

Supplementary information

Transparent polycrystalline cubic silicon nitride

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This PDF file includes:

1. Pressure and temperature calibrations; *P-T* path of the synthesis runs
2. XRD patterns of the recovered samples
3. Chemical composition measurements
4. Sample preparation for TEM observations
5. Grain size distribution
6. Brillouin spectroscopy
7. Fracture surface observation
8. Numerical data for Fig. 4c to update the H_V - G plot for hard materials

1. Pressure and temperature calibrations; *P-T* path of the synthesis runs:

Pressure was calibrated at room temperature using a fixed point of pressure: the semiconductor to metal transition of ZnS at 15.6 GPa. Temperature was calibrated in a separate run using a W5%Re/W26%Re thermocouple (C-type). For synthesis runs, a pressure of 15.6 GPa was applied first at room temperature for ~2 h and temperature was then increased at a constant load corresponding to this pressure. Temperature was increased to the target values in two steps: first, temperature was increased to 450°C with a heating rate of ~50°C/min; secondly, temperature was rapidly increased to the target value within 10 seconds. The target temperature was maintained for 30 min. Then temperature was gradually decreased to 400°C for ~10 min. Decompression at ~400°C took 3 h. After the decompression, the sample was recovered.

2. XRD patterns of the recovered samples:

Figure S1 shows synchrotron XRD patterns of polycrystalline c-Si₃N₄ samples synthesized at 1700 and 1800°C. These two patterns look almost identical and all the diffracted peaks are explained by the presence of the single phase of c-Si₃N₄. The peaks look very sharp, indicating that no residual stress exists in the sintered bodies.

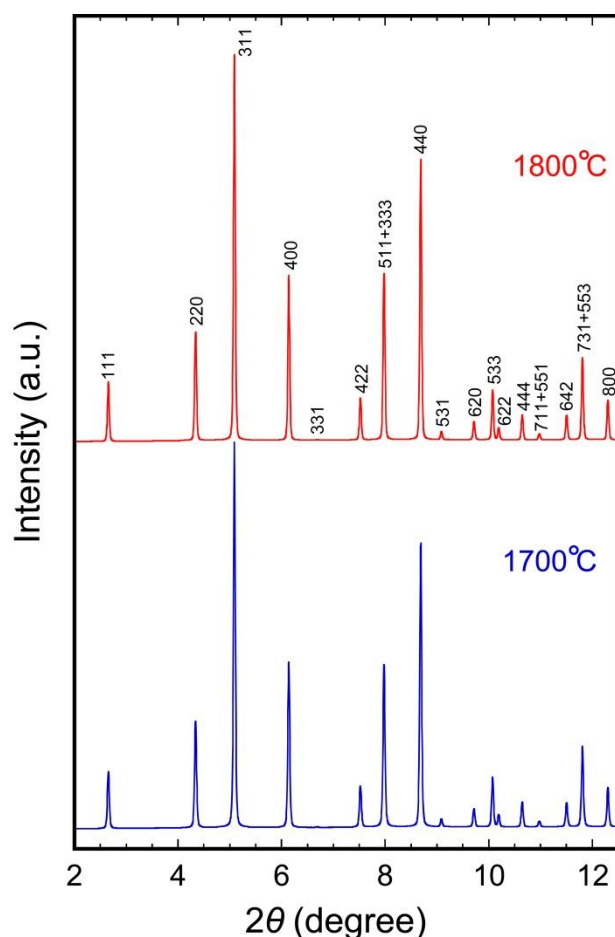


Figure S1. Synchrotron XRD patterns of bulk nanocrystalline forms of c-Si₃N₄ synthesized at 1700 and 1800°C under a fixed pressure of 15.6 GPa. Energy of the X-rays was about 60 keV.

3. Chemical composition measurements:

Figure S2 shows an example of energy-dispersive X-ray spectrum obtained from the sample synthesized at 15.6 GPa and 1800°C. We can see the presence of silicon, nitrogen, and oxygen; no other element was detected.

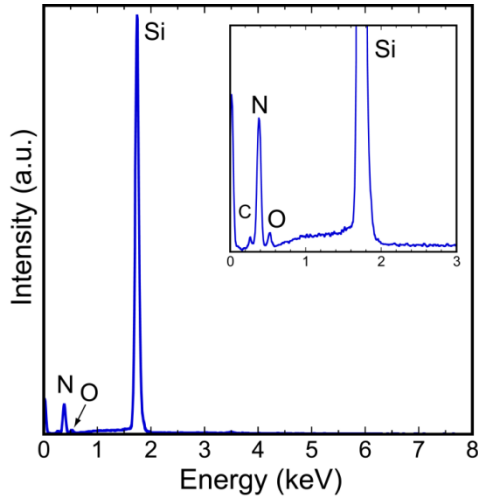


Figure S2. An example of EDS spectra of c-Si₃N₄ synthesized at 15.6 GPa and 1800°C. The presence of Si, N, and O can be seen. The calculated chemical composition is shown in the main text.

4. Sample preparation for TEM observations:

The electron transparent thin specimen was prepared by the following procedure: cutting the pellet sample, mechanical polished the sample less than 50 μm thickness, and a conventional Ar ion beam thinning method. We note that, to suppress Ar ion beam damages, the sample was cooled down to liquid nitrogen temperature.

5. Grain size distribution:

Figure S3 shows grain size distribution of the transparent c-Si₃N₄ polycrystalline sample synthesized at 15.6 GPa and 1800°C. We directly measured 100 grains that appear black (satisfying the Bragg condition) in a BF-TEM image³².

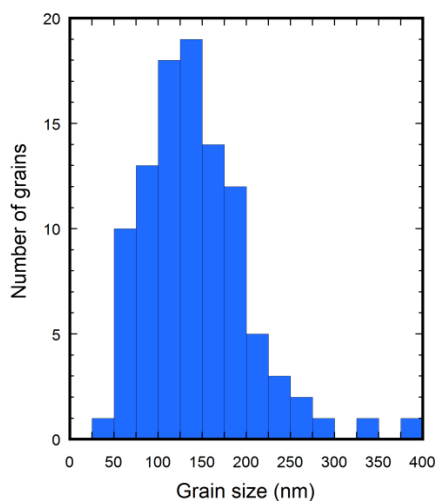


Figure S3. Grain size distribution of the transparent c-Si₃N₄ polycrystalline sample synthesized at 15.6 GPa and 1800°C. The average grain size is 143 ± 59 nm.

6. Brillouin spectroscopy:

Brillouin spectra collected in several positions on the polycrystalline sample as well as with varying phonon propagation directions showed that the sample is elastically isotropic as expected (less than 50 m/s deviation of velocities from the average velocity) because of the randomly oriented nanocrystalline c-Si₃N₄. Figure S4 shows an example of the spectra.

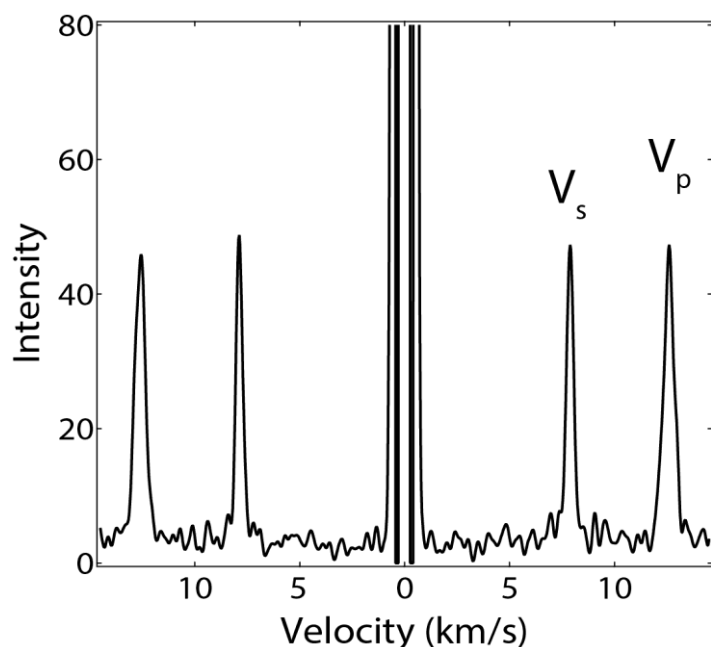


Figure S4. An example of Brillouin spectra of a bulk nanocrystalline form of c-Si₃N₄ synthesized at 15.6 GPa and 1800°C.

7. Fracture surface observation:

A field emission scanning electron microscope (JSM-7000F, JEOL) operating at 15 kV was used for secondary electron image observations of a fracture surface of a polycrystalline c-Si₃N₄ synthesized at 15.6 GPa and 1800°C. The fracture surface was coated with osmium. Figure S5 shows an example of secondary electron images of a fracture surface. This image clearly shows that the intergranular fracture is dominant in this sample.

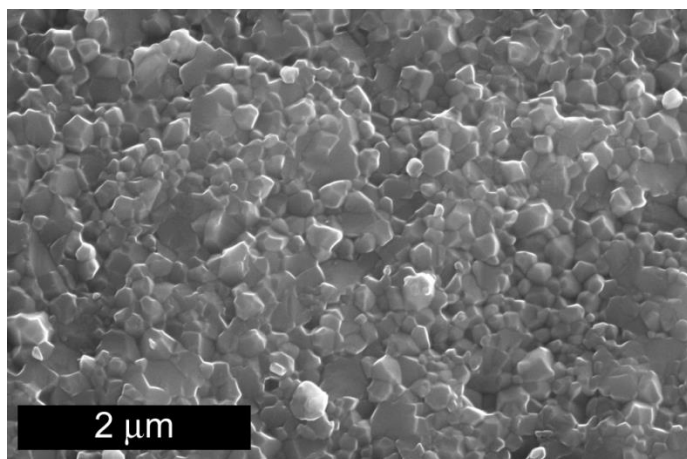


Figure S5. An example of secondary electron images of a fracture surface of a polycrystalline c-Si₃N₄.

8. Numerical data for Fig. 4c to update the H_V - G plot for hard materials:

Table S1 shows numerical data for the Fig. 4c. Other data are the same as those used by Teter²⁶. Recently, synthesis of some potential superhard materials ($H_V > 40$ GPa), such as hetrodiamonds (c-BC₂N^[S15] and c-BC₅^[S17]) and a high-pressure phase of boron (γ -B₂₈^[S18]), was reported, but the mechanical properties of these materials have been still under debate. Thus, at the moment, these materials are not included in the plot. Diamond and cBN are the well-established hardest and second hardest materials, respectively.

Table S1. H_V and G_0 values for Figure 4.

Material	H_V (GPa)	Reference	G_0 (GPa)	Reference
B ₄ C	37.8	[S1] Ji et al. 2015	193	[S2] Dodd et al. 2002
c-Si ₃ N ₄	34.9	Present study	247.5	Present study
B ₆ O	34.8	[S3] Kleebe et al. 2008	206	[S4] Petrak et al. 1974
TiB ₂	30.4	[S5] Rizzo et al. 1962	262.4	[S6] Spoor et al. 1997
SiO ₂ -stishovite	29	[S7] Nishiyama et al. 2014	228	[S8] Jiang et al. 2009
BP	28	[S9] Mukhanov et al. 2013	136.4	[S10] Wettling and Windscheif 1984
γ -AlON	17.4	[S11] Klement et al. 2008	135.5	[S12] Graham et al. 1988
MgAl ₂ O ₄	15.4	[S13] Morita et al. 2009	109	[S14] Chopelas 1996
c-BC ₂ N	76	[S15] Solozhenko et al. 2001	238	[S16] Tkachev et al., 2003
c-BC ₅	71	[S17] Solozhenko et al. 2009	-	-
γ -B ₂₈	58	[S18] Zarechnaya et al. 2009	227.2	[S19] Qin et al., 2012

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