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Oxygenation as a driver of the Great Ordovician Biodiversification Event

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2 Biodiversification Event

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5 **Supplementary Information**

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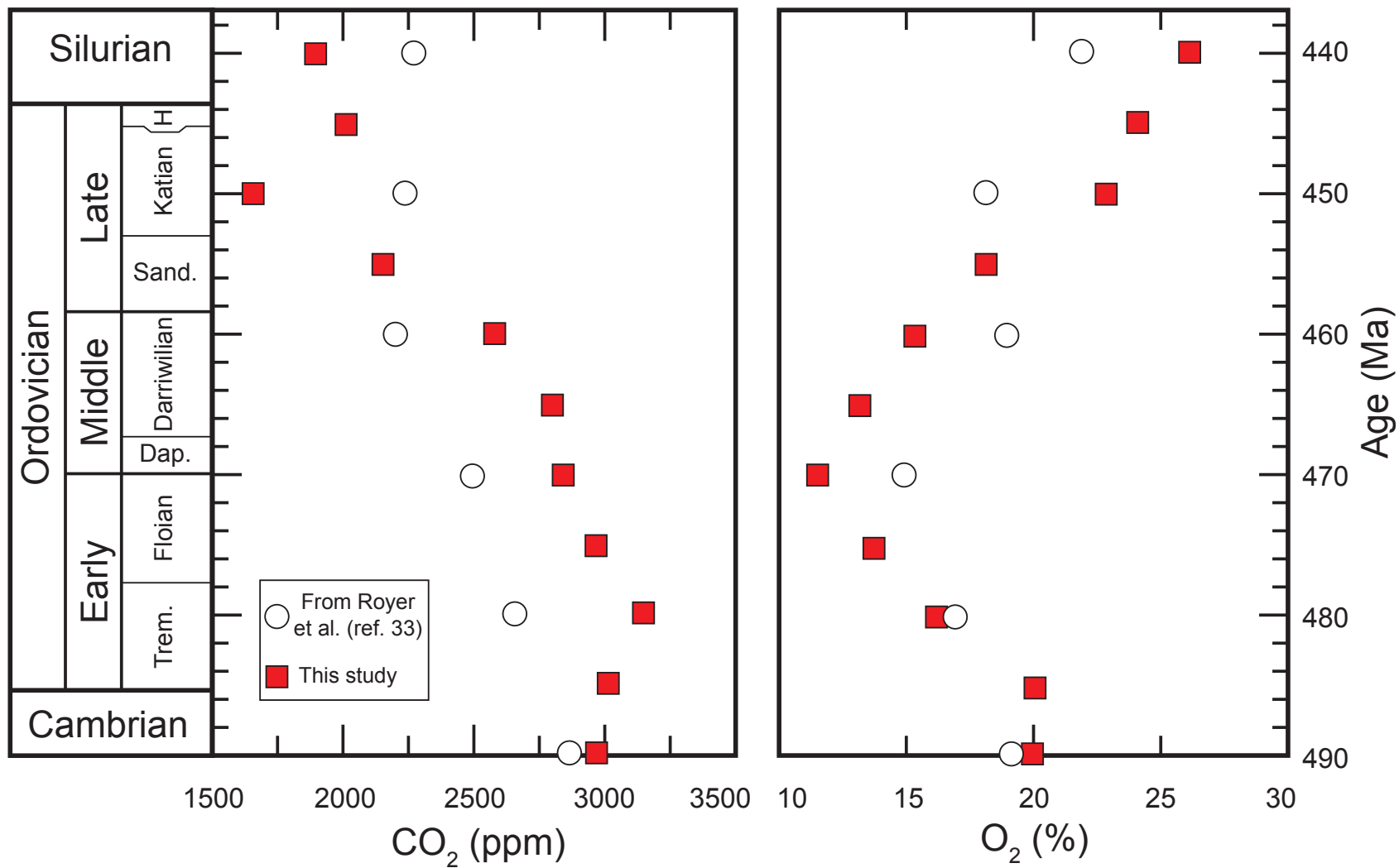
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9 **Model sensitivity testing**

10 *GEOCARB*

11 New O₂ estimates using the GEOCARB model are broadly similar to previous
12 GEOCARB results³³ with the important exception of an O₂ increase during the Middle
13 and Late Ordovician (Fig. S1). We established reasonable uncertainties for the three new
14 isotope datasets used in our standard GEOCARB run. For each dataset (Sr, C, and S) and
15 the other 65 inputs, we reduce the measured uncertainties (Fig. 1) in all time bins
16 proportionately, resampling only one dataset at a time (and leaving all other inputs fixed),
17 until failure rates in all time bins are less than 5%, following ref. 33. Our reported
18 uncertainties (light gray band in Fig. 2) are likely too large because the model failure rate
19 in the fully propagated run is high (~85%), meaning that many combinations of input
20 value choices are incompatible with the model³³. Uncertainties for Sr, C, and S required
21 reductions of 0, 83, and 88%, respectively, to achieve <5% failure for all time bins. One
22 exception, however, is that time bin 465 Ma consistently has failure rates of 65–99%
23 caused by a low $\delta^{13}\text{C}$ mean value of -1.42‰ in the 470 Ma time bin. Increasing $\delta^{13}\text{C}$ to
24 $\sim -1.2\text{‰}$ causes model failure rates to be <5% for all time bins. After this $\delta^{13}\text{C}$
25 adjustment, when all inputs are resampled simultaneously, the highest failure rates in the
26 standard GEOCARB run occur in the 465 Myr time bin (89%; Table S1). Because
27 failure rates are generally high in our standard run, we carry out 100,000 Monte Carlo
28 simulations in order to ensure > 10,000 successful individual runs.

29 The sulfur isotope data used in the standard GEOCARB run are the least
30 constrained with respect to data density and age resolution relative to the $\delta^{13}\text{C}_{\text{carb}}$ and
31 $^{87}\text{Sr}/^{86}\text{Sr}$ records (Fig. 1). Our standard GEOCARB run (Fig. 2) uses high-resolution $\delta^{34}\text{S}$



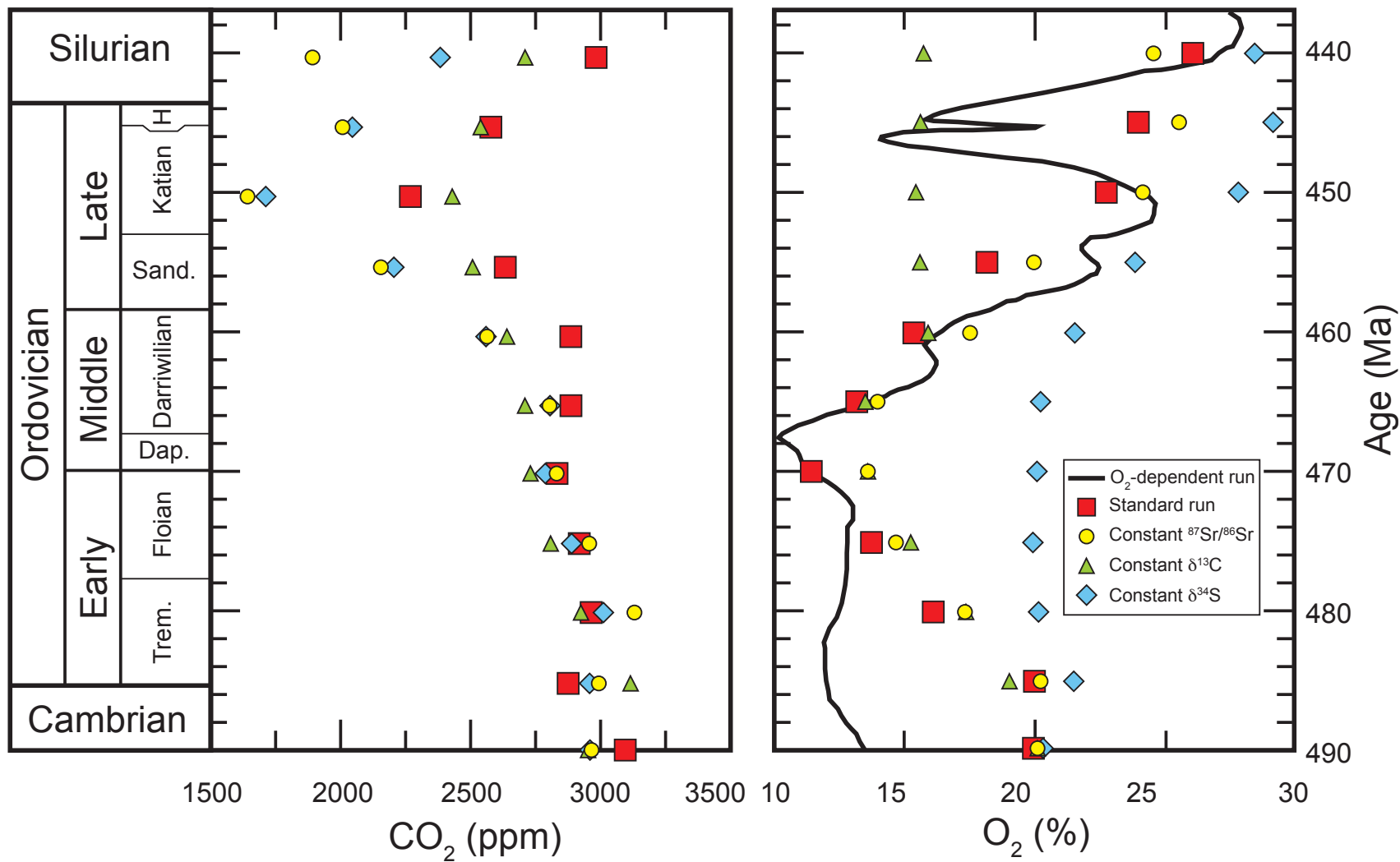
Edwards et al., Figure S1

data from a single stratigraphic section (Shingle Pass, NV⁴⁹) with values that are similar to what is thought to be the global $\delta^{34}\text{S}$ trend⁴⁷.

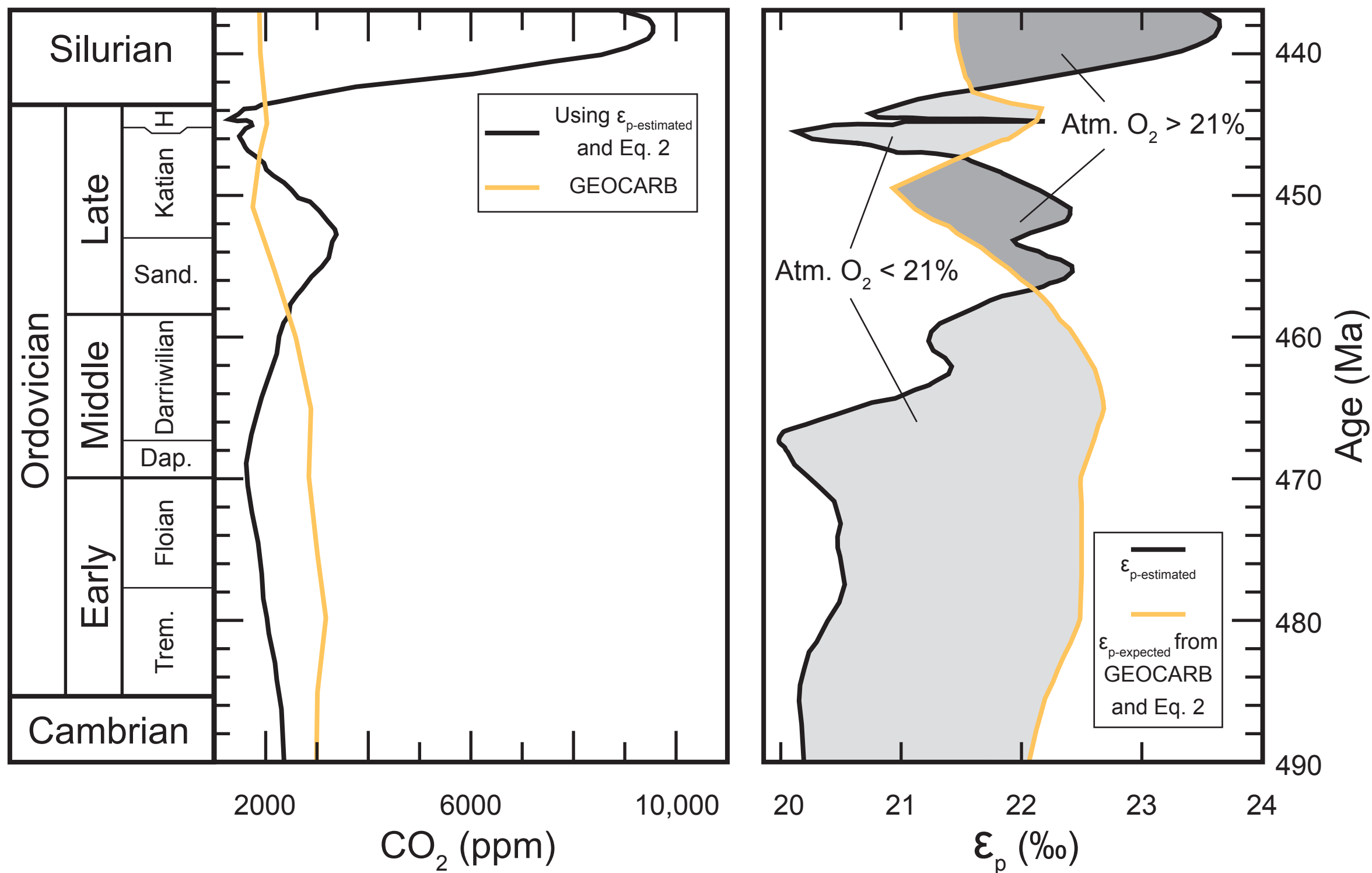
Sensitivity testing of GEOCARB by holding each updated dataset ($\delta^{13}\text{C}$, $\delta^{34}\text{S}$, and $^{87}\text{Sr}/^{86}\text{Sr}$) constant individually shows that constant $\delta^{34}\text{S}$ values result in overestimates of O_2 during the Middle Ordovician compared to other model runs, but the Middle–Late Ordovician increase is still evident (Fig. S2). Our model run with constant $\delta^{13}\text{C}$ does estimate that atmospheric O_2 was <15% during the Middle Ordovician, but in this case the Middle–Late Ordovician oxygenation is not evident. These sensitivity tests suggest that the combination of observed $\delta^{13}\text{C}$ and $\delta^{34}\text{S}$ trends does in fact play an important role in O_2 estimates where neither input alone is responsible for driving the GEOCARB O_2 estimates. Likewise, holding each of these parameters constant does not appreciably affect CO_2 estimates (Fig. 2), which is particularly important because $\delta^{13}\text{C}_{\text{carb}}$ data are used in both approaches and does not appear to affect the non-independence of the models. In the model's base run the GEOG parameter increases during the time of the CO_2 drop, which accounts for changes in temperature relative to today³³ and should impact silicate weathering rates (a CO_2 sink). Holding this term constant at 490 Ma does decrease the magnitude of the CO_2 drop, suggesting that weathering rates and temperature are largely what are driving this CO_2 drop and not variations in the updated $\delta^{13}\text{C}$ and $^{87}\text{Sr}/^{86}\text{Sr}$ records.

Photosynthetic fractionation effect model

Using the $\epsilon_{\text{p-estimated}}$ trend and Eq. 3, atmospheric O_2 levels can be calculated after removing the effect of variations in CO_2 ($\epsilon_{\text{p-expected}}$) and scaled using an empirically



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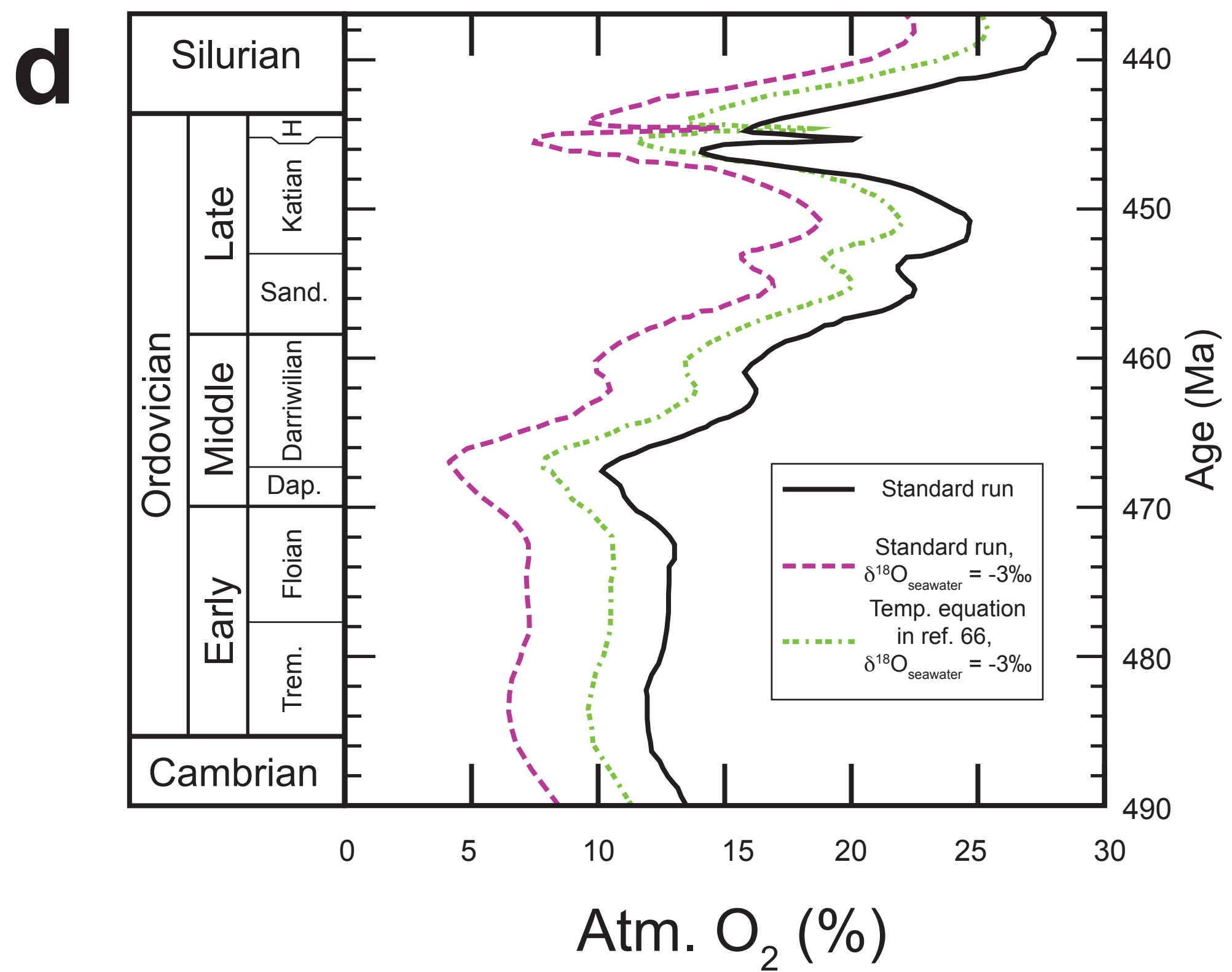
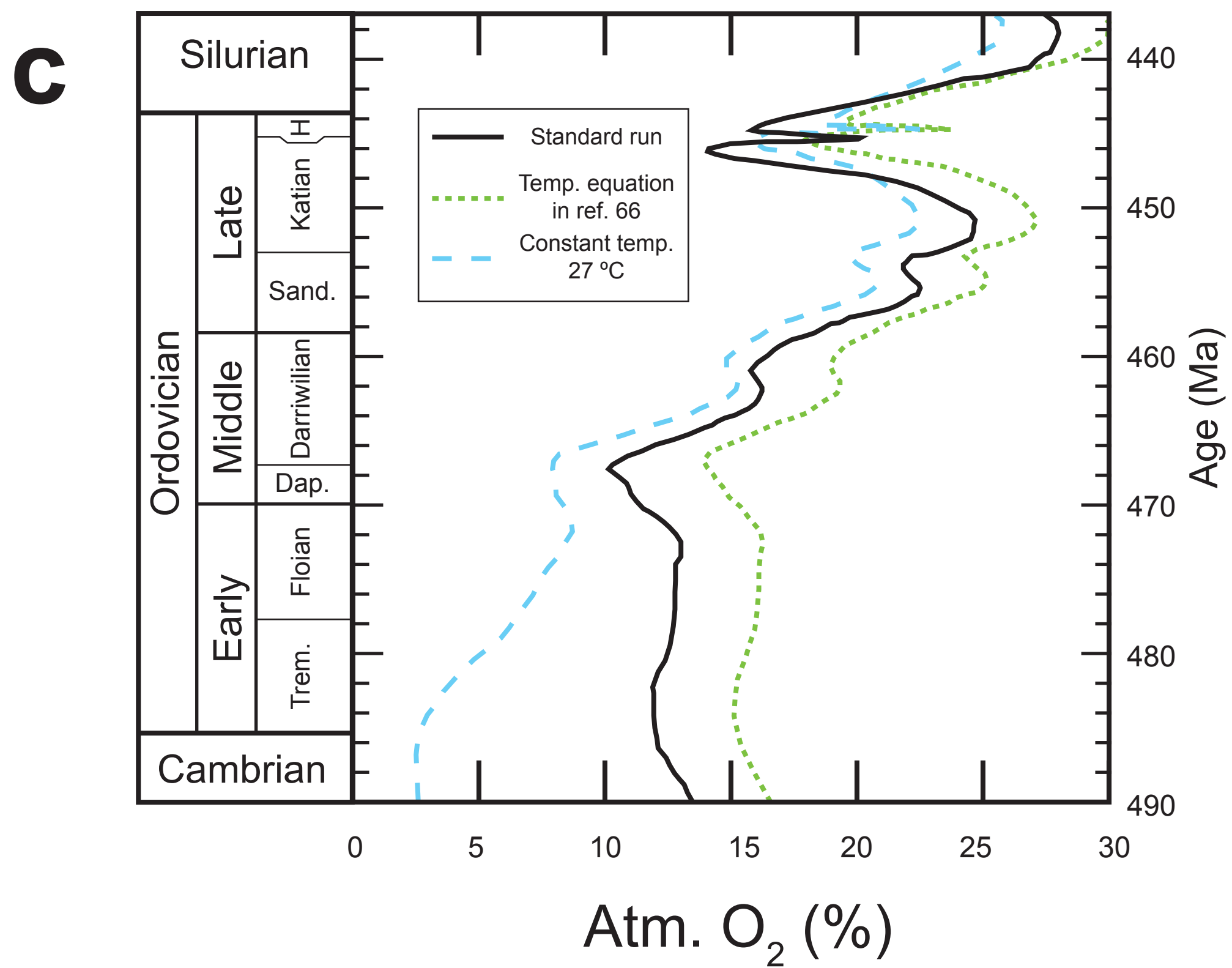
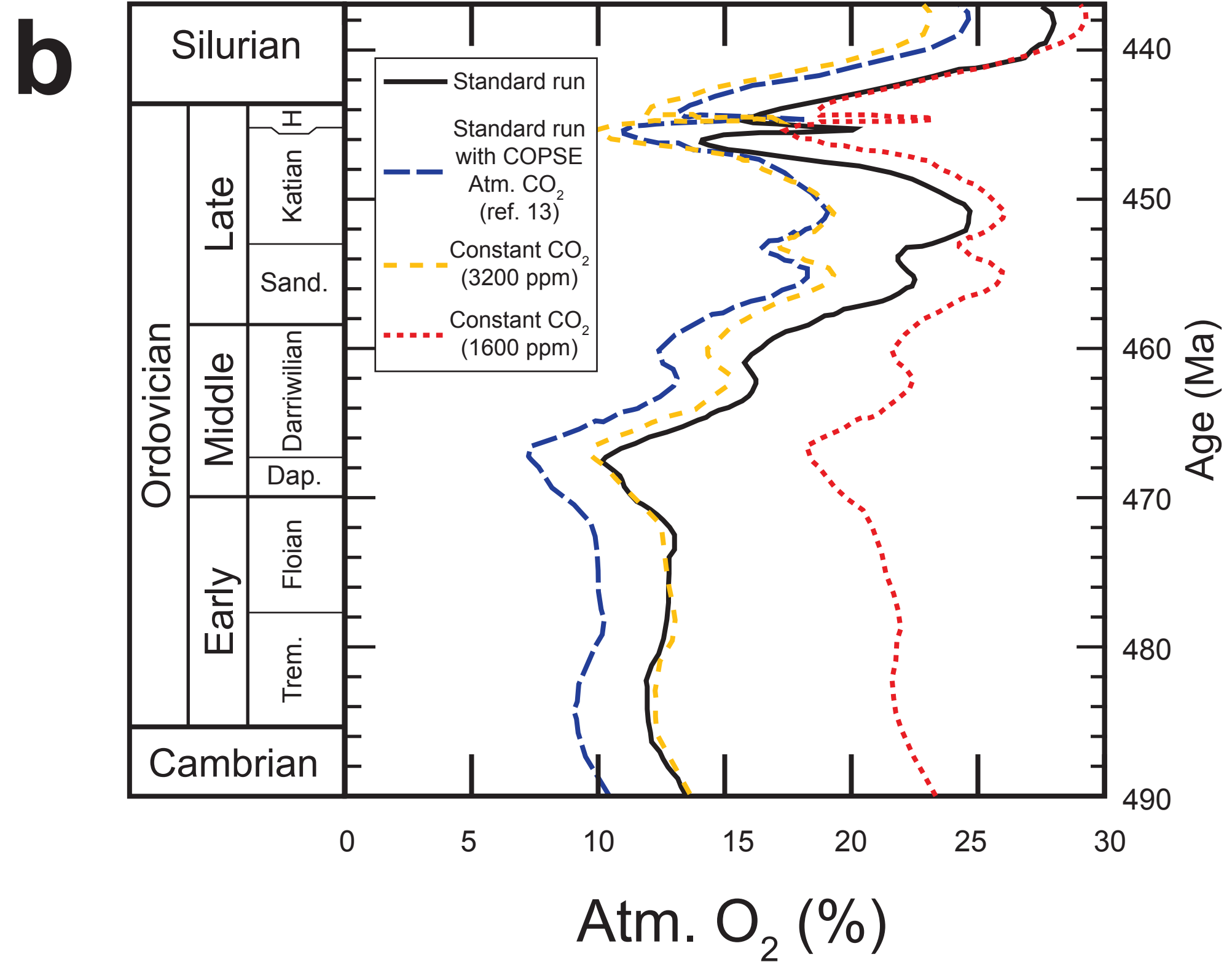
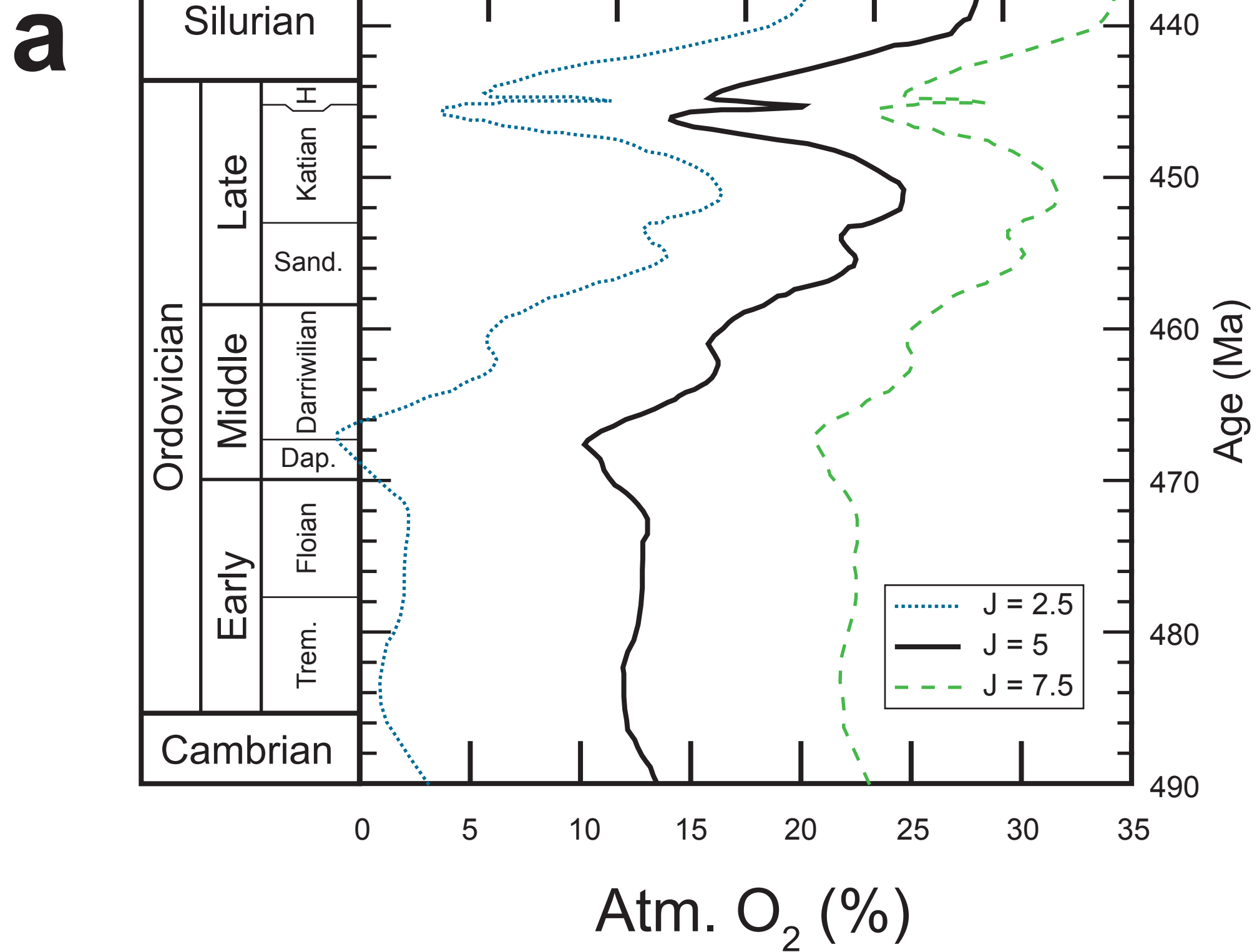


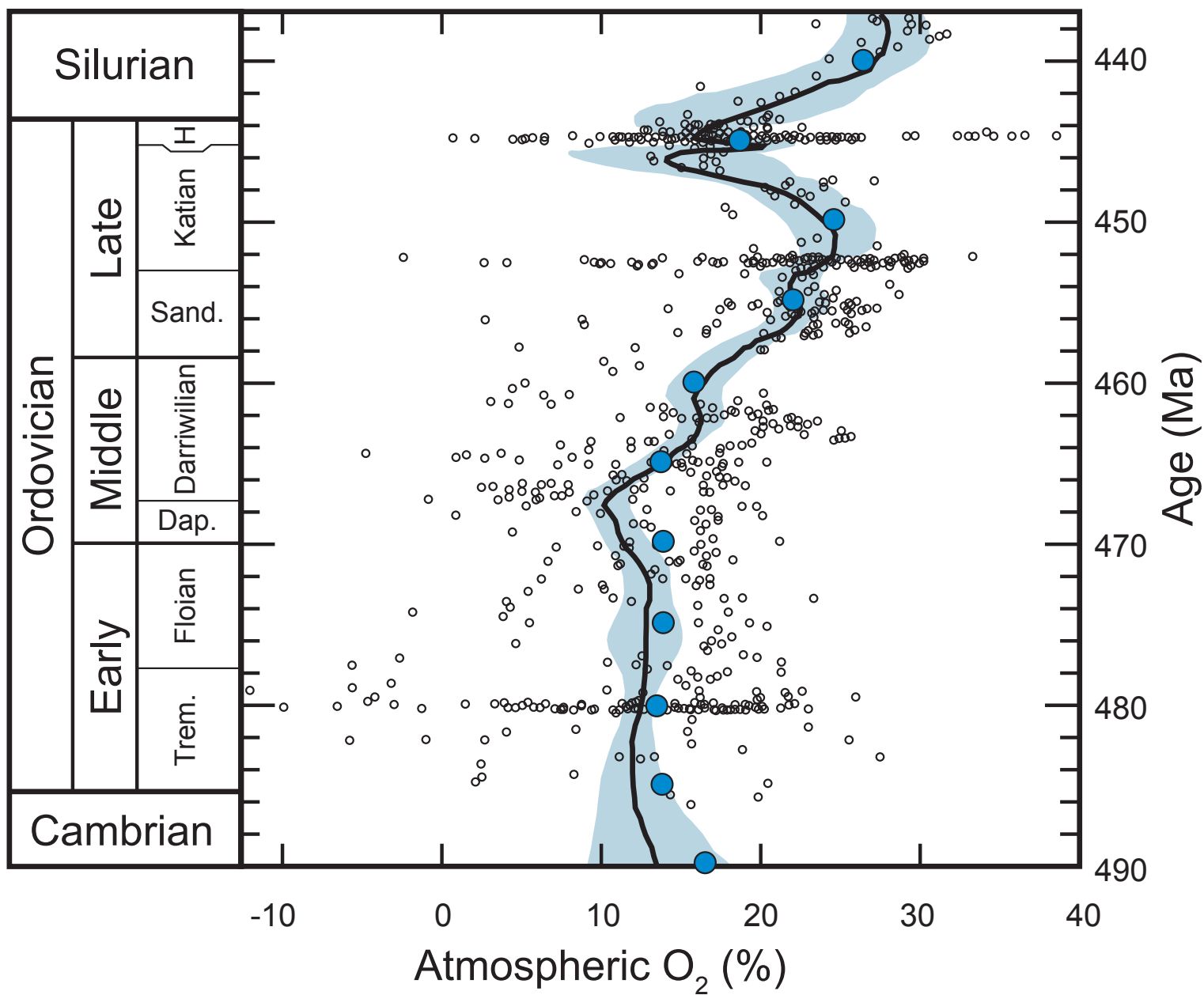
Edwards et al., Figure S3

55 derived value (J). The parameter J describes the dependence of isotopic fractionation on
56 estimated O_2 . Using $J = 2.5$, as in ref. 22, yields very low O_2 values $<5\%$ during the
57 Early–Middle Ordovician (~ 485 – 467 Myr) and very high O_2 ($>25\%$) just 15 Myr later
58 (Fig. S4A). Saltzman et al. (ref. 58) also found that small J values yield unrealistic
59 atmospheric O_2 values during the late Cambrian. When a larger value is used ($J = 7.5$),
60 the same O_2 patterns are present but they are dampened (Fig. S4A). We favor using a
61 value for $J = 5$ because it is close to what other studies have used to model Paleozoic
62 atmospheric O_2 (2.5 – 5)^{16,58,59} and because it does not lead to highly improbable swings in
63 estimated O_2 .

64 Despite using a similar J value, individual calculated O_2 values exhibit significant
65 scatter between successive measurements, including some negative O_2 values (Fig. S5).
66 We pass a LOESS smoothing line throughout all calculations to capture the overall O_2
67 trend. When O_2 calculations are binned in a similar resolution to GEOCARB estimates
68 (blue circles in Fig. S5), the Hirnantian O_2 drop is still present, suggesting that either the
69 O_2 -dependent fractionation approach is capturing a short-term change in the carbon cycle
70 not capable of being resolved in the GEOCARB model (consistent with the <1 Myr
71 estimated duration of the Hirnantian $\delta^{13}C$ excursion), or this drop is an artifact related to
72 uncertainties in $\delta^{13}C$ correlation across several datasets with poor biostratigraphic age
73 resolution.

74 A close examination of the data used to derive O_2 estimates using the
75 photosynthetic fractionation approach shows that a small, but resolvable increase in
76 $\delta^{13}C_{org}$ is driving lower $\epsilon_{p-estimated}$ values (Fig. S6). $\delta^{13}C$ data collected from Anticosti
77 Island^{42,43}, calibrated to the 2012 Geologic Time Scale using graptolite and chitinozoan



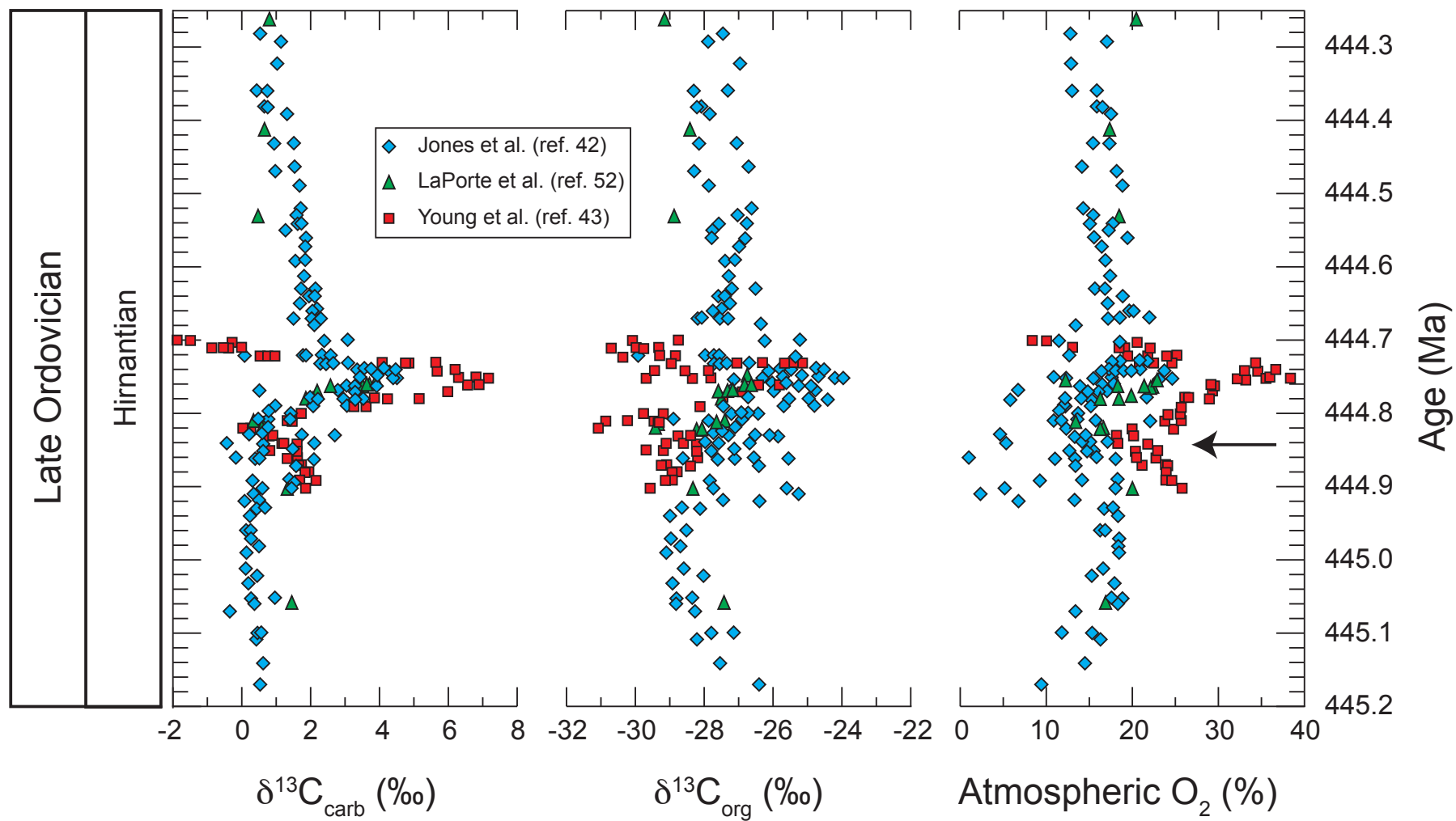


Edwards et al., Figure S5

biozones, both indicate that O₂ decreased prior to the HICE event. O₂ estimates based on $\delta^{13}\text{C}$ data from Nevada⁵², using just graptolite biozones for age calibration, also shows an O₂ drop prior to the HICE event. However, this drop occurs prior to the O₂ drop using Anticosti Island data (Fig. S6) and does not drop much below near-modern O₂ levels (21%). It may be possible that all three sections record the same O₂ drop prior to the Hirnantian glaciation and mass extinction event, but until more highly resolved biostratigraphic information becomes available from these (and other) Hirnantian sections we only want to point out the possibility that an oxygen crisis may have occurred, which we hope will encourage future investigation into this time period.

Modeling atmospheric O₂ during other periods of Earth history: limitations and directions for continued study

The models used in this study could be used to explore variations in atmospheric O₂ levels during other times of Earth history, such as the Mesozoic ocean anoxic events (OAEs). During Mesozoic OAEs $\delta^{13}\text{C}$ and $\delta^{34}\text{S}$ have been shown to fluctuate over the course of 1–2 Myr, interpreted to transiently record marine anoxia or euxinia in shallow environments^{60,61}. These isotopic excursions would be averaged out to a single input parameter in a 5 Myr time bin of GEOCARB, which is unlikely to show any significant change in O₂ and prevents a meaningful comparison of O₂ estimates between the models used here. Conceptually, the photosynthetic effect approach has the potential to record changes in atmospheric O₂, provided high-resolution paired $\delta^{13}\text{C}$ measurements and atmospheric CO₂ estimates exist. Nevertheless, given that these OAEs occurred during times of major change to biogeochemical cycles, it is possible that local algal



Edwards et al., Figure S6

assemblages were also physiologically affected (if not completely replaced by new populations). Thus, caution should be used when interpreting any short-term changes in $\epsilon_{p\text{-estimated}}$ values as they may instead record temporary changes to biological controls (growth rates, cell size, new dominant groups that have unique ‘vital effects’) instead of atmospheric O₂ variation.

Applying this photosynthetic fractionation approach to the post-Eocene interval is also not likely to yield atmospheric O₂ levels similar to GEOCARB (or other independent means to estimate O₂ levels). The alkenone community has struggled to use ϵ_p measurements from Cenozoic coccolithophore biomass as a proxy for paleoclimate trends, without factoring in cell size and vital effects^{62,63}. These algae, among others, are known to utilize carbon-concentrating mechanisms (CCMs) during periods of low atmospheric CO₂, which discriminate ¹²C differently compared other algal groups and will yield different ϵ_p values under similar CO₂ and O₂ levels. Although Cenozoic strata contain high-resolution $\delta^{13}\text{C}$ trends from well-preserved bulk and compound-specific organic molecules, because CCMs were likely used by some algae included in bulk records $\epsilon_{p\text{-estimated}}$ values are not likely to strictly record changes in atmospheric CO₂ or O₂. Thus, using $\epsilon_{p\text{-estimated}}$ values from the post-Eocene to modern in our photosynthetic fractionation approach will not produce similar O₂ estimates with GEOCARB.

Error analysis and error propagation calculations

Error analysis and uncertainties for atmospheric CO₂ and O₂ values using GEOCARB (Table S1) are calculated using the original code, discussed in detail in Royer et al. (ref. 33). Uncertainties for O₂ estimates using the O₂-dependent fractionation effect approach

are determined using standard error propagation equations for each calculation. Table S5 shows an example of error propagation calculations, which includes a data value and its error estimate (measured or calculated using a particular error propagation calculation; the example shown is the first row in Table S2). Error propagation calculations were compared against Monte Carlo simulations (bottom portion of table) to check that reasonable uncertainties were calculated. Error analysis using Monte Carlo simulations was done by generating 10,000 random values using the inverse of the cumulative normal distribution (NORMINV function in Excel®) and the value's associated uncertainty. The median value of these 10,000 randomly generated values was treated as the calculated value (e.g. 1845 for 'Interpolated $p\text{CO}_2$ (ppm)'), and 1σ was calculated by subtracting the median value from the 84th percentile value. The 95% confidence interval of the smoothed LOESS line in Figure 2 was produced using 'predict function' in R Studio®, which does not depict the error associated with each data point from Table S2. This is not meant to dismiss the large uncertainties associated with each value; rather we emphasize the confidence of the long-term trend based on the large dataset used here.

Controls on $\Delta^{13}\text{C}$:

The main control on $\Delta^{13}\text{C}$ is the biological fractionation effect (ϵ_p), but temperature during carbonate formation and secondary heterotrophy of organic matter (Δ_2) can also play an important role (Fig. S7; ref. 25). Temperature has a strong effect on the isotopic fractionation (0.12‰/°C) between aqueous CO_2 ($[\text{CO}_2]_{\text{aq}}$) in the surface ocean and the dissolved inorganic carbon (DIC) reservoir (represented by the term Δ_{carb} in Fig. S7). There is a small temperature effect (0.01‰/°C) associated with the formation of

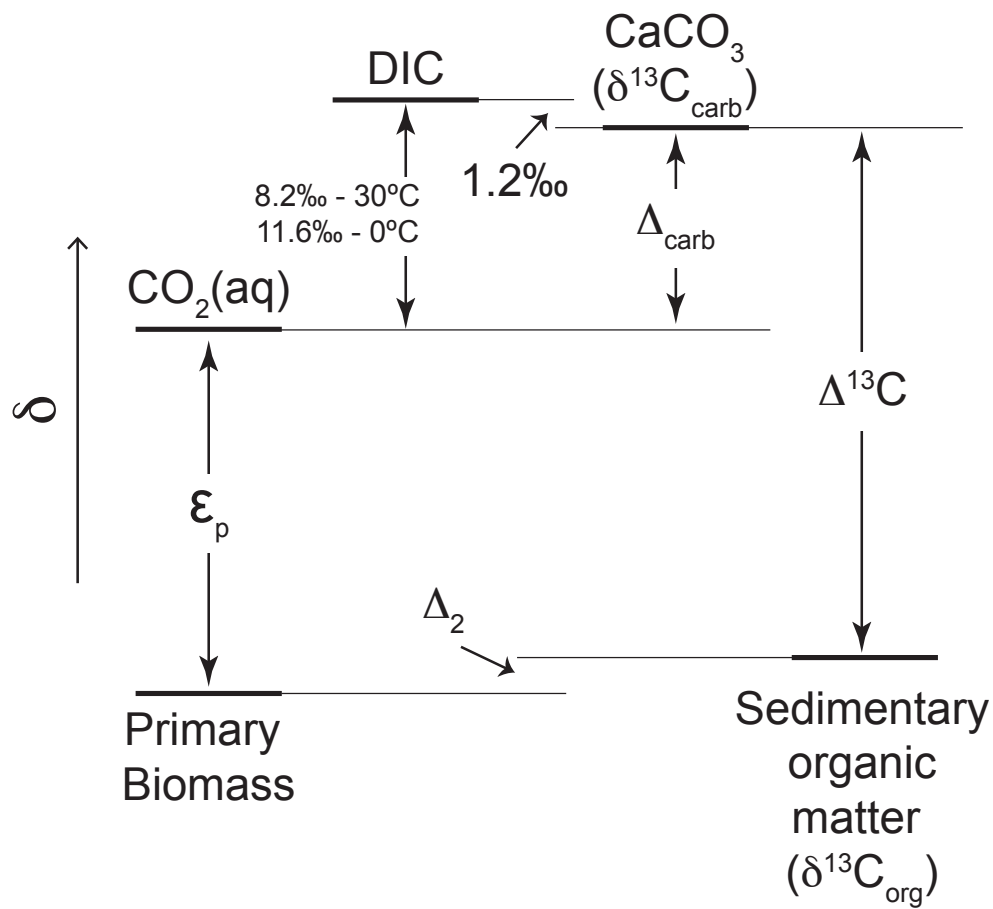
carbonate from seawater, but this effect is outweighed by the 1‰ depletion between carbonate formation from the DIC²⁵. In most marine environments Δ_{carb} can vary by up to 3.4‰ due to temperature effects alone between 30 °C (8.2‰) and 0 °C (11.6‰).

Secondary processes (Δ_2) associated with the metabolic breakdown (heterotrophy) of organic matter from primary producers can also influence $\Delta^{13}\text{C}$. Studies of modern environments underlying upwelling zones show that sedimentary organic matter can be enriched in ^{13}C by as much as 3.5‰ compared to primary biomass, with a typical mean value of +1.5‰ based on observations from a range of oxic environments (ref. 25 and references therein). Negative Δ_2 values can result when a portion of the original biomass is replaced by biomass from anaerobic chemoautotrophic organisms. This difference can be as much as 3‰ between the $\delta^{13}\text{C}$ of biomass from photosynthesizers ($\delta^{13}\text{C}_{\text{org}}$: $\sim -22.5\text{‰}$) versus sulfur-oxidizers ($\delta^{13}\text{C}_{\text{org}}$: $\sim -25.5\text{‰}$) as measured in the modern dysoxic Black Sea⁶⁴. However, our study areas are interpreted as well-oxygenated shallow marginal and epicontinental seas (ref. 25 and refs. therein); thus, we exclude the possibility for negative Δ_2 values. Together, we do not consider that changes in the secondary alteration of organic matter over the duration of the Early–Middle Ordovician can explain a 3‰ increase in $\Delta^{13}\text{C}$ (26 to 29‰), but we correct all $\delta^{13}\text{C}_{\text{org}}$ values by -1.5‰ to more accurately estimate ϵ_p (eq. S3, see below). Hayes et al. (ref. 25) estimate that a ^{13}C -enrichment of 1.5‰ of original biomass (δ_p) is representative of oxic marine settings up to 100 cm deep into the sediment profile where a balance of secondary processes favors a slight ^{13}C enrichment of bulk organic matter. Using these assumptions about secondary microbial processes and carbonate formation, we can use paired $\delta^{13}\text{C}_{\text{carb}}$

and $\delta^{13}\text{C}_{\text{org}}$ measurements to estimate the $\delta^{13}\text{C}$ of dissolved CO_2 ($\delta^{13}\text{C}_{\text{CO}_2}$) and primary biomass ($\delta^{13}\text{C}_{\text{biomass}}$), respectively, in order estimate ϵ_p (Fig. S7).

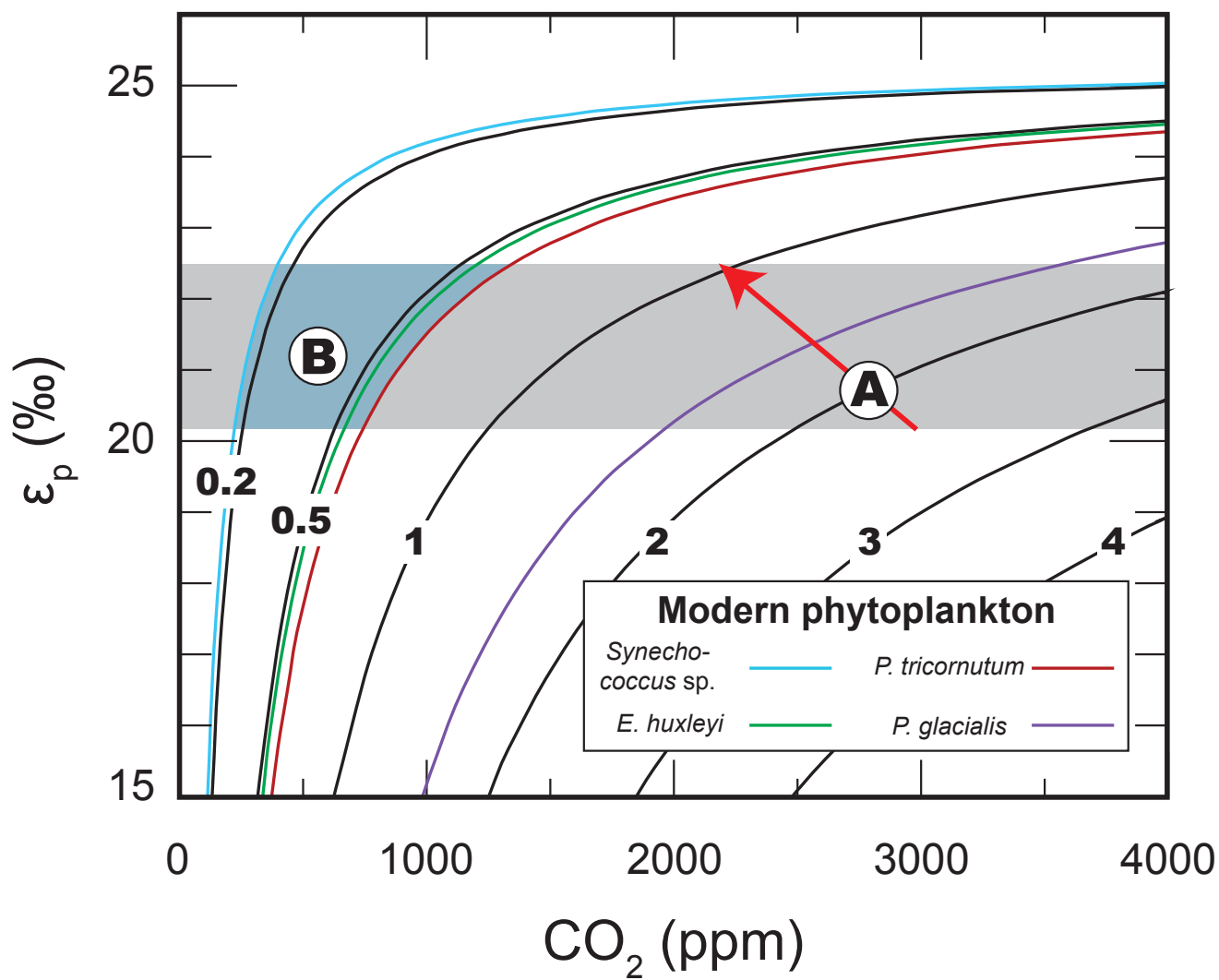
Determining ϵ_p values

Experimental work with modern phytoplankton shows that ϵ_p can be affected by biological effects and CO_2 partial pressure²². Cellular growth rates (μ) and volume/surface area (V/S) ratios influence ϵ_p such that the highest ϵ_p values are produced with slow growth rates or small V/S ratios. If these biological factors were solely responsible for the long-term increase in ϵ_p from 20.2‰ to 22.5‰ (Fig. S3), using Eq. 1 in Naafs et al. (ref. 31) would require the μ -V/S product to decline from 2.45 to 1.79 (where atmospheric CO_2 concentrations decrease from 2978 ppm to 2148 ppm during the increase in ϵ_p from 20.2‰ to 22.5‰) given the range of atmospheric CO_2 estimates from GEOCARB (A in Fig. S8). A typical μ -V/S product, however, of small modern marine algae only varies between 0.54 and 0.56 in two diatom species to as much as 0.88 in some haptophytes with slow growth rates (ref. 21 and refs. within). Some of the highest observed μ -V/S products (1.1–2.2) occur in highly productive upwelling zones where phytoplankton have large V/S ratios (2.2) but also have high growth rates (i.e. high μ : 0.5–1). Therefore it seems unrealistic that the 2.3‰ increase in ϵ_p was caused by slowing growth rates or decreasing cell sizes from values more than four times those that are typical of most modern environments, which would need to be sustained globally for millions of years during the Early Ordovician before decreasing during the Middle Ordovician.



$$\Delta^{13}\text{C} = \epsilon_p + \Delta_{\text{carb}} - \Delta_2$$

Edwards et al., Figure S7



Edwards et al., Figure S8

Determining Ordovician ϵ_p values

Existing fossil evidence severely under-represents the true diversity of size and shape of Ordovician phytoplankton, but some soft-bodied acritarchs (dinoflagellate cysts) preserved in Silurian limestone are 5 to 10 μm in diameter⁶⁵ and similar in size to modern planktic coccolithophores and diatoms originally studied by Popp et al. (ref. 21). If Ordovician seas were dominantly populated by small phytoplankton forms, of which almost no evidence exists to confirm or refute this, factors such as carbon concentrating mechanisms or other ‘vital effects’ could have affected ϵ_p and $\delta^{13}\text{C}_{\text{org}}$. It is also likely that bulk $\delta^{13}\text{C}_{\text{org}}$ data used herein includes components of bulk organic matter from different algal and bacterial assemblages, including benthic mat-forming algae and cyanobacteria in addition to planktonic forms. However, because this dataset includes data from several ocean basins that overlap in time, we assume that any anomalous ϵ_p -estimated values due to algal assemblages, and not atmospheric O_2 change, will be averaged out by the large dataset and LOESS smoothing line.

Ordovician cooling based on conodont apatite $\delta^{18}\text{O}$:

Trotter et al. (ref. 6) inferred that global Ordovician ocean temperatures cooled based on oxygen isotope ($\delta^{18}\text{O}$) paleothermometry from conodont apatite. Sampled conodonts were collected from several sedimentary basins around the globe and from a range of time bins in the Ordovician and Silurian. Each of these bins represents several million years such that ten discrete time slices comprise the Early–Middle Ordovician. Based on the observed $\delta^{18}\text{O}$ increase, sea surface temperatures were estimated to cool from 41 °C to 30 °C during the Early to Middle Ordovician (Tremadocian–Darriwilian),

then cooling further to ~23 °C by the end-Ordovician glaciation⁶. This cooling occurs at approximately the same time as the observed $\Delta^{13}\text{C}$ increase (Fig. 1). An 11°C drop in surface ocean temperatures from 41 °C would result in a 1.3‰ increase in the fractionation effect between DIC and carbonate formation (see above). This would translate into larger $\Delta^{13}\text{C}$ values (if Δ_2 remained constant) during the start of the cooling trend at the base of the Ordovician. Therefore in order to estimate changes in ϵ_p using the $\Delta^{13}\text{C}$ trend ($\delta^{13}\text{C}_{\text{carb}}$ and $\delta^{13}\text{C}_{\text{org}}$), temperature must be considered to account for the effect on the formation of carbonate as well as the solubility of CO_2 into seawater.

Sensitivity testing using an alternative equation⁶⁶ to calculate paleotemperature using conodont-based $\delta^{18}\text{O}$ data shows a similar O_2 trend as the standard run but with slightly higher O_2 estimates overall (Fig. S4C). This alternate O_2 trend also predicts that oxygenation occurred during the Middle–Late Ordovician. Further sensitivity testing using a constant $\delta^{18}\text{O}$ value for seawater of -3‰ instead of -1‰⁶ used in the standard run shows a similar pattern where the timing of the oxygenation does not change but only the magnitude of O_2 levels (Fig. S4D).

Calculation of ϵ_p

Variations in both atmospheric O_2 and CO_2 can influence the fractionation effect produced by primary producers during photosynthesis^{20,59,67,68}. This is because the RuBisCO enzyme (where the major fractionation effect occurs during carbon fixation) has a dual carboxylase-oxygenase function⁵⁹. Increased atmospheric O_2 can produce larger ϵ_p values as observed in various groups of photosynthetic organisms. To solve for the size of the atmospheric oxygen reservoir we use estimates of ϵ_p using Eq. 1 of the

main text and $\delta^{13}\text{C}_{\text{carb}}$ and $\delta^{13}\text{C}_{\text{org}}$ data from Edwards and Saltzman (ref. 23) and new data (Fig. 1; Table S2) as $\delta^{13}\text{C}$ proxies of the DIC (δ_{d} , in the form of dissolved CO_2) and biomass (δ_{p}). $\delta^{13}\text{C}_{\text{carb}}$ can be used to solve for the $\delta^{13}\text{C}$ of dissolved CO_2 (δ_{d}) from the relationship (from ref. 31):

$$\delta_{\text{d}} = \delta^{13}\text{C}_{\text{carb}} - 1\text{‰} + \varepsilon_{\text{b(a)}} \quad [\text{S1}]$$

where 1‰ ($\pm 0.2\text{‰}$) represents the ^{13}C enrichment associated with the formation of calcite from seawater⁶⁹, and $\varepsilon_{\text{b(a)}}$ = the temperature effect of $\delta^{13}\text{C}$ between dissolved CO_2 and HCO_3^- in seawater⁷⁰. $\varepsilon_{\text{b(a)}}$ is calculated using the equation:

$$\varepsilon_{\text{b(a)}} = 24.12 + 9866/T \quad [\text{S2}]$$

where T = temperature in Kelvin, which we base on temperatures estimated from conodont apatite $\delta^{18}\text{O}$ (Table S2)⁶.

To adjust for secondary alteration of the biomass (δ_{p}), we subtract 1.5‰ from $\delta^{13}\text{C}_{\text{org}}$ values:

$$\delta_{\text{p}} = \delta^{13}\text{C}_{\text{org}} - 1.5\text{‰} \quad [\text{S3}]$$

based on the average offset between organic matter in the sediment compared to what has been measured in the overlying water column²⁵ (Δ_2). These δ_{d} and δ_{p} values can now be used to solve for ε_{p} using Eq. 2 of the main text.

Expected ε_{p}

Experimental studies^{21,22} on modern phytoplankton (at fixed O_2 levels) show that the overall photosynthetic fractionation effect (ε_{p}) is a function of both biological factors (growth rate and cell geometry) and the concentration of dissolved CO_2 ($[\text{CO}_2]_{\text{aq}}$) (Eq. 2 of the main text). If the biological factors remain relatively constant for the

phytoplankton groups that make up the majority of biomass, then ϵ_p can be calculated based only on variations in $[\text{CO}_2]_{\text{aq}}$. To solve for $[\text{CO}_2]_{\text{aq}}$ we use the CO_2 results from GEOCARB and Henry's law:

$$[\text{CO}_2]_{\text{aq}} = K_o \times \text{CO}_2 \text{ (ppm)} \quad [\text{S4}]$$

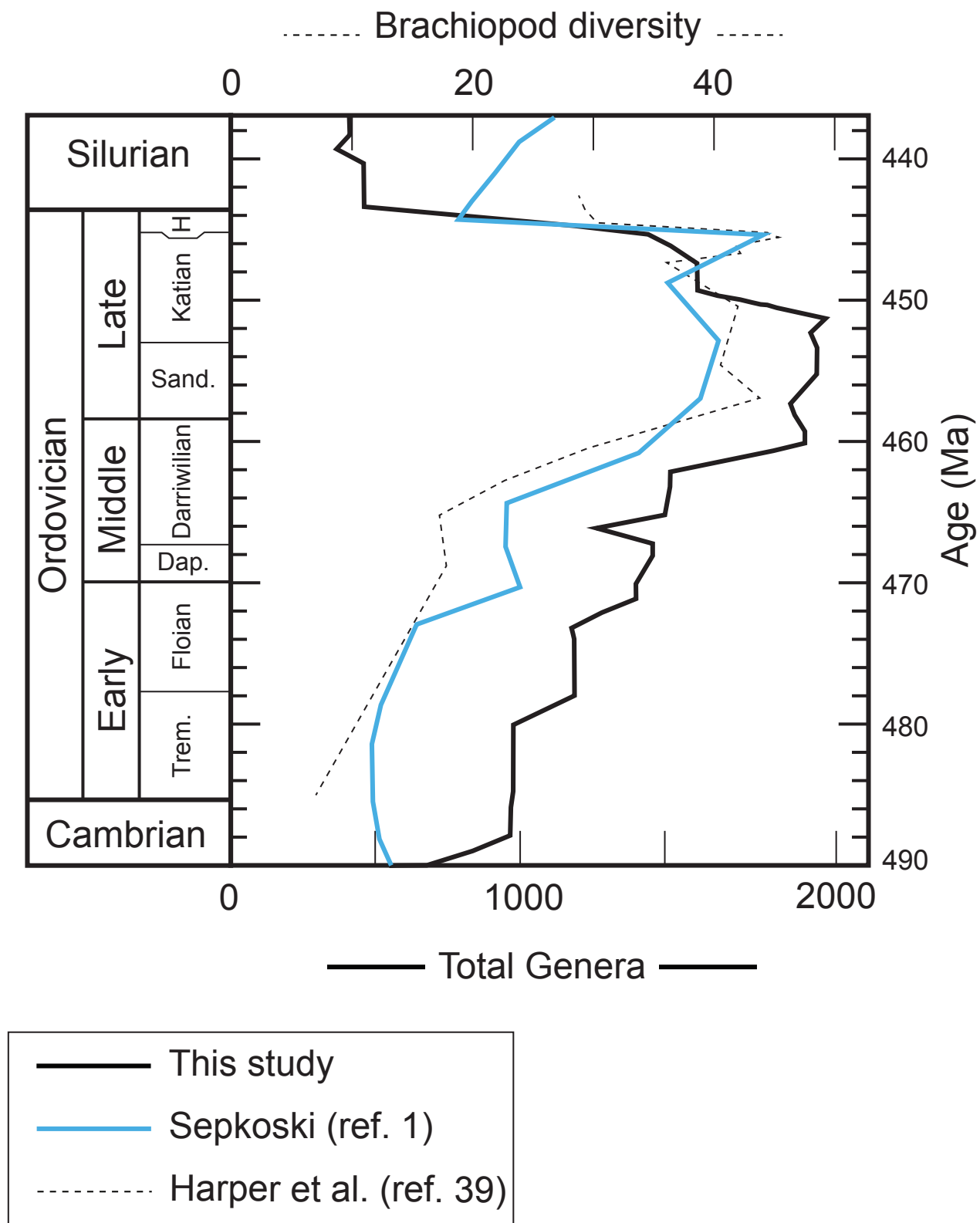
where K_o is the solubility constant in seawater dependent on salinity (we use 35 ppt) and temperature⁷¹ (Table S2). The difference between calculated ϵ_p (from $\Delta^{13}\text{C}$ data; Eq. 1) vs. expected ϵ_p (GEOCARB CO_2 ; Eq. 2) is assumed to reflect changes in atmospheric O_2 relative to our modern atmosphere, where negative values represent atmospheric O_2 levels less than the modern (21%) and vice versa for positive values (Fig. S3). We solve for the size of atmospheric O_2 reservoir (M_{O_2} , 10^{18} mol) by rearranging Eq. 3:

$$M_{\text{O}_2} = 38 \times ((\epsilon_1 - \epsilon_2) + J)/J \quad [\text{S5}]$$

where 38 represents the size of the modern atmospheric O_2 reservoir ($\times 10^{18}$ mol).

Biodiversity curves using the Paleobiology Database

The biodiversity curve used in this study (Fig. 3 of the main text) was constructed using data⁷² from the Paleobiology Database (www.fossilworks.org) and methods similar to other biodiversity studies⁷³, producing a trend similar to others where brachiopods^{39,40} and all taxa¹ diversity rapidly increased during the Middle Ordovician (Fig. S9). The database was queried to provide all taxa at the genus taxonomic rank that had been entered at the time of acquisition (June 2015) extant between ages between 510 Ma and 430 Ma (this is to avoid edge effects when looking at biodiversity trends during the Ordovician: 490–440 Ma). Our diversity curve is constructed using first and last occurrences of taxa (without singletons⁷⁴) and time bins of 1 Myr. For time bins where



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taxa have either a first or last occurrence half a unit is assigned for that time bin, whereas bins receive a full unit where taxa range through the entire time bin.

The new curve has improved age resolution compared to some of the original work of Sepkoski (ref. 1 and refs. therein) as well as incorporating taxonomic occurrences on a more global scale. This curve relies less on the fossil record described from the intensely studied carbonate strata that were deposited under epicontinental seas covering Laurentia that may not be representative of marine ecosystems globally⁴. Several assumptions and biases are inherent in interpreting trends using this kind of database (e.g. sampling and preservation, reported fossil occurrences that have not yet been entered, poor age assignments), but these errors are thought to be random and do not significantly change biodiversity trends when thoroughly screened⁷⁵ (this approach was not conducted in this study). Incorrect taxonomic identification in the Sepkoski compendium has been shown to be non-random among Ordovician and Silurian crinoid genera⁷⁶; however, the corrected compilation made by Ausich and Peters (ref. 76) shows the same overall trend in biodiversity. Because the scope of this study is to compare the overall trend in Ordovician biodiversity with atmospheric O₂, we do not consider that errors resulting from incorrect taxonomic identification would drastically change our conclusions. We use the generic taxonomic level instead of species to minimize overestimates of actual biodiversity that might result from ambiguous species assignment⁴ or poor preservation of some species-specific morphologic structures.

A total of 5335 different genera from 94,177 occurrences were used (Table S4). The dataset was checked to remove errors, often identified as a single occurrence, such as trace fossils (ichnogenera, n=17) and spelling mistakes (n=208), both of which would

308 result in overestimates of actual diversity. Spelling errors occur either as mistakes made
309 by the original author, which were entered into the database as originally reported, or as
310 transcription errors during database entry. Despite this thorough vetting process there
311 were still 73 entries (out of 94,177) that could not be verified, and removing them from
312 the overall database made no noticeable change in the overall biodiversity curve. The
313 taxonomic affiliation of these unverified entries was assigned based on likely groups
314 from the Latin derivatives of their generic name, and if doubt remained they were listed
315 as ‘unclassified’ when assigning them to one of the three evolutionary faunas.

SUPPLEMENTAL REFERENCES

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SUPPLEMENTAL FIGURES

Figure S1. Atmospheric CO₂ and O₂ estimates using GEOCARB, comparing results between a previously published run³³ and this study with a 5 Myr time-step and new carbon, sulfur, and strontium isotope data.

Figure S2. Sensitivity testing of new isotope data used in the GEOCARB approach. Each parameter (colored symbols) is held constant, starting at the 490 Myr time bin, and compared with the standard run (red squares, from Fig. 2). The mid-Ordovician CO₂ drop is seen among all model runs, particularly with a constant $\delta^{13}\text{C}$, suggesting that a combination of parameters is responsible for this drop. The mid-Ordovician O₂ drop in the standard run is reproduced when $\delta^{13}\text{C}$ and $^{87}\text{Sr}/^{86}\text{Sr}$ values are constant, indicating that this O₂ drop is driven by the $\delta^{34}\text{S}$ record. However, the Middle–Late Ordovician O₂ increase is reproduced in the constant $\delta^{34}\text{S}$ run but not in the constant $\delta^{13}\text{C}$ run, indicating that this increase is driven by the $\delta^{13}\text{C}$ record. Constant $^{87}\text{Sr}/^{86}\text{Sr}$ values seems to have no appreciable difference on atmospheric O₂ compared to the standard run.

Figure S3. Estimates of atmospheric CO₂ (left) and ϵ_p (right) using $\delta^{13}\text{C}$ data and GEOCARB model results. Atmospheric CO₂ from GEOCARB (yellow line) is compared with expected CO₂ (black line) using calculated ϵ_p values (from the black line in the right panel, which represents a smoothed LOESS line of $\epsilon_{p\text{-estimated}}$ using $\Delta^{13}\text{C}$ data presented in Figure 1, which have been adjusted for temperature effects), and Eq. 2. The difference between these trends suggests that some factor other than CO₂ played a role in the

fractionation of ^{13}C into biomass. Solving for expected ϵ_p using CO_2 from GEOCARB, adjusted for temperature effects, and Eq. 2 (yellow line in the right panel) shows a similar relationship where there is a mismatch between expected and estimated ϵ_p . This mismatch is interpreted here as the effect of O_2 on the fractionation effect imparted by RuBisCO²⁰ and the difference between the trends (gray regions) reflects changes in atmospheric O_2 compared to modern levels (21%).

Figure S4. Sensitivity tests of atmospheric O_2 model estimates using the photosynthetic fractionation approach by A) varying the dependence of O_2 on fractionation (J parameter in Eq. 3), B) using atmospheric CO_2 estimates from the COPSE model¹³ and constant high and low CO_2 levels, C) using a different equation used to estimate temperature from phosphate oxygen isotopes⁶⁶ and assuming a constant sea surface temperature of 27 °C, and D) using an average $\delta^{18}\text{O}$ value of -3‰ for seawater instead of -1‰ used in the standard run.

Figure S5. Calculated atmospheric O_2 values for paired $\delta^{13}\text{C}$ values (Table S2) using the photosynthetic fractionation effect approach, with LOESS smoothed line of mean atmospheric O_2 values and 95% CI envelope from Fig. 2 (not the error associated with each data point; see supplement for further discussion). Large blue circles represent averaged values in 5 Myr bins for comparison of O_2 trends made using data at 5 Myr (e.g. GEOCARB).

Figure S6. Late Ordovician (Hirnantian) $\delta^{13}\text{C}$ data used to estimate O_2 using the photosynthetic fractionation approach. Data from Anticosti Island (green triangles and blue diamonds) show a slight increase in $\delta^{13}\text{C}_{\text{org}}$ prior to the positive $\delta^{13}\text{C}_{\text{carb}}$ excursion known as the HICE event. This decrease in the difference between the reservoir ($\delta^{13}\text{C}_{\text{carb}}$) and biomass ($\delta^{13}\text{C}_{\text{org}}$) is expressed as a decrease in atmospheric O_2 using the O_2 -dependent approach. $\delta^{13}\text{C}$ data from Nevada (red squares) also predict a decrease in atmospheric O_2 (black arrow), but graptolite biostratigraphic information used to assign ages to these data may be incomplete and give the appearance of two O_2 drops (one at 444.84 Ma and one at 444.80 Ma) that may in fact record the same O_2 crisis prior to the HICE.

Figure S7. Schematic diagram of the isotopic relationships involved in the production of sedimentary organic matter from seawater. Modified from Hayes et al. (ref. 25).

Figure S8. Relationship between atmospheric CO_2 concentration (ppm) and the isotopic fractionation (ϵ_p) imparted during primary production at varying growth rates (μ ; black contoured lines) at $V/S = 1 \mu\text{m}$. Examples of modern phytoplankton are shown in colored lines²². Gray region shows range of possible ϵ_p increase from 20.2‰ to 22.5‰ (Fig. S3). Red arrow (A) represents overall trend of atmospheric CO_2 estimates from GEOCARB during the observed increase in ϵ_p , which requires unrealistic μ or V/S values far higher than modern phytoplankton. The light blue region (B) represents range of growth conditions typical of modern small phytoplankton likely present during the Ordovician (*P. glacialis* is considered a large diatom compared to the others listed²²), but

in order for ϵ_p to increase from 20.2‰ to 22.5‰ atmospheric CO₂ would have to increase from levels significantly lower than CO₂ model estimates. *E. huxleyi* = *Emiliana huxleyi*, *P. tricornutum* = *Phaeodactylum tricornutum*, *P. glacialis* = *Porosira glacialis*.

Figure S9. Comparison of biodiversity curves of all marine taxa (solid lines) from this study (black) and a recent compilation (blue¹), and global brachiopod diversity (dashed black line³⁹).

Table S1. Binned isotope data* used for GEOCARB and model results of O₂ and CO₂ with error estimates.

Table S2. Isotope data and atmospheric O₂ using the photosynthetic fractionation effect approach.

Table S3. New $\delta^{13}\text{C}^*$ and $\delta^{34}\text{S}$ data[#] used in the photosynthetic fractionation effect and GEOCARB models.

Table S4. Taxonomic data* used to construct the biodiversity curve (Fig. 3) with Evolutionary Faunas, First Appearance Datum (FAD), and Last Appearance Datum (LAD).

Table S5. Example calculations for error propagation using the O₂-dependent photosynthetic fractionation approach.