

Supplementary Material

# Accurate Formation Enthalpies of Solids Using Reaction Networks

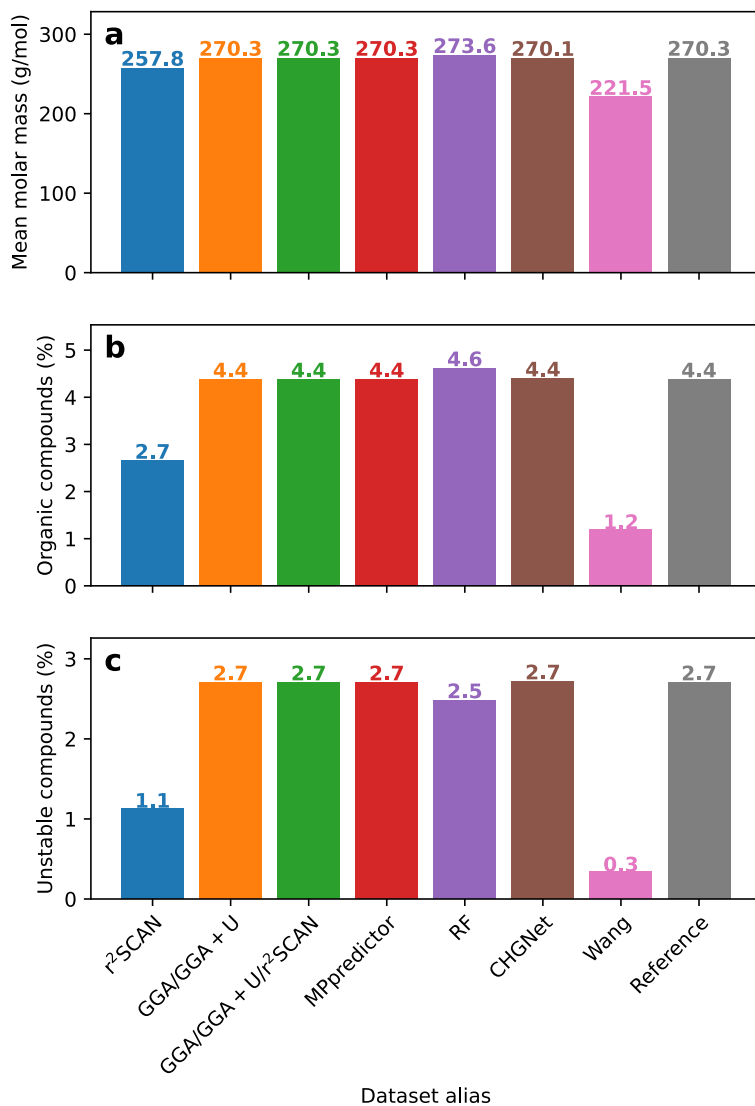
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# 1 Datasets

## 1.1 Statistics



Supplementary Figure 1: Various statistics of the datasets considered in the paper (see Table 1, for further information about the individual datasets). For a compound in a given dataset, the compound is considered unstable if  $\Delta_f H_{\text{ref}} > 0$  (ref is Reference) for this compound.

## 1.2 Data-validation

Supplementary Table 1: Significant disagreements, between experimental values of  $\Delta_f H$  from the NBS reference dataset ( $\Delta_f H_{\text{ref}}$ ) and from values in the Wang data-set [1]. All values are reported in eV atom<sup>-1</sup>. These compounds are not included in the main data-sets.

| likely_mpid | Formula                                        | Reference | Wang   |
|-------------|------------------------------------------------|-----------|--------|
| mp-1018028  | TiS                                            | -1.233    | -1.410 |
| mp-12765    | GdC <sub>2</sub>                               | -0.363    | -0.257 |
| mp-1336     | PdO                                            | -0.443    | -0.599 |
| mp-1479     | BP                                             | -0.409    | -0.599 |
| mp-18511    | NdF <sub>3</sub>                               | -4.293    | -4.438 |
| mp-19760    | Na <sub>2</sub> Mo <sub>2</sub> O <sub>7</sub> | -2.115    | -2.225 |
| mp-1984     | TaS <sub>2</sub>                               | -1.603    | -1.223 |
| mp-2071     | Zn <sub>3</sub> P <sub>2</sub>                 | -0.980    | -0.330 |
| mp-21075    | HfC                                            | -1.303    | -1.171 |
| mp-21856    | Co <sub>2</sub> SiO <sub>4</sub>               | -2.187    | -2.071 |
| mp-22205    | InN                                            | -0.091    | -0.716 |
| mp-22216    | In <sub>2</sub> S <sub>3</sub>                 | -0.885    | -0.737 |
| mp-22856    | Bi <sub>2</sub> S <sub>3</sub>                 | -0.297    | -0.418 |
| mp-22870    | InBr                                           | -0.908    | -0.106 |
| mp-23162    | ZrCl <sub>2</sub>                              | -1.734    | -1.489 |
| mp-27719    | CuAl <sub>2</sub> O <sub>4</sub>               | -2.684    | -2.883 |
| mp-2815     | MoS <sub>2</sub>                               | -0.812    | -0.954 |
| mp-29555    | MoBr <sub>2</sub>                              | -0.902    | -0.701 |
| mp-438      | Nd <sub>2</sub> S <sub>3</sub>                 | -2.463    | -2.333 |
| mp-540828   | FeOCl                                          | -1.302    | -1.411 |
| mp-542613   | GeS <sub>2</sub>                               | -0.655    | -0.542 |
| mp-684638   | Al <sub>2</sub> S <sub>3</sub>                 | -1.501    | -1.349 |
| mp-700      | GeSe                                           | -0.477    | -0.358 |
| mp-9813     | WS <sub>2</sub>                                | -0.722    | -0.896 |

Supplementary Table 2: Significant disagreements between experimental values of  $\Delta_f H$  from the NBS reference dataset ( $\Delta_f H_{\text{ref}}$ ) and from various literature sources ( $\Delta_f H_{\text{lit}}$ ).  $\Delta_f H_{\text{lit}}$  is reported as a single value or a range of values. In some cases no value is reported but estimate(s) of a closely related property, related compound or supporting calculations suggest that there is significant uncertainty/inaccuracy in  $\Delta_f H$  for the compound in question. All values are reported in eV atom<sup>-1</sup>. These compounds are not included in the main datasets.

| likely_mpid | Formula                          | Reference | Literature                    | Literature source |
|-------------|----------------------------------|-----------|-------------------------------|-------------------|
| mp-1188188  | Mn <sub>3</sub> C                | 0.012     | -0.103                        | [2]               |
| mp-1274     | CoS                              | -0.429    | — <sup>§</sup>                | [3]               |
| mp-1347     | Th <sub>3</sub> P <sub>4</sub>   | -1.691    | — <sup>§</sup>                | [4]               |
| mp-13862    | B <sub>13</sub> P <sub>2</sub>   | -0.115    | — <sup>§</sup>                | [4]               |
| mp-1415     | CaSe                             | -1.908    | -2.145                        | [5]               |
| mp-1550     | AlP                              | -0.863    | -0.671 <sup>†</sup>           | [4]               |
| mp-1598     | Na <sub>3</sub> P                | -0.240    | -0.523 to -0.257 <sup>‡</sup> | [4]               |
| mp-15668    | Na <sub>2</sub> Se <sub>2</sub>  | -0.970    | — <sup>§</sup>                | [5]               |
| mp-2011     | UP                               | -1.389    | -1.689 <sup>†</sup>           | [4]               |
| mp-22488    | Co <sub>2</sub> C                | 0.145     | 0.010                         | [6]               |
| mp-2296     | Ni <sub>3</sub> P                | -0.544    | -0.919 to -0.309 <sup>†</sup> | [4]               |
| mp-23162    | ZrCl <sub>2</sub>                | -1.734    | -1.834                        | [7]               |
| mp-24290    | S <sub>7</sub> NH                | -0.325    | 0.103                         | [8]               |
| mp-2441     | Cd <sub>3</sub> P <sub>2</sub>   | -0.238    | — <sup>§</sup>                | [4]               |
| mp-2490     | GaP                              | -0.912    | — <sup>§</sup>                | [4]               |
| mp-2662     | MnP                              | -0.586    | -0.713 <sup>†</sup>           | [4]               |
| mp-2758     | SrSe                             | -1.999    | -2.104                        | [5]               |
| mp-27440    | ZrCl                             | -1.368    | -1.555                        | [7]               |
| mp-29014    | P <sub>2</sub> S <sub>3</sub>    | -0.167    | -0.324                        | [9]               |
| mp-413      | UP <sub>2</sub>                  | -1.054    | — <sup>§</sup>                | [4]               |
| mp-560553   | IrF <sub>6</sub>                 | -0.858    | -1.316                        | [10]              |
| mp-560613   | Na <sub>2</sub> TeO <sub>4</sub> | -1.660    | -1.445                        | [11]              |
| mp-561511   | FeAsS                            | -0.145    | — <sup>§</sup>                | [12]              |
| mp-568264   | SiSe <sub>2</sub>                | -0.100    | -0.613                        | [5]               |
| mp-739      | TiP                              | -1.466    | — <sup>§</sup>                | [4]               |
| mp-7463     | Cu <sub>3</sub> P                | -0.393    | -0.396 to -0.176 <sup>†</sup> | [4]               |
| mp-787      | U <sub>3</sub> P <sub>4</sub>    | -1.239    | -1.840 <sup>†</sup>           | [4]               |
| mp-927      | CuP <sub>2</sub>                 | -0.418    | -0.498 to -0.260 <sup>†</sup> | [4]               |
| mp-931      | ThP                              | -1.804    | — <sup>§</sup>                | [4]               |
| mp-972      | MnSe                             | -0.553    | -0.922                        | [5]               |

<sup>†</sup> The reference state that has been for P used in the source (red V) has been converted to the reference state used in the reference data ( $\alpha$ -white), using the suggested enthalpy difference in [4].

<sup>‡</sup> Unknown which reference state(s) are used for P.

<sup>§</sup> No value is reported in the source, but estimate(s) of a closely related property, related compound or supporting calculations suggest that there is significant uncertainty/inaccuracy in  $\Delta_f H$  for this compound.

Compounds for which  $\Delta_f H_{\text{ref}}$  is considered unreliable with reference to the literature values of Wang et al. are given in Table 1. Compounds for which  $\Delta_f H_{\text{ref}}$  is considered unreliable with reference to miscellaneous literature values are given in Table 2. If the same compound was found in both cases, the entry belonging to Table 2 is reported.  $\Delta_f H_{\text{ref}}$  is considered unreliable if it differs from the literature values  $\Delta_f H_{\text{lit}}$  by  $\geq 0.1$  eV atom<sup>-1</sup> or if no explicit value is provided for  $\Delta_f H_{\text{ref}}$  but there is reason to believe that the reference value is highly uncertain.

Supplementary Table 3: Mismatches between Materials Project’s (MP) composition/formula and the matched ICSD entry. These compounds are not included in the main data-sets.

| likely_mpid | Formula (MP)                                     | Formula (ICSD)                                                                                                    | ICSD identifier |
|-------------|--------------------------------------------------|-------------------------------------------------------------------------------------------------------------------|-----------------|
| mp-1057216  | LiS                                              | LiSH                                                                                                              | 25590           |
| mp-1180691  | KVO <sub>4</sub>                                 | KVO <sub>3</sub> · H <sub>2</sub> O                                                                               | 24728           |
| mp-1193254  | Sr(CO <sub>2</sub> ) <sub>2</sub>                | Sr(HCO <sub>2</sub> ) <sub>2</sub>                                                                                | 151246          |
| mp-1197831  | LiBrO <sub>3</sub>                               | LiBrO <sub>2</sub> · H <sub>2</sub> O                                                                             | 4323            |
| mp-1204504  | U(CO <sub>3</sub> ) <sub>2</sub>                 | (UO <sub>2</sub> ) <sub>2</sub> (OH) <sub>2</sub> (C <sub>4</sub> O <sub>4</sub> )(H <sub>2</sub> O) <sub>2</sub> | 248209          |
| mp-643934   | CuH <sub>2</sub> (CO <sub>2</sub> ) <sub>2</sub> | Cu(HCOO) <sub>2</sub> · 4 H <sub>2</sub> O                                                                        | 109965          |
| mp-727410   | K <sub>2</sub> SnCl <sub>4</sub> O               | K <sub>2</sub> SnCl <sub>4</sub> · H <sub>2</sub> O                                                               | 76169           |
| mp-726693   | TlCl <sub>3</sub>                                | TlCl <sub>3</sub> · 4 H <sub>2</sub> O                                                                            | 20760           |

Supplementary Table 4: Compounds for which a more likely materials project id was found (likely\_mpid) during the data processing, after initial assignments (assigned mpid) had been made. These compounds are not included in the main datasets.

| assigned mpid | likely_mpid | Formula                                         | Reference |
|---------------|-------------|-------------------------------------------------|-----------|
| mp-1017565    | mp-612740   | In <sub>2</sub> Se <sub>3</sub>                 | [5]       |
| mp-10264      | mp-203      | NiTe                                            | [13]      |
| mp-1045       | mp-2763     | Nd <sub>2</sub> O <sub>3</sub>                  | [14]      |
| mp-1078295    | mp-3671     | NaNbO <sub>3</sub>                              | [15]      |
| mp-1195863    | mp-1192688  | Zn(N <sub>3</sub> ) <sub>2</sub>                | [16]      |
| mp-15651      | mp-662      | NiSe                                            | [5]       |
| mp-16705      | mp-2063     | Pr <sub>2</sub> O <sub>3</sub>                  | [14]      |
| mp-19443      | mp-619461   | WO <sub>3</sub>                                 | [17]      |
| mp-20530      | mp-780712   | CeCrO <sub>3</sub>                              | [18]      |
| mp-218        | mp-1745     | Sm <sub>2</sub> O <sub>3</sub>                  | [14]      |
| mp-2292       | mp-1968     | La <sub>2</sub> O <sub>3</sub>                  | [14]      |
| mp-24834      | mp-23790    | NH <sub>4</sub> NO <sub>3</sub>                 | [19]      |
| mp-294        | mp-1198490  | UO <sub>3</sub>                                 | [14]      |
| mp-504886     | mp-643084   | Gd <sub>2</sub> O <sub>3</sub>                  | [14]      |
| mp-505531     | mp-2099     | FeS                                             | [20]      |
| mp-551905     | mp-540783   | OsO <sub>4</sub>                                | [21]      |
| mp-552185     | mp-669466   | AgNO <sub>3</sub>                               | [22]      |
| mp-554278     | mp-2657     | TiO <sub>2</sub>                                | [9]       |
| mp-571453     | mp-23499    | K <sub>2</sub> SnCl <sub>6</sub>                | [23]      |
| mp-571486     | mp-488      | CuSe                                            | [5]       |
| mp-583213     | mp-23192    | HgI <sub>2</sub>                                | [24]      |
| mp-696658     | mp-557263   | CH <sub>3</sub> NH <sub>3</sub> NO <sub>3</sub> | [25]      |
| mp-707535     | mp-1181584  | CH <sub>4</sub> N <sub>4</sub> O <sub>2</sub>   | [26]      |
| mp-9813       | mp-224      | WS <sub>2</sub>                                 | [27]      |

Supplementary Table 5: Compounds for which the crystal structure was found to be discrepant or unclear with respect to the NBS reference data (see Comment column), after initial assignments (assigned mpid) had been made. These compounds are not included in the main datasets.

| assigned mpid | Formula                                               | Comment             | Reference |
|---------------|-------------------------------------------------------|---------------------|-----------|
| mp-1077078    | KCN                                                   | wrong polymorph     | [28]      |
| mp-1106356    | AlH <sub>3</sub>                                      | wrong polymorph     | [9]       |
| mp-1179716    | RbHCO <sub>3</sub>                                    | wrong compound      | [29]      |
| mp-1195133    | Ca(CH <sub>2</sub> OHCO <sub>2</sub> ) <sub>2</sub>   | wrong compound      | [30]      |
| mp-1393       | AuSn <sub>4</sub>                                     | wrong symmetry      | [31]      |
| mp-18905      | FeO                                                   | uncertain structure | [32]      |
| mp-1984       | TaS <sub>2</sub>                                      | uncertain polymorph | [33]      |
| mp-29687      | KC <sub>60</sub>                                      | uncertain polymorph | [34]      |
| mp-30053      | NaCN                                                  | wrong polymorph     | [28]      |
| mp-32539      | WO <sub>2</sub> Cl <sub>2</sub>                       | uncertain structure | [35]      |
| mp-554867     | Ta <sub>2</sub> O <sub>5</sub>                        | wrong polymorph     | [36]      |
| mp-560282     | K <sub>2</sub> Ba(NO <sub>2</sub> ) <sub>4</sub>      | wrong polymorph     | [37]      |
| mp-567366     | CsCuCl <sub>3</sub>                                   | wrong polymorph     | [38]      |
| mp-570634     | Cd <sub>3</sub> As <sub>2</sub>                       | wrong polymorph     | [39]      |
| mp-643099     | CNH(NH <sub>2</sub> ) <sub>2</sub> · HNO <sub>3</sub> | wrong polymorph     | [40]      |
| mp-680506     | PtCl <sub>4</sub>                                     | uncertain structure | [41]      |
| mp-721084     | NH <sub>4</sub> I · NH <sub>3</sub>                   | wrong polymorph     | [42]      |

Supplementary Table 6: Compounds with  $\Delta_{\text{hull}}H_{\text{MP}} > 0.2 \text{ eV atom}^{-1}$ ,  $|\Delta_{\text{hull}}H_{\text{MP}} - \Delta_{\text{f}}H_{\text{MP}}| > 0.2 \text{ eV atom}^{-1}$  and not satisfying  $0.85V_{\text{ICSD}} < V_{\text{MP}} < 1.1V_{\text{ICSD}}$ .  $\Delta_{\text{hull}}H_{\text{MP}}$ ,  $\Delta_{\text{f}}H_{\text{MP}}$ ,  $V_{\text{MP}}$  are the default Materials Project GGA/GGA+U/r<sup>2</sup>SCAN values for the energy above hull, formation enthalpy and unit cell volume, respectively.  $V_{\text{ICSD}}$  is the experimental volume and  $\Delta_{\%}V$  is the percentage difference between  $V_{\text{MP}}$  and  $V_{\text{ICSD}}$ . Enthalpies/Energies are in  $\text{eV atom}^{-1}$  and volumes are in  $\text{\AA}^3/\text{atom}$ .

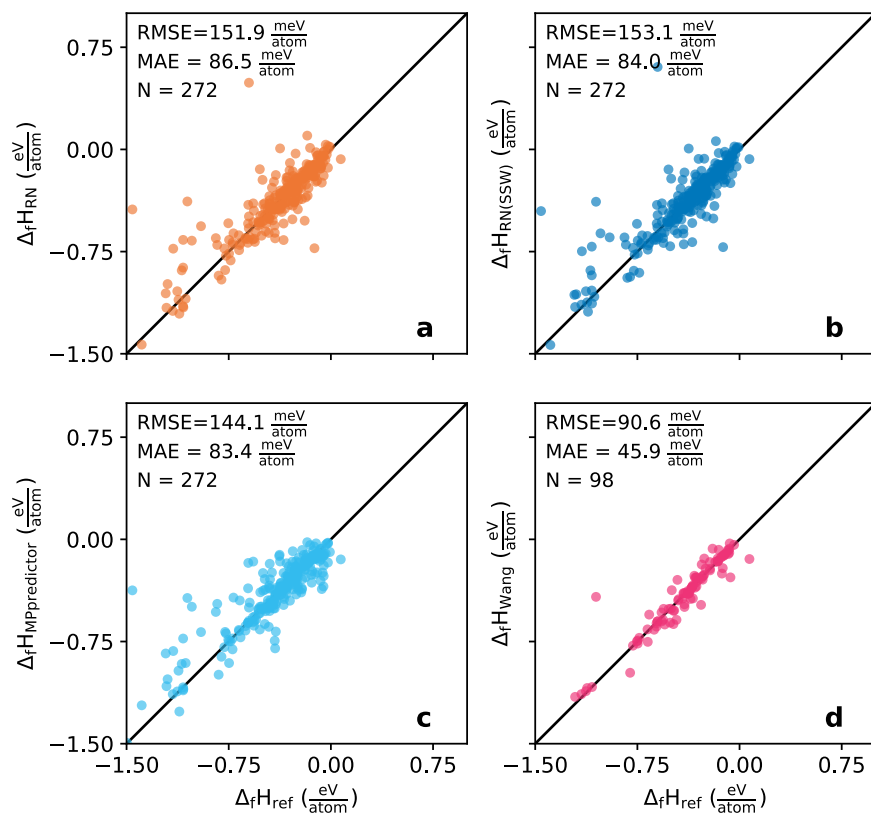
| likely_mpid | Formula                                                                                     | $\Delta_{\text{hull}}H_{\text{MP}}$ | $\Delta_{\text{f}}H_{\text{MP}}$ | $V_{\text{MP}}$ | $V_{\text{ICSD}}$ | $\Delta_{\%}V$ | ICSD ID |
|-------------|---------------------------------------------------------------------------------------------|-------------------------------------|----------------------------------|-----------------|-------------------|----------------|---------|
| mp-13862    | B <sub>13</sub> P <sub>2</sub>                                                              | 0.96                                | 0.65                             | 10.8            | 8.15              | 32.0           | 23923   |
| mp-5282     | Ag <sub>2</sub> C <sub>2</sub> O <sub>4</sub>                                               | 0.22                                | -1.10                            | 14.0            | 12.4              | 12.9           | 109601  |
| mp-560574   | Hg(CN) <sub>2</sub> · HgO                                                                   | 0.49                                | 0.11                             | 20.6            | 18.7              | 10.4           | 25708   |
| mp-642662   | SrCl <sub>2</sub> · H <sub>2</sub> O                                                        | 0.29                                | -1.80                            | 19.0            | 16.7              | 13.8           | 60883   |
| mp-653562   | Ag <sub>3</sub> I(NO <sub>3</sub> ) <sub>2</sub>                                            | 0.23                                | -0.41                            | 17.2            | 15.4              | 11.8           | 14248   |
| mp-696849   | NaBO <sub>2</sub> · 4 H <sub>2</sub> O                                                      | 0.33                                | -1.29                            | 11.1            | 9.4               | 18.0           | 34652   |
| mp-699496   | K <sub>2</sub> Cd(SO <sub>4</sub> ) <sub>2</sub> · 2 CdSO <sub>4</sub> · 5 H <sub>2</sub> O | 0.46                                | -1.24                            | 13.2            | 11.6              | 14.0           | 96214   |
| mp-707471   | ZnI <sub>2</sub> · 4 NH <sub>3</sub>                                                        | 0.59                                | 0.06                             | 15.6            | 13.4              | 16.2           | 78972   |
| mp-707786   | Ba(OH) <sub>2</sub> · 8 H <sub>2</sub> O                                                    | 0.61                                | -0.81                            | 11.4            | 8.6               | 32.2           | 33741   |
| mp-721035   | Cu(NH <sub>3</sub> ) <sub>4</sub> (NO <sub>3</sub> ) <sub>2</sub>                           | 0.88                                | -0.04                            | 11.9            | 8.9               | 33.9           | 369     |
| mp-721144   | Na <sub>4</sub> P <sub>2</sub> O <sub>7</sub> · 10 H <sub>2</sub> O                         | 0.37                                | -1.31                            | 5.7             | 4.7               | 19.5           | 15389   |
| mp-836079   | Mn <sub>2</sub> (CO) <sub>10</sub>                                                          | 0.61                                | -0.79                            | 8.8             | 7.8               | 13.4           | 415718  |

Supplementary Table 7: Compounds for which no prediction could be made.

| likely_mpid | Formula                                                               |
|-------------|-----------------------------------------------------------------------|
| mp-1201095  | H <sub>4</sub> Fe(CN) <sub>6</sub>                                    |
| mp-1427     | Lu <sub>2</sub> O <sub>3</sub>                                        |
| mp-24445    | Nd <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> · 8 H <sub>2</sub> O  |
| mp-571448   | Cs <sub>2</sub> PuCl <sub>6</sub>                                     |
| mp-642318   | (Ba(NO <sub>2</sub> ) <sub>2</sub> ) <sub>2</sub> · TiNO <sub>2</sub> |

Table 7 lists the compounds for which no reactions could be found. For Lu<sub>2</sub>O<sub>3</sub>, Nd<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> · 8 H<sub>2</sub>O and Cs<sub>2</sub>PuCl<sub>6</sub>, the exclusion of compound-sets through data-validation removed all possible reactions. For H<sub>4</sub>Fe(CN)<sub>6</sub> and (Ba(NO<sub>2</sub>)<sub>2</sub>)<sub>2</sub> · TiNO<sub>2</sub> no reactions were found at all with the current settings.

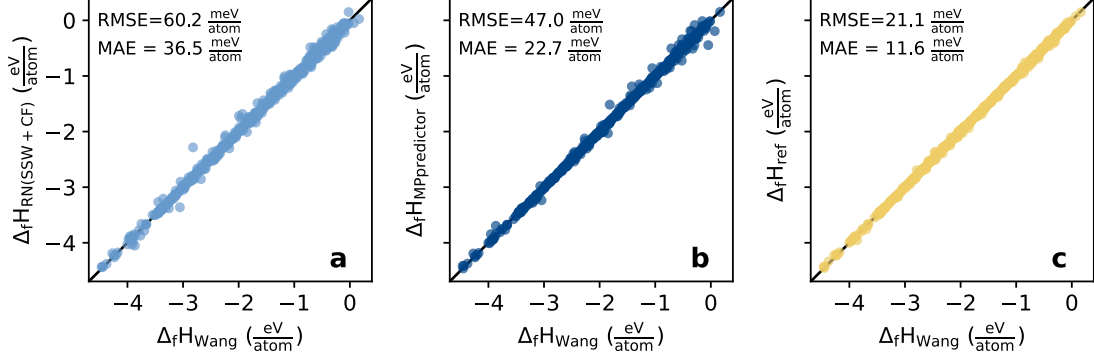
### 1.3 The intermetallic compounds



Supplementary Figure 2: Formation enthalpies predicted using several methods compared to the experimental reference data for the intermetallic compounds: **(A)**  $r^2\text{SCAN}(\text{GGA}/\text{GGA}+\text{U})\text{-RN}$  predictions. **(B)**  $r^2\text{SCAN}(\text{GGA}/\text{GGA}+\text{U})\text{-RN(SSW)}$  predictions. **(C)** MPpredictor predictions. **(D)** Wang experimental data. RMSE is the root mean squared error, MAE is the mean absolute error and N is the number of compounds considered.

We define an intermetallic compound as a solid phase containing two or more metallic elements. Metalloids are included in the definition of metallic elements. For predictions of the intermetallics all intermetallics were included in the reaction space. Figure 2 shows some analysis of the intermetallic compounds in the context of other methods/data-sets. Only the SSW variant is shown because the chemistry-guided filter has not been optimized to account for intermetallic chemistry.

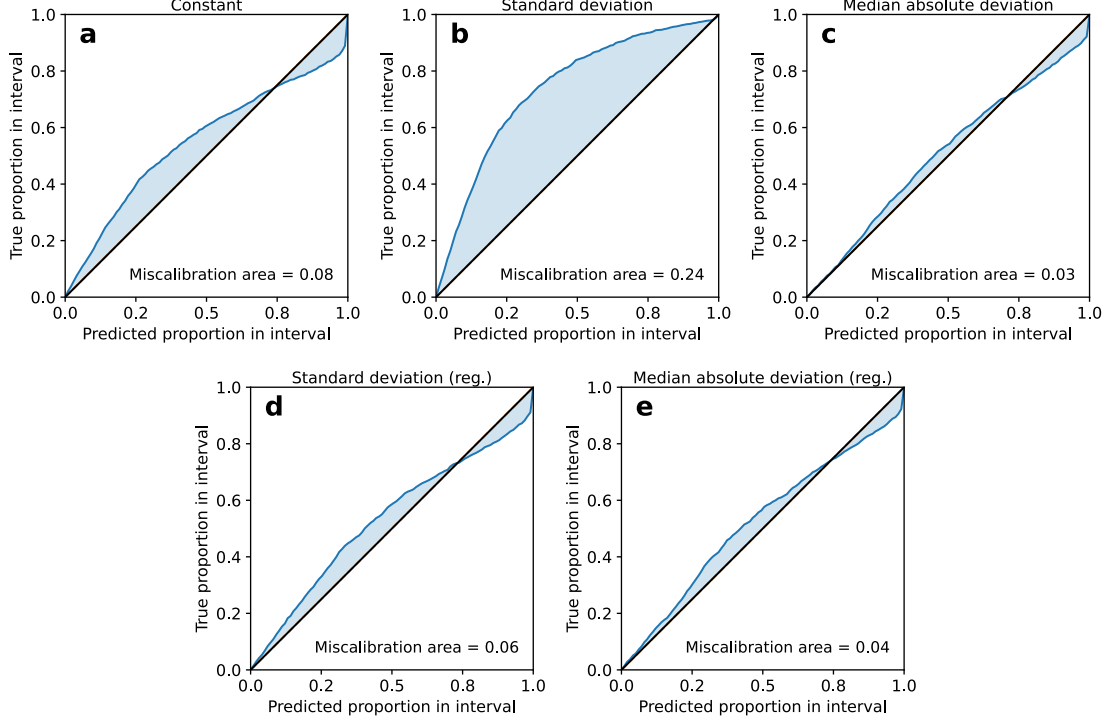
## 2 Comparison with Wang et al. experimental data



Supplementary Figure 3: Formation enthalpies predicted using several methods compared to experimental data from Wang [1] for 586 compounds part of the Table 1 data-sets: **(A)**  $r^2\text{SCAN}(\text{GGA}/\text{GGA}+\text{U})\text{-RN}(\text{SSW}+\text{CF})$  predictions. **(B)** MPpredictor predictions. **(C)** Reference data. RMSE is the root mean squared error, MAE is the mean absolute error. Note that outliers have been removed (Table 1 and Table 2 and this data does not contain the intermetallics).

In Figure 3 the Wang experimental dataset[1] is considered against other data. MPpredictor performs better than  $r^2\text{SCAN}(\text{GGA}/\text{GGA}+\text{U})\text{-RN}(\text{SSW}+\text{CF})$  when compared to the Wang data. In this comparison it seems quite likely that MPpredictor benefits from having been trained on the Wang data. This is further demonstrated when we compare the NBS experimental data to the Wang data; a non-negligible discrepancy is seen. This emphasizes that the reference data chosen is an important factor to consider when using, testing and studying a predictive method.

### 3 Uncertainty estimation



Supplementary Figure 4: Calibration plots of the uncertainty estimators used in the paper for the estimated absolute uncertainties (predicted) against the observed absolute errors (true) for 1550  $\Delta_f H_{r^2\text{SCAN}(GGA/GGA+U)-RN(SSW+CF)}$  predictions. The miscalibration area is represented by the shaded area.

Another correction to the predicted uncertainties can be obtained by minimization of the calibration area via a uniform scaling of all the predicted uncertainties (calibration) [43]. The calibration (cal.) provides significant improvement to the evaluation measures for the SD, but only slight improvement for the MAD. Finally, we can test the serial application of first regression and then calibration (reg.+cal.). Notably, this seems to improve on the individual corrections from regression and calibration w.r.t the MAE, while the miscalibration area is only altered very slightly, in comparison.

Supplementary Table 8: Evaluating uncertainty estimators for 1550  $\Delta_f H_{\text{r}^2\text{SCAN}(\text{GGA}/\text{GGA}+\text{U})-\text{RN}(\text{SSW}+\text{CF})}$  predictions. The mean absolute error of the estimated uncertainty (MAEU) w.r.t the true absolute errors ( $|\Delta_f H_{\text{RN}} - \Delta_f H_{\text{ref}}|$ ) is given in meV atom<sup>-1</sup>; a value of zero is ideal. The miscalibration area quantifies systematic under or over-confidence of the predicted uncertainty; a value of zero is ideal. The proportion of reference values within a prediction  $\pm 1$ -,  $\pm 2$ - and  $\pm 3$ -times the predicted uncertainty are given under  $\pm 1\text{U}$ ,  $\pm 2\text{U}$  and  $\pm 3\text{U}$ , respectively.  $\rho$  is Spearman’s rank correlation coefficient of the predicted uncertainty and the true absolute errors; a value of one is ideal. The abbreviation reg. means regressed and cal. means calibrated.

| Method          | MAEU | Miscalibration area | $\pm 1\text{U}$ | $\pm 2\text{U}$ | $\pm 3\text{U}$ | $\rho$ |
|-----------------|------|---------------------|-----------------|-----------------|-----------------|--------|
| Constant        | 27.5 | 0.08                | 71%             | 85%             | 92%             | -      |
| SD (cal.)       | 24.2 | 0.04                | 64%             | 84%             | 92%             | 0.49   |
| MAD (cal.)      | 23.2 | 0.03                | 66%             | 87%             | 93%             | 0.57   |
| SD (reg.+cal.)  | 22.3 | 0.05                | 65%             | 82%             | 91%             | 0.49   |
| MAD (reg.+cal.) | 21.1 | 0.03                | 64%             | 85%             | 92%             | 0.57   |

# References

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