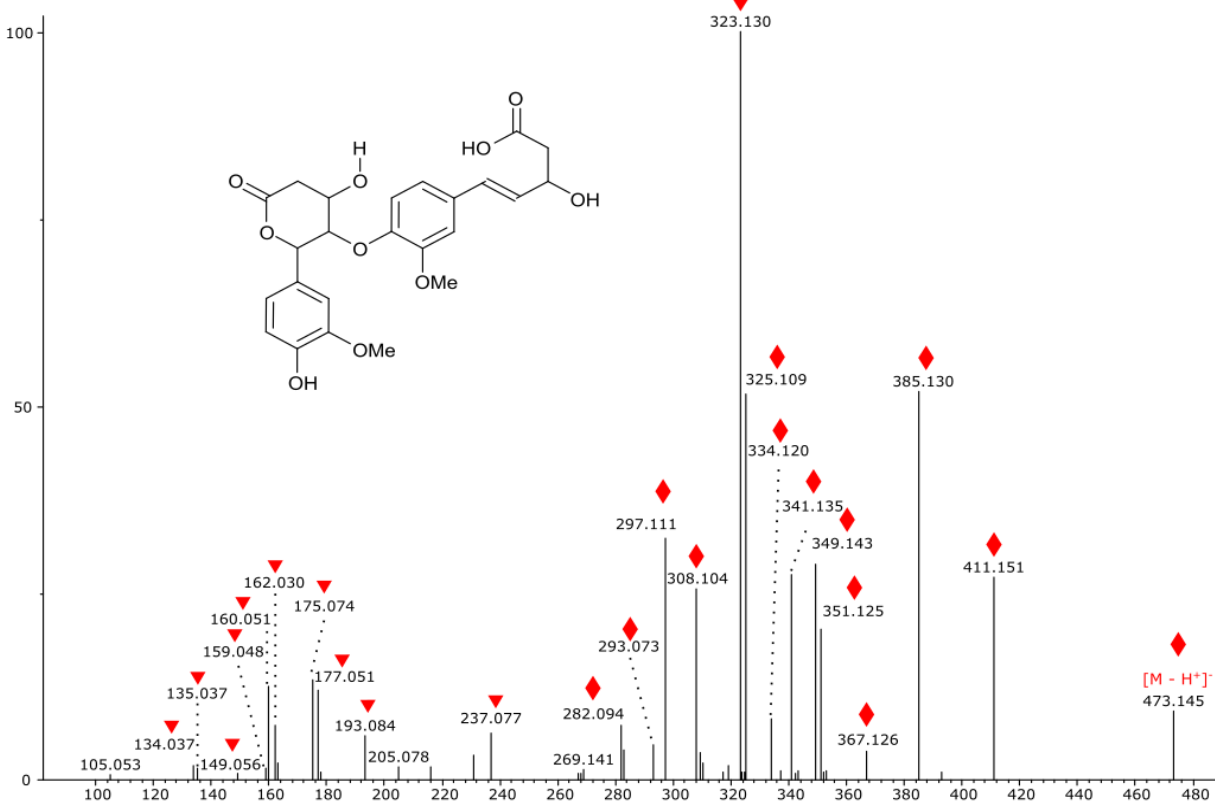


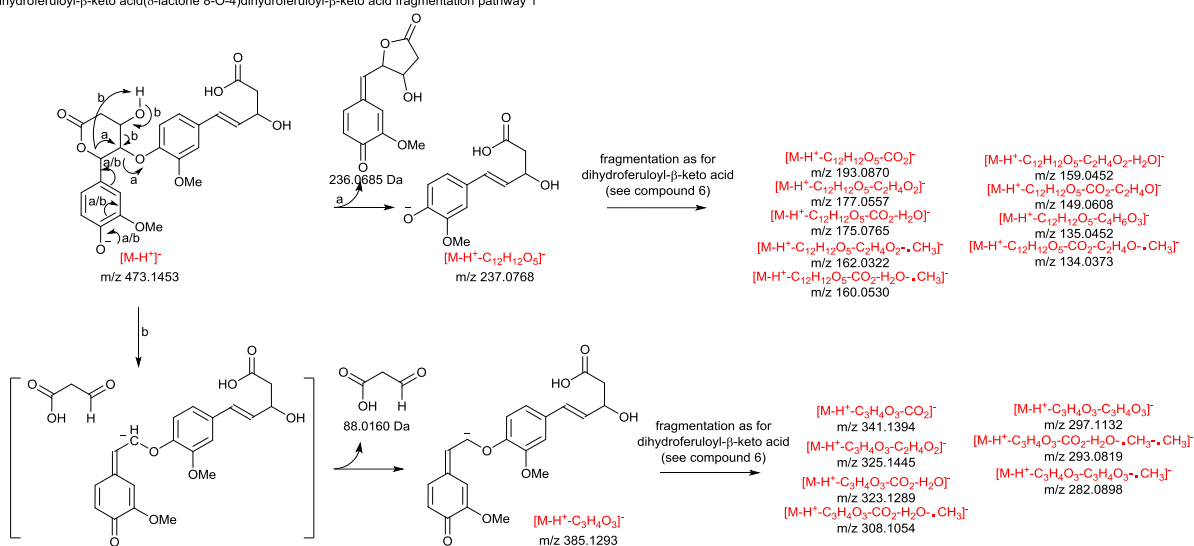


31. MS/MS dihydroferuloyl- β -keto acid(8-5)G + hexose
(8.10 min, m/z 577.1927)

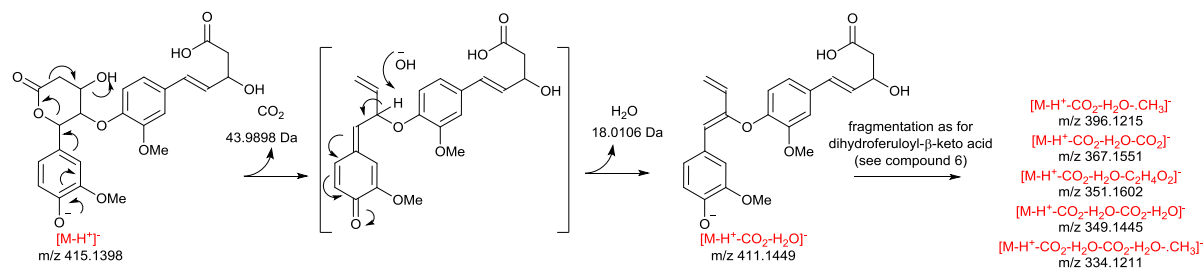


32. MS/MS dihydroferuloyl- β -keto acid(δ -lactone 8-O-4)
dihydroferuloyl- β -keto acid (11.09 min, m/z 473.1475)

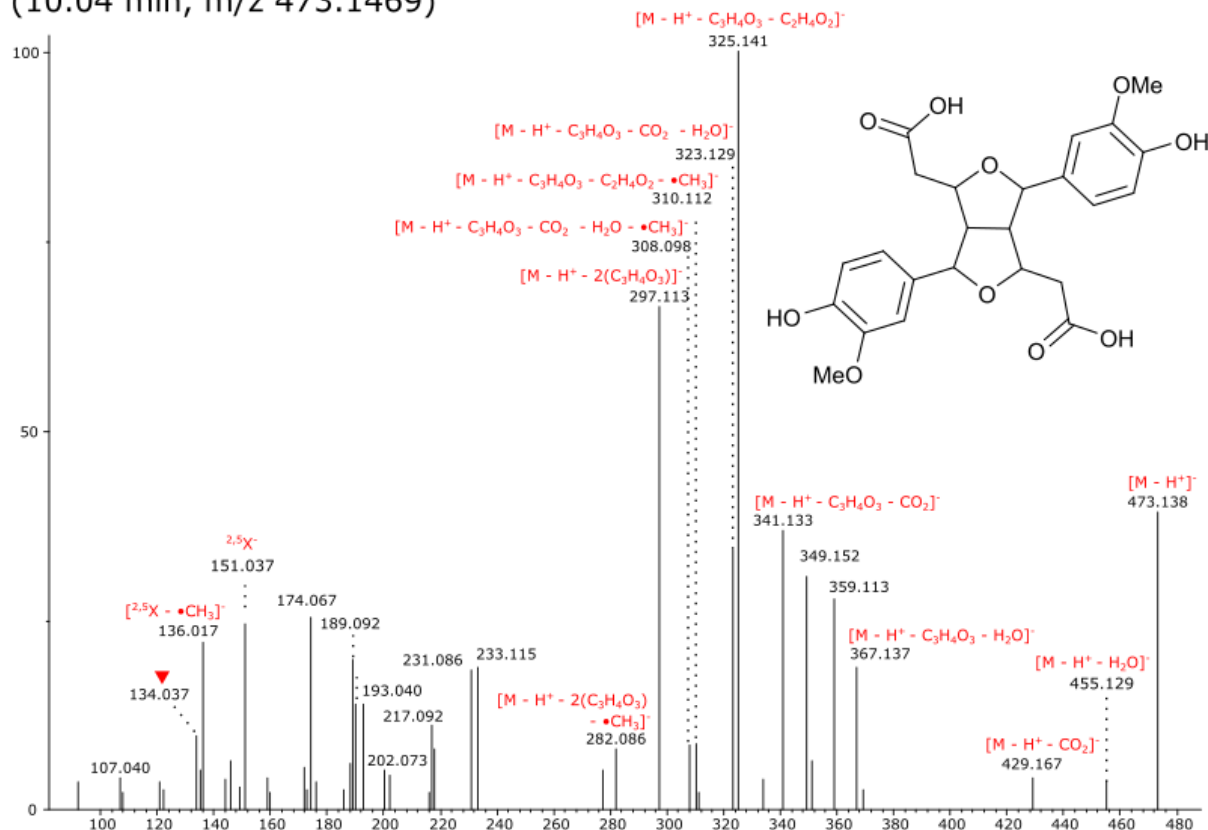


dihydroferuloyl- β -keto acid(δ -lactone 8-O-4)dihydroferuloyl- β -keto acid fragmentation pathway 1

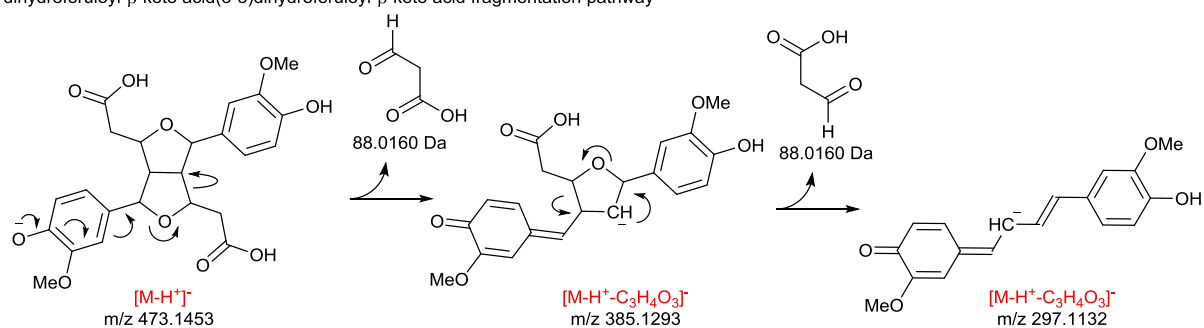
dihydroferuloyl-β-keto acid(δ-lactone 8-O-4)dihydroferuloyl-β-keto acid fragmentation pathway 2



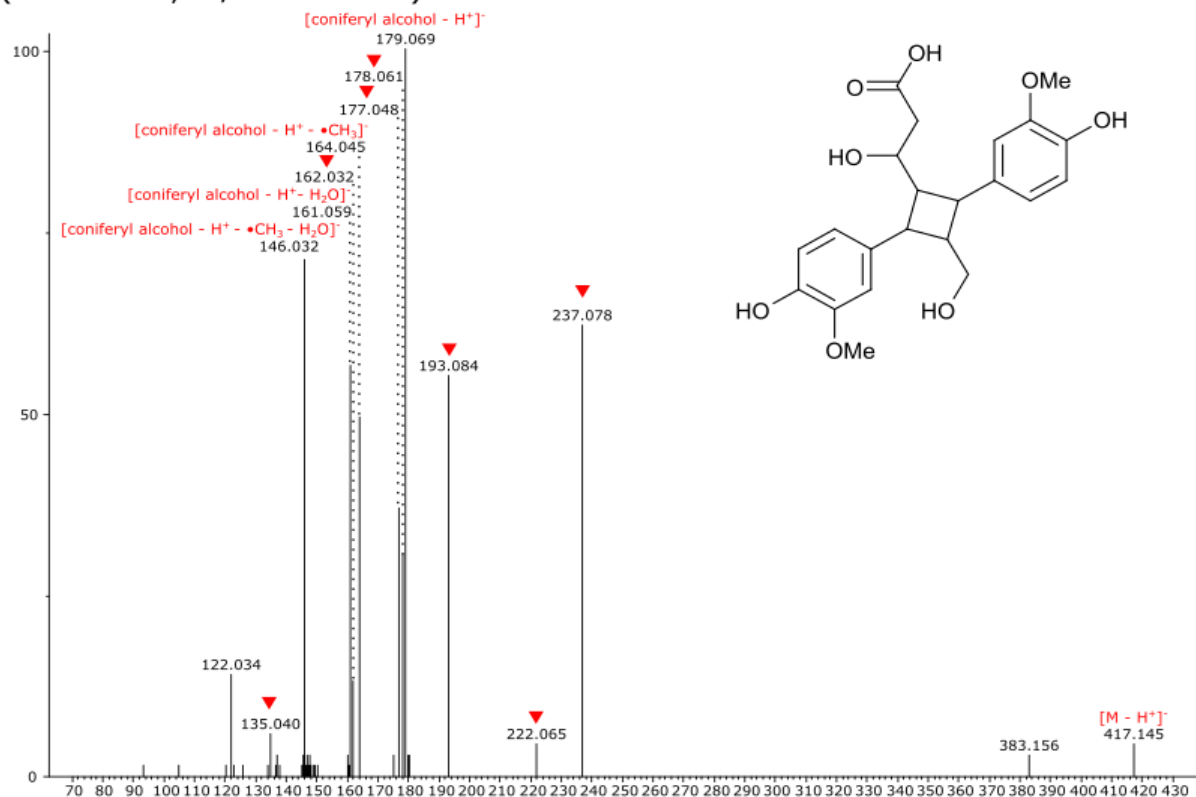
33. MS/MS dihydroferuloyl-β-keto acid(8-8)dihydroferuloyl-β-keto acid
(10.04 min, m/z 473.1469)



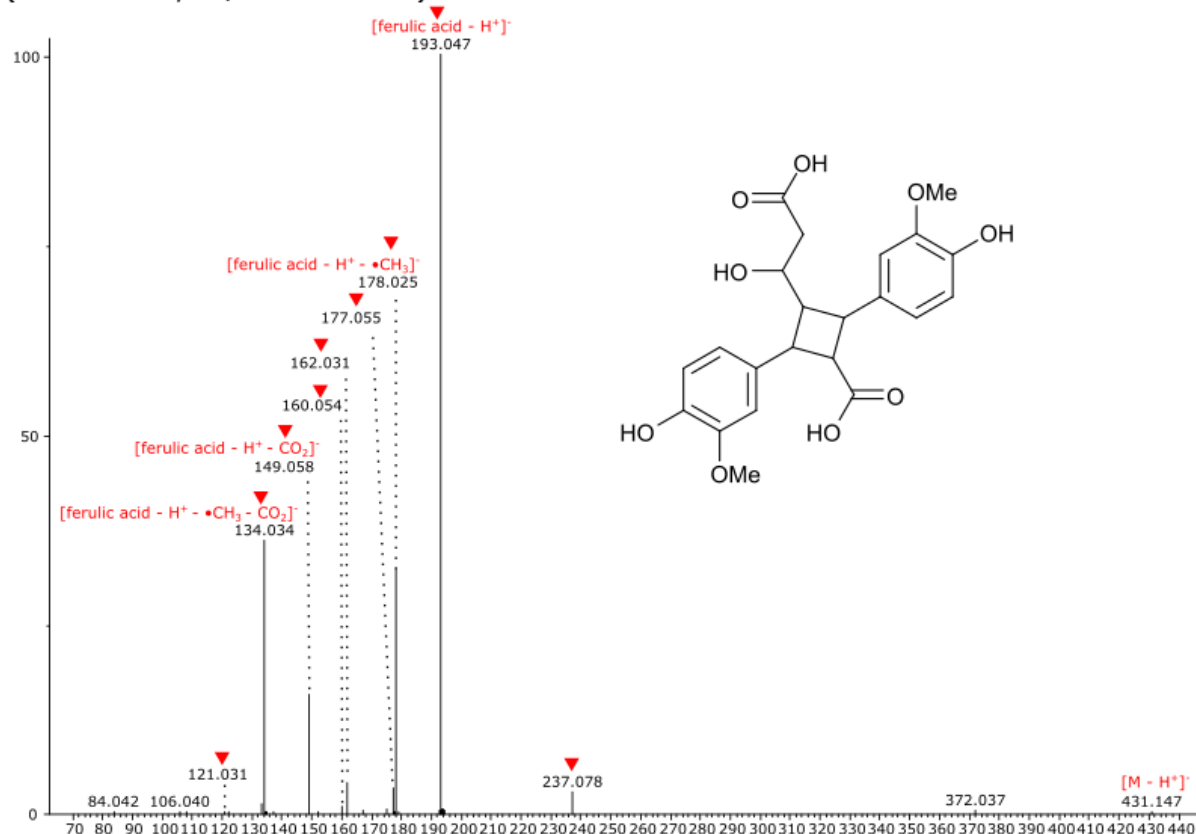
dihydroferuloyl-β-keto acid(8-8)dihydroferuloyl-β-keto acid fragmentation pathway



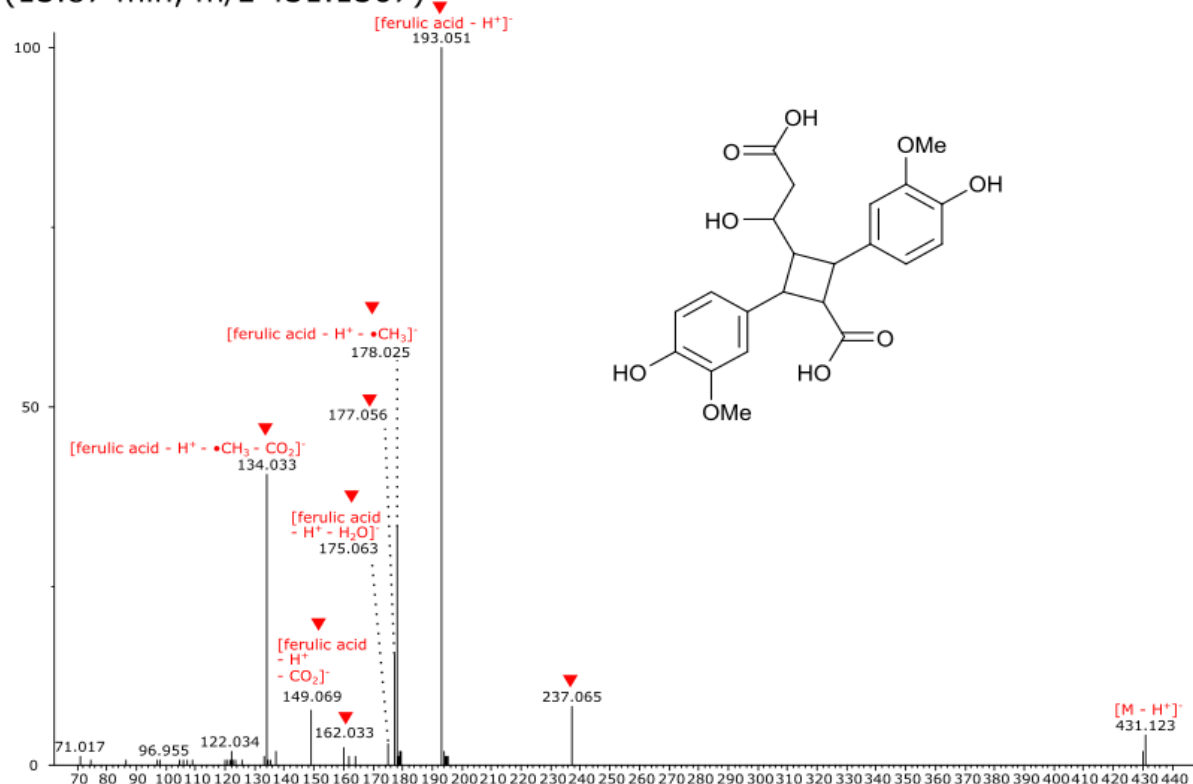
34. MS/MS dihydroferuloyl- β -keto acid + coniferyl alcohol cyclobutane dimer (12.47 min, m/z 417.1570)



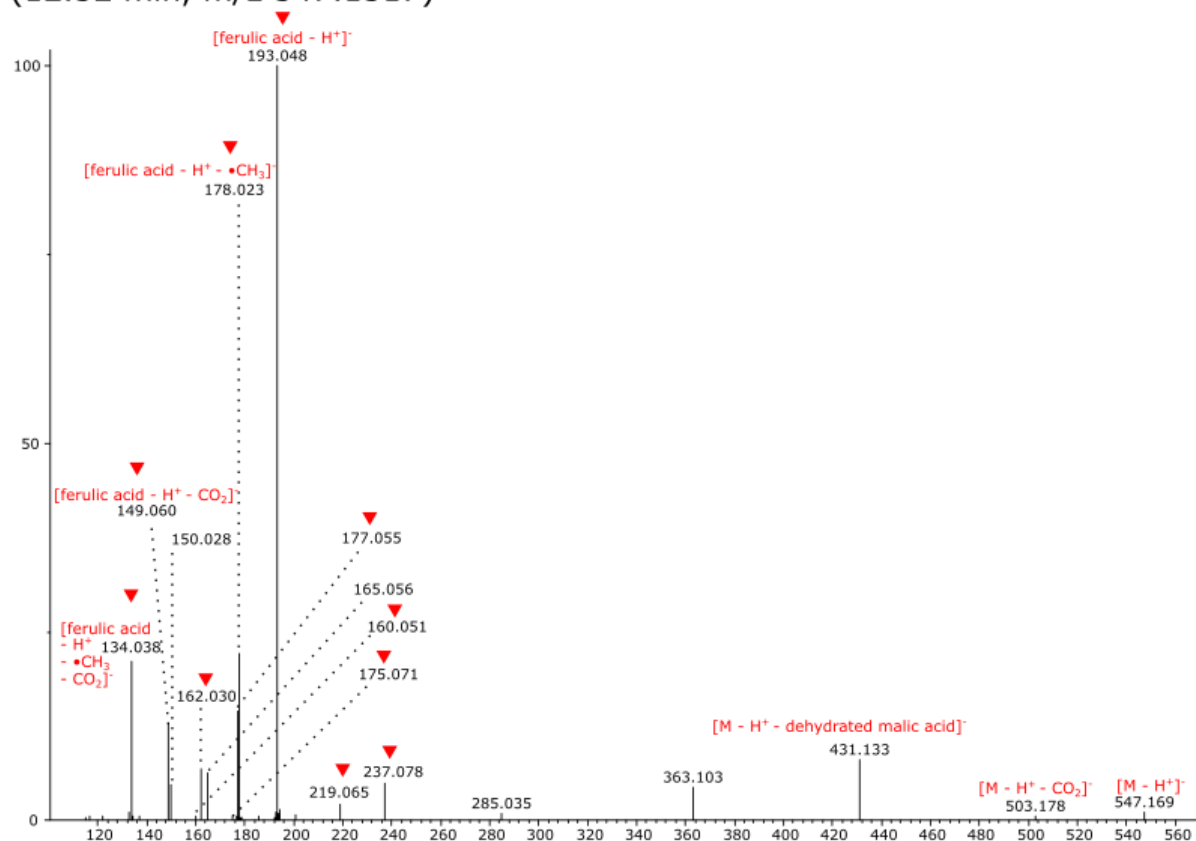
35. MS/MS dihydroferuloyl- β -keto acid ferulic acid cyclobutane dimer 1 (12.55 min, m/z 431.1362)



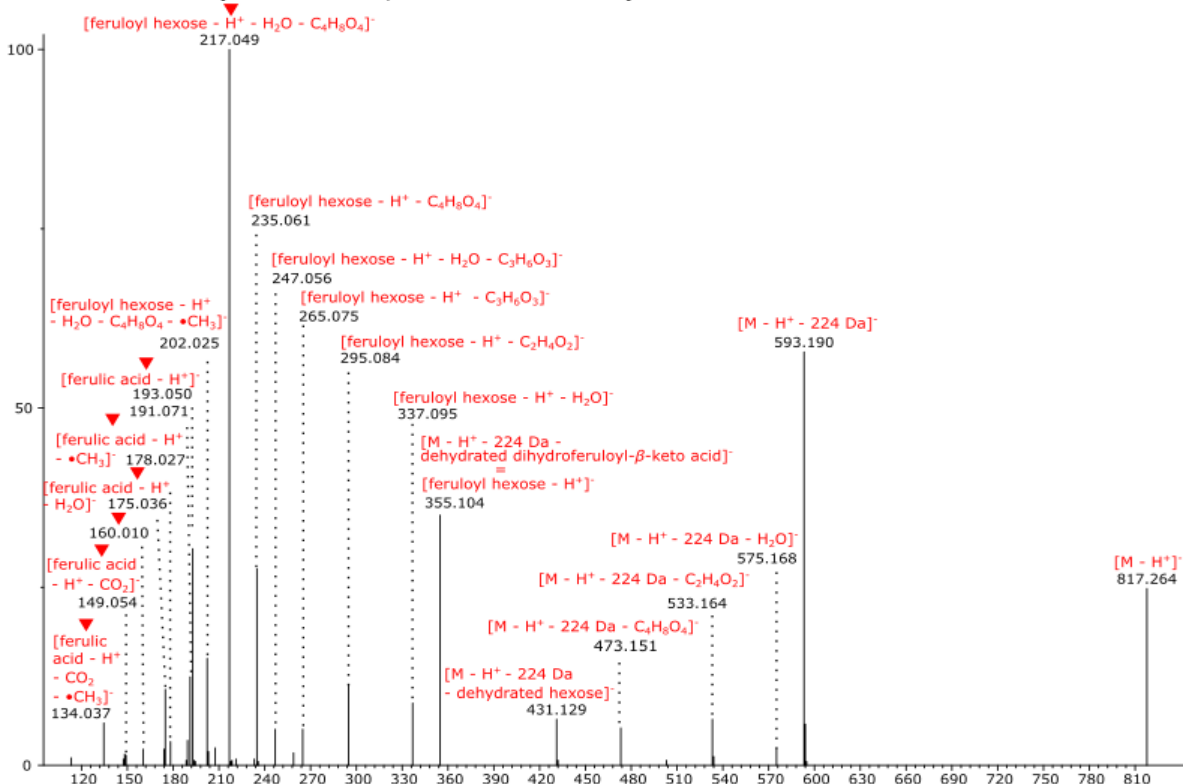
36. MS/MS dihydroferuloyl- β -keto acid ferulic acid cyclobutane dimer 2
(13.87 min, m/z 431.1367)



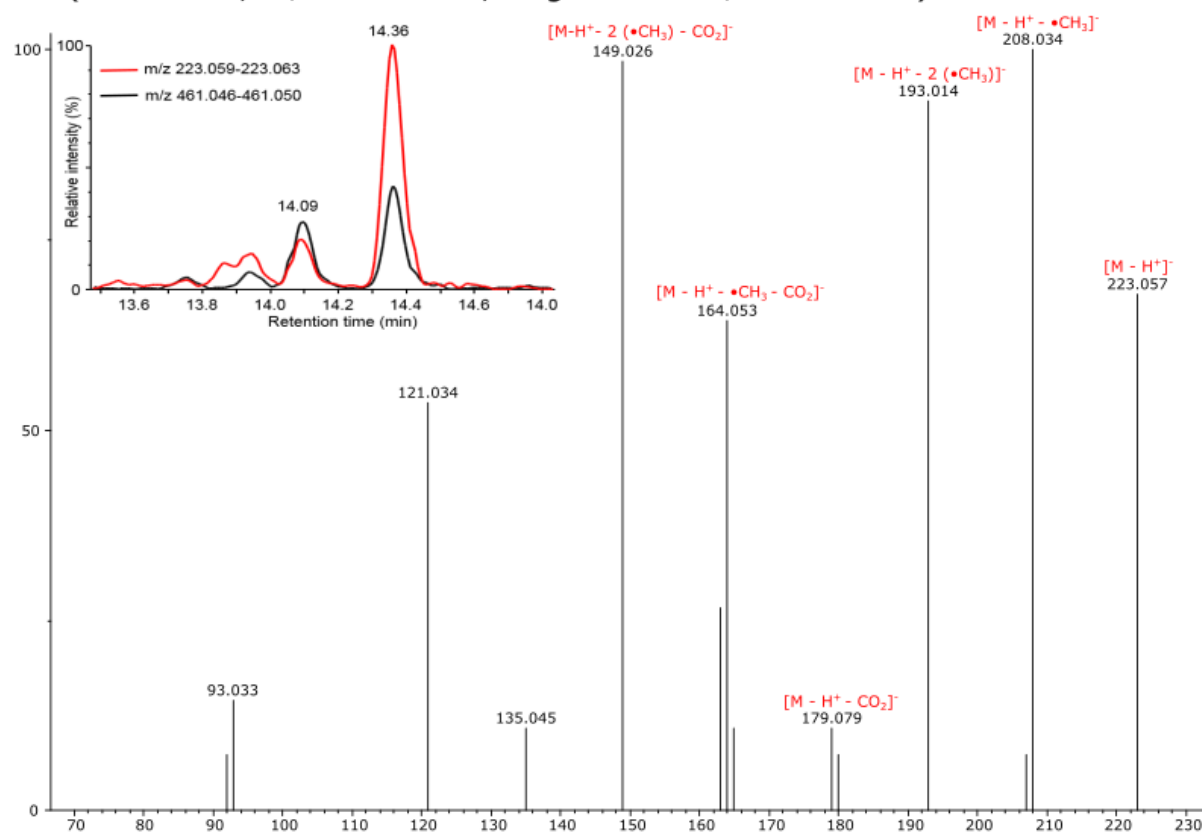
37. dihydroferuloyl- β -keto acid ferulic acid cyclobutane dimer + malate
(12.52 min, m/z 547.1517)



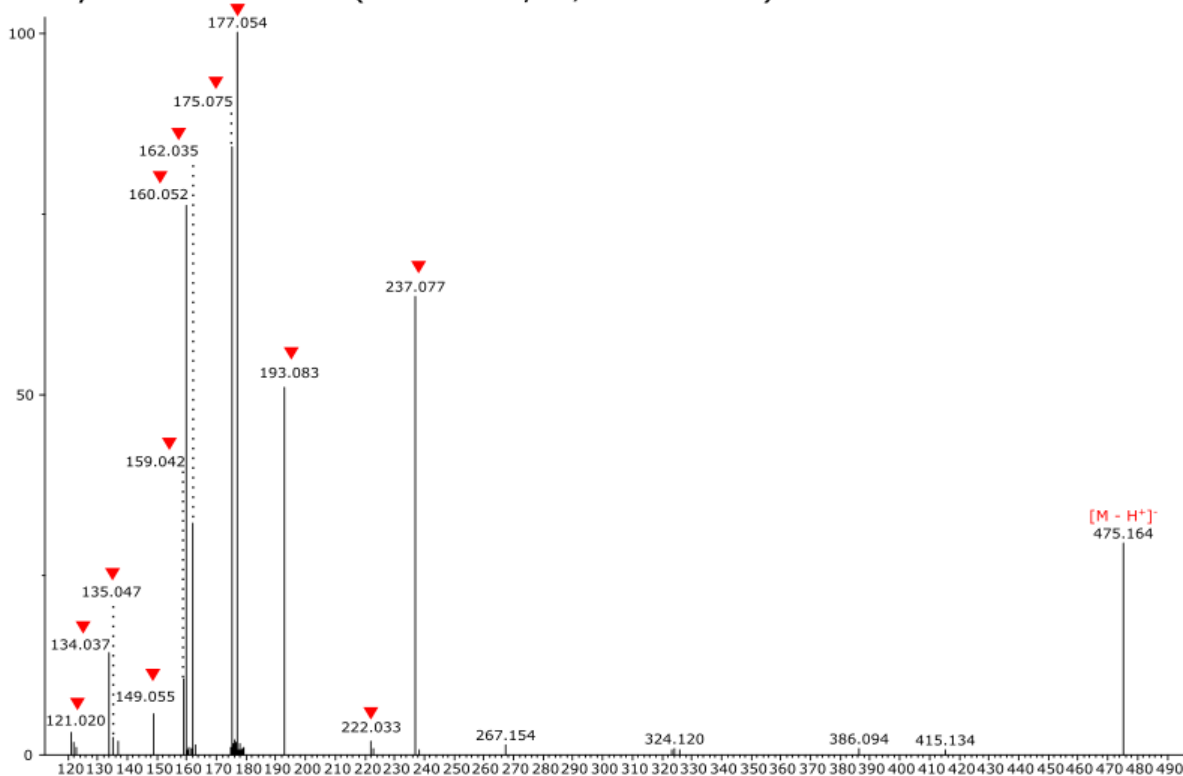
38. MS/MS dihydroferuloyl- β -keto acid feruloyl hexose cyclobutane dimer + 224 Da (14.16 min, m/z 817.2557)



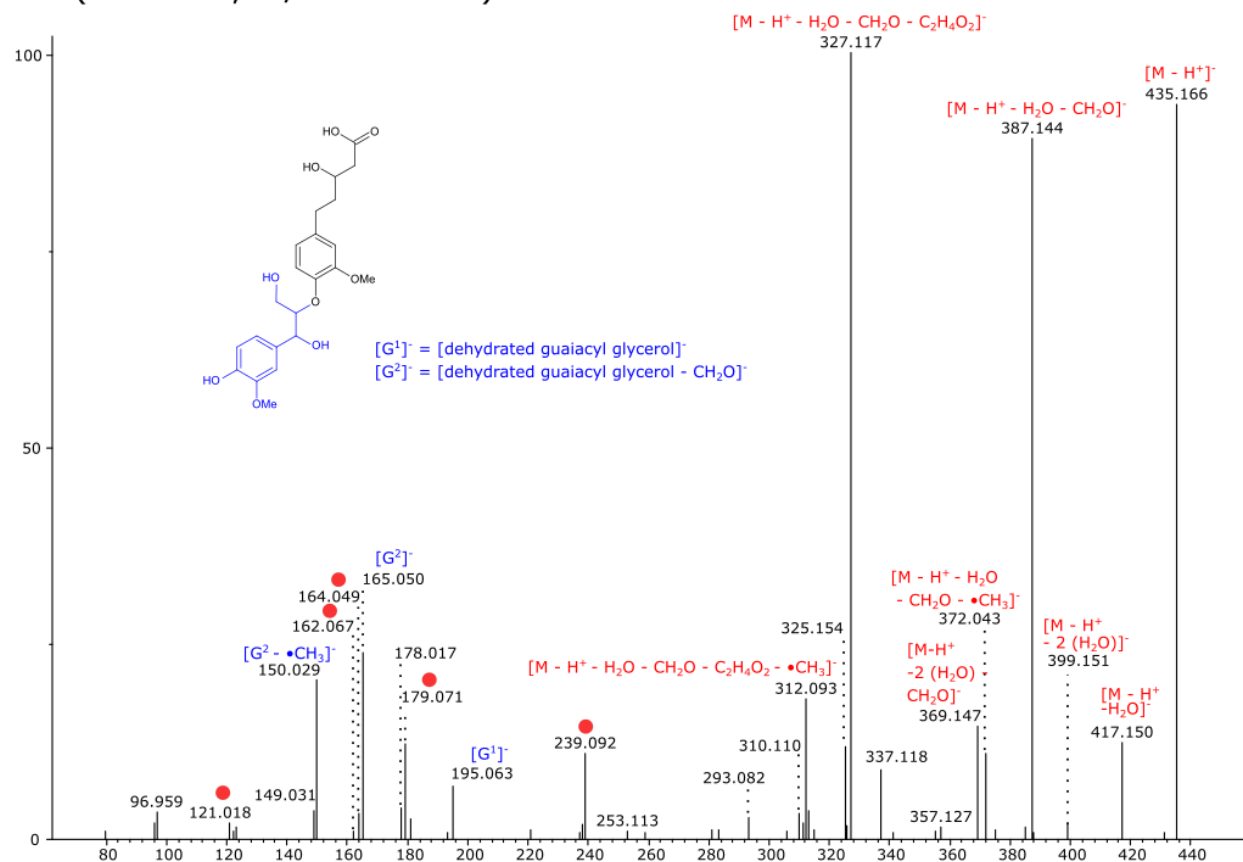
39. MS/MS dihydroferuloyl- β -keto acid sinapic acid cyclobutane dimer (14.36 min, m/z 223.0614, fragment of m/z 461.1476)



40. MS/MS dihydroferuloyl- β -keto acid dihydroferuloyl- β -keto acid cyclobutane dimer (11.23 min, m/z 475.1626)



41. MS/MS tetrahydroferuloyl- β -keto acid(4-O-8)G (8.82 min, m/z 435.1676)



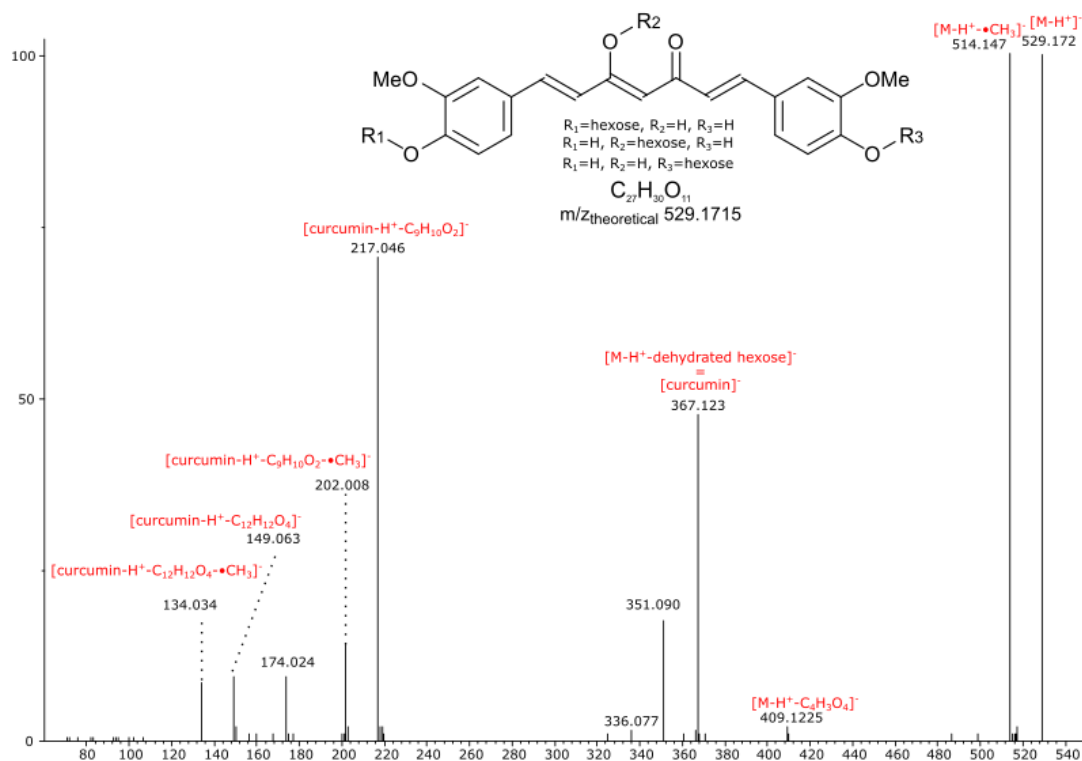
Supplementary Figure 5. Targeted search for curcuminoid metabolites in the *pCesA4:DCS_CURS2* lines. (a) Peak area of curcumin-*O*-hexoside and a curcumin-containing trimer in which coniferyl alcohol is linked to curcumin(8-8)G. Because their respective peak areas were below 1000, the latter two compounds did not end up as one of the 49 compounds in **Table 1**. *n* = 8 independent biological replicates per line. stdev: standard deviation; R.T.: retention time; inf: infinity; ratio: average *pCesA4:DCS_CURS2* / average WT. (b) MS/MS-based structural elucidation of curcumin-*O*-hexoside. (c) MS/MS-based structural elucidation of the trimer in which coniferyl alcohol is linked to curcumin(8-8)G. The compound is either curcumin(8-8)G(4-*O*-8)G, G(8-*O*-4)curcumin(8-8)G, or G(8-*O*-4')curcumin(8-8)G. For (b) and (c), a representative MS/MS spectrum is shown from at least four independent MS/MS spectra.

a

compound name	<i>m/z</i>	R.T. (min)	Wild type	<i>pCesA4:DCS_CURS2_1</i>		<i>pCesA4:DCS_CURS2_2</i>	
			mean ± stdev	mean ± stdev	ratio	mean ± stdev	ratio
curcumin- <i>O</i> -hexoside	529.173	19.59	n.d.	168 ± 130	inf	214 ± 140	inf
curcumin(8-8)G(8- <i>O</i> -4)G	741.255	22.78	n.d.	367 ± 368	inf	941 ± 733	inf

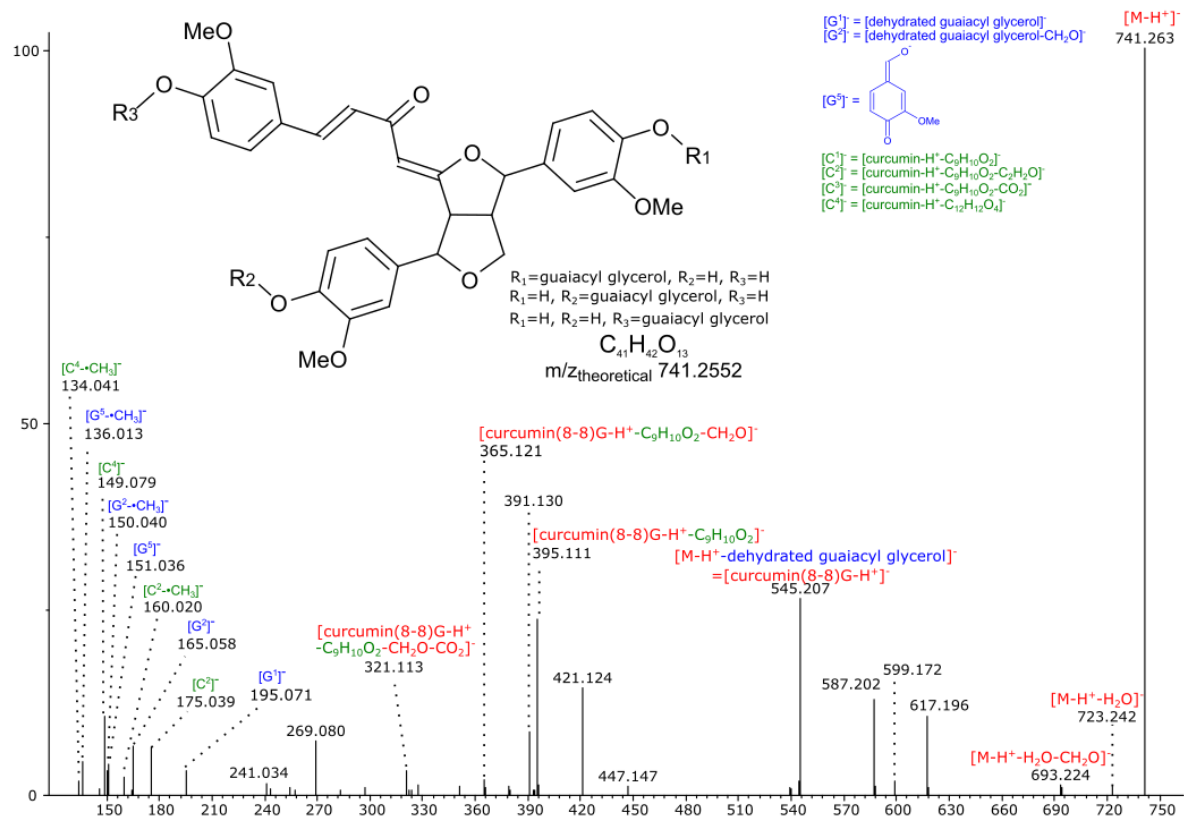
b

curcumin-*O*-hexoside (19.59 min, *m/z* 529.1729)

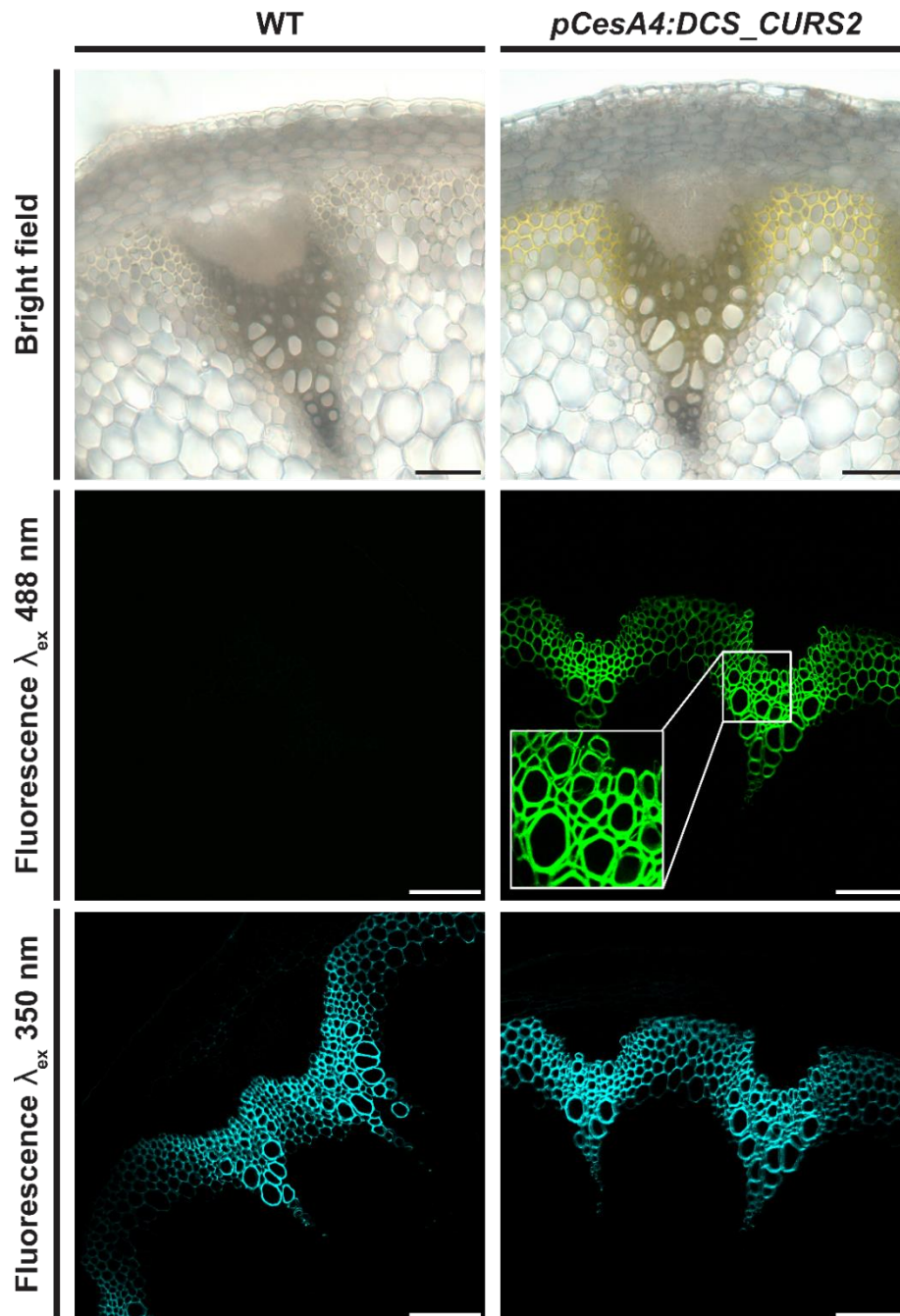


c

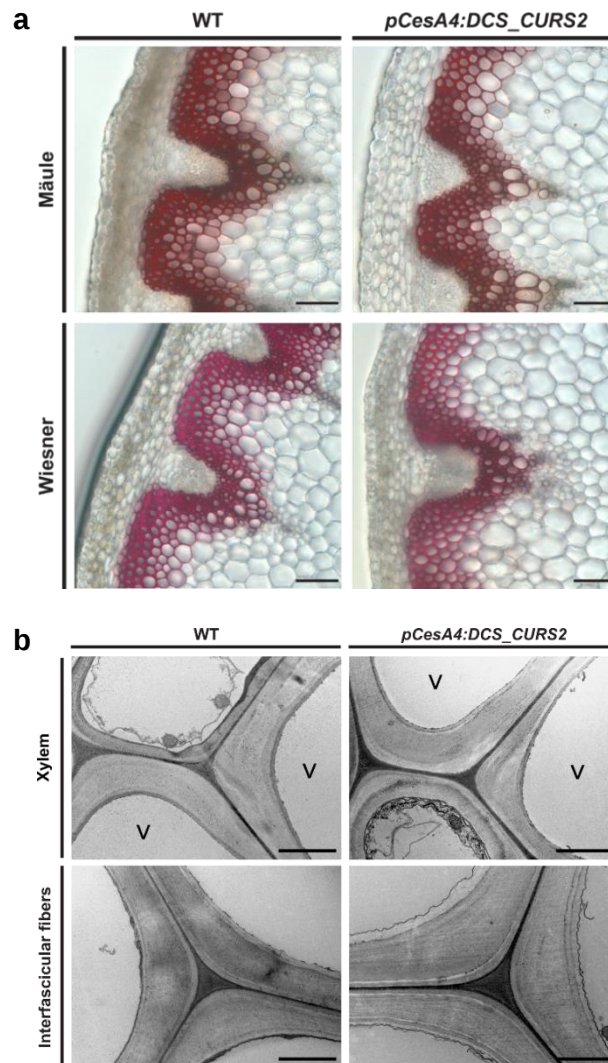
curcumin(8-8)G(4-O-8)G or G(8-O-4/4')curcumin(8-8)G (22.78 min, m/z 741.2545)



Supplementary Figure 6. Bright field and fluorescence microscopy of WT and *pCesA4:DCS_CURS2* cross sections. The yellow colour and fluorescence at an excitation wavelength of 488 nm remained in the secondary cell walls of cross sections of *pCesA4:DCS_CURS2* inflorescence stems after a 2 h incubation in acetone. Representative images are shown from six independent biological replicates per line, eight sections for each biological replicate and three regions for each section. Scale bar, 100 μ m.

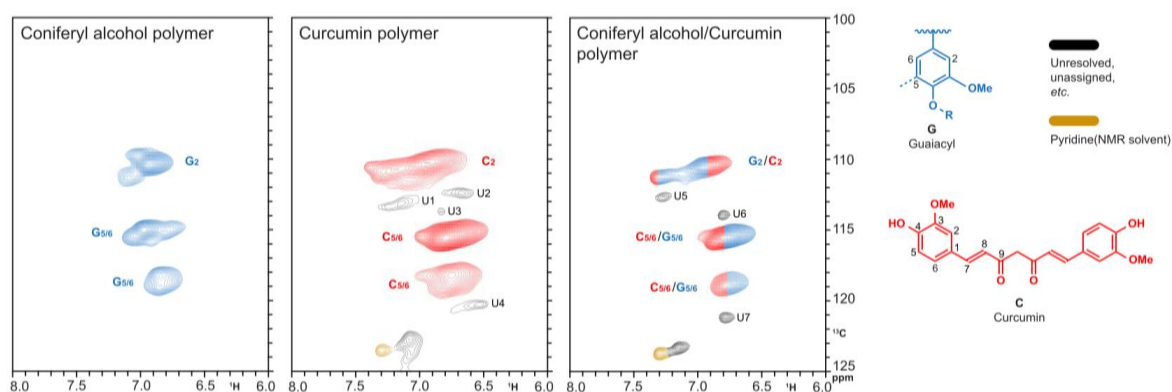


Supplementary Figure 7. Lignin staining and TEM of WT and *pCesA4:DCS_CURS2* stem cross sections. (a) Mäule stains S units red and G units light brown and is generally used to visualize the presence of S units in lignin, whereas the Wiesner reagent colors cinnamaldehyde end-groups in lignin and is used to visualize the presence of lignin in a concentration-dependent manner. Representative images are shown of three independent biological replicates per line, eight sections from each biological replicate and three regions for each section. With both Mäule and Wiesner staining, no differences were observed between *pCesA4:DCS_CURS2* and WT. Scale bar, 100 μ m. (b) Transmission electron microscopy (TEM) demonstrates the ultrastructure of the interfascicular fibers and xylem regions. Representative images are shown from three independent biological replicates per line, one section for each biological replicate and a visual screen of the complete xylem and interfascicular fiber region for each section. No differences between *pCesA4:DCS_CURS2* and WT were observed. v, xylary vessel. Scale bar, 2 μ m.

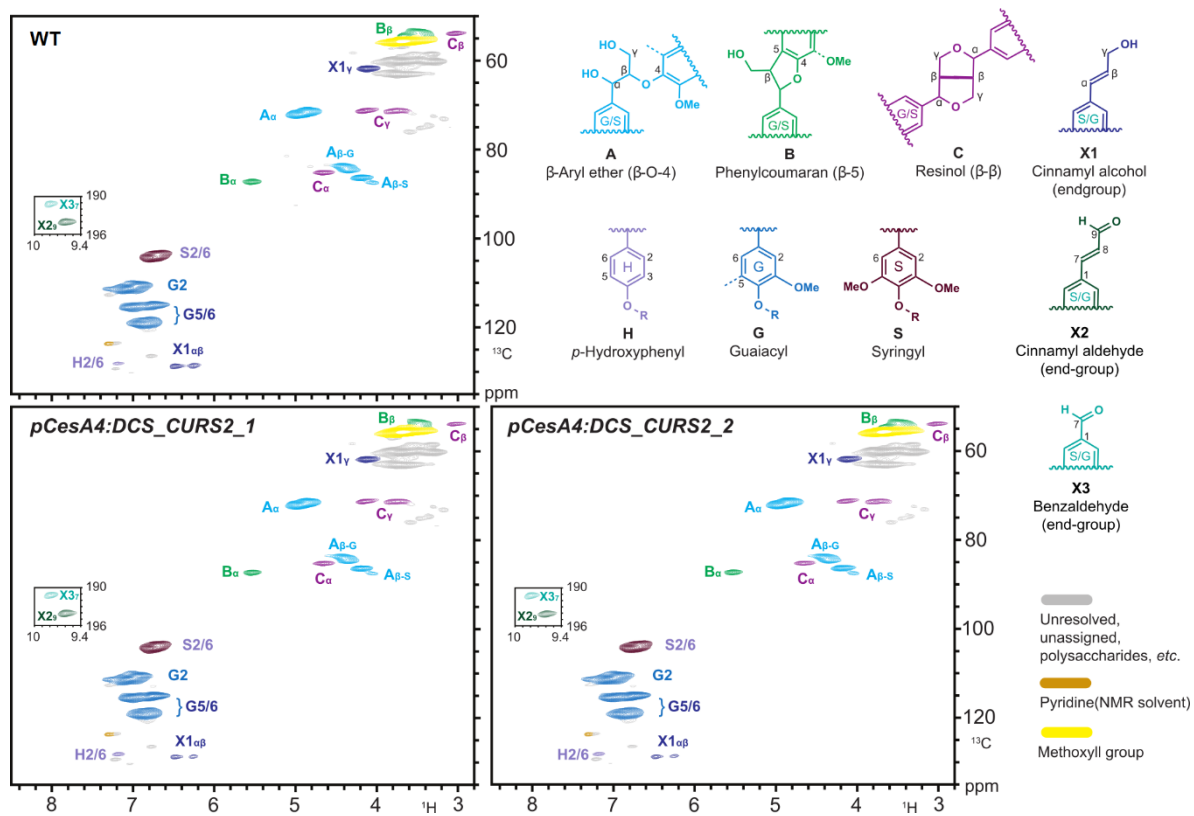


Supplementary Figure 8. Structural characterization of dehydrogenation polymers (DHPs) and lignins by 2D HSQC NMR. It was investigated which 2D HSQC NMR signals were specific for curcumin, anticipating that curcumin might provide signatures of its incorporation into lignins. **(a)** Pure coniferyl alcohol DHP; pure curcumin DHP; 50% coniferyl alcohol and 50% curcumin DHP ($n=1$). In the DHPs made from pure curcumin and coniferyl alcohol/curcumin, the main signals originated from the aromatic rings, i.e. the C₂, C₅ and C₆ signals and the G₂/C₂, G₅/C₅ and G₆/C₆ signals, respectively. However, similar signals were also present in DHPs made from pure coniferyl alcohol (G₂, G₅ and G₆) and can therefore not be used as specific markers for curcumin. In the pure curcumin DHP, signals (U1-U4) with relatively low intensity were detected that were not present in the pure coniferyl alcohol DHP. However, these signals remain unassigned. Also in the coniferyl alcohol/curcumin DHP, different signals (U5-U7) with relatively low intensity were detected that were not present in the pure coniferyl alcohol DHP. These signals also remained unassigned. **(b)** HSQC spectra of the aromatic regions and the oxygenated aliphatic regions of whole cell walls from stems of WT and the *pCesA4:DCS_CURS2* lines. A representative spectrum from three independent biological repeats is shown; each biological repeat is a pool of four or five inflorescence stems. Integrated values for each monomeric unit H, G and S and the α -C/H correlation peaks from the major lignin interunit structures A-C are provided in Table 2. The colors of the substructures shown match those of the corresponding signals in the HSQC spectra (where they are resolved). None of the U1-U7 signals were present in 2D HSQC NMR spectra of lignin samples from the *pCesA4:DCS_CURS2* lines. No other signals appeared in the *pCesA4:DCS_CURS2* spectra that were different to those of WT.

a

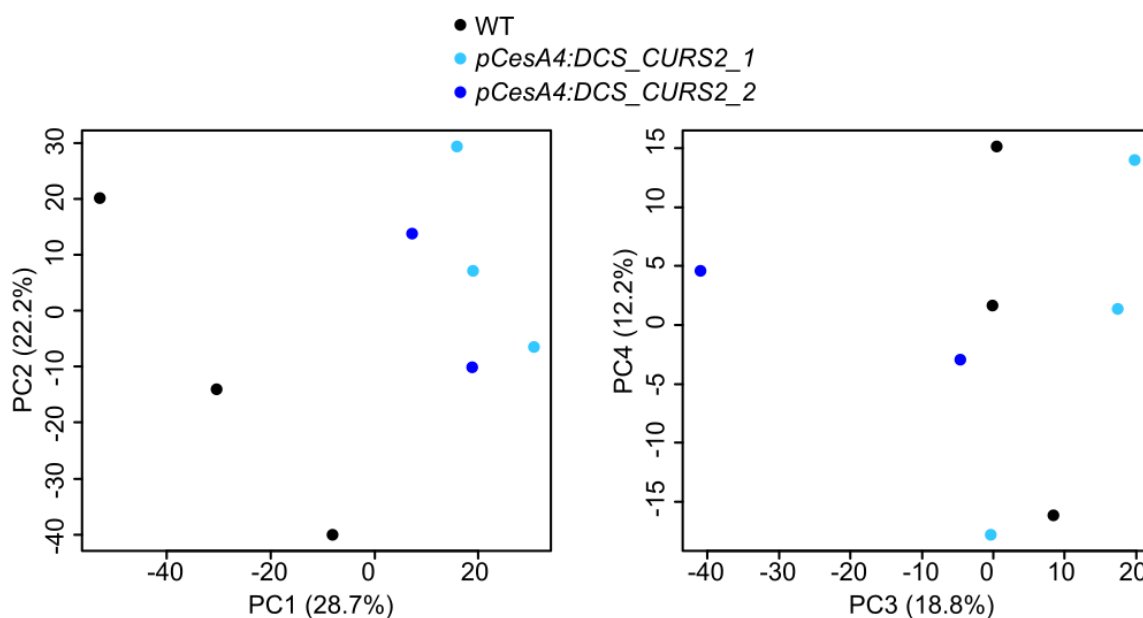


b



Supplementary Figure 9. Characterization of catalytic hydrogenolysis lignin oil from *pCesA4:DCS_CURS2* inflorescence stems. (a) PCA showing the separation between UHPLC-MS profiles of *pCesA4:DCS_CURS2* and WT lignin oils based on PC1. For WT, *pCesA4:DCS_CURS2_1* and *pCesA4:DCS_CURS2_2*, $n=3$, 3 and 2, respectively. (b) Peak list of 21 compounds that accumulated significantly 10-fold in lignin oils of *pCesA4:DCS_CURS2* ($n=5$) as compared with lignin oils from WT ($n=3$). Names given are based on high-resolution m/z and on MS/MS spectra given in panel (c). Ratio: average *pCesA4:DCS_CURS2* / average WT. SD, standard deviation. (c) MS/MS spectra of compounds given in (b). Because of the lack of appropriate authentic chemical standards for these compounds, the structural characterization remains tentative. Two independent MS/MS spectra were generated for each compound, of which one MS/MS spectrum is shown. The signals for which the m/z value is given were common in the two independent MS/MS spectra. The detection limit was set at a peak intensity of 200 counts. n.d., not detected.

a

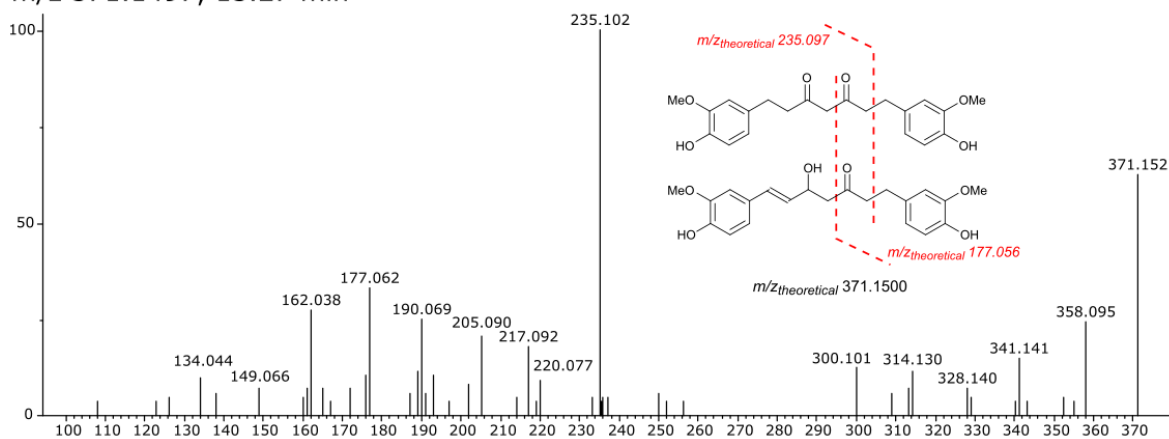


b

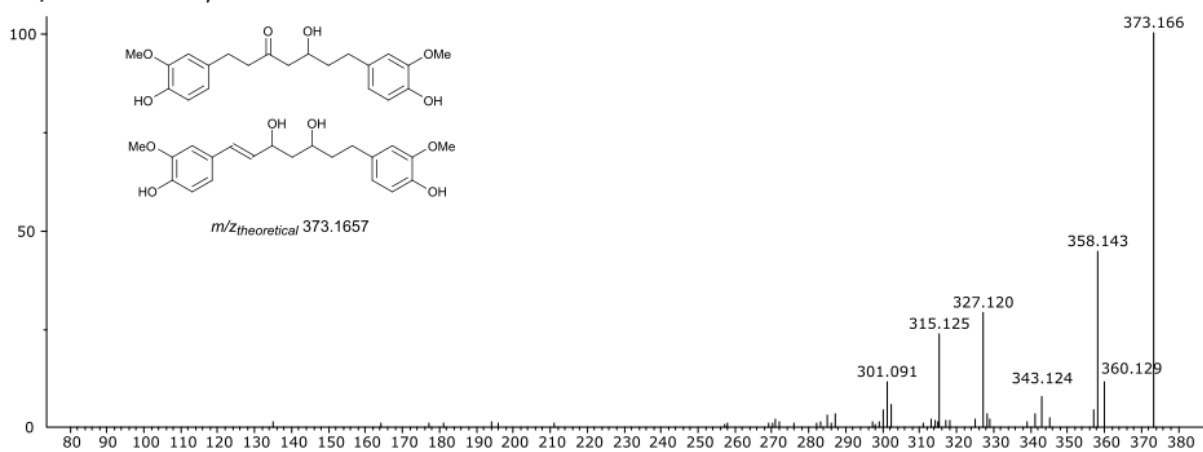
no.	m/z	R.T. (min)	compound name	WT	<i>pCesA:DCS_CURS</i>	ratio
				Average $\pm$ SD	Average $\pm$ SD	<i>pCesA:DCS_CURS2/WT</i>
1	249.0772	8.71	-	366 $\pm$ 187	3679 $\pm$ 452	10
2	263.0941	10.19	-	n.d.	1089 $\pm$ 301	inf
3	371.1497	13.27	tetrahydrocurcumin	n.d.	1159 $\pm$ 501	inf
4	373.1651	15.82	hexahydrocurcumin 1	n.d.	8200 $\pm$ 1433	inf
5	389.1938	20.81	dideoxyhexahydroferuloyl- β -keto acid(red5-8)G	n.d.	4193 $\pm$ 1145	inf
6	187.0761	21.63	-	278 $\pm$ 77	2909 $\pm$ 410	10
7	417.4103	22.28	-	n.d.	1083 $\pm$ 142	inf
8	387.1800	22.56	dideoxyhexahydroferuloyl- β -keto acid(5-8)G	n.d.	2370 $\pm$ 558	inf
9	387.1524	23.00	-	n.d.	1938 $\pm$ 178	inf
10	537.2445	23.94	deoxyoctahydrocurcumin(5-8)G	n.d.	1053 $\pm$ 315	inf
11	537.2466	24.69	deoxyoctahydrocurcumin(5-8)G	n.d.	2983 $\pm$ 484	inf
12	201.0915	24.83	-	n.d.	1204 $\pm$ 332	inf
13	343.1549	25.14	-	n.d.	3207 $\pm$ 672	inf
14	359.1515	26.71	-	n.d.	2260 $\pm$ 305	inf
15	309.2075	26.77	-	n.d.	1015 $\pm$ 248	inf
16	359.3899	27.34	-	n.d.	1089 $\pm$ 110	inf
17	357.1711	28.18	deoxyhexahydrocurcumin (no MSMS)	n.d.	3809 $\pm$ 645	inf
18	431.2073	28.98	-	n.d.	3475 $\pm$ 549	inf
19	373.1666	29.08	hexahydrocurcumin 2	n.d.	2942 $\pm$ 362	inf
20	509.2207	29.95	-	n.d.	3795 $\pm$ 403	inf
21	415.2133	32.47	-	n.d.	1479 $\pm$ 228	inf

c

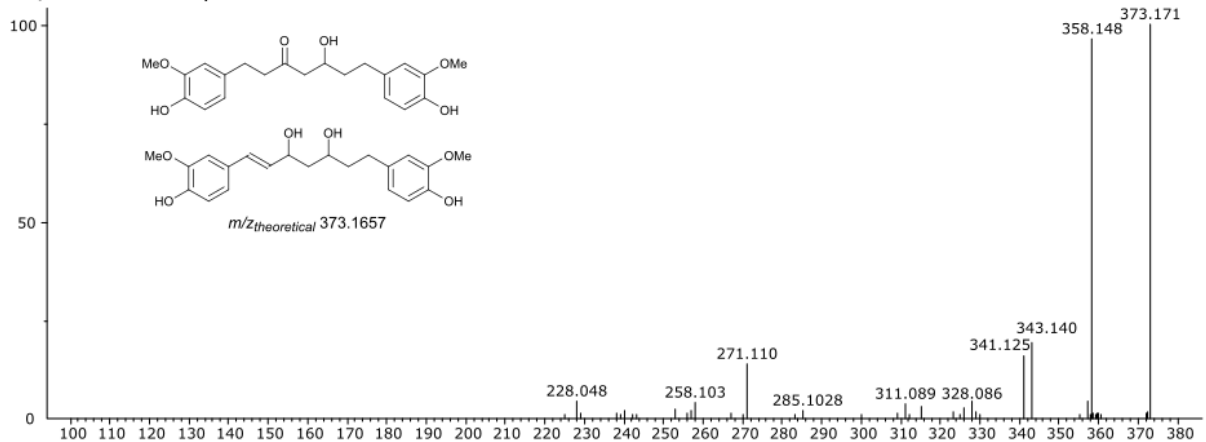
tetrahydrocurcumin
m/z 371.1497, 13.27 min



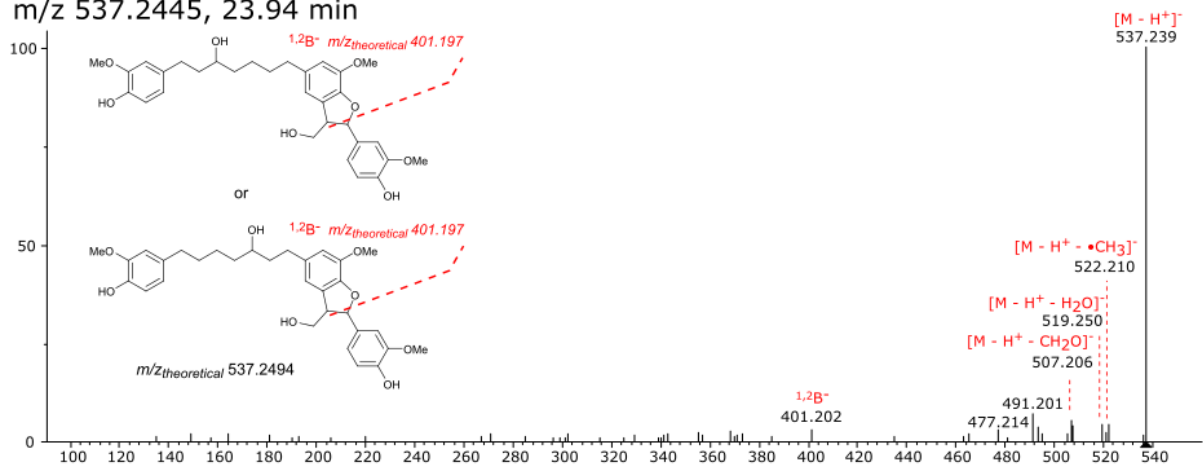
hexahydrocurcumin 1
m/z 373.1651, 15.82 min



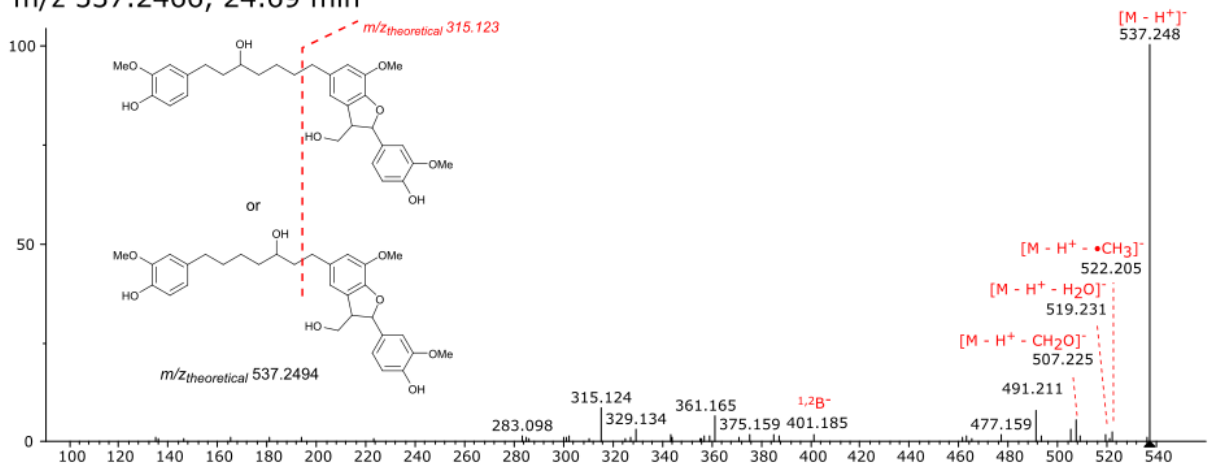
hexahydrocurcumin 2
m/z 373.1667, 29.08 min



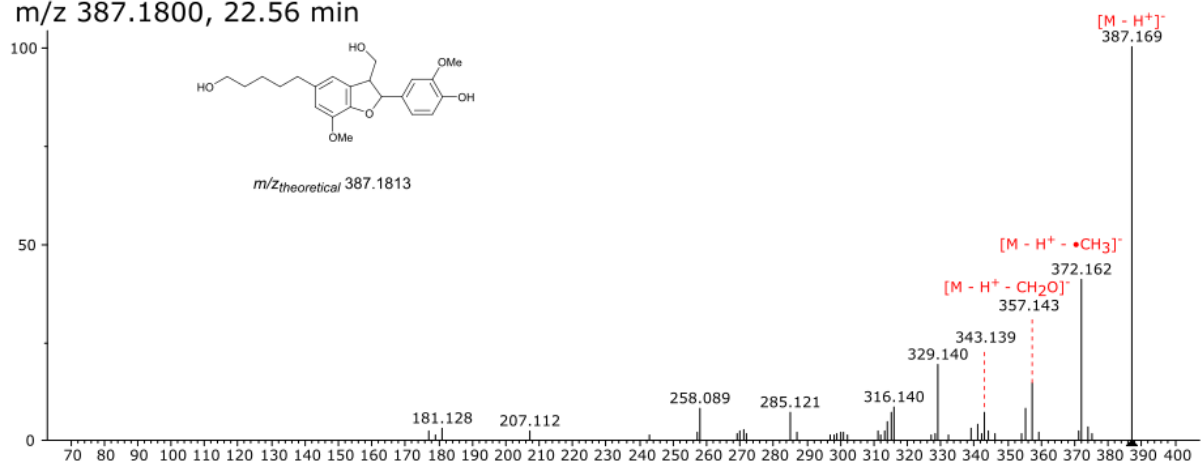
deoxyoctahydrocurcumin(5-8)G 1
m/z 537.2445, 23.94 min



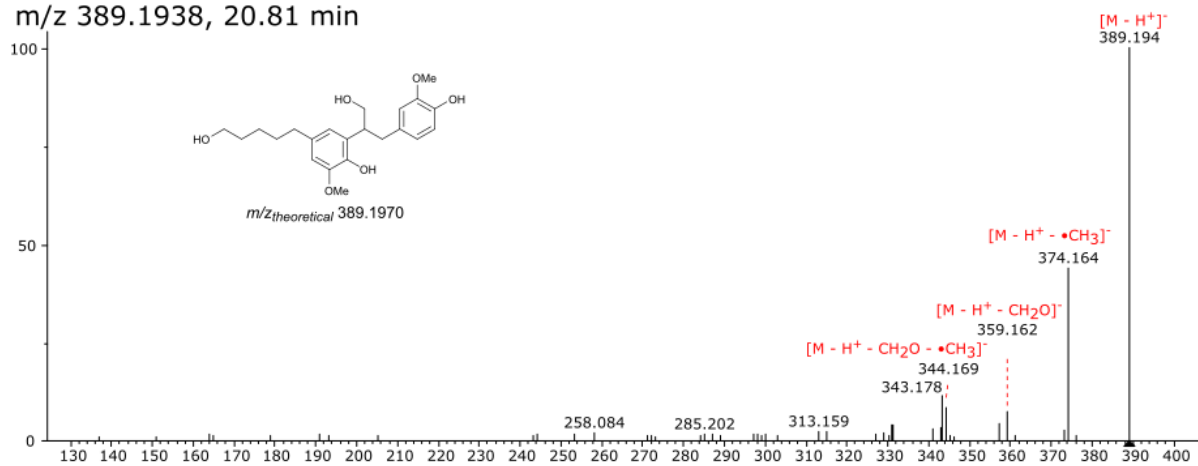
deoxyoctahydrocurcumin(5-8)G 2
m/z 537.2466, 24.69 min



dideoxyhexahydroferuloyl- β -keto acid(5-8)G
[4-hydroxy-3-methoxybenzenepentanol(5-8)G]
m/z 387.1800, 22.56 min

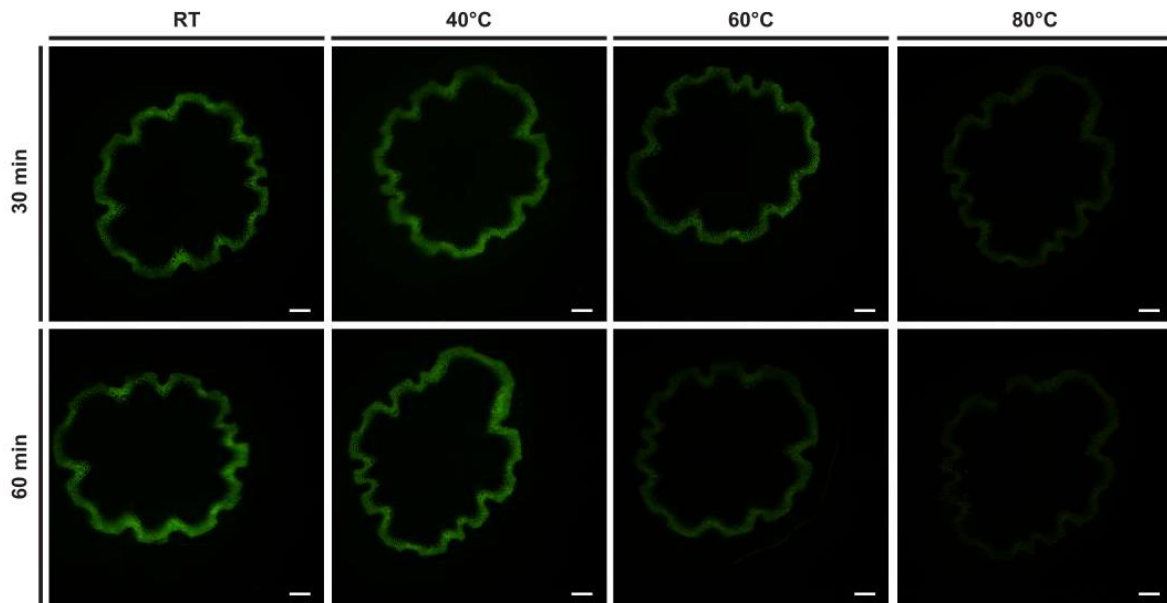


dideoxyhexahydroferuloyl- β -keto acid(red5-8)G
 [4-hydroxy-3-methoxybenzenepentanol(red5-8)G]
 m/z 389.1938, 20.81 min

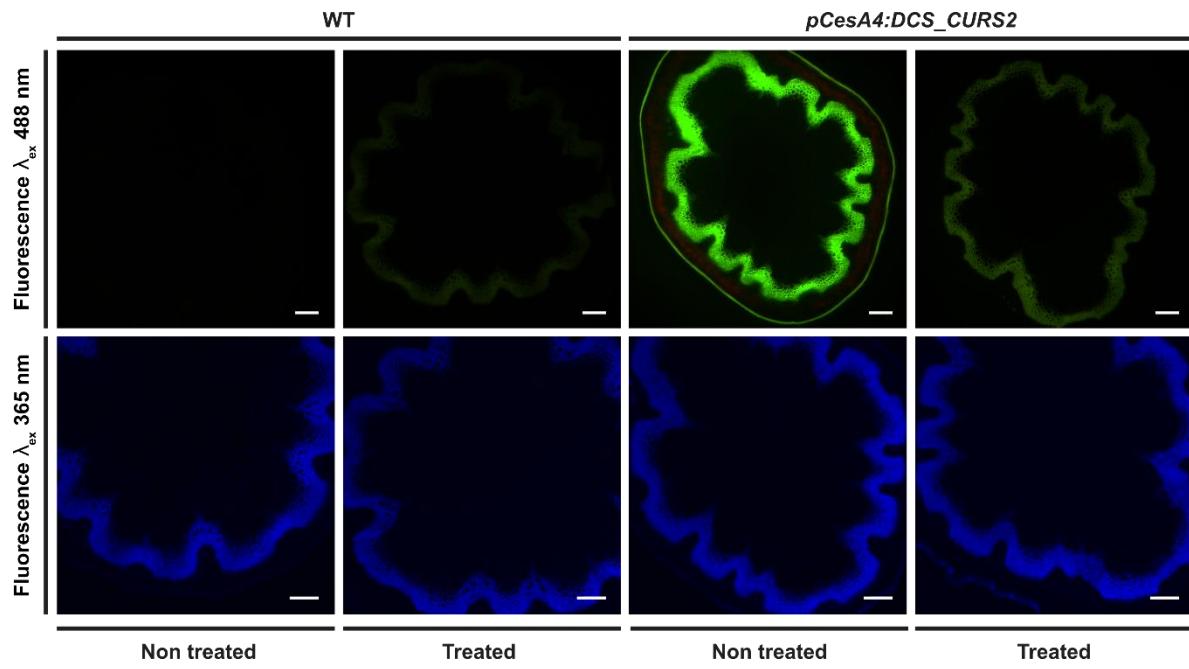


Supplementary Figure 10. Curcumin breakdown in alkaline conditions. (a) Transgenic *pCesA4:DCS_CURS2* stem cross sections were incubated in 62.5 mM NaOH (pH 12.8) for 30 and 60 min at four different temperatures: room temperature (RT), 40°C, 60°C and 80°C. Images (excitation filter of 460-500 nm) taken after extensive washing show representative pictures from 20 to 30 replicates originating from 3 independent biological replicates. Scale bars, 200 μ m. (b) WT and *pCesA4:DCS_CURS2* stem cross sections were incubated in 62.5 mM NaOH (pH 12.8, 90°C, 30 min). Fluorescence of *pCesA4:DCS_CURS2* at an excitation wavelength of 488 nm decreased after the treatment. WT cross sections were not fluorescent at a wavelength of 488 nm before treatment, but were slightly fluorescent after the alkaline treatment. The autofluorescence of lignin at an excitation wavelength of 365 nm was similar for WT and *pCesA4:DCS_CURS2* cross sections and appeared not to be affected by the alkaline treatment. Representative images are shown from three replicates derived from a visual screening of 24 stem sections (eight stem sections of three independent biological replicates per line) for each condition tested. Scale bars, 200 μ m.

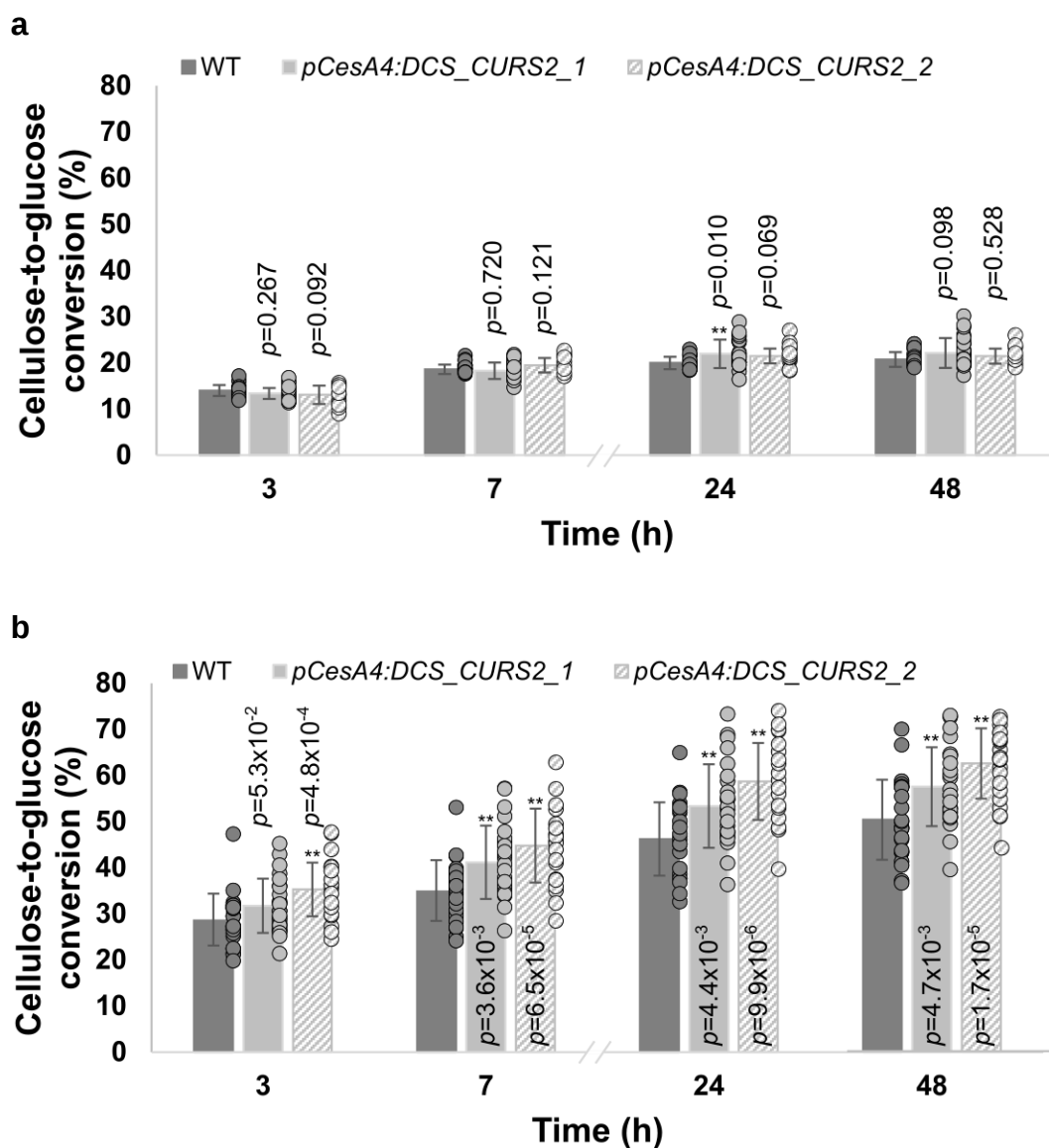
a



b



Supplementary Figure 11. Cellulose-to-glucose conversion during saccharification of WT and *pCesA4:DCS_CURS2* cell wall material. Samples were saccharified using (a) no or (b) alkaline pretreatment (62.5 mM NaOH, 90°C, 3 h). Individual values (dots) and means (bars) of 24 biological replicates for all lines are shown. *****P* < 0.01** (two-tailed Student's *t*-test) the exact *P* value is shown. Error bars indicate standard deviation.



Supplementary Table 1. Cell wall composition of the transgenic lines and WT plants. CWR was determined as the fraction of material obtained after washes relative to the original dry weight. Lignin content was determined by acetyl bromide (AcBr) and Klason, and expressed as a percentage of CWR ($n = 8$ independent biological replicates for all lines). Lignin composition was determined in enzyme lignin by 2D HSQC NMR ($n = 3$ pools of each 5 independent biological replicates for all lines). Crystalline cellulose content was determined by the Updegraff method, while alpha-cellulose (i.e. the sum of crystalline and amorphous cellulose) was determined via an acetic acid/nitric acid-based protocol (see Methods) ($n = 8$ pools of each 6 independent biological replicates for all lines). Both are expressed as a percentage of CWR. No statistical differences were observed between transgenic and control lines ($P > 0.05$; two-tailed Student's t -test). stdev, standard deviation.

	WT mean $\pm$ stdev	<i>pCesA4:DCS_CURS2_1</i> mean $\pm$ stdev (P value)	<i>pCesA4:DCS_CURS2_2</i> mean $\pm$ stdev (P value)
CWR (% dry weight)	77.5 $\pm$ 3.2	77.0 $\pm$ 2.4 (0.747)	76.9 $\pm$ 1.2 (0.643)
Lignin			
AcBr lignin (% CWR)	11.2 $\pm$ 1.7	11.6 $\pm$ 1.1 (0.577)	11.2 $\pm$ 0.3 (0.967)
Klason lignin (% CWR)	16.1 $\pm$ 0.4	17.1 $\pm$ 2.0 (0.442)	15.8 $\pm$ 0.9 (0.644)
Aromatic units (%)			
%S	24.2 $\pm$ 1.2	24.2 $\pm$ 0.6 (0.551)	24.8 $\pm$ 0.3 (0.911)
%G	74.8 $\pm$ 1.1	74.5 $\pm$ 0.8 (0.681)	73.9 $\pm$ 0.4 (0.534)
%H	1.0 $\pm$ 0.2	1.3 $\pm$ 0.3 (0.214)	1.3 $\pm$ 0.1 (0.132)
Inter-unit linkages (%)			
β -aryl ether	83.0 $\pm$ 1.7	84.7 $\pm$ 1.2 (0.316)	85.7 $\pm$ 0.5 (0.163)
Phenylcoumaran	10.5 $\pm$ 1.4	9.5 $\pm$ 0.9 (0.436)	8.4 $\pm$ 0.2 (0.131)
Resinol	6.5 $\pm$ 0.4	5.8 $\pm$ 0.2 (0.076)	5.9 $\pm$ 0.3 (0.299)
Cellulose			
Crystalline cellulose (%CWR)	37.7 $\pm$ 2.1	39.1 $\pm$ 1.7 (0.174)	37.8 $\pm$ 1.2 (0.885)
Alpha-cellulose (%CWR)	45.6 $\pm$ 2.6	45.8 $\pm$ 2.4 (0.875)	44.9 $\pm$ 1.8 (0.521)

Supplementary Table 2. GPC chromatographic comparison of acetylated lignin from *pCesA4:DCS_CURS2* and WT inflorescence stems. Mean $\pm$ standard deviation of number-average molecular weight (Mn), weight-average molecular weight (Mw), and polydispersity index (PDI) estimated from the GPC curves (integrated between 80,000 and 600 Da). n = 3 pools of each 5 independent biological replicates for all lines. No significant differences were observed between the lines ($P > 0.05$; two-tailed Student's *t*-test).

Line	Mw (Da)	(<i>P</i> value)	Mn (Da)	(<i>P</i> value)	PDI (Mw/Mn)	(<i>P</i> value)
WT	30,019 $\pm$ 1,330		10,249 $\pm$ 604		2.93 $\pm$ 0.06	
<i>pCesA4:DCS_CURS2_1</i>	28,582 $\pm$ 3,340	(0.602)	9,636 $\pm$ 630	(0.377)	2.96 $\pm$ 0.15	(0.828)
<i>pCesA4:DCS_CURS2_2</i>	31,071 $\pm$ 1,292	(0.468)	10,430 $\pm$ 206	(0.708)	2.98 $\pm$ 0.07	(0.516)