

Time-course relationship between environmental factors and microbial diversity in tobacco soil

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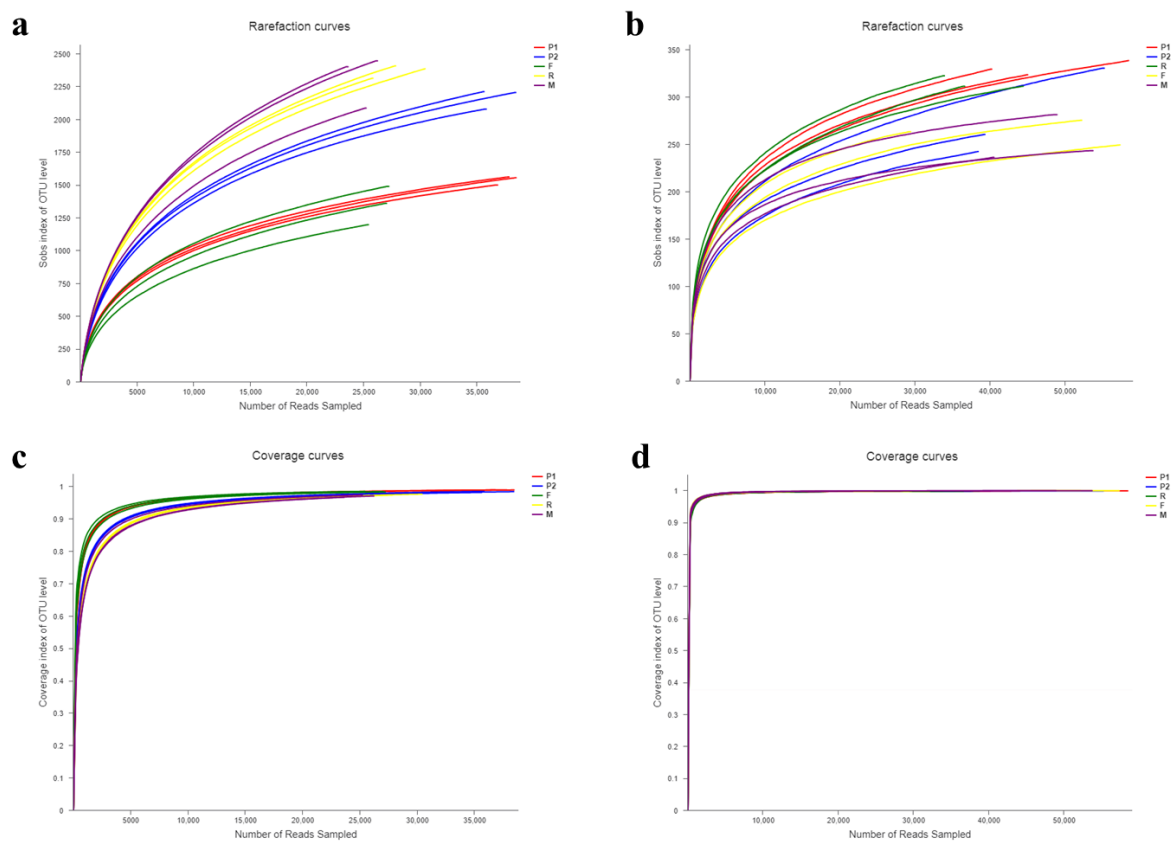


Figure S1. The rarefaction curves (**a** for bacteria, **b** for fungi) and coverage curves (**c** for bacteria, **d** for fungi) of the samples.

Table S1 The soil samples used in this study.

Location	Site name	Pretransplant stage	Root extending stage	Flourishing stage	Mature stage
Xinhua, Yuxi	YX-1*	YX-p1*	/	/	/
	YX-2*	P*	R*	F*	M*

* Two sites were selected in the previous work, named YX-1 and YX-2. The samples in the pretransplant stage of the two sites were named as YX-p1 and P, respectively. The samples of root extending, flourishing, and mature stages in the optimum site (YX-2) were named as R, F, and M, respectively.

File S2. The detailed procedures of enzyme activities determination.

Measure of soil enzyme activities

Determination of urease activity. A sample of 1 g soil producing 1 μg ammonium nitrogen in 24 h was set as 1 U. 0.25 ml toluene, 0.75 ml citrate buffer (pH, 6.7), and 1 ml of urea substrate solution were added to the 1 g soil sample and the samples were incubated. The formation of ammonium was determined spectrophotometrically at 578 nm and results were expressed as $\mu\text{g N g}^{-1}$ dry soil. All determinations of urease activities were performed in triplicate, and all values reported were averages of the 3 determinations expressed on an oven-dried soil basis (105 $^{\circ}\text{C}$).

Determination of nitrate reductase activity. Nitrate reductase activity was calculated by the concentration of nitrite ions catalyzed by soil from nitrate. A sample of 1 g soil producing 1 μmol nitrite nitrogen in 24 h was set as 1 U. A sample of soil (5 g) was treated with 2 ml of absolute ethanol containing 2,4-dinitrophenol (DNP). The DNP concentrations desired was prepared by diluting an aliquot of the stock DNP solution (25 mg/ml) with ethanol. The alcohol was evaporated for 2 h, and the soil was treated with 10 ml of solution containing 500 $\mu\text{g NO}_3^- \text{-N}$ or 10 ml of 5 mM KNO_3 . The sample and the contents were mixed by swirling for a few seconds to mix, stoppered, and held in the dark at 25 $^{\circ}\text{C}$ for 24 h. Then 40 ml 2.5 M KCl solution was added, following the sample was shaken for 30 min. Then the soil suspension was filtered. An aliquot (1 ml) of the resulting soil filtrate was transferred into a 50 ml volumetric flask and the $\text{NO}_3^- \text{-N}$ was determined by the calorimetric method of Griess-Ilosvay.

Determination of sucrase activity. A sample of 1 g soil producing 1 mg glucose in 24 h was set as 1 U. A sample of soil added with 8% sucrose solution, phosphate buffer (PBS, pH 5.5) and methylbenzene was incubated at 37 $^{\circ}\text{C}$ for 24 h. Part of filtered supernatant was boiled for 5 min after 3,5-dinitrosalicylic acid was added. The formation of product was determined spectrophotometrically at 510 nm.

Determination of polyphenol oxidase activity. A sample of 1 g soil producing 1 μg purpurogallin in 1 h was set as 1 U. A sample of soil mixed with 1% pyrogallol solution was incubated at 30 $^{\circ}\text{C}$ for 2 h, after which citric acid-phosphate buffer (pH

4.5) was added. Then the mixture was extracted using ethyl acetate. The purpurogallin contained in the extraction was determined spectrophotometrically at 430 nm.

Determination of catalase activity. A sample of 1 g soil decomposing 1 g hydrogen peroxide in 1 h was set as 1 U. A sample of soil added with distilled water and 0.3% H₂O₂ was incubated in dark for 2 h, after which citric acid-phosphate buffer (pH 4.5) was added. Then saturated KAl(SO₄)₂ was added into the mixture, following it was filtrated into 1.5 M H₂SO₄ solution. The concentration of final product contained in the filtrate was determined spectrophotometrically at 240 nm.

Determination of acid phosphatase activity. A sample of 1 g soil producing 1 µmol phosphorus in 24 h was set as 1 U. A sample of soil was incubated for 1 h at 37 °C with 4 ml of modified universal buffer (MUB) (pH 6.5) and 1 ml of 0.115 M *p*-nitrophenylphosphate solution. After incubation, 1 ml of 0.5 M CaCl₂ and 4 ml of 0.5 M NaOH were added to the sample, which was followed by filtration. The concentration of final product was determined spectrophotometrically at 400 nm.